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Supporting Information for the manuscript

Biguanide and squaric acid as pH-dependent building blocks in crystal engineering.

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<i>D</i> —H···· <i>A</i>	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H···· A
		1a		
O(1)-H(1)···O(6)	1.048(4)	1.421(4)	2.447(3)	164.6(4)
O(5)-H(5)···O(2)	1.022(4)	1.569(4)	2.568(3)	164.5(4)
$N(2)-H(2)\cdots O(4)^{i}$	1.034(4)	1.779(4)	2.742(2)	153.2(3)
N(5)-H(5B)····O(3)	1.019(4)	1.788(4)	2.743(2)	154.5(3)
$N(3)-H(3B)\cdots O(4)^i$	1.017(4)	1.893(4)	2.789(2)	145.2(3)
$N(5)-H(5A)\cdots O(7)^{ii}$	1.023(4)	1.838(4)	2.822(2)	160.5(4)
$N(3)-H(3A)\cdots O(8)^{ii}$	1.014(4)	1.849(4)	2.824(2)	160.1(3)
$N(4)-H(4A)\cdots O(3)^{iii}$	1.015(4)	1.900(4)	2.856(2)	155.7(4)
N(4)-H(4A) ···N(5)	1.015(4)	2.456(5)	2.9061(16)	106.1(3)
N(5)-H(5B)···N(4)	1.019(4)	2.479(4)	2.9061(16)	104.5(2)
$N(4)-H(4B)\cdots O(7)^{iv}$	1.010(4)	2.068(5)	3.008(2)	154.1(4)

Table S1. Complete table of hydrogen bonds (Å ,°); non-classical contacts with $D \cdots A > 3.1$ Å have not been included.

i = -1 + x, -1 + y, z; ii = x, y, 1 + z; iii = 2 - x, 1 - y, 2 - z; iv = 2 - x, 1 - y, 1 - z

D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	<i>D</i> —H··· <i>A</i>
		1b		
$O(5)-H(151)\cdots O(1)^{i}$	0.86(3)	1.92(4)	2.664(3)	144(4)
$O(5)-H(150)\cdots O(6)^{ii}$	0.83(2)	1.87(2)	2.697(3)	177(4)
N(2)–H(20)····O(5)	0.91(2)	1.82(2)	2.717(3)	166(2)
N(5)-H(50)····O(4)	0.90(2)	1.88(3)	2.763(3)	164(2)
O(6)–H(161)····O(2)	0.85(3)	1.92(3)	2.768(3)	171(3)
O(6)–H(160)····O(2) ⁱⁱⁱ	0.86(3)	1.93(3)	2.783(3)	173(3)
N(3)-H(30)····O(3)	0.92(2)	1.89(2)	2.792(3)	165(2)
N(4)–H(41)····O(3) ^{iv}	0.893(18)	1.942(19)	2.804(3)	162(3)
$N(3)-H(31)\cdots O(4)^{\nu}$	0.894(19)	1.964(19)	2.833(3)	164(3)
$N(5)-H(51)\cdots O(1)^{i}$	0.90(3)	1.99(3)	2.866(3)	164(2)
$N(4)-H(40)\cdots O(4)^{iii}$	0.87(2)	2.04(2)	2.882(3)	162(3)
N(4)-H(40)···N(3)	0.87(2)	2.55(3)	2.932(3)	107(2)
N(3)-H(31)····N(4)	0.894(19)	2.56(3)	2.932(3)	106(2)

i=2-x, 1-y, -1/2+z; ii=x, 1+y, z; iii=3/2-x, y, -1/2+z; iv=1-x, 1-y, -1/2+z; v=-1/2+x, 1-y, z.

D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H···A	
		1c			
N(7)-H(70)···O(3) ^{<i>i</i>}	0.908(15)	1.738(16)	2.609(2)	159.9(19)	
$N(2)-H(20)\cdots O(1)^{ii}$	0.949(16)	1.692(16)	2.621(2)	165.3(19)	
$O(9)-H(109)\cdots O(7)^{iii}$	0.89(4)	2.05(3)	2.700(2)	129(3)	
$O(10)-H(210)\cdots O(5)^{iv}$	0.901(17)	1.825(17)	2.723(2)	175.2(18)	
$N(10)-H(100)\cdots O(9)^{\nu}$	0.880(16)	1.898(17)	2.738(3)	159(2)	
$N(4)-H(41)\cdots O(8)^{iii}$	0.934(18)	1.832(18)	2.759(2)	171.3(17)	
N(3)-H(30)···O(10) ^{vi}	0.913(16)	1.893(17)	2.764(3)	159(2)	
$N(5)-H(51)\cdots O(2)^{ii}$	0.935(17)	1.859(17)	2.772(2)	165(2)	
N(8)–H(81)····O(5) ⁱⁱⁱ	0.94(2)	1.85(2)	2.777(2)	167.1(19)	
$N(9)-H(90)\cdots O(6)^{vii}$	0.937(19)	1.860(19)	2.792(2)	172.7(19)	
$N(8) - H(80) \cdots O(4)^{i}$	0.842(16)	1.981(16)	2.807(2)	166.9(16)	
O(9)−H(209)····O(3)	0.876(19)	1.944(19)	2.812(3)	171(2)	
$N(5)-H(50)\cdots O(7)^{vi}$	0.894(16)	2.040(16)	2.871(2)	153.9(14)	
N(10)–H(101)····O(6) ⁱⁱⁱ	0.87(2)	2.17(2)	2.876(2)	137.3(16)	
O(10)–H(110)····O(1)	0.82(2)	2.14(2)	2.878(2)	151(3)	
$N(4)-H(40)\cdots O(4)^{iii}$	0.92(2)	2.02(2)	2.898(2)	160.7(19)	
$N(9)-H(91)\cdots O(2)^{vi}$	0.917(18)	2.073(17)	2.905(2)	150.2(18)	
N(3)-H(31)····O(7) ^{vi}	0.91(2)	2.23(2)	2.974(2)	139.5(18)	
$N(3)-H(31)\cdots O(8)^{vi}$	0.91(2)	2.46(3)	3.099(2)	128.1(18)	
C(9)–H(9C)···N(4)	0.98	2.45	2.793(3)	100	
C(10)-H(10C)···N(2)	0.98	2.38	2.785(3)	104	
C(14)−H(14A)…N(7)	0.98	2.41	2.785(3)	102	

Table S1.Complete table of hydrogen bonds (Å ,°) - continued

i = -1 + x, -1 + y, z; ii = -x, -1/2 + y, 1 - z; iii = 1 - x, -1/2 + y, 1 - z; iv = 1 - x, 1/2 + y, 1 - z; v = 1 - x, -1/2 + y, 2 - z;

 $v_i = x, -1+y, z; v_{ii} = x, -1+y, 1+z.$

D—H···A	<i>D</i> —Н	H····A	$D \cdots A$	D—H···A
		1d		
O(3)–H(3A)····O(1)	0.857(16)	1.923(18)	2.7455(18)	161(2)
$N(3)-H(30)\cdots O(3)^{i}$	0.894(17)	1.956(18)	2.809(2)	159.1(15)
$O(3)-H(3B)\cdots N(2)^{ii}$	0.864(18)	1.997(18)	2.8517(18)	170(2)
N(5)-H(50)···O(1)	0.877(17)	2.010(18)	2.877(2)	169.8(15)
N(3)–H(31)···O(2) ⁱⁱⁱ	0.864(16)	2.104(16)	2.9036(17)	153.7(14)
$N(4) - H(40) \cdots O(2)^{i}$	0.886(17)	2.080(18)	2.9447(18)	164.9(15)
N(4)–H(41)···O(2) ^{iv}	0.877(16)	2.164(17)	2.9779(18)	154.1(14)
N(5)–H(51)····O(3) ⁱⁱⁱ	0.879(19)	2.368(18)	3.204(2)	159.1(19)
C(3)-H(3E)···N(2)	0.98	2.38	2.756(2)	102

Table S1. Complete table of hydrogen bonds (Å ,°) - continued

 i = x,1+y,z; ii = 1-x,1-y,1-z; iii = -x,1-y,1-z; iv = -x,1-y,-z.

D—H····A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H···A
		2a		
$O(1)-H(1)\cdots O(6)^{i}$	0.87(2)	1.64(2)	2.496(3)	168(4)
$N(6)-H(60)\cdots O(6)^{ii}$	0.91(2)	1.80(2)	2.706(3)	172(2)
N(7)–H(70)····O(8)	0.85(3)	1.91(2)	2.706(3)	155(2)
N(9)–H(90)····O(4) ^{<i>iii</i>}	0.92(2)	1.85(2)	2.744(3)	163(2)
N(10)–H(101)····O(3) ^{<i>iv</i>}	0.945(19)	1.857(18)	2.774(3)	163(3)
N(10)-H(100)···O(7)	0.94(2)	1.886(19)	2.810(3)	169(2)
N(8)–H(81)····O(2) ^{iv}	0.92(2)	1.91(2)	2.832(3)	172(2)
$O(30)-H(30C)\cdots O(6)^{\nu}$	0.84	2.49	2.845(8)	107
N(1)-H(10)····O(7)	0.92(2)	1.94(2)	2.846(3)	165(3)
$N(5)-H(51)\cdots O(8)^{vi}$	0.93(2)	1.963(19)	2.873(3)	165(3)
N(3)–H(31)····O(5) ^{vi}	0.93(2)	1.98(2)	2.892(3)	166(2)
N(3)–H(30)····O(2) ^{vii}	0.93(2)	2.05(3)	2.931(3)	158(3)
N(5)–H(50)····O(4) ^{iv}	0.90(2)	2.05(2)	2.946(3)	173(2)
N(4)–H(40)····O(1) ^{vii}	0.90(2)	2.21(2)	2.972(3)	142(2)
N(8)–H(80)····O(5) ^{<i>ii</i>}	0.90(2)	2.11(2)	2.978(3)	161(2)
N(4)-H(41)····N(3)	0.88(3)	2.55(2)	2.991(4)	112(2)
N(9)–H(91)····O(3)	0.91(3)	2.13(3)	3.007(3)	162(2)
O(30)–H(30C)····O(1) ^{viii}	0.84	2.36	3.178(7)	163
C(10)–H(10A)····N(2)	0.95	2.32	2.894(4)	118

Table S1. Complete table of hydrogen bonds (Å , $^{\circ}$) - continued

i = 1-x, 1/2+y, 1/2-z; ii = x, 3/2-y, 1/2+z; iii = 1-x, 2-y, 1-z; iv = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; v = -x, -1/2+y, 1/2-z; vi = 1-x, 1-y, 1-z; vi = 1-x, 1-z; vi = 1-

= x, -1+y, z; vii = 1-x, -1/2+y, 1/2-z; viii = -1+x, -1+y, z.

<i>D</i> —H··· <i>A</i>	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	<i>D</i> —H…A
		2b		
N(6)-H(160)····O(2)	0.940(18)	1.854(19)	2.766(2)	163.1(18)
$N(4) - H(141) \cdots O(4)^{i}$	0.933(18)	1.879(19)	2.790(2)	164.9(18)
N(10)-H(201)····O(2) ^{<i>ii</i>}	0.887(18)	1.936(18)	2.811(2)	168.8(18)
N(5)-H(150)···O(4) ⁱⁱⁱ	0.931(18)	1.929(18)	2.835(2)	163.7(16)
N(8)-H(180)···O(1) ^{<i>ii</i>}	0.93(2)	1.94(2)	2.856(2)	167.8(18)
N(3)-H(131)····O(3) ⁱⁱⁱ	0.909(19)	2.025(19)	2.911(2)	164.5(17)
$N(5)-H(151)\cdots N(7)^{iv}$	0.847(17)	2.079(17)	2.921(3)	173.1(17)
$N(3)-H(130)\cdots O(1)^{i}$	0.880(18)	2.15(2)	2.952(2)	151.2(18)
$N(4)-H(140)\cdots O(1)^{\nu}$	0.903(19)	2.137(19)	2.954(2)	150.2(19)
N(8)-H(181)····O(3)	0.90(2)	2.09(2)	2.968(2)	165.1(19)
N(9)-H(190)···O(3) ^{v_i}	0.91(2)	2.073(19)	2.973(2)	172(2)
N(8)-H(181)····N(6)	0.90(2)	2.58(2)	3.013(3)	109.9(17)
N(3)-H(130)···N(4)	0.880(18)	2.59(2)	3.013(3)	110.2(16)
N(10)–H(200)····N(2) ^{iv}	0.943(19)	2.076(19)	3.018(3)	177.3(16)
N(9)-H(191)····O(2)	0.910(19)	2.50(2)	3.147(2)	128.8(17)
C(9)–H(9)…N(2)	0.95	2.34	2.923(3)	120
C(17)−H(17)····N(7)	0.95	2.57	3.081(3)	114

Table S1. Complete table of hydrogen bonds (Å ,°) - continued

i = -x, 1-y, 1-z; ii = 1+x, y, z; iii = 1-x, 1-y, 1-z; iv = 1-x, 2-y, 1-z; v = x, y, -1+z; vi = 1-x, 2-y, 2-z.

D—H···A	<i>D</i> —Н	H····A	$D \cdots A$	D—H···A
		3 a		
N(2)-H(120)····O(3)	0.906(15)	1.751(15)	2.6410(19)	166.5(16)
N(1)-H(110)····O(5)	0.873(17)	1.872(17)	2.711(2)	160.5(17)
$N(4)-H(141)\cdots O(1)^{i}$	0.898(18)	1.874(18)	2.7358(19)	160.3(16)
O(5)–H(5A)····O(1)	0.838(18)	1.947(18)	2.7842(18)	178(2)
$O(5)-H(5B)\cdots O(1)^{ii}$	0.860(18)	1.97(2)	2.8060(18)	163(2)
$N(4) - H(140) \cdots O(2)^{i}$	0.936(17)	2.164(18)	2.813(2)	125.5(15)
N(3)-H(130)····O(2) ⁱⁱⁱ	0.899(18)	1.974(18)	2.847(2)	163(2)
N(3)-H(131)····O(4)	0.915(19)	1.974(19)	2.8856(19)	175(2)
N(5)-H(150)····O(4) ^{<i>iv</i>}	0.907(17)	2.001(17)	2.888(2)	165.5(17)
$N(5)-H(151)\cdots O(5)^{\nu}$	0.885(17)	2.133(17)	2.908(2)	145.9(15)
N(4)-H(141)····N(5)	0.898(18)	2.612(17)	2.951(2)	103.3(13)

Table S1. Complete table of hydrogen bonds (Å , $^{\circ}$) - continued

i = x, 1/2-y, -1/2+z; ii = 2-x, -y, 1-z; iii = 1+x, 1/2-y, -1/2+z; iv = x, 1/2-y, 1/2+z; v = 2-x, 1/2+y, 1/2-z.

D—H···A	D—H	H····A	$D \cdots A$	<i>D</i> —H···A
		3b		
$N(12)-H(112)\cdots O(11)^{i}$	0.892(16)	1.780(16)	2.651(3)	165(2)
N(7)-H(107)···O(8)	0.91(2)	1.77(2)	2.655(3)	166(3)
N(2)–H(102)···O(10) ^{<i>ii</i>}	0.899(19)	1.797(19)	2.691(3)	173(2)
N(14)–H(214)····O(3)	0.92(2)	1.85(2)	2.717(3)	158(2)
N(9)–H(209)····O(5) ⁱⁱⁱ	0.93(2)	1.84(2)	2.748(3)	167(2)
N(8)-H(108)····O(6)	0.90(3)	1.91(2)	2.753(3)	157(2)
$N(1)-H(101)\cdots O(9)^{ii}$	0.88(2)	1.89(2)	2.754(3)	166(2)
O(14)–H(24O)····O(9)	0.894(18)	1.879(18)	2.758(2)	168(3)
N(8)–H(208)···· O(7)	0.92(2)	1.85(2)	2.762(3)	173(3)
N(5)–H(205)···· O(14) ^{<i>iv</i>}	0.89(3)	1.88(3)	2.764(3)	174(3)
N(11)-H(111)····O(13)	0.89(2)	1.98(2)	2.778(3)	150(2)
$N(13)-H(213)\cdots O(12)^{i}$	0.91(2)	1.89(2)	2.785(3)	169(2)
$N(4)-H(104)\cdots O(1)^{\nu}$	0.90(2)	2.06(2)	2.793(3)	138(2)
$N(3)-H(203)\cdots O(1)^{ii}$	0.91(3)	2.03(2)	2.801(3)	142(2)
N(9)–H(109)····O(6) ⁱⁱⁱ	0.91(3)	2.18(3)	2.805(3)	126(2)
N(6)-H(106)O(15)	0.90(2)	2.00(2)	2.811(3)	150(2)
$N(13)-H(113)\cdots O(4)^{ii}$	0.92(2)	1.97(2)	2.820(3)	153(2)
$N(5)-H(105)\cdots O(2)^{vi}$	0.90(3)	1.94(3)	2.823(3)	163(2)
O(13)–H(23O)····O(5) ^{vii}	0.88(3)	1.96(3)	2.824(2)	168(3)
O(13)–H(13O)····O(5)	0.89(3)	1.94(3)	2.825(2)	173(3)
N(10)-H(210)···O(12)	0.89(2)	1.95(2)	2.831(3)	174(2)
O(15)–H(15O)····O(3) ^{viii}	0.85(2)	1.99(2)	2.834(3)	172(3)
N(14)-H(114)····O(4)	0.91(3)	2.13(3)	2.839(3)	134(2)
$N(15)-H(115)\cdots O(10)^{ii}$	0.89(2)	1.99(2)	2.853(3)	162(2)
O(14)–H(14O)····O(8)	0.859(18)	2.06(2)	2.862(2)	155(2)
$N(4)-H(204)\cdots O(2)^{\nu}$	0.93(3)	2.06(3)	2.870(3)	146(2)
$O(15)-H(25O)\cdots O(1)^{ix}$	0.85(3)	2.04(3)	2.888(3)	176(3)
$N(15)-H(215)\cdots O(4)^{ii}$	0.90(2)	2.15(3)	2.952(3)	149(2)
N(15)-H(115)···N(14)	0.89(2)	2.53(2)	2.956(4)	109.9(17)

Table S1.Complete table of hydrogen bonds (Å ,°) - continued

N(14)-H(214)····N(15)	0.92(2)	2.59(3)	2.956(4)	104.4(17)
$N(10)-H(110)\cdots O(13)^{x}$	0.92(2)	2.25(2)	2.996(3)	138(2)
N(3)-H(103)····O(3)	0.890(19)	2.20(2)	3.004(3)	151(2)
N(4)-H(104)····N(5)	0.90(2)	2.62(3)	3.048(4)	109.8(17)
$N(11)-H(111)\cdots O(11)^{i}$	0.89(2)	2.54(2)	3.066(3)	119(2)
C(31)–H(31B)····N(11)	0.98	2.42	2.880(3)	108

i = 1+x, 1+y, z; ii = 1+x, y, z; iii = -1+x, y, z; iv = 1-x, 1-y, -z; v = 1-x, 2-y, -z; vi = 2-x, 2-y, -z; vii = 2-x, 2-x, -z; vii = 2-x, 2-x, -z; vii = 2-x, 2-x, -

x,1-y,1-z; $v_{iii} = -1+x, -1+y, z; ix = x, -1+y, z; x = 1-x, 1-y, 1-z.$

D—H····A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H···A
		3c		
$O(1)-H(1)\cdots O(6)^{i}$	0.906(16)	1.607(16)	2.5120(14)	177(3)
N(7)–H(70)····O(8)	0.892(16)	1.890(17)	2.7432(15)	159.6(18)
$N(9)-H(90)\cdots O(4)^{ii}$	0.885(16)	2.042(16)	2.7865(17)	141.0(15)
N(10)-H(101)····O(7)	0.903(16)	1.902(16)	2.7940(15)	169.1(19)
N(8)–H(80) \cdots O(2) ^{<i>iii</i>}	0.908(15)	1.906(15)	2.8072(15)	171.3(16)
N(4)–H(40A)····O(7) ^{<i>iv</i>}	0.899(17)	1.966(17)	2.8223(17)	158.6(17)
$N(6)-H(60)\cdots O(6)^{\nu}$	0.887(17)	1.983(17)	2.8245(15)	158.1(17)
$N(10)-H(100)\cdots O(3)^{iii}$	0.891(16)	1.964(16)	2.8424(16)	168.8(16)
$N(8)-H(80A)\cdots O(5)^{\nu}$	0.886(16)	2.008(15)	2.8486(15)	157.8(14)
$N(3)-H(30A)\cdots O(2)^{i}$	0.890(18)	2.011(18)	2.8775(17)	164.3(17)
N(9)-H(90A)····O(3)	0.889(17)	2.032(18)	2.8775(17)	158.6(17)
N(3)-H(30)···O(5)	0.895(16)	2.012(16)	2.8826(16)	164.0(16)
N(5)–H(50A)····O(8)	0.890(16)	2.023(15)	2.9019(17)	169.0(18)
$N(4)-H(40)\cdots O(5)^{i}$	0.878(14)	2.118(14)	2.9646(18)	161.7(16)
$N(5)-H(50)\cdots O(4)^{ii}$	0.884(16)	2.204(16)	3.0521(17)	160.8(18)
N(9)–H(90)···O(6) ^{v}	0.885(16)	2.499(18)	3.1005(17)	125.8(13)
$N(1)-H(10)\cdots O(1)^{i}$	0.875(17)	2.352(17)	3.1289(17)	148.0(17)
C(18)–H(18A)…N(6)	0.98	2.58	2.930(3)	101

Table S1. Complete table of hydrogen bonds (Å ,°) - continued

i = 1-x, 1/2+y, 1/2-z; ii = 1-x, 2-y, 1-z; iii = 1-x, 1-y, 1-z; iv = x, 1+y, z; v = x, 3/2-y, 1/2+z.

D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	<i>D</i> —Н…А
		3d		
N(4)-H(41)····N(1)	0.929(14)	1.923(14)	2.6541(17)	134.0(12)
$N(2)-H(20)\cdots O(1)^{i}$	0.878(15)	1.912(15)	2.7602(15)	162.1(14)
$N(4)-H(40)\cdots O(2)^{ii}$	0.903(13)	1.947(13)	2.8399(16)	169.7(14)
$N(3)-H(30)\cdots O(1)^{iii}$	0.892(14)	1.992(14)	2.8609(16)	164.2(13)
$N(3)-H(31)\cdots O(2)^{i}$	0.908(14)	2.003(14)	2.8990(16)	169.0(13)
$N(5)-H(50)\cdots O(1)^{i}$	0.872(14)	2.217(15)	2.9663(18)	143.8(12)

Table S1. Complete table of hydrogen bonds (Å ,°) - continued

i = 1/2-x, 1/2+y, 1/2-z; ii = -x, 2-y, -z; iii = x, 1+y, z.

Fig. S1. ADP plot of the asymmetric unit in **1b**



Fig. S2. ADP plot of the asymmetric unit in 1c





Fig. S4. ADP plot of the asymmetric unit in **2a**





Fig. S6. ADP plot of the asymmetric unit in **3a**





Fig. S8. ADP plot of the asymmetric unit in **3c**





Fig. S10. Experimental and simulated powder patterns for 2a



Fig. S11. Experimental and simulated powder patterns for 2b



Fig. S12. Experimental and simulated powder patterns for 3c





Fig. S13. Experimental and simulated powder patterns for 3d