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Supporting Information

Calcium Oxalate Crystallization in the Presence of Amphiphilic

Phosphoproteins

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Fig. S1. SEM images of CaC_2O_4 obtained in the presence of 4.0 g L^{-1} casein kept in the air for five months



Fig. S2. FTIR spectra of CaC_2O_4 obtained in the presence of 4.0 g L⁻¹ casein kept in the air for five months, $[CaCl_2] = 1$ mM.



Fig. S3. XRD patterns of CaOx samples obtained at different reaction conditions. (A) $[CaCl_2] = 1$ mM, casein concentrations: 0.2 g L⁻¹ (a), 1 g L⁻¹ (b), and 6.0 g L⁻¹ (c), 25 ± 1 °C, pH 6.9; (B) $[CaCl_2] = 1$ mM, 4.0 g L⁻¹ casein, pH 6.9, temperatures: 10 ± 1 °C (a), and 37 ± 1 °C (b); (C) $[CaCl_2] = 1$ mM, 4.0 g L⁻¹ casein, 25 °C, pH 10.5; (D) $[CaCl_2] = 2$ mM, 4.0 g L⁻¹ casein, 25 °C, pH 6.9.



Fig. S4. FTIR spectra of CaOx samples obtained in the presence of 4.0 g L^{-1} casein at pH 3.5 (a) and 10.5 (b).



Fig. S5. (a, c) SEM image of CaOx obtained in the absence of casein at 25 ± 1 °C, [CaCl₂] = 1 mM, showing flowerlike COM agglomerates and hexagonal COM at pH 3.5 (a) and showing mainly tabular COT and flowerlike COM agglomerates at pH 10.5 (c); (b, d) XRD patterns of the corresponding CaOx obtained in the absence of casein at pH 3.5 (b) and 10.5 (d).



Fig. S6. FT-IR spectra of CaOx samples at the early reaction stages. The reaction time is (a) 3, (b) 8, (c) 12 and (d) 24 h.