Supporting Information

Temperature-induced racemic compounds and chiral conglomerates based on polyoxometalates and lanthanides: syntheses, structures and catalytic properties

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I. Supplementary Structure Figures



Fig. S1 Ball-stick representation of the tricapped trigonal prism coordination modes of Sm(1) and Sm(2) cations in **1**.



Fig. S2 Ball-stick representation of the monocapped square antiprismatic coordination mode of Sm(1) cation in **3**.



Fig. S3 Schematic representation of the 4-connected diamond topology of 3.

II. Supplementary Physical Characterizations



Fig. S4a IR spectrum for compound 1.



Fig. S4b IR spectrum for compound 2.



Fig. S4c IR spectrum for compound 3.



Fig. S4d IR spectrum for compound 4.



Fig. S5a TG curve for compound 1.



Fig. S5b TG curve for compound 2.



Fig. S5d TG curve for compound 4.



Fig. S6a The calculated and experimental PXRD patterns for compound 1.



Fig. S6b The calculated and experimental PXRD patterns for compound 2.



Fig. S6c The calculated and experimental PXRD patterns for compound 3.



Fig. S6d The calculated and experimental PXRD patterns for compound 4.



Fig. S7a Powder X-ray diffraction (PXRD) patterns of **1**: calculated pattern from crystal data (blue line); experimental pattern before catalysis (red line); recovered catalyst **1** after 3 catalytic runs of the cyanosilylation of benzaldehyde (black line).



Fig. S7b Powder X-ray diffraction (PXRD) patterns of **3**: calculated pattern from crystal data (blue line); experimental pattern before catalysis (red line); recovered catalyst **3** after 3 catalytic runs of the cyanosilylation of benzaldehyde (black line).



Fig. S8a IR spectrum for (a) as-synthesized compound 1 and (b) recovered catalyst after catalysis reaction.



Fig. S8b IR spectrum for (a) as-synthesized compound 3 and (b) recovered catalyst after catalysis reaction.



Fig. S9a IR spectra of 1 (bottom), benzaldehyde (top), and 1 obtained after the absorption of benzaldehyde (middle).



Fig. S9b IR spectra of 3 (bottom), benzaldehyde (top), and 3 obtained after the absorption of benzaldehyde (middle).

III. Supplementary Tables

		Compound 1 ^a		
Mo(9)-O(12)	1.681(6)	Mo(4)-O(28)	1.713(6)	
Mo(6)-O(14)	1.725(6)	Mo(9)-O(27)	1.728(6)	
Mo(6)-O(17)	1.874(6)	Mo(9)-O(4)	2.229(5)	
Mo(2)-O(9)	1.972(6)	Mo(8)-O(32)	2.288(6)	
Mo(5)-O(20)	1.989(5)	Mo(1)-O(5)	2.222(5)	
Mo(9)-O(3)	2.363(5)	Mo(5)-O(5)	2.333(5)	
Co(1)-O(5)	1.871(5)	Co(2)-O(4)	1.958(5)	
Sm(1)-O(4W)	2.399(6)	Sm(2)-O(11W)	2.430(7)	
Sm(1)-O(3W)	2.405(7)	Sm(2)-O(9W)	2.450(7)	
Sm(1)-O(1W)	2.437(8)	Sm(2)-O(10W)	2.457(7)	
Sm(1)-O(10)	2.448(5)	Sm(2)-O(12W)	2.461(8)	
Sm(1)-O(5W)	2.450(6)	Sm(2)-O(28)	2.483(6)	
Sm(1)-O(27)	2.456(6)	Sm(2)-O(8W)	2.483(7)	
Sm(1)-O(34)#1	2.511(6)	Sm(2)-O(6W)	2.489(8)	
Sm(1)-O(2W)	2.557(8)	Sm(2)-O(38)#2	2.526(6)	
Sm(1)-O(11)	2.628(6)	Sm(2)-O(7W)	2.539(8)	
O(8)-Co(2)-O(5)	83.5(2)	O(5)-Co(2)-O(32)	175.8(2)	
Compound 3				
Mo(2)-O(15)	1.682(5)	Mo(1)-O(35)	1.727(5)	
Mo(5)-O(28)	1.730(5)	Mo(2)-O(31)	1.740(5)	
Mo(10)-O(22)	1.847(5)	Mo(2)-O(23)	2.233(5)	
Mo(2)-O(7)	1.987(5)	Mo(8)-O(21)	2.287(5)	
Mo(5)-O(2)	1.980(5)	Mo(1)-O(14)	2.210(5)	
Mo(7)-O(2)	2.369(5)	Mo(5)-O(12)	2.363(5)	
Co(1)-O(8)	1.857(4)	Co(2)-O(38)	1.936(5)	
Sm(1)-O(6W)	2.437(6)	Sm(1)-O(35)	2.483(5)	
Sm(1)-O(10)	2.446(5)	Sm(1)-O(31)	2.491(5)	
Sm(1)-O(1W)	2.465(5)	Sm(1)-O(2W)	2.491(6)	
Sm(1)-O(4W)	2.473(6)	Sm(1)-O(5W)	2.523(6)	
Sm(1)-O(3W)	2.483(7)			
O(23)-Co(1)-O(6)	83.8(2)	O(12)-Co(2)-O(38)	175.9(2)	
^a Symmetry transformations used to generate equivalent atoms:			#1 x-1,y,z	#2 -

 Table S1 Selected bond lengths (Å) and angles (°) for 1 and 3.

x+1,-y+1,-z+1

compound	Entry	Efficiency (%)	
1	Round 1	98.2	
	Round 2	97.8	
	Round 3	96.9	
3	Round 1	98.4	
	Round 2	97.1	
	Round 3	96.4	

Table S2 Study on recycling of catalysts 1 and 3 for the heterogeneouscyanosilylation of benzaldehyde under the similar condition.