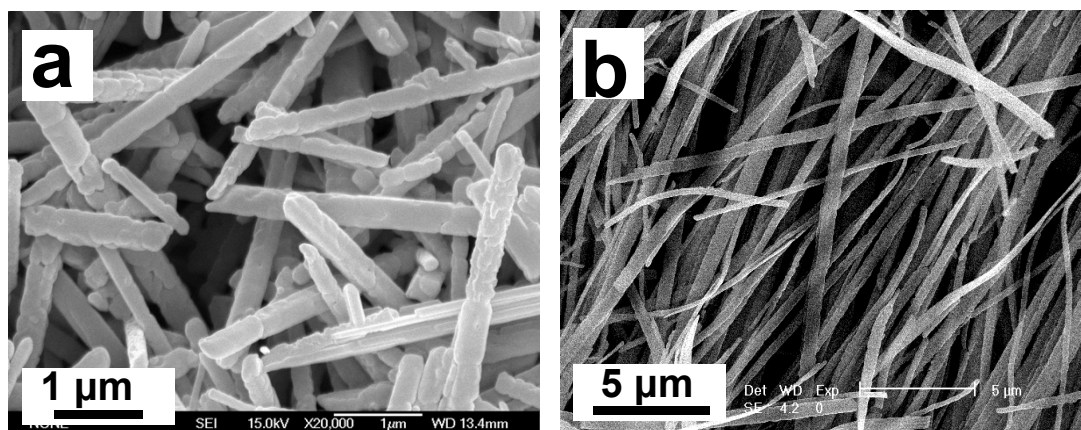


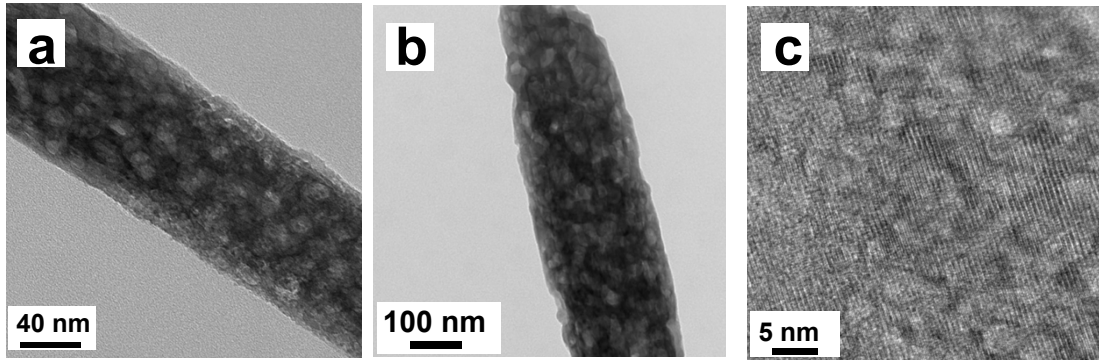
## Hydrothermal Growth and Characterization of Length Tunable Porous Iron Vanadates One-dimensional Nanostructures

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**Fig. S1** SEM images of the A-FeV<sub>x</sub>O<sub>y</sub>-4 (a) and A-FeV<sub>x</sub>O<sub>y</sub>-6 (b) 1-D nanostructures.



**Fig. S2** HRTEM images of the A-FeV<sub>x</sub>O<sub>y</sub>-4 (a), A-FeV<sub>x</sub>O<sub>y</sub>-5 (b) and A-FeV<sub>x</sub>O<sub>y</sub>-6 (c) 1-D nanostructures.

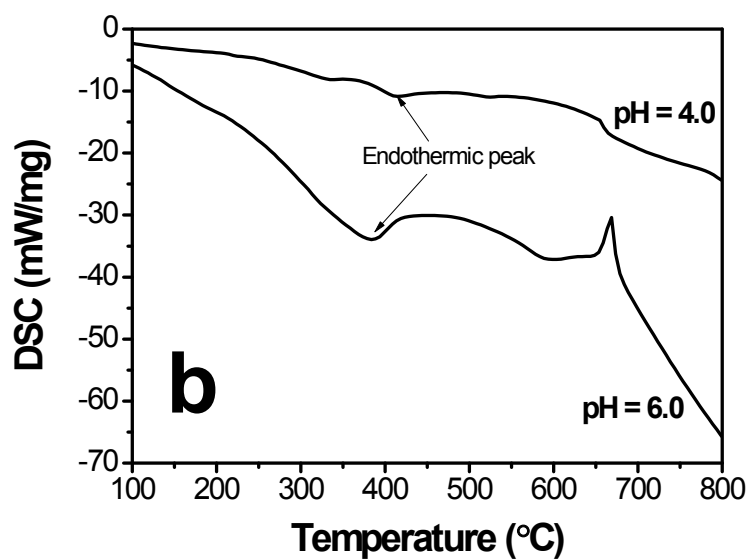
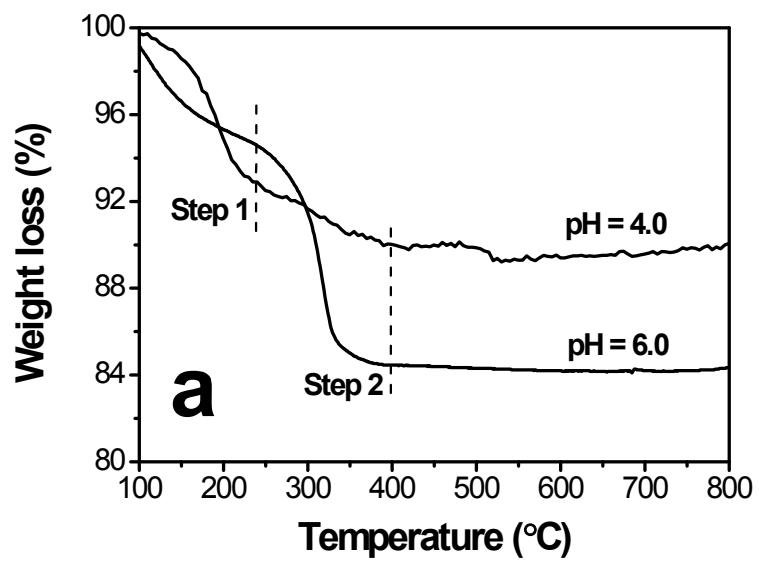


Fig. S3 Thermal analysis (a) and DSC analysis (b) of  $\text{FeV}_x\text{O}_y$ -4 and  $\text{FeV}_x\text{O}_y$ -6.

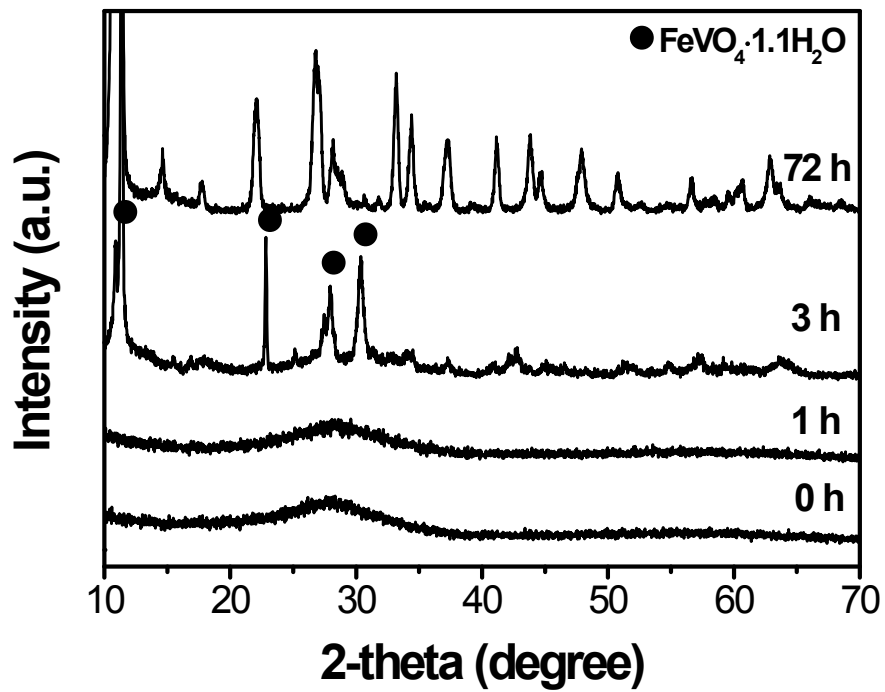
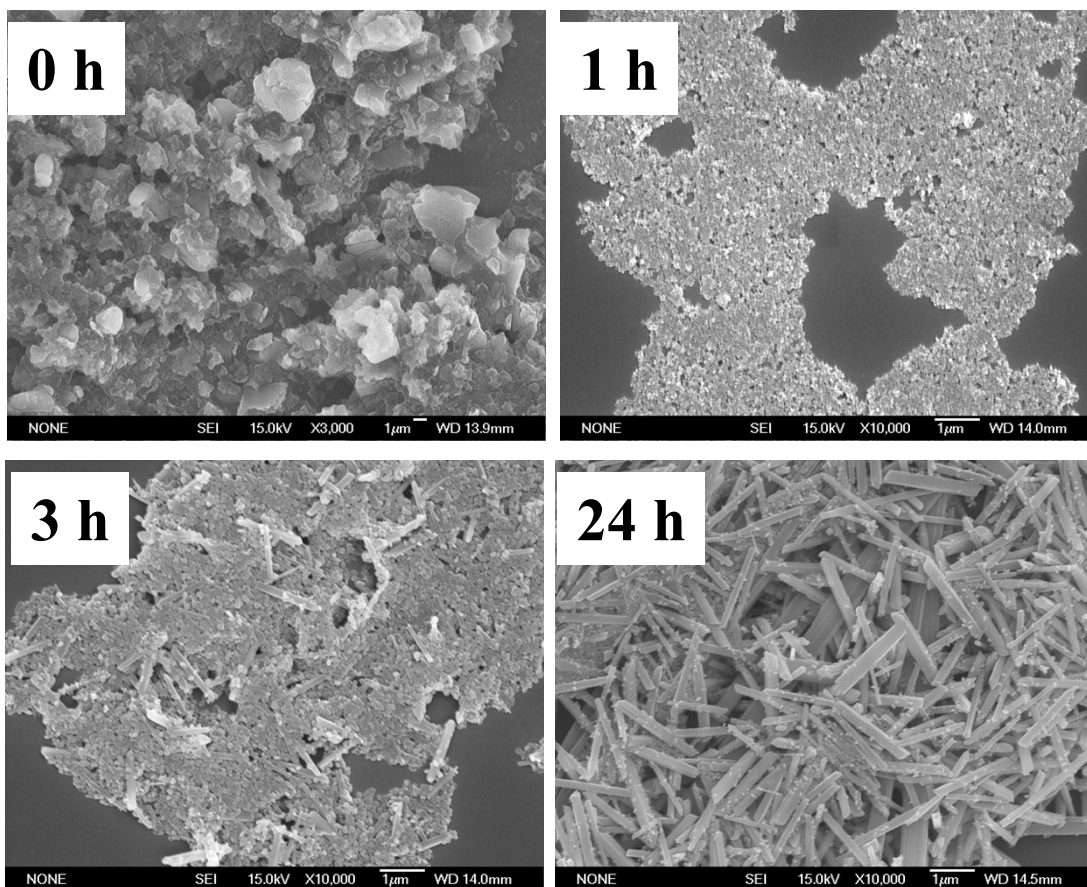
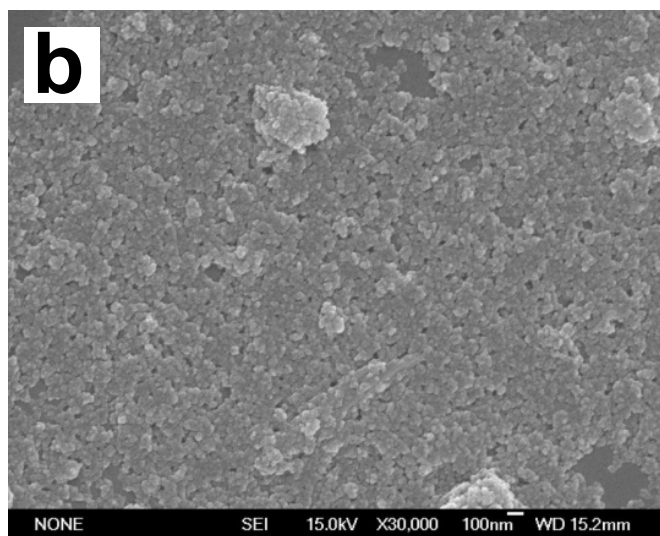
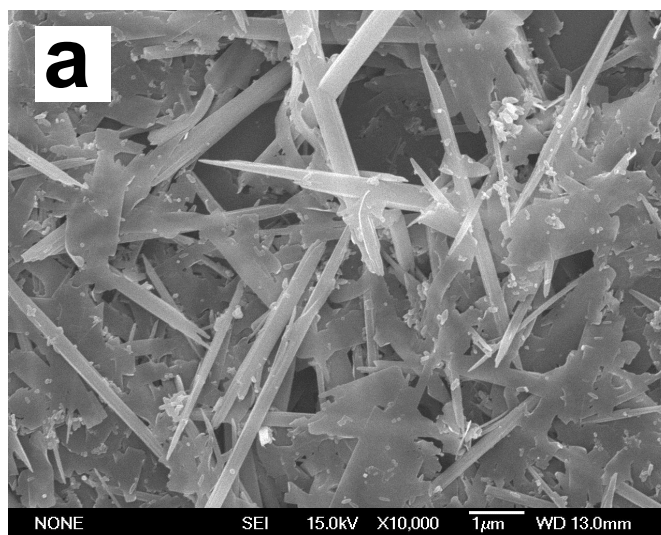


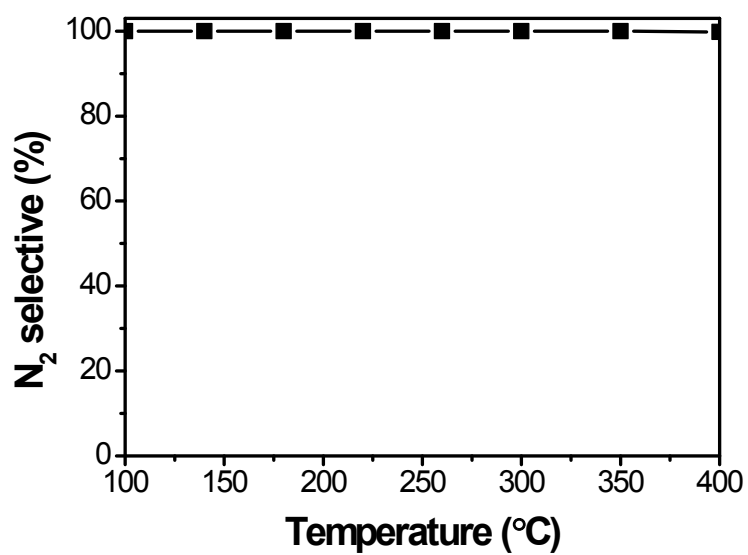
Fig. S4 XRD patterns evolution of FeV<sub>x</sub>O<sub>y</sub>-6 at different synthesis stages.



**Fig. S5** Morphological evolution of FeV<sub>x</sub>O<sub>y</sub>-4 1-D nanostructure at different synthesis stages, starting from the precursor nanoparticles (0h) and followed by 1h, 3h and 24 h hydrothermal treatment of the precursors.



**Fig. S6** FESEM images of FeV<sub>x</sub>O<sub>y</sub> 1-D nanostructures prepared at pH values of 2.0 (a) and 7.0 (b).



**Fig. S7** Selectivity towards N<sub>2</sub> as a function of temperature in the feed gas of 250 mL/min total rate, 500ppm NO, 500ppm NH<sub>3</sub>, 3%O<sub>2</sub>, and N<sub>2</sub> balance, GHSV=20000h<sup>-1</sup>.