

Supporting Information

Interplay between hydrogen bonding and metal coordination in alkali metal tartrates and hydrogen tartrates

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1. Analysis of structure similarity relationships

1.1. General

Crystal packing comparisons were carried out using version 2.0 of the program *XPac*¹ and quantitative dissimilarity parameters were generated in the previously described manner.² All comparisons were based on geometrical parameters generated from all C and O atoms of the anion, whereas metal centres and water molecules were not included in these calculations.

1.2. Supramolecular constructs

1.2.1. Connectivity motif I: chain type A-syn

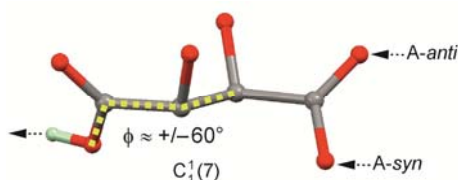


Figure S1. Definition of the geometries A-syn and A-anti chains of (carbonyl)O–H...O=C(carboxylate) bonded HTart[−] ions.

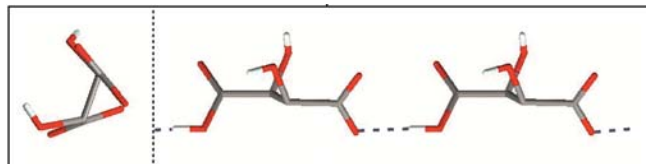


Figure S2. Chain type A-syn.

Table S1. Lattice vectors (*t*) associated with the chain type A-syn.

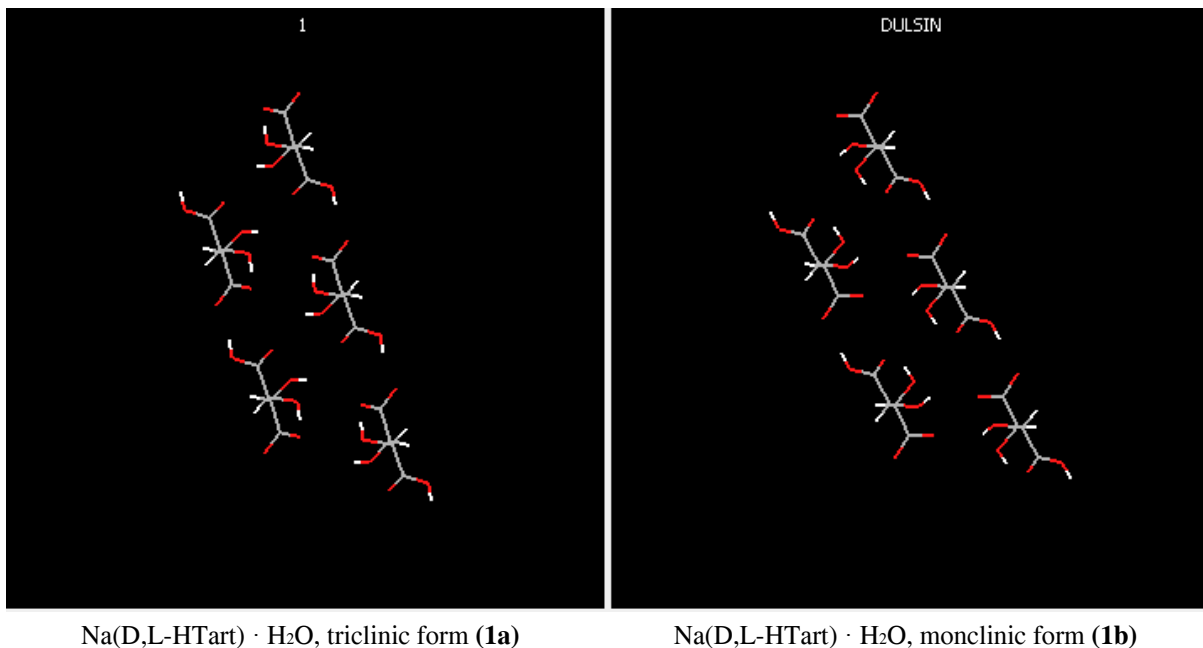
Structure	CSD	Ref.	<i>t</i>	Length (Å)
1a	.	.	010	7.065
1b	DULSIN	3	0 $\bar{1}$ 0	7.146
15	ZZZSS01	4	100	7.242

Table S2. *XPac* dissimilarity parameters *x* for pairwise structure comparisons involving a cluster of three anions which represents the A-syn chain.

Structure 1	Structure 2	<i>x</i>	Notes
1a	1b	2.2	<i>a</i>
1a	15	5.2	.
1b	15	5.8	<i>b</i>

a = The structures **1a** and **2b** have a double-stranded chain of H-bonded HTart[−] ions in common, whose two strands are A-syn chains (see Figure S3 and Figure 2c).

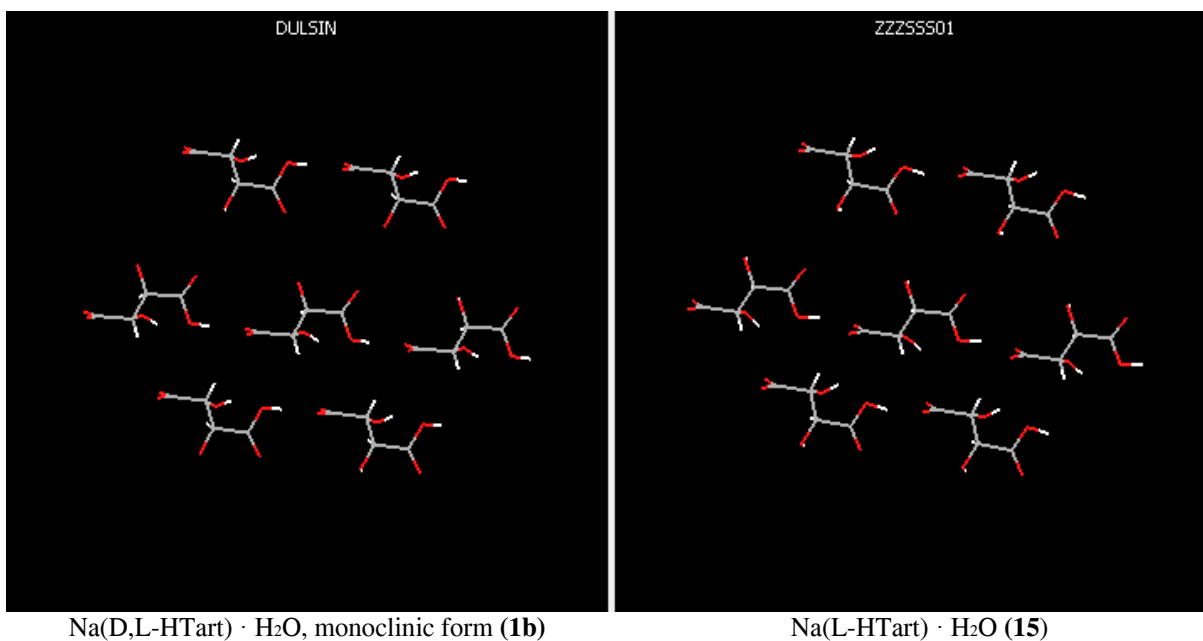
b = The structures **1b** and **15** have a layer in common which is composed of A-syn chains related by 2₁ symmetry (see Figure S4).



Na(D,L-HTart) · H₂O, triclinic form (**1a**)

Na(D,L-HTart) · H₂O, monoclinic form (**1b**)

Figure S3. Common 1D SC (supramolecular construct)¹ of structures **1a** and **1b**: double-stranded chain of H-bonded HTart⁻ ions, whose two strands are A-*syn* chains ($x = 2.2$ for the cluster of five HTart⁻ ions representing this SC).



Na(D,L-HTart) · H₂O, monoclinic form (**1b**)

Na(L-HTart) · H₂O (**15**)

Figure S4. Common 2D SC of structures **1b** and **15**: layer of H-bonded HTart⁻ ions composed of A-*syn* chains ($x = 8.1$ for the cluster of seven HTart⁻ ions representing this SC). Corresponding lattice parameters: **1b**: 010 / 7.146 Å, 101 / 10.887 Å, 90°; **15**: $\bar{1}00$ / 7.242 Å, 001 / 10.592, 90°.

1.2.2. Connectivity motif I: chain type A-anti

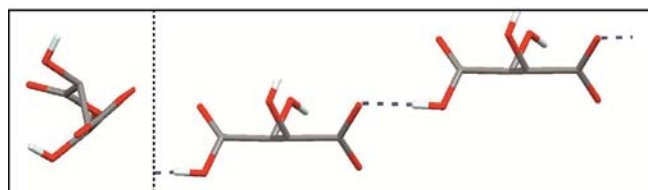


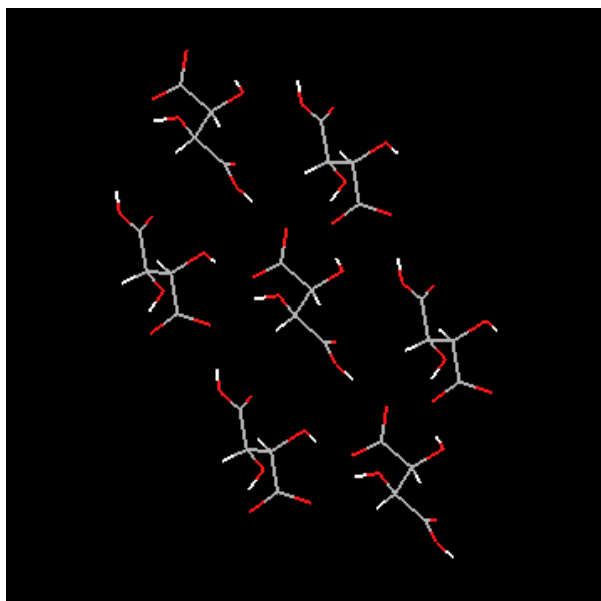
Figure S5. Chain type A-anti.

Table S3. Lattice vectors (t) associated with the chain type A-anti.

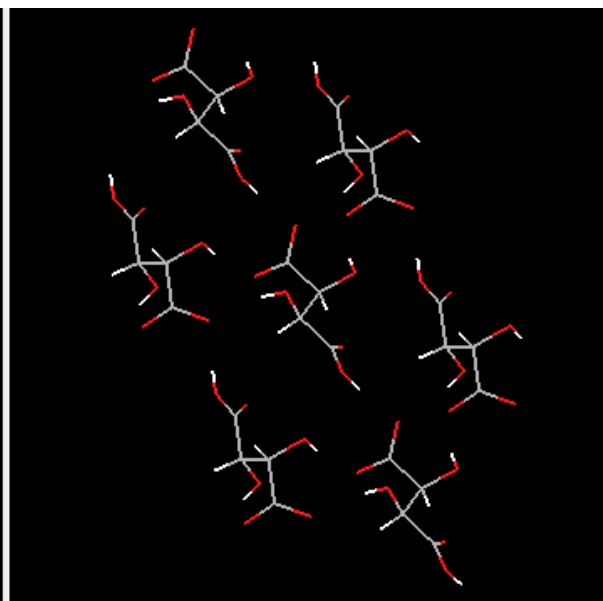
Structure	CSD	Ref.	t	Length (Å)
2b	XAHZIQ	5	100	7.683
5	XAHZAI	5	100	7.579
6	XAHZEM	5	100	7.626
13	YEKYIW	6	$\bar{1}00$	7.615
14	YELNIM	7	$00\bar{1}$	7.594
16	ZZZRZW01	8	001	7.604
17	KAMBIJ	9	001	7.653
18	CSHTAR10	10	001	7.692

Table S4. *XPac* dissimilarity parameters x for pairwise structure comparisons involving a cluster of three anions which represents the A-*syn* chain. The dissimilarity parameters x' refer to larger structure fragments, either the complete substructure of HTart⁻ ions (a) or to a layer of H-bonded HTart⁻ ions (b) represented by six anions (see Figure S6).

Structure 1	Structure 2	x	x'
13	14	9.8	.
13	16	3.3	.
13	17	3.5	.
13	18	4.2	.
13	5	3.4	.
13	6	3.6	.
13	2b	4.3	.
14	16	9.4	.
14	17	9.8	.
14	18	10.9	.
14	5	9.9	.
14	6	10.5	.
14	2b	11.0	.
16	17	1.2	1.6 ^(a)
16	18	2.7	3.7 ^(a)
16	5	2.4	2.1 ^(b)
16	6	1.8	2.1 ^(b)
16	2b	2.5	3.6 ^(b)
17	18	1.6	2.2 ^(a)
17	5	3.0	1.7 ^(b)
17	6	1.8	1.7 ^(b)
17	2b	1.7	2.5 ^(b)
18	5	4.1	4.1 ^(b)
18	6	2.5	2.6 ^(b)
18	2b	1.4	1.4 ^(b)
5	6	1.7	1.8 ^(a)
5	2b	3.5	3.9 ^(a)
6	2b	1.8	2.3 ^(a)



K(L-HTart) (**16**)



K(D,L-HTart) (**5**)

Figure S6. Common 2D SC present in the structures **2b**, **5**, **6** and **16 – 18**: layer of H-bonded HTart⁻ ions containing A-syn chains.

1.2.3. Connectivity motif II

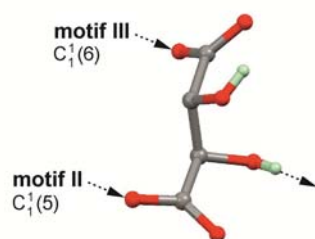


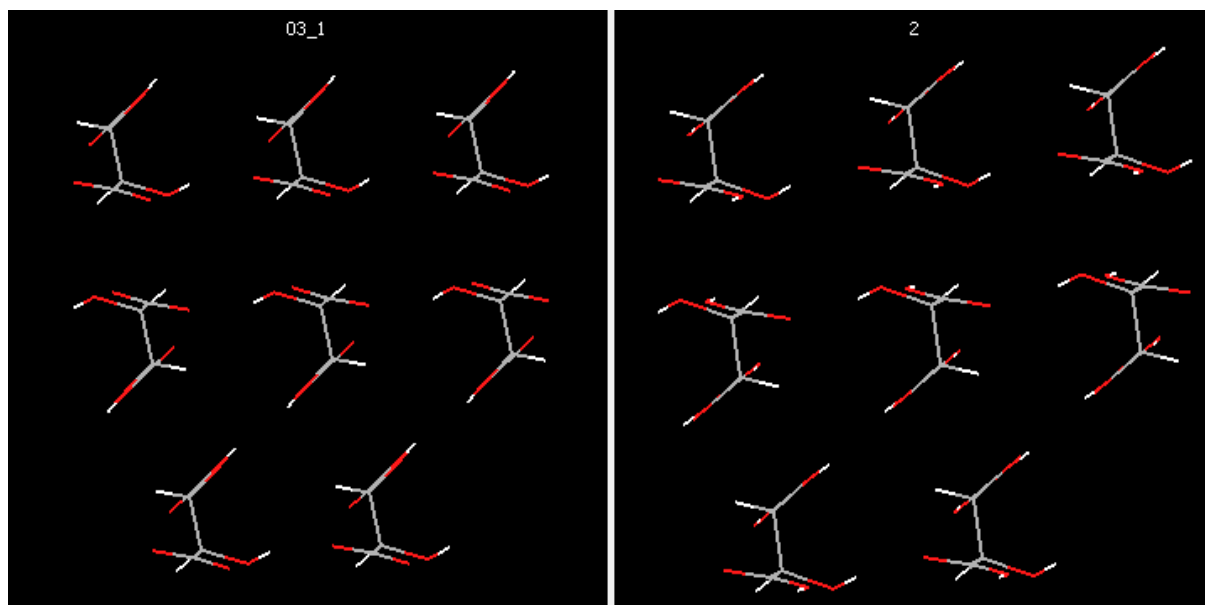
Figure S7. Definition of the connectivity motifs **II** and **III**.

Table S5. Lattice vectors (t) associated with the translation of the chain motif **II** and the geometry subsets **IIa** and **IIb**.

Structure	CSD	Ref.	t	Length (\AA)	Geometry
2a	.	.	100	5.065	IIa
3	.	.	100	5.073	IIa
12	UNIROZ	11	100	5.064	IIb
14	YELNIM	7	100	5.251	IIb
19a	UNIRUF	11	100	5.044	IIb
19c	.	12	010	5.014	.
20a	UNISIU	11	010	5.019	IIb
20b	.	12	0 $\bar{1}$ 0	4.970	IIb
23	NADTRT	13	001	4.959	IIa
28	ZZZLZE01	14	0 $\bar{1}$ 0	5.014	.

Table S6. *XPac* dissimilarity parameters x for pairwise structure comparisons involving a cluster of three anions. This cluster represents the geometry of the H-bonded chain which is based on motif **II**. The dissimilarity parameters x' refer to the larger 2D structure fragments shown in Figure S8 and Figure 5 (a) or in Figure S9 (b).

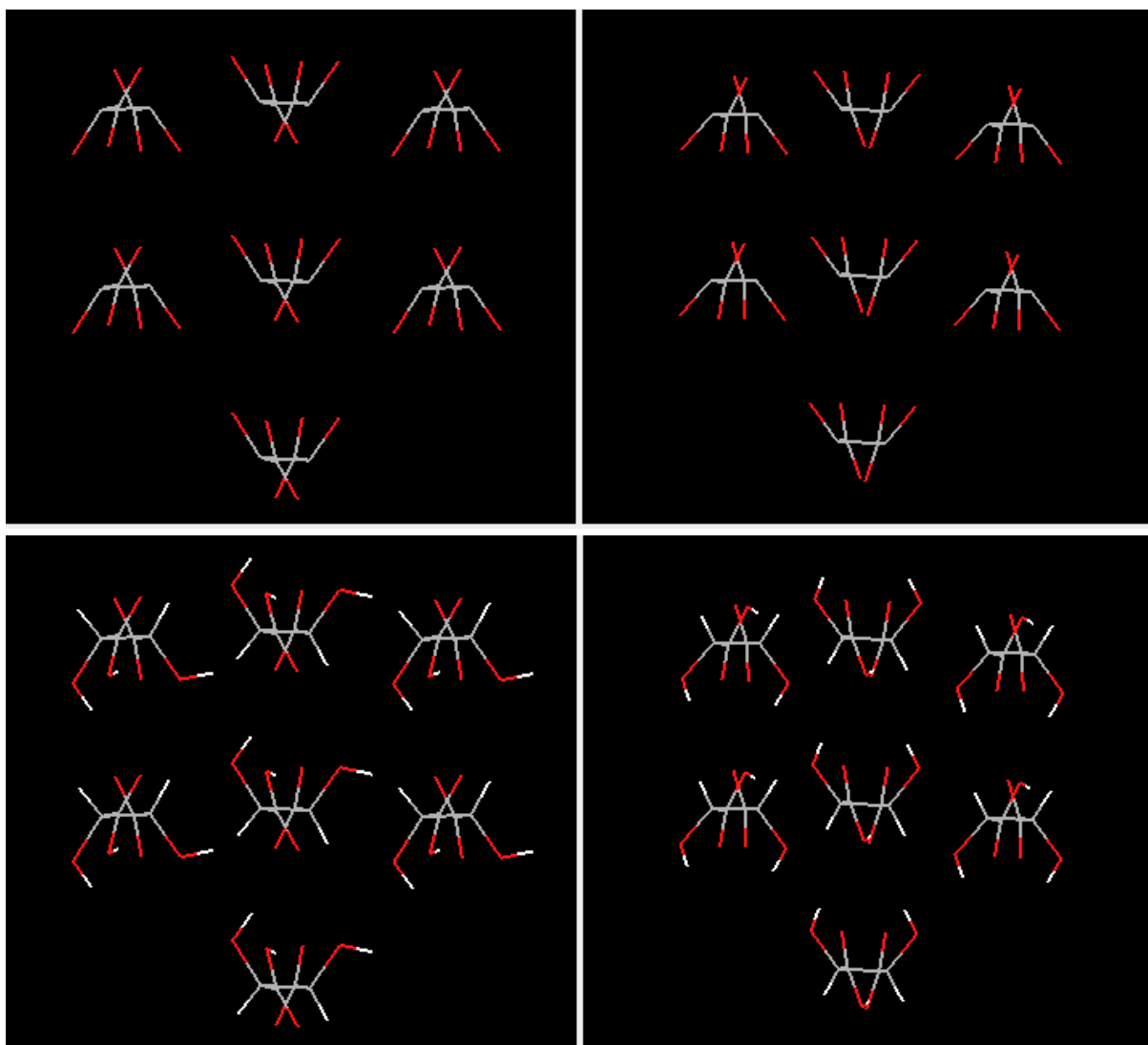
Structure 1	Structure 2	x	x'
Geometry subset IIa			
2a	3	4.4	3.4 ^(a)
2a	23	3.1	.
3	23	5.6	.
Geometry subset IIb			
12	14	7.0	8.3 ^(b)
12	19a	{ 8.8 10.7	.
12	20a	10.8	.
12	20b	11.5	.
14	19a	{ 4.7 6.2	.
14	20a	6.1	.
14	20b	9.3	.
19a	20a	{ 1.9 5.1	.
19a	20b	{ 5.8 10.9	.
20a	20b	10.5	.



LiCs(D,L-Tart) · 2H₂O (**3**)

Cs(D,L-HTart), triclinic form (**2a**)

Figure S8. Common 2D SC of structures **3** and **2a**: layer of H-bonded HTart⁻ ions composed of chains of the **IIa** geometry type. Corresponding lattice parameters: **3**: 100 / 5.073 Å, 00 $\bar{1}$ / 10.509 Å, 81.2°; **2a**: 100 / 5.064 Å, 001 / 9.957 Å, 85.5°.



Li(L-HTart) (**12**)

Na(L-HTart) (**14**)

Figure S9. Common 2D SC of structures **12** and **14**: layer of H-bonded HTart⁻ ions composed of chains of the **IIb** geometry type. Corresponding lattice parameters: **12**: 100 / 5.064 Å, 010 / 8.330 Å, 90°; **14**: 100 / 5.064 Å, 010 / 7.782 Å, 90°. The top and bottom diagrams show the same layer of anions without and with H atoms. Note the differences in the H positions between the two structures.

2. Crystal packing arrangement in the series $M(\text{D,L-HTart})$

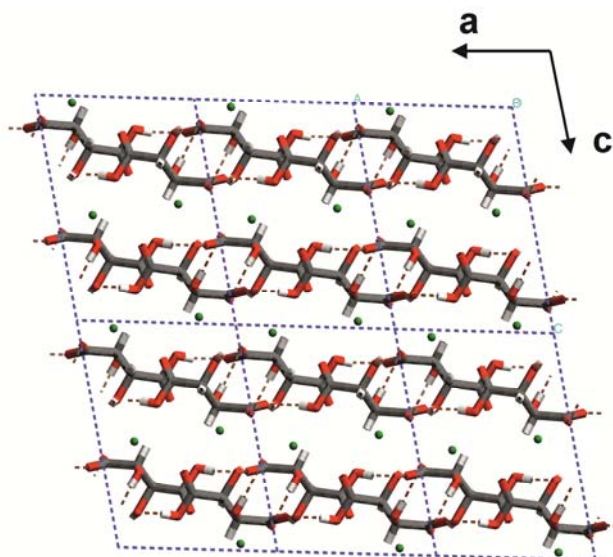


Figure S10. The packing of HTart^- (rods) and M^+ (balls) ions in the crystal structure of the series $M(\text{D,L-HTart})$ with $M = \text{Cs}$ (**2b**), K (**5**) and Rb (**6**) (view along $[010]$).⁵ The M^+ ions (green balls) are arranged between two layers which are composed exclusively of H-bonded D-HTart^- or L-HTart^- ions and which lie parallel to (110) .

3. Crystallisation experiments

Table S7. Overview of crystalline phases obtained from mixtures of 0.2 molar aqueous solutions of alkali metal or ammonium hydroxides (*MOH*) and D,L-tartaric acid (D,L-H₂Tart) in the wells of a 96 well plate (crystallisation at room temperature). No. = number of crystallization experiment (multiple phases were identified for nos. 8, 9, 15, 16, 23, 27); Rat. = ratio *MOH* : D,L-H₂Tart or ratio *MOH* : *M'*OH : D,L-H₂Tart; *d* = number of days from the preparation of the mixture to the harvest of the crystal; SXR / UC: phase identification on the basis of SXR unit cell data; SXR / FS = phase identification on the basis of a full structure determination; # = label of the phase in Table 5.

No.	<i>M</i>	<i>M'</i>	Rat.	<i>d</i>	SXR / UC	FS	Phase	#	CSD refcode	Ref.
1	Li	-	1:1	15	+	+ ^c	Li(D,L-HTart) · H ₂ O	13	YEKYIW01	5
2	Na	-	1:1	7	+	+ ^d	Na(D,L-HTart) · H ₂ O ^e	1a	.	<i>this work</i>
3	K	-	1:1	7	+	+ ^b	K(D,L-HTart)	5	XAHZAI	5
4	Rb	-	1:1	7	+	+ ^b	Rb(D,L-HTart)	6	XAHZEM	5
5	Cs	-	1:1	7	+	+ ^d	Cs(D,L-HTart) ^e	2a	.	<i>this work</i>
6	NH ₄	-	1:1	7	+	+ ^b	NH ₄ (D,L-HTart)	.	PUXKAU01	5
7	Li	-	2:1	20	+	+ ^c	Li ₂ (D,L-Tart) · 3H ₂ O	7	CEGPEK	15
8a	Na	-	2:1	30	+		Na ₂ (D or L-Tart) · 2H ₂ O	23	NADTRT	13
8b	Na	-	2:1	365	+	+	Na ₂ (D,L-Tart)	24	COZGED	16
9a	K	-	2:1	20	+		K(D,L-HTart)	5	XAHZAI	5
9b	K	-	2:1	365	+	+	D,L-Tart · H ₂ O	.	TARTDL01	17
10	Rb	-	2:1	30	+		Rb ₂ (D or L-Tart)	29	ZZZVZO02	18
11	Cs	-	2:1	15	+	+	Cs ₂ (D or L-Tart)	30	SOFJOM	18
12	NH ₄	-	2:1	20	+	+	NH ₄ (D,L-Tart)	.	ZZZJH01	19
13	Li	Na	1:1:1	365	+	+ ^c	LiNa(D,L-Tart) · 2H ₂ O	8	CEGPI0	15
14	Li	K	1:1:1	20	+	+ ^c	LiK(D,L-Tart) · H ₂ O	9	JEFVIA	15
15a	Li	Rb	1:1:1	20	+	+ ^c	LiRb(D,L-Tart) · H ₂ O	10	JEFVOG	15
15b	Li	Rb	1:1:1	300	+	+	Rb(D or L-HTart)	17	KAMBIJ	9
16a	Li	Cs	1:1:1	20	+	+ ^c	LiCs(D,L-Tart) · H ₂ O	11	CEGPOU	15
16b	Li	Cs	1:1:1	20	+	+ ^d	LiCs(D,L-Tart) · 2H ₂ O	3	.	<i>this work</i>
17	Li	NH ₄	1:1:1	20	+	+ ^c	LiNH ₄ (D,L-Tart) · H ₂ O	.	ZZZKDE01	15
18a	Na	K	1:1:1	20	+		K(D,L-HTart)	5	XAHZAI	5
18b	Na	K	1:1:1	20	+		Na(D,L-HTart) · H ₂ O	1a	.	<i>this work</i>
19	Na	Rb	1:1:1	^a		
20	Na	Cs	1:1:1	15	+	+	Na ₂ CO ₃ · H ₂ O	.	.	20
21	Na	NH ₄	1:1:1	300	+	+	Na ₂ CO ₃ · NaHCO ₃ · 2H ₂ O	.	.	21
22	K	Rb	1:1:1	7	+	+ ^b	K _{0.5} Rb _{0.5} (D,L-HTart)	.	XAHZAI01	5
23b	K	Cs	1:1:1	20	+		K(D,L-HTart)	5	XAHZAI	5
23a	K	Cs	1:1:1	20	+	+	CsHCO ₃	.	.	22
24	K	NH ₄	1:1:1	7	+	+ ^b	K _{0.56} (NH ₄) _{0.44} (D,L-HTart)	.	XAHZAI02	5
25	Rb	Cs	1:1:1	20	+	+ ^b	(Cs _{0.5} Rb _{0.5}) ₂ (D or L-Tart)	4	.	<i>this work</i>
26	Rb	NH ₄	1:1:1	20	+	+ ^b	Rb _{0.61} (NH ₄) _{0.39} (D,L-HTart)	.	XAHZEM01	5
27a	Cs	NH ₄	1:1:1	20	+	+ ^b	Cs(D,L-HTart) ^f	2b	XAHZIQ	5
27b	Cs	NH ₄	1:1:1	20	+		NH ₄ (D,L-HTart)	.	PUXKAU01	5

^a No crystalline product was obtained.

^b Crystal structure published in the first report.⁵

^c Crystal structure published in the second report.¹⁵

^d Crystal structure contained in this report.

^e Triclinic polymorph.

^f Monoclinic polymorph.

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