Supplementary Information

Coordination of Ln³⁺ in Ortho-tetramethyl Substituted Cucurbituril Supramolecular Assemblies Formed in the Presence of Cadmium Nitrate: Potential Applications for Isolation of Heavier Lanthanides

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SI-Figure 1 (a) ITC profile of *o*-TMeQ[6] with Ce(NO₃)₃ at 298.15 K: ΔH° = -8.903 kJ mol⁻¹, $T\Delta S^{\circ}$ = 28.145 kJ mol⁻¹; (b) ITC profile of *o*-TMeQ[6] with Eu(NO₃)₃ at 298.15 K: ΔH° = -5.861 kJ mol⁻¹, $T\Delta S^{\circ}$ = 28.363 kJ mol⁻¹; (c) ITC profile of *o*-TMeQ[6] with Yb(NO₃)₃ at 298.15 K: ΔH° = -7.912 kJ mol⁻¹, $T\Delta S^{\circ}$ = 24.863 kJ mol⁻¹; (d) ITC profile of *o*-TMeQ[6]+Cd(NO₃)₂ with Ce(NO₃)₃ at 298.15 K: ΔH° = -6.824 kJ mol⁻¹, $T\Delta S^{\circ}$ = 28.232 kJ mol⁻¹; (e) ITC profile of *o*-TMeQ[6]+Cd(NO₃)₂ with Eu(NO₃)₃ at 298.15 K: ΔH° = -5.621 kJ mol⁻¹, $T\Delta S^{\circ}$ = 30.411 kJ mol⁻¹; (f) ITC profile of *o*-TMeQ[6]+Cd(NO₃)₂ with Yb(NO₃)₃ at 298.15 K: ΔH° = -5.621 kJ mol⁻¹, $T\Delta S^{\circ}$ = 30.411 kJ mol⁻¹; (f) ITC profile of *o*-TMeQ[6]+Cd(NO₃)₂ with Yb(NO₃)₃ at 298.15 K: ΔH° = -7.048 kJ mol⁻¹, $T\Delta S^{\circ}$ = 26.595 kJ mol⁻¹.



SI-Figure 2 the electron spectroscopies of o-TMeQ[6]/NO₃⁻ systems with 1:1 ratio of La:Yb, Ce:Yb, Pr:Yb and corresponding crystals with a general yields



SI-Figure 3 the electron spectroscopy of o-TMeQ[6]/NO₃⁻ systems with 1:1 ratios of light : heavy lanthanides



SI-Figure 4 the electron spectroscopy of *o*-TMeQ[6]/NO₃⁻ systems with 10:1, 5:1, 1:1, 1:5 and 1:10 ratios of Eu:Yb



SI-Figure 5 Powder X-ray diffraction (PXRD) of the representative crystals from the two isomorphous groups respectively



SI-Figure 6 DTA (left) and TG (right) curves of the representative crystals from the two isomorphous groups with a comparison o-TMeQ[6] in N₂ respectively



SI-Figure 7 FT-IR spectra of the representative crystals from the two isomorphous groups with a comparison of *o*-TMeQ[6] powders respectively.