

Electronic supplementary information (ESI)

Synthesis, characterization and selective hysteretic sorption property of metal-organic frameworks with 3,5-di(pyridine-4-yl)benzoate

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Table S1. Selected bond lengths (Å) and angles (°) for 1 - 3.

Compound 1			
Zn1-N2	2.083(2)	Zn1-N1 ⁱ	2.087(2)
Zn1-O4	2.098(2)	Zn1-O1 ⁱⁱ	2.1688(18)
Zn1-O1W	2.176(2)	Zn1-O2 ⁱⁱ	2.230(2)
N2-Zn1-N1 ⁱ	109.12(8)	N2-Zn1-O4	94.16(9)
N1 ⁱ -Zn1-O4	86.31(9)	N2-Zn1-O1 ⁱⁱ	154.21(7)
N1-Zn1-O1 ⁱⁱ	96.12(8)	O4-Zn1-O1 ⁱⁱ	92.64(8)
N2-Zn1-O1W	86.86(9)	N1 ⁱ -Zn1-O1W	92.43(9)
O4-Zn1-O1W	178.59(8)	O1 ⁱⁱ -Zn1-O1W	86.85(8)
N2-Zn1-O2 ⁱⁱ	95.10(7)	N1 ⁱ -Zn1-O2 ⁱⁱ	155.77(8)

O4-Zn1-O2 ⁱⁱ	92.77(9)	O1 ⁱⁱ -Zn1-O2 ⁱⁱ	59.70(6)
O1W-Zn1-O2 ⁱⁱ	88.11(8)		
Symmetry codes: (i) $x-1/2, -y+3/2, z-1/2$; (ii) $x-1/2, -y+5/2, z-1/2$; (iii) $x+1/2, -y+5/2, z+1/2$; (iv) $x+1/2, -y+3/2, z+1/2$.			
Compound 2			
Cd1-N2 ⁱ	2.317(3)	Cd1-O3 ⁱⁱ	2.327(2)
Cd1-O2 ⁱⁱⁱ	2.365(3)	Cd1-N3	2.375(3)
Cd1-N1	2.396(3)	Cd1-O1 ⁱⁱⁱ	2.459(3)
Cd1-O4 ⁱⁱ	2.542(3)		
N2 ⁱ -Cd1-O2 ⁱⁱⁱ	136.73(9)	N2 ⁱ -Cd1-O2 ⁱⁱⁱ	137.40(9)
O3 ⁱⁱ -Cd1-O2 ⁱⁱⁱ	85.80(9)	N2 ⁱ -Cd1-N3	90.84(11)
O3 ⁱⁱ -Cd1-N3	86.57(10)	O2 ⁱⁱⁱ -Cd1-N3	94.95(10)
N2 ⁱ -Cd1-N1	95.81(10)	O3 ⁱⁱ -Cd1-N1	84.78(9)
O2 ⁱⁱⁱ -Cd1-N1	83.91(9)	N3-Cd1-N1	171.34(10)
N2 ⁱ -Cd1-O1 ⁱⁱⁱ	82.89(9)	O3 ⁱⁱ -Cd1-O1 ⁱⁱⁱ	140.02(9)
O2 ⁱⁱⁱ -Cd1-O1 ⁱⁱⁱ	54.54(8)	N3-Cd1-O1 ⁱⁱⁱ	100.40(11)
N1-Cd1-O1 ⁱⁱⁱ	85.97(10)	N2 ⁱ -Cd1-O4 ⁱⁱ	82.68(9)
O3 ⁱⁱ -Cd1-O4 ⁱⁱ	54.05(8)	O2 ⁱⁱⁱ -Cd1-O4 ⁱⁱ	139.79(8)
N3-Cd1-O4 ⁱⁱ	85.58(11)	N1-Cd1-O4 ⁱⁱ	89.75(10)
O1 ⁱⁱⁱ -Cd1-O4 ⁱⁱ	164.45(9)		
Symmetry codes: (i) $-x+3/2, y-1/2, z$; (ii) $x, -y+1, z+1/2$; (iii) $-x+5/2, -y+1/2, z-1/2$; (iv) $x, -y+1, z-1/2$; (v) $-x+5/2, -y+1/2, z+1/2$; (vi) $-x+3/2, y+1/2, z$.			
Compound 3			
Co1-O3	2.009(3)	Co1-N1 ⁱ	2.116(3)

Co1-N2 ⁱⁱ	2.124(3)	Co1-O1W	2.134(3)
Co1-O2	2.169(3)	Co1-O1	2.232(3)
O3-Co1- N1 ⁱ	99.52(13)	O3-Co1-N2 ⁱⁱ	85.55(12)
N1 ⁱ -Co1-N2 ⁱⁱ	112.77(13)	O3-Co1-O1W	171.52(11)
N1 ⁱ -Co1-O1W	84.85(12)	N2 ⁱⁱ -Co1-O1W	86.05(12)
O3-Co1-O2	94.01(12)	N1 ⁱ -Co1-O2	94.66(12)
N2 ⁱⁱ -Co1-O2	152.29(12)	O1W-Co1-O2	92.87(11)
O3-Co1-O1	90.76(13)	N1-Co1-O1	153.30(12)
N2 ⁱⁱ -Co1-O1	92.43(12)	O1W-Co1-O1	88.40(11)
O2-Co1-O1	59.86(9)		
Symmetry codes: (i) $x+1/2, y-1/2, z$; (ii) $x+1/2, y+1/2, z$; (iii) $-x+1/2, -y+1/2, -z+2$; (iv) $x-1/2, y+1/2, z$; (v) $x-1/2, y-1/2, z$.			

Table S2. The parameters of hydrogen bonds for 1 – 3.

Complex 1				
$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O1W—H1WA \cdots O1	0.85	2.02	2.868 (3)	171
O1W—H1WB \cdots O5	0.84	1.94	2.755 (4)	165
C1—H1 \cdots O2W	0.93	2.53	3.231(10)	132
C1—H1 \cdots O5	0.93	2.54	3.271(4)	135
Symmetry codes: (i) $1/2-x, 5/2-y, 1-z$; (ii) $x, 2-y, 1/2+z$; (iii) $1-x, 1-y, 1-z$; (iv) $1/2+x, -1/2+y, 1+z$.				
Compound 2				
$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C4—H4A \cdots O2	0.93	2.55	3.484(4)	176
C5—H5A \cdots O3	0.93	2.32	3.023(4)	133
C13—H13A \cdots O4	0.93	2.42	3.048(4)	125
C14—H14A \cdots O2	0.93	2.48	3.389(5)	167
C17—H17A \cdots O1	0.93	2.43	3.090(4)	128
C18—H18A \cdots O3	0.93	2.47	3.121(5)	127
Symmetry codes: (i) $-1/2+x, 1/2+y, 3/2-z$; (ii) $x, 1-y, 1/2+z$; (iii) $3/2-x, 3/2-y, 1/2+z$; (iv) $-1/2+x, 1/2+y, 3/2-z$; (v) $-1+x, 1-y, -1/2+z$; (vi) $x, 1-y, 1/2+z$.				
Compound 3				
$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O1W—H1WA \cdots O4	0.85	2.12	2.696 (4)	125
O1W—H1WB \cdots O1	0.85	2.04	2.812(4)	151
C12—H12 \cdots O2	0.93	2.58	3.133(6)	118

C13—H13 ···O1	0.93	2.47	3.082(6)	124
Symmetry codes: (i) $x, -y, -1/2+z$; (ii) $1/2-x, 1/2-y, 1-z$; (iii) $-1/2+x, 1/2+y, z$; (iv) $-1/2+x, -1/2+y, z$.				

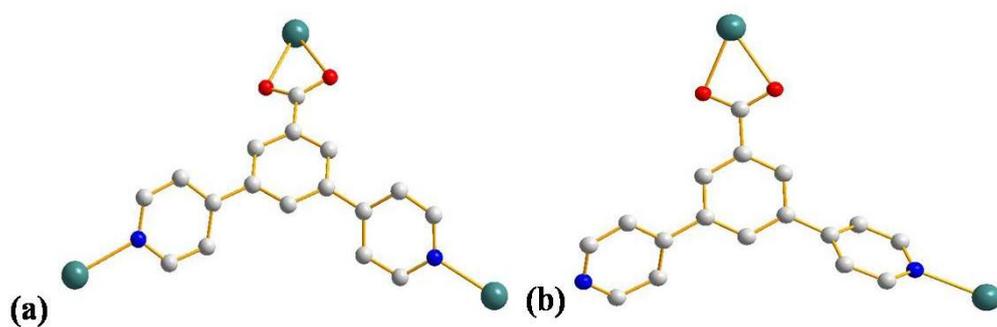


Fig.S1 Coordination modes of L⁻: (a) mode I and (b) mode II in complex **2**.

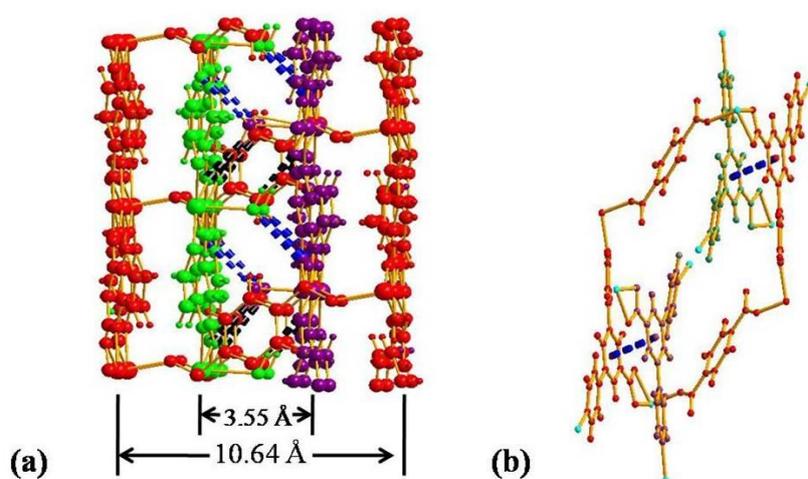


Fig.S2 (a) Hydrogen bonding interactions of complex **3** between different layers (O1W-H1WA \cdots O4 are shown in blue dash lines, O1W-H1WB \cdots O1 are shown in black dash lines). (b) The $\pi \cdots \pi$ interactions between the layers in **3** indicated by blue dash lines.

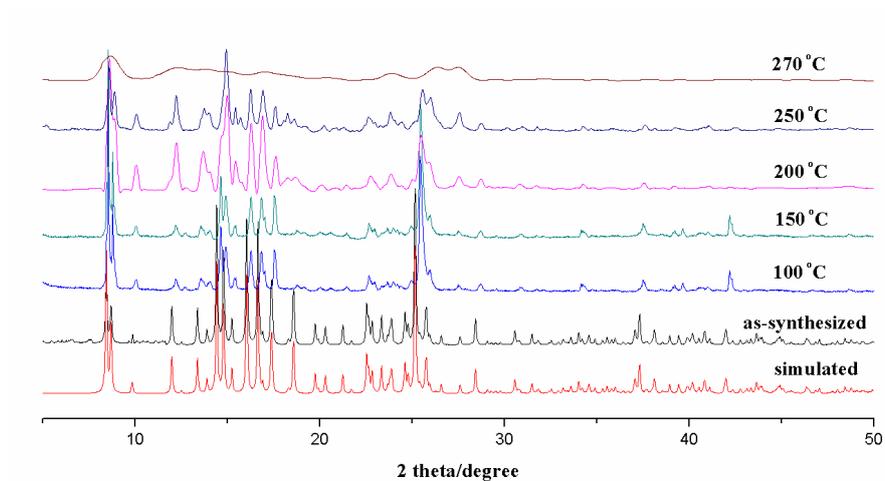


Fig.S3 The PXRD patterns of complex **1** at different temperature.

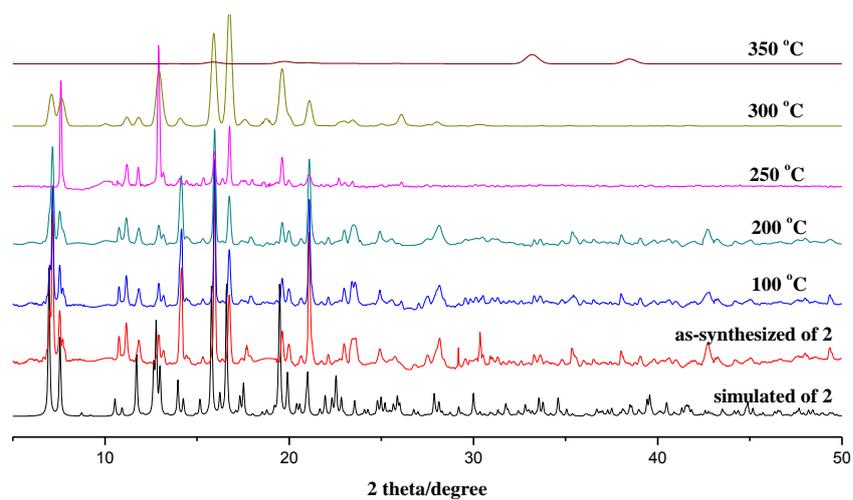


Fig.S4 The PXRD patterns of complex **2** at different temperature.

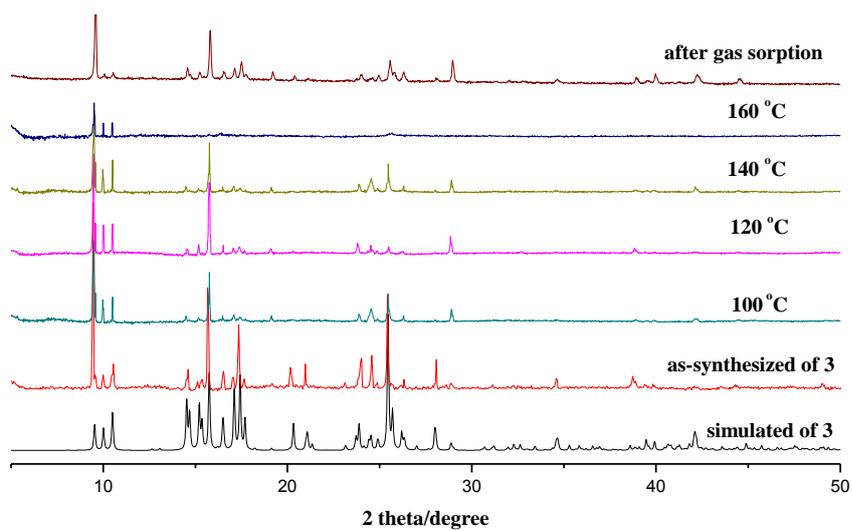


Fig.S5 The PXRD patterns of complex **3** at different temperature and after gas sorption.

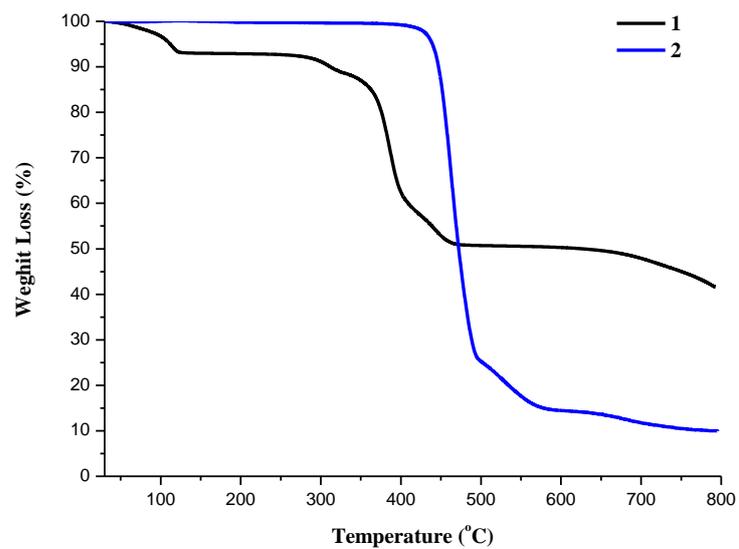


Fig.S6 The TGA curves of complexes **1** and **2**.

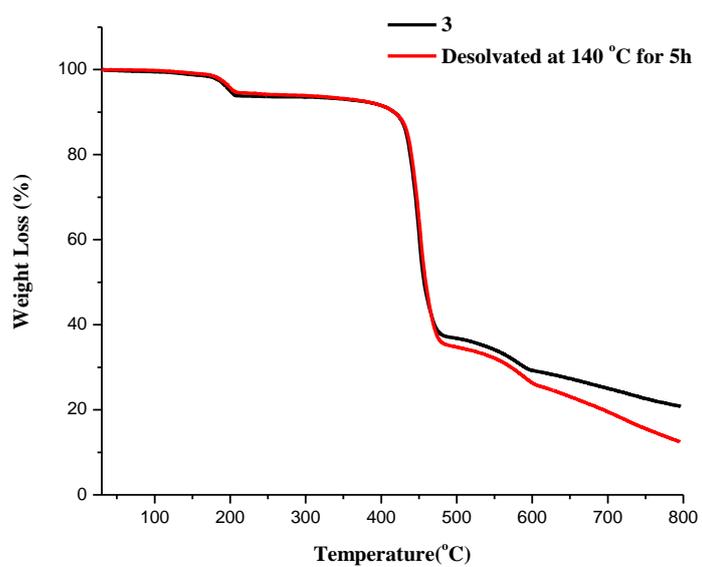
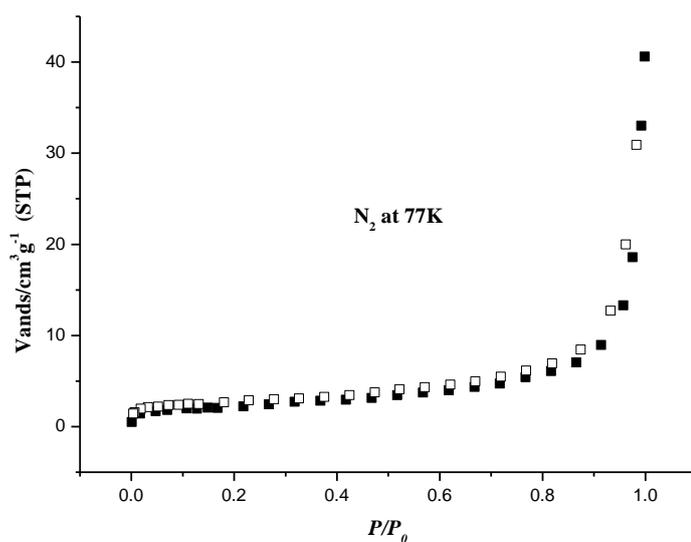
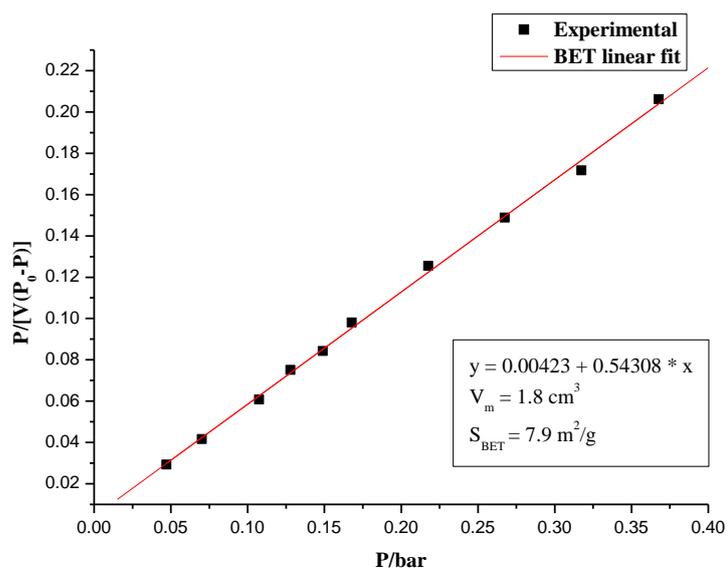


Fig.S7 The TGA curves of complex **3**.



(a)



(b)

Fig.S8 (a) N₂ sorption isotherm at 77 K for **3** (filled shape, adsorption; open shape, desorption). (b) The BET plot calculated from N₂ isotherm at 77K of **3**.