Electronic Supplementary Material (ESI) for CrystEngComm.

A series of new 3-D boratopolyoxovanadates based on different K-O-K building blocks

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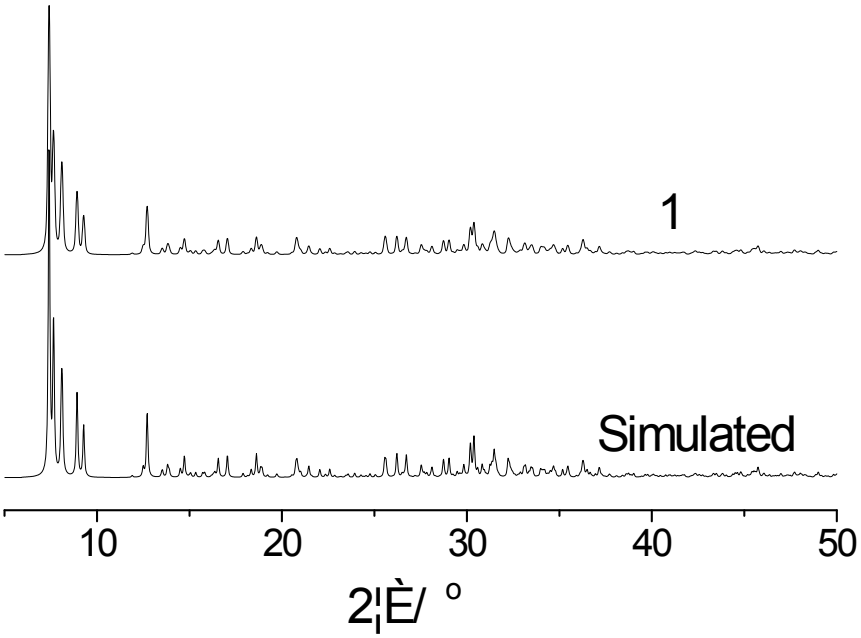
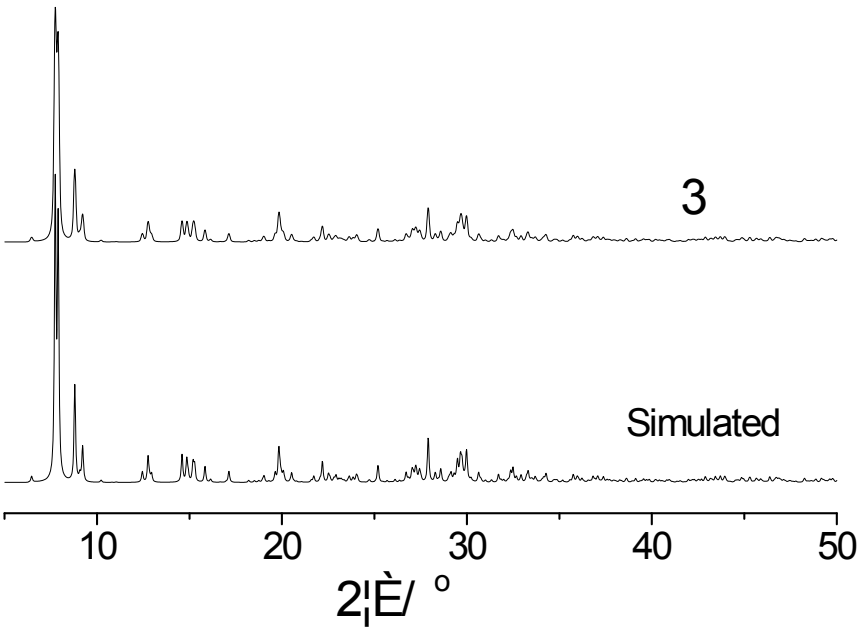
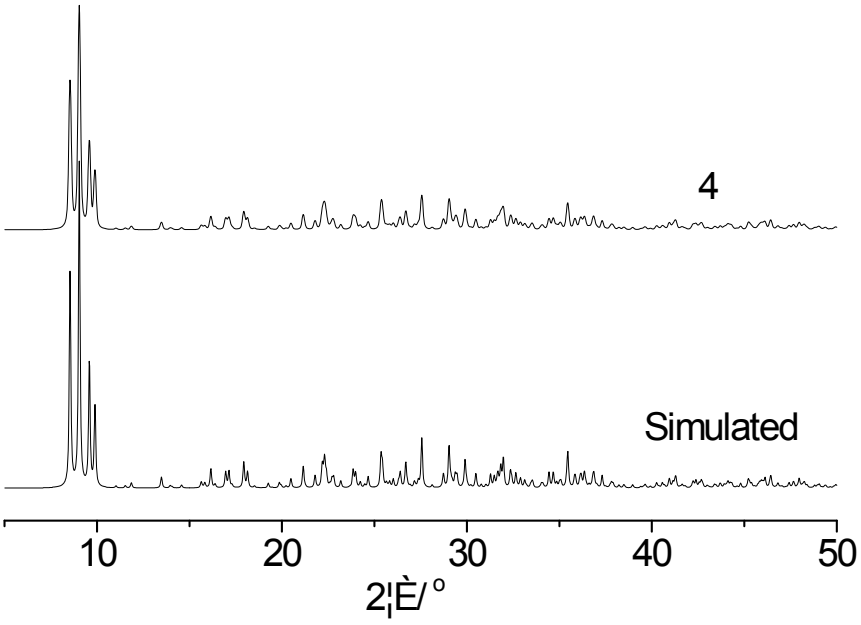



3
Simulated
10
20
30
40
50
2¦È/ °


4
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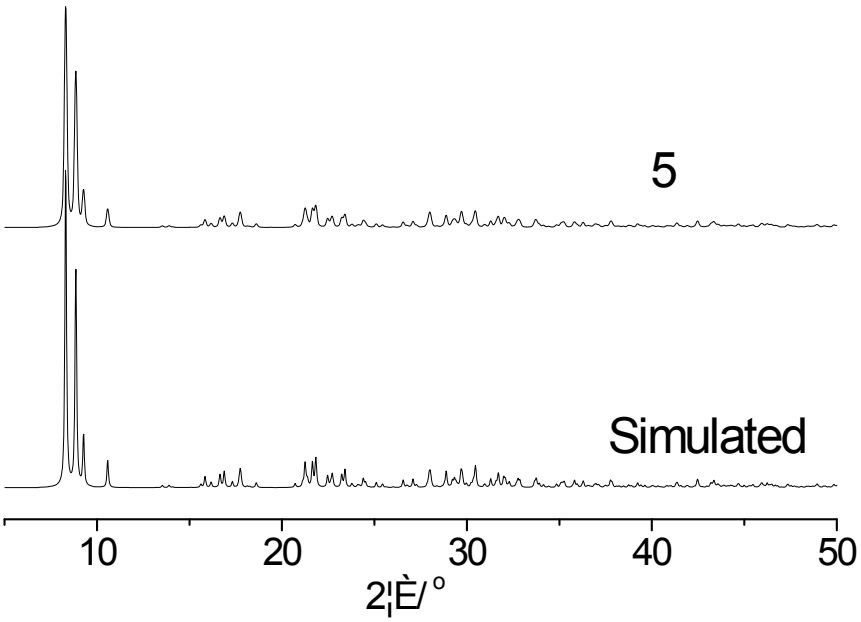


Fig. S1 Simulated and experimental powder XRD patterns of **1, 3, 4 and 5**.

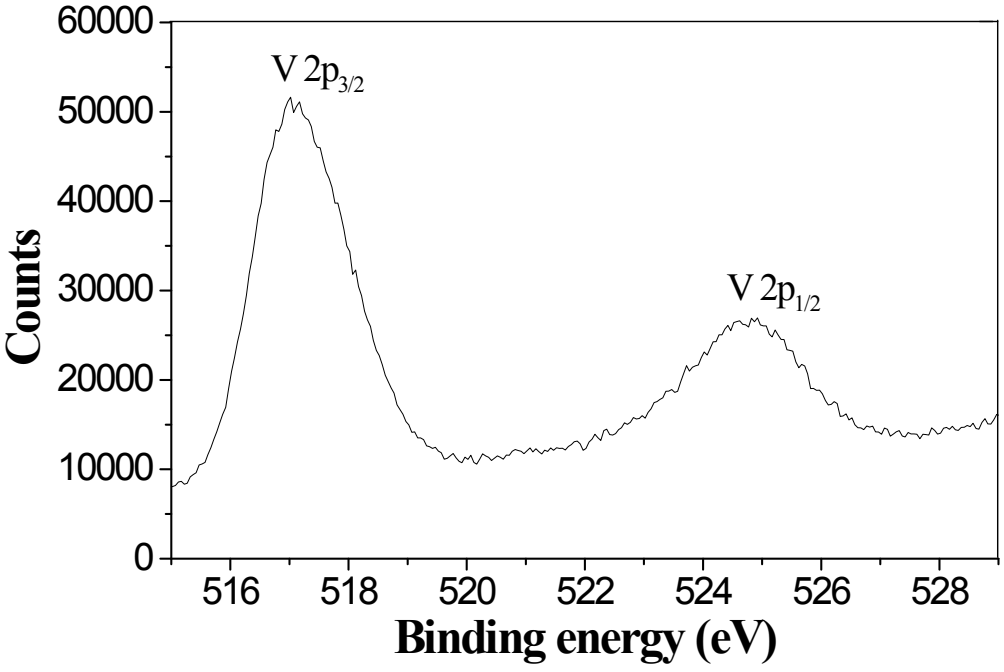


Fig. S2 Vanadium 2p XPS spectrum of **1**.

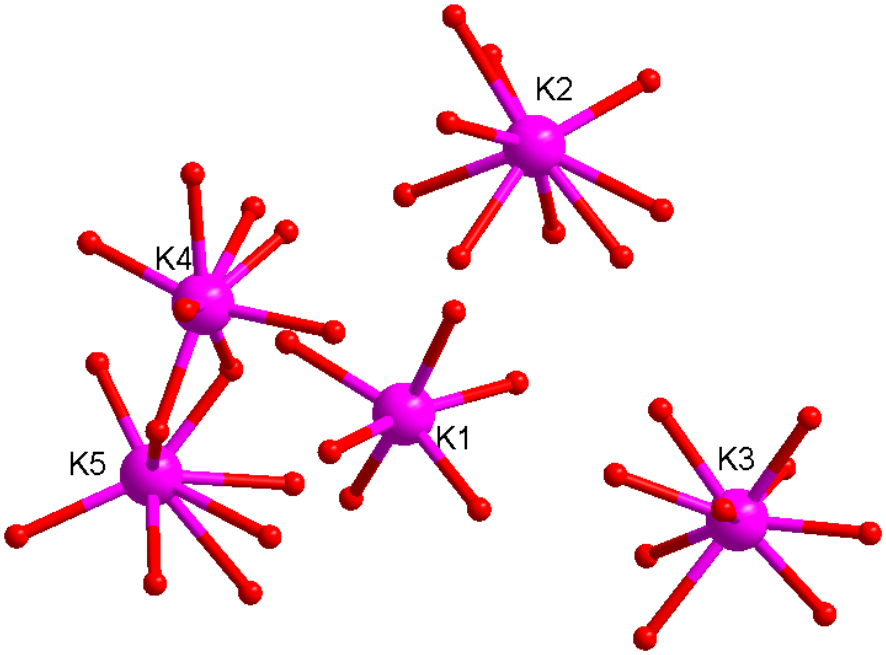


Fig. S3 The coordination environment of K^+ ion in **2**.

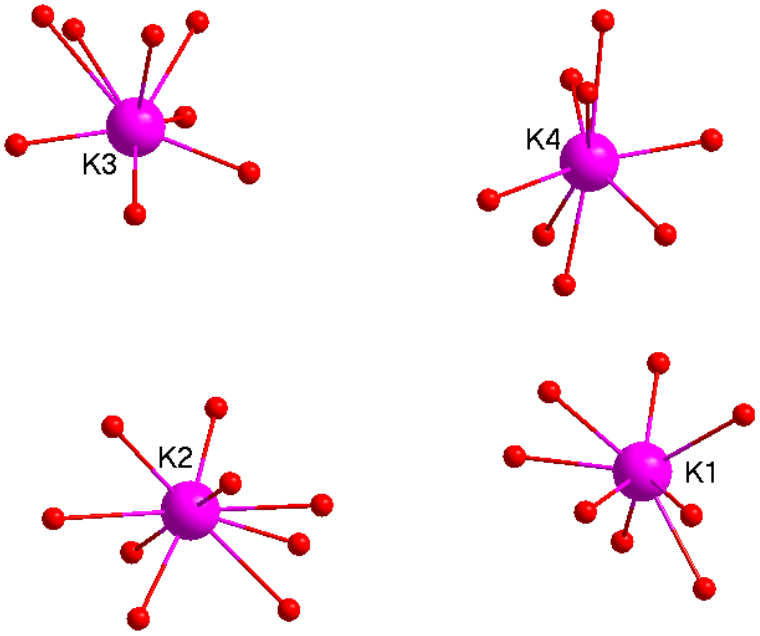


Fig. S4 The coordination environment of K^+ ion in **3**.

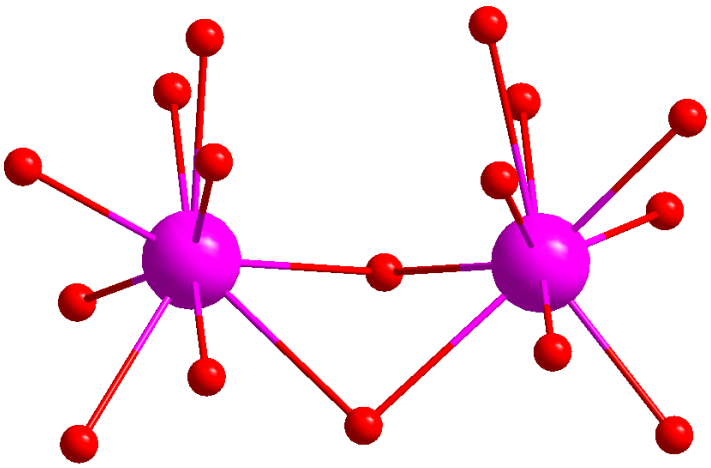
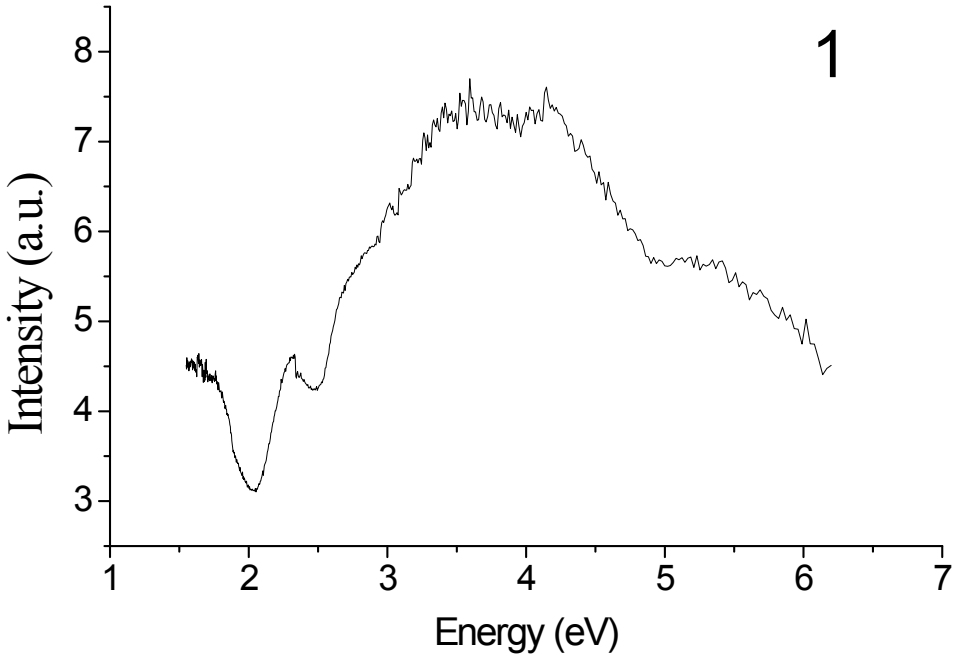


Fig. S5 Structure of dimeric [K_2O_{16}] unit in **5**.



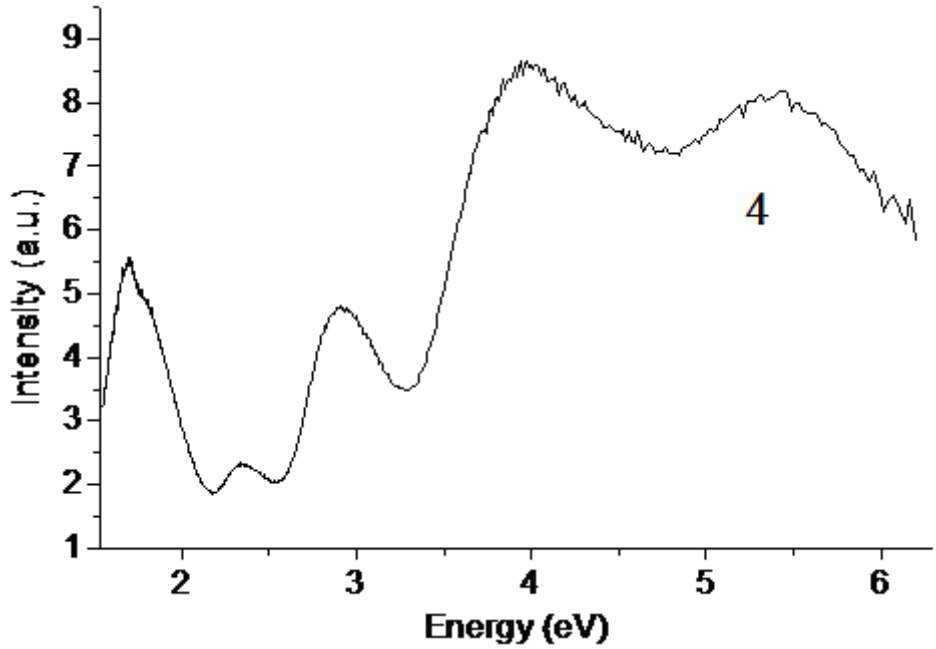


Fig. S6 the solid-state reflectance spectra of **1** and **4**.

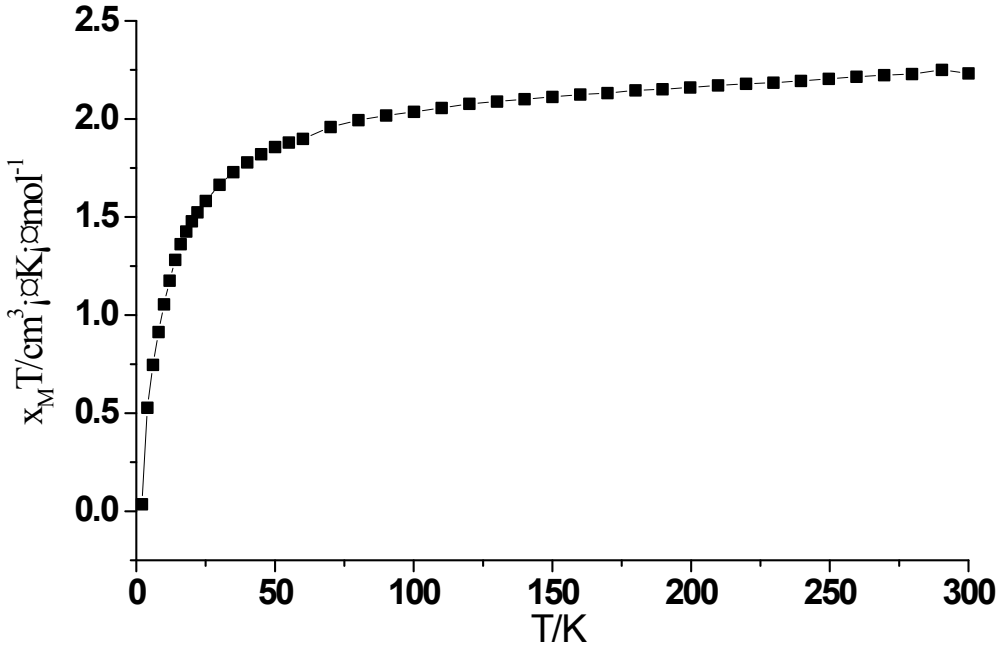


Fig. S7 Plot of $\chi_M T$ vs T for **4**.

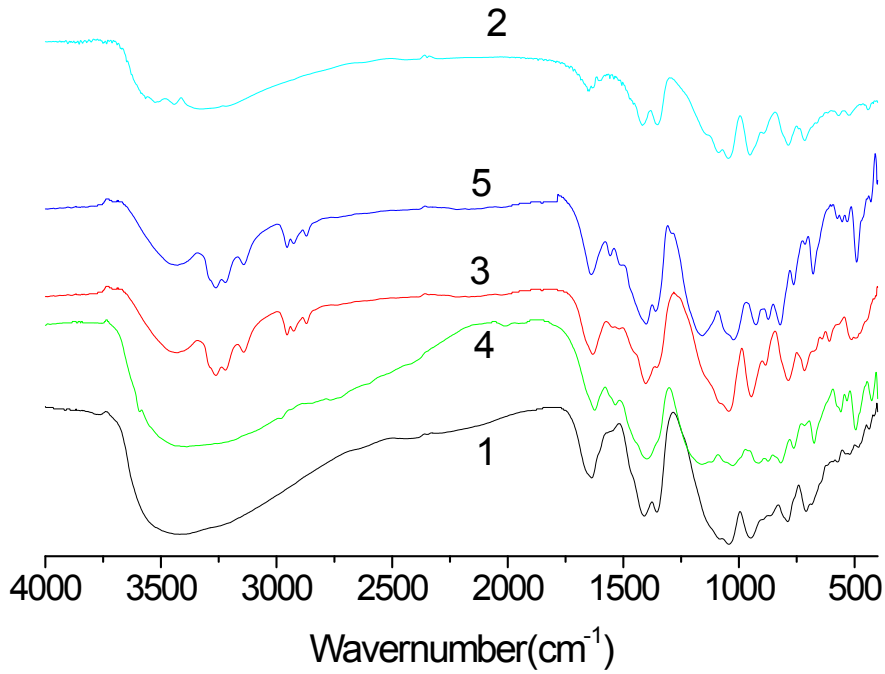


Fig. S8 The IR spectra of **1**-**5**.

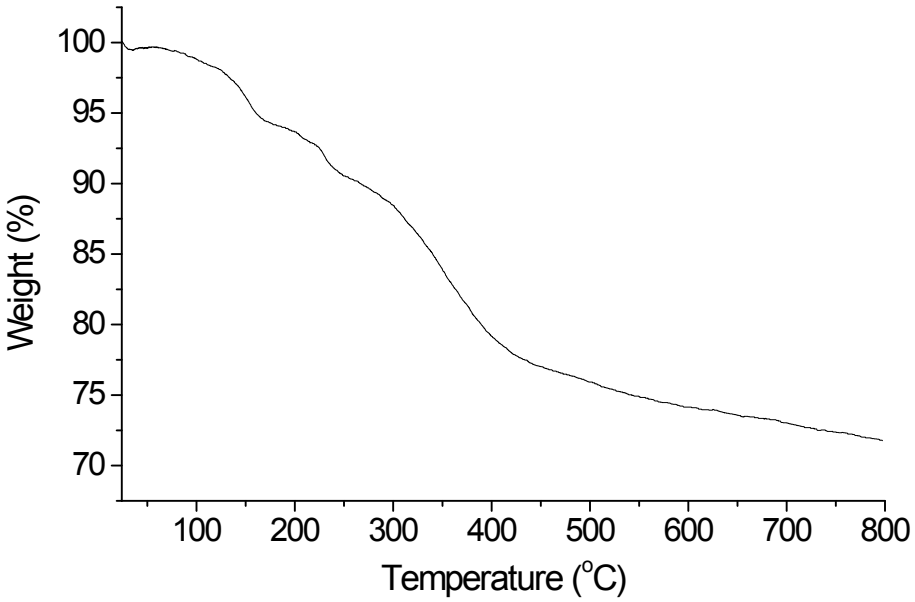


Fig. S9 Thermogravimetric analyses curves of **5**.

Thermogravimetric analysis of **5** was carried out in N_2 from 25 to 800 ºC with a heating rate of 10 ºC·min^{-1}. A three-step weight loss occurs in the range of 25-700 ºC which is related to the removal of free water molecules, coordinated water molecules, -OH groups and the H_3dien^{3+} ions (found, 27.0 %; calcd, 26.5 %).