Supporting information

(Ionic liquid)-derived morphology control of Nb₂O₅ materials and their photocatalytic properties

Lei Du, Zhouyang Long, Haimeng Wen, Weilin Ge, Yu Zhou*, Jun Wang* State Key Laboratory of Materials-Oriented Chemical Engineering, College of Chemistry and Chemical Engineering, Nanjing Tech University, Nanjing 210009, P. R. China

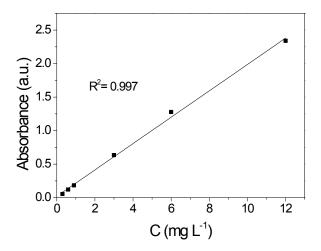


Figure S1. The calibration curve of methyene bule concentration. When the concentration of the methyene blue was above $12 \text{ mg} \cdot \text{L}^{-1}$, it was diluted by adding water to fit the scope of the standard curve.

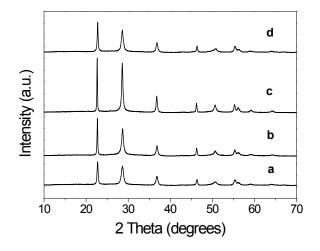


Figure S2. XRD patterns of Nb_2O_5 synthesized with the [Nb]:[[BMIm]Br] ratios of (a) 1:0.1, (b) 1:0.5, (c) 1:1, and (d) 1:3.

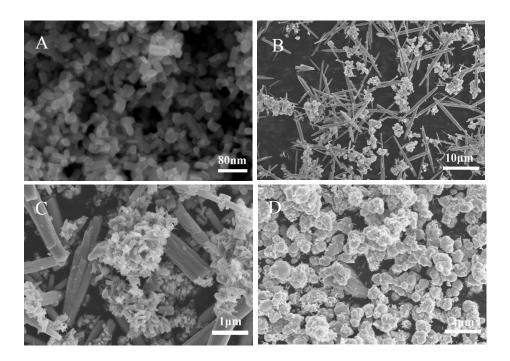


Figure S3. SEM images of the samples synthesized with the [Nb]:[[BMIm]Br] ratios of (A) 1:0.1, (B) 1:0.5, (C) 1:1, and (D) 1:3.

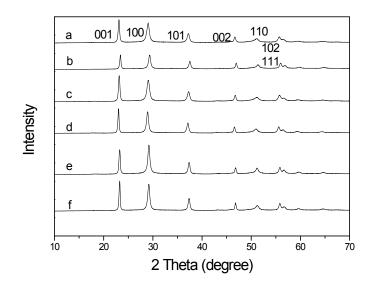


Figure S4. XRD patterns of Nb_2O_5 synthesized with molar ratio of [Nb]: [[BMIm]Br]=1:1.5 under different hydrothermal time: (a) 0, (b) 1, (c) 3, (d) 6, (e) 12 and (f) 24 h.

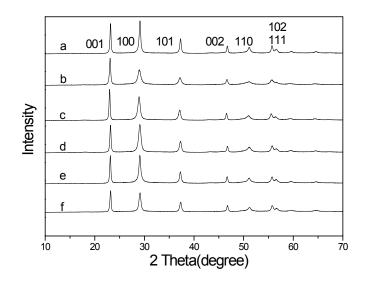


Figure S5. XRD patterns of Nb_2O_5 synthesized with molar ratio of [Nb]: [[BMIm]Br]=1:4 under different hydrothermal time: (a) 0, (b) 1, (c) 3, (d) 6, (e) 12 and (f) 24 h.

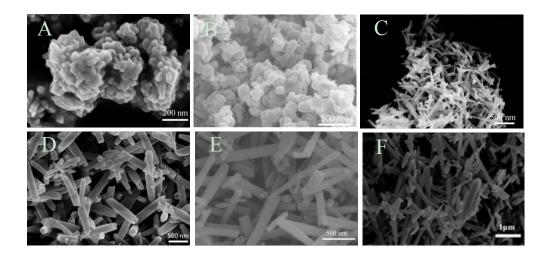


Figure S6. SEM images of Nb_2O_5 synthesized with molar ratio of [Nb]: [[BMIm]Br]=1:1.5 under different hydrothermal time: (a) 0, (b) 1, (c) 3, (d) 6, (e) 12 and (f) 24 h.

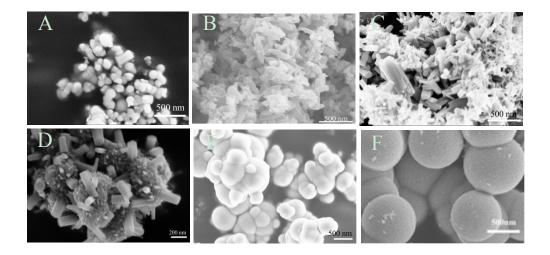


Figure S7. SEM images of Nb_2O_5 synthesized with molar ratio of [Nb]: [[BMIm]Br]=1:4 under different hydrothermal time: (a) 0, (b) 1, (c) 3, (d) 6, (e) 12 and (f) 24 h.

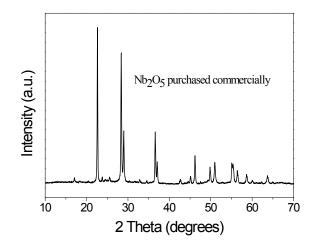


Figure S8. XRD pattern of the Nb_2O_5 raw material purchased commercially which was used to synthesize the Nb_2O_5 products with different morphologies in this work.