# A Series of Hybrids with a Framework Constructed from $\left\{\text { SiW }_{11}-\mathrm{Ln}-\mathrm{SiW}_{11^{-}}\right\}_{\mathrm{n}}$ Chains and $\{\mathbf{C u} /$ bimpy $\}$ Ribbons 



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## Electronic Supplementary Information



Fig .S1 IR spectra of compounds $\mathbf{1 - 5}$.


Fig. S2 The TG curves of compounds $\mathbf{1}$ (red) and $\mathbf{2}$ (black).


Fig. S3 The TG curves of compounds $\mathbf{3}$ (red), $\mathbf{4}$ (black) and $\mathbf{5}$ (green).


Fig. S4 The experimental (red line) and simulated (black line) PXRD patterns obtained from compound 1 .


Fig. S5 The experimental (red line) and simulated (black line) PXRD patterns obtained from compound 2.


Fig. S6 The experimental (red line) and simulated (black line) PXRD patterns obtained from compound 3 .


Fig. S7 The experimental (red line) and simulated (black line) PXRD patterns obtained from compound 4 .


Fig. S8 The experimental (red line) and simulated (black line) PXRD patterns
obtained from compound 5 .


Fig. S9 The cyclic voltammograms of 1-CPE in $1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ solution at different scan rates (from inner to outer: $0.08,0.12,0.14,0.16 \mathrm{~V} \cdot \mathrm{~S}^{-1}$ ).


Fig. S10 The cyclic voltammograms of 2-CPE in $1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ solution at different scan rates (from inner to outer: $0.08,0.12,0.14,0.16 \mathrm{~V} \cdot \mathrm{~S}^{-1}$ ).


Fig. S11 The cyclic voltammograms of 3-CPE in $1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ solution at different scan rates (from inner to outer: $0.08,0.12,0.14,0.16 \mathrm{~V} \cdot \mathrm{~S}^{-1}$ ).


Fig. S12 The cyclic voltammograms of 4-CPE in $1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ solution at different scan rates (from inner to outer: $0.08,0.12,0.14,0.16 \mathrm{~V} \cdot \mathrm{~S}^{-1}$ ).


Fig. S13 The cyclic voltammograms of 5-CPE in 1M H2SO4 solution at different scan rates (from inner to outer: $0.08,0.12,0.14,0.16 \mathrm{~V} \cdot \mathrm{~S}-1$ ).

Table S1 Bond valence sum (BVS) calculations of all $\mathrm{Ln}, \mathrm{Cu}$ and W atoms in compounds 1-5.

| Atoms | BVS values (1) | BVS values (2) | BVS values (3) | BVS values (4) | BVS values (5) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Ln | 3.07 | 3.08 | 2.98 | 3.06 | 3.08 |
| Cu1 | 2.05 | 2.04 | 2.12 | 2.02 | 2.00 |
| Cu2 | 2.04 | 2.05 | 2.04 | 2.02 | 1.99 |
| Cu3 | 2.00 | 2.01 | 2.00 | 2.01 | 2.00 |
| W1 | 6.25 | 6.05 | 6.29 | 6.21 | 5.91 |
| W2 | 6.34 | 5.96 | 6.11 | 6.33 | 6.19 |
| W3 | 6.22 | 6.20 | 6.33 | 6.10 | 6.15 |
| W4 | 6.09 | 6.21 | 6.25 | 6.15 | 6.33 |
| W5 | 6.32 | 6.19 | 6.24 | 6.19 | 6.24 |
| W6 | 6.28 | 6.30 | 6.30 | 6.31 | 6.20 |
| W7 | 6.20 | 6.11 | 6.14 | 6.49 | 6.21 |
| W8 | 6.25 | 5.95 | 6.20 | 6.03 | 6.24 |
| W9 | 6.37 | 6.32 | 6.39 | 6.03 | 6.38 |
| W10 | 6.27 | 6.06 | 6.08 | 6.16 | 6.25 |
| W11 | 6.18 | 6.04 | 6.10 | 6.19 | 6.25 |

