

Electronic Supplementary Information

Positional Isomeric Effect on Structural Variation of Cd(II) Coordination Polymers Based on Flexible Linear/V-Shaped Bipyridyl Benzene Ligands

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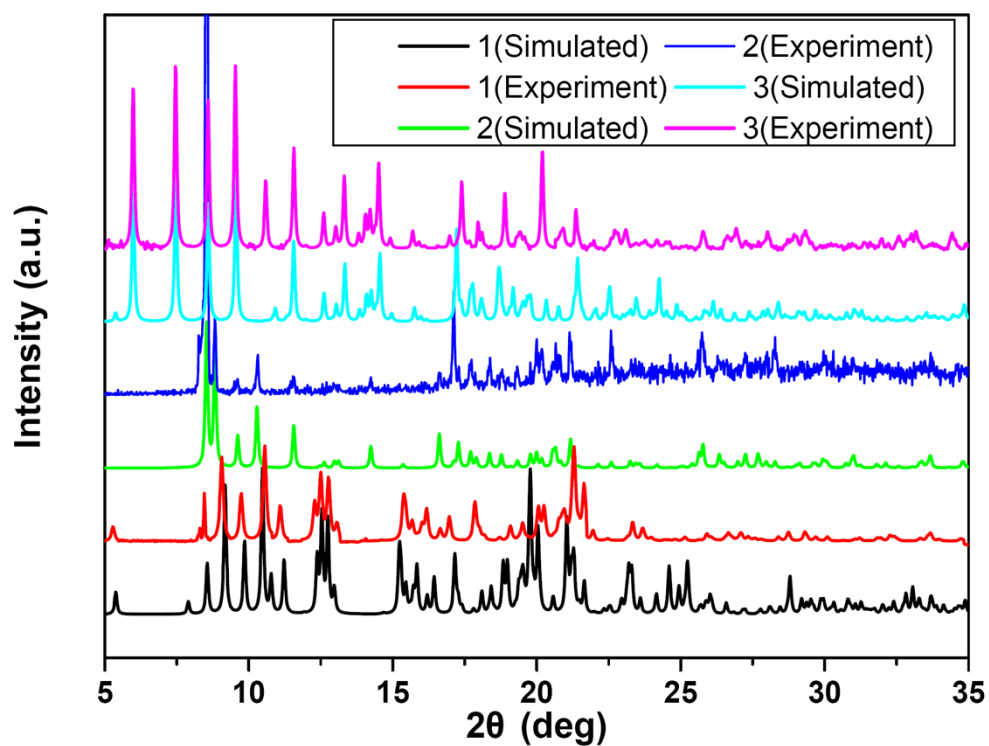


Figure S1. Experimental and simulated PXRD patterns for **1–3**. The obtained PXRD signals of the compound **1** had a little shifts comparing with the simulated one maybe due to preferred orientation of the crystals.

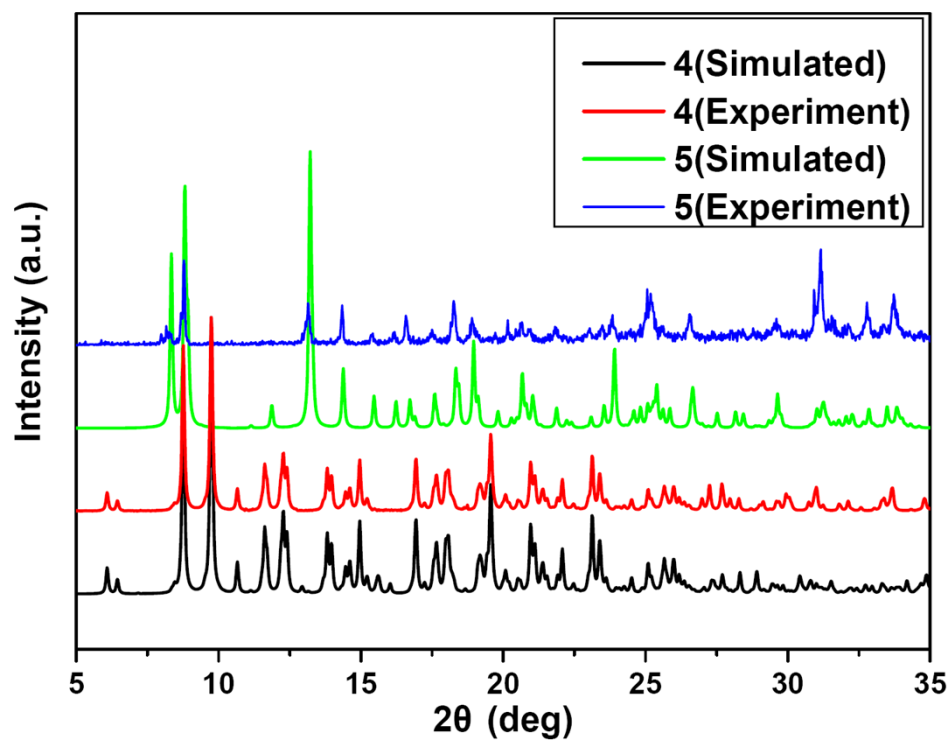


Figure S2. Experimental and simulated PXRD patterns for **4–5**.

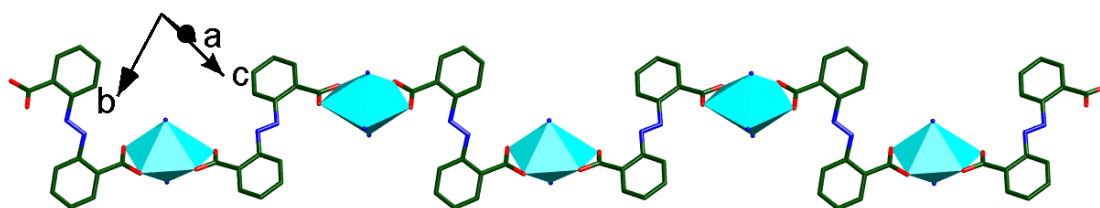
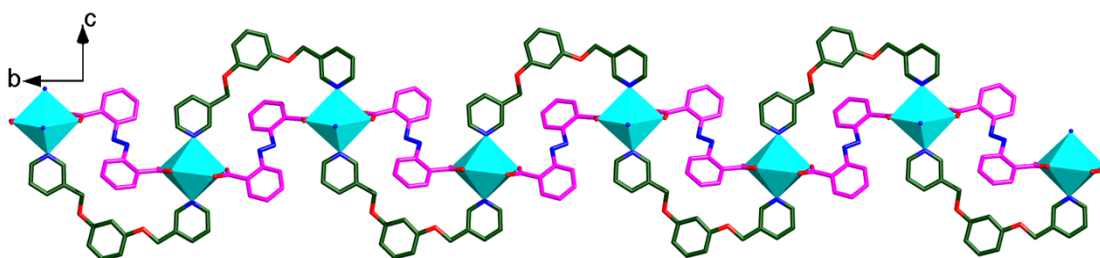
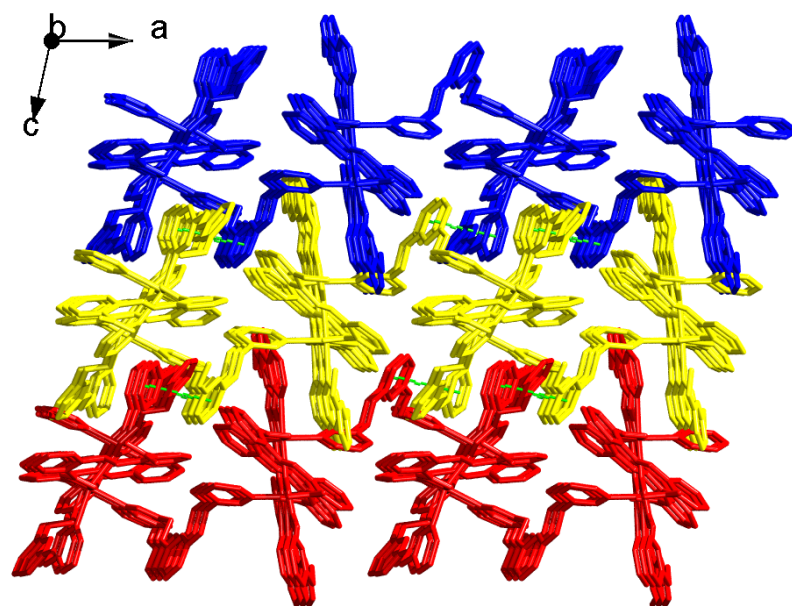


Figure S3. View of the 1D $[\text{CdL}]_n$ chain of **3**.

(a)



(b)



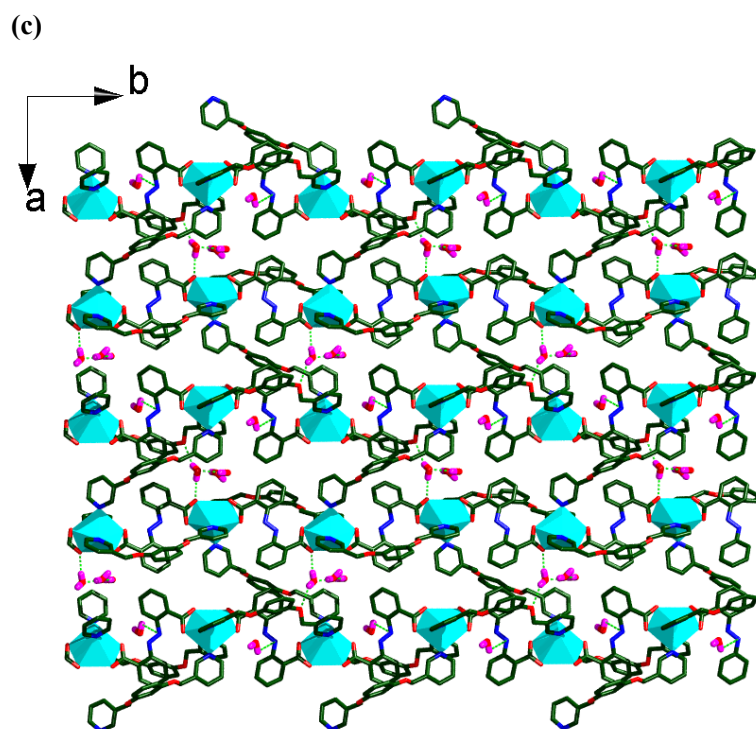


Figure S4. (a) View of the 1D chain of **4** extending along the *b* axis. (b) View of a 3D supramolecular structure in **4** looking down the *b* axis, formed *via* π - π stacking interactions. Green dashed lines represent the π - π interactions. (c) View of a 2D network in **4** extending along the *ab* plane. Green dashed lines represent the hydrogen-bonded interactions.

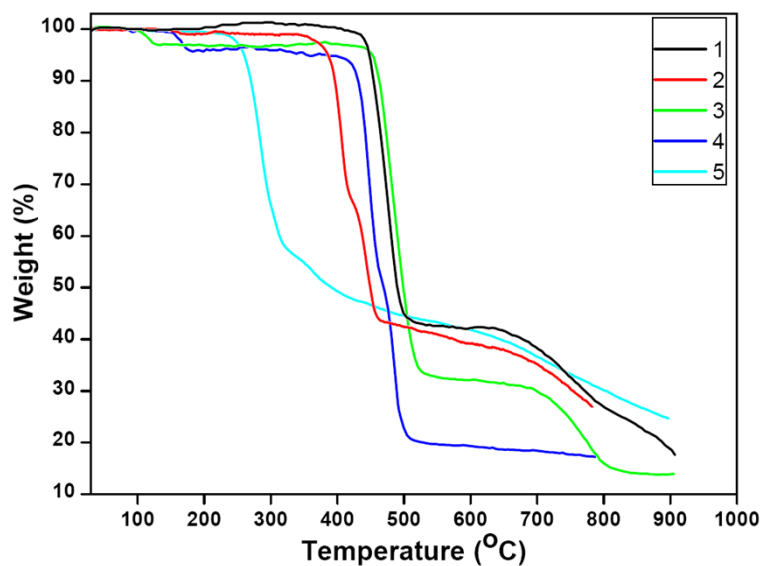


Figure S5. The TGA curves for **1–5**.

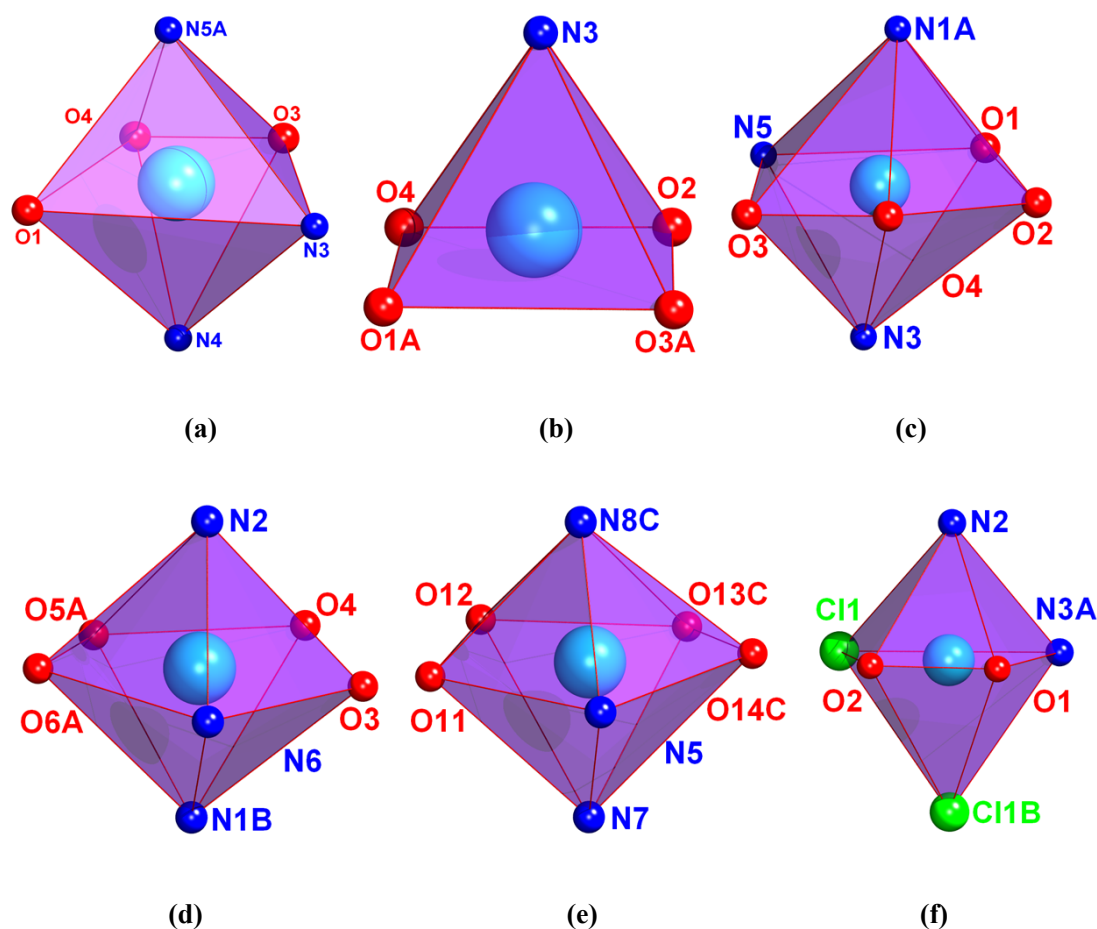


Figure S6. (a) CdN_3O_3 polyhedron showing the coordination geometry of Cd(II) atom in **1**. Symmetry codes: (A) $x + 1, -y + 3/2, z + 1/2$. (b) CdNO_4 polyhedron showing the coordination geometry of Cd(II) atom in **2**. Symmetry codes: (A) $-x + 1, -y + 1, -z + 1$. (c) CdN_3O_4 polyhedron showing the coordination geometry of Cd(II) atom in **3**. Symmetry codes: (A) $-x + 1, -y + 2, -z + 2$. (d-e) CdN_3O_4 polyhedron showing the coordination geometry of Cd(II) atoms in **4**. Symmetry codes: A: $-x + 1, y - 1/2, -z + 3/2$; B: $-x + 1, y + 1/2, -z + 3/2$; C: $-x + 2, y - 1/2, -z + 3/2$. (f) $\text{CdCl}_2\text{N}_2\text{O}_2$ polyhedron showing the coordination geometry of Cd(II) atom in **5**. Symmetry codes: A: $-x + 1, -y + 1, -z + 1$; B: $-x, -y + 1, -z + 1$.

Table S1. Selected angles (°) for **1–5^a**.

Compound 1			
O(1)–Cd(1)–N(5A)	91.87(12)	O(1)–Cd(1)–O(3)	149.20(12)
N(5A)–Cd(1)–O(3)	89.16(11)	O(1)–Cd(1)–N(4)	86.22(12)
N(5A)–Cd(1)–N(4)	177.78(11)	O(3)–Cd(1)–N(4)	91.91(11)
O(1)–Cd(1)–N(3)	125.55(12)	N(5A)–Cd(1)–N(3)	97.67(11)
O(3)–Cd(1)–N(3)	84.70(10)	N(4)–Cd(1)–N(3)	84.38(12)
O(1)–Cd(1)–O(4)	95.37(11)	N(5A)–Cd(1)–O(4)	90.70(10)
O(3)–Cd(1)–O(4)	53.83(9)	N(4)–Cd(1)–O(4)	88.36(10)
N(3)–Cd(1)–O(4)	137.63(10)		
Compound 2			
O(3A)–Cd(1)–O(1)	97.79(7)	O(3A)–Cd(1)–N(3)	114.57(6)
O(1)–Cd(1)–N(3)	116.47(7)	O(3A)–Cd(1)–O(2A)	84.01(7)
O(1)–Cd(1)–O(2A)	153.28(6)	N(3)–Cd(1)–O(2A)	86.15(7)
O(3A)–Cd(1)–O(4)	152.95(6)	O(1)–Cd(1)–O(4)	82.65(6)
N(3)–Cd(1)–O(4)	88.66(6)	O(2A)–Cd(1)–O(4)	83.98(7)
Compound 3			
O(4)–Cd(1)–O(2)	85.91(8)	O(4)–Cd(1)–N(5)	135.53(8)
O(2)–Cd(1)–N(5)	138.55(8)	O(4)–Cd(1)–N(6A)	88.04(9)
O(2)–Cd(1)–N(6A)	89.65(9)	N(5)–Cd(1)–N(6A)	92.40(9)
O(4)–Cd(1)–N(3)	100.57(9)	O(2)–Cd(1)–N(3)	90.14(9)
N(5)–Cd(1)–N(3)	82.07(9)	N(6A)–Cd(1)–N(3)	171.34(8)

O(4)–Cd(1)–O(1)	138.42(8)	O(2)–Cd(1)–O(1)	53.57(8)
N(5)–Cd(1)–O(1)	85.57(8)	N(6A)–Cd(1)–O(1)	83.34(9)
N(3)–Cd(1)–O(1)	89.58(9)	O(4)–Cd(1)–O(3)	52.00(8)
O(2)–Cd(1)–O(3)	137.14(8)	N(5)–Cd(1)–O(3)	83.80(8)
N(6A)–Cd(1)–O(3)	95.81(9)	N(3)–Cd(1)–O(3)	90.23(9)
O(1)–Cd(1)–O(3)	169.29(7)		

Compound 4

O(5A)–Cd(1)–N(6)	140.45(9)	O(5A)–Cd(1)–O(4)	84.52(9)
N(6)–Cd(1)–O(4)	135.01(9)	O(5A)–Cd(1)–N(1B)	84.95(10)
N(6)–Cd(1)–N(1B)	91.27(11)	O(4)–Cd(1)–N(1B)	92.34(11)
O(5A)–Cd(1)–N(2)	96.63(9)	N(6)–Cd(1)–N(2)	86.42(10)
O(4)–Cd(1)–N(2)	89.53(10)	N(1B)–Cd(1)–N(2)	177.66(10)
O(5A)–Cd(1)–O(6A)	54.22(8)	N(6)–Cd(1)–O(6A)	86.46(9)
O(4)–Cd(1)–O(6A)	138.38(9)	N(1B)–Cd(1)–O(6A)	89.47(9)
N(2)–Cd(1)–O(6A)	90.07(9)	O(5A)–Cd(1)–O(3)	136.93(9)
N(6)–Cd(1)–O(3)	82.04(9)	O(4)–Cd(1)–O(3)	53.30(9)
N(1B)–Cd(1)–O(3)	88.18(9)	N(2)–Cd(1)–O(3)	91.82(9)
O(6A)–Cd(1)–O(3)	168.21(9)	O(13C)–Cd(2)–N(7)	95.77(10)
O(13C)–Cd(2)–N(8C)	101.30(9)	N(7)–Cd(2)–N(8C)	160.73(10)
O(13C)–Cd(2)–O(12)	82.57(9)	N(7)–Cd(2)–O(12)	97.98(10)
N(8C)–Cd(2)–O(12)	93.07(10)	O(13C)–Cd(2)–N(5)	137.23(9)
N(7)–Cd(2)–N(5)	81.69(11)	N(8C)–Cd(2)–N(5)	79.72(10)
O(12)–Cd(2)–N(5)	140.18(9)	O(13C)–Cd(2)–O(11)	135.53(8)

N(7)–Cd(2)–O(11)	85.45(9)	N(8C)–Cd(2)–O(11)	88.56(9)
O(12)–Cd(2)–O(11)	53.44(8)	N(5)–Cd(2)–O(11)	87.05(9)
O(13C)–Cd(2)–O(14C)	53.96(8)	N(7)–Cd(2)–O(14C)	93.92(10)
N(8C)–Cd(2)–O(14C)	88.99(9)	O(12)–Cd(2)–O(14C)	135.89(9)
N(5)–Cd(2)–O(14C)	83.49(9)	O(11)–Cd(2)–O(14C)	170.51(8)

Compound 5

N(3A)–Cd(1)–O(1)	103.01(7)	N(3A)–Cd(1)–N(2)	88.93(7)
O(1)–Cd(1)–N(2)	87.51(7)	N(3A)–Cd(1)–O(2)	158.32(6)
O(1)–Cd(1)–O(2)	55.42(6)	N(2)–Cd(1)–O(2)	88.02(7)
N(3A)–Cd(1)–Cl(1)	103.21(5)	O(1)–Cd(1)–Cl(1)	153.76(5)
N(2)–Cd(1)–Cl(1)	91.69(6)	O(2)–Cd(1)–Cl(1)	98.33(4)
N(3A)–Cd(1)–Cl(1B)	88.70(6)	O(1)–Cd(1)–Cl(1B)	93.46(5)
N(2)–Cd(1)–Cl(1B)	177.59(5)	O(2)–Cd(1)–Cl(1B)	94.35(5)
Cl(1)–Cd(1)–Cl(1B)	88.42(4)		

^a Symmetry codes for **1**: A: $x + 1, y, z + 1$; Symmetry codes for **2**: A: $-x + 1, -y + 1, -z + 1$; Symmetry codes for **3**: A: $-x + 1, -y + 2, -z + 2$; Symmetry codes for **4**: A: $-x + 1, y - 1/2, -z + 3/2$; B: $-x + 1, y + 1/2, -z + 3/2$; C: $-x + 2, y - 1/2, -z + 3/2$; Symmetry codes for **5**: A: $-x + 1, -y + 1, -z + 1$; B: $-x, -y + 1, -z + 1$.

Table S2. Selected dihedral angles (°) for **1–5^a**.

Compound 1			
Cg1–Cg2	30.715	Cg1–Cg3	40.166
Cg4–Cg5	61.167	Cg4–Cg5A	61.167
Cg1 is the C30–C35 benzene ring; Cg2 is the N5/C37–C41 pyridine ring; Cg3 is the N4/C24–C28 pyridine ring; Cg4 is the C21–C23/C21A–C23A benzene ring; Cg5 is the N3/C15–C19 pyridine ring; Cg5A is the N3A/C15A–C19A pyridine ring.			
Compound 2			
Cg6–Cg7	46.217	Cg6–Cg7A	46.217
Cg6 is the C21–C23/C21A–C23A benzene ring; Cg7 is the N3/C15–C19 pyridine ring; Cg7A is the N3A/C15A–C19A pyridine ring.			
Compound 3			
Cg8–Cg9	89.220	Cg8–Cg10	13.191
Cg12–Cg11	74.683	Cg12–Cg13	59.872
Cg8 is the C39–C44 benzene ring; Cg9 is the N5/C33–C37 pyridine ring; Cg10 is the N6/C46–C50 pyridine ring; Cg11 is the N3/C15–C19 pyridine ring; Cg12 is the C21–C26 benzene ring; Cg13 is the N4/C28–C32 pyridine ring.			
Compound 4			
Cg14–Cg15	89.788	Cg14–Cg16	82.081
Cg17–Cg18	25.608	Cg17–Cg19	8.667
Cg14 is the C39–C44 benzene ring; Cg15 is the N5/C33–C37 pyridine ring; Cg16 is the N6/C46–C50 pyridine ring; Cg17 is the C7–C12 benzene ring; Cg18 is the N2/C14–C18 pyridine ring; Cg19 is the N1/C1–C5 pyridine ring.			
Compound 5			
Cg20–Cg21	77.160	Cg20–Cg22	20.199

Cg20 is the C14-C19 benzene ring; Cg21 is the N3/C21-C25 pyridine ring; Cg22 is the N2/C8-C12 pyridine ring;

^a Symmetry codes for **1**: A: $-x, -y, -z + 2$; Symmetry codes for **2**: A: $-x, -y + 1, -z + 2$.

Table S3. Selected torsion angles (°) for **1–5**.

Compound 1			
C(18)–C(20)–O(5)–C(21)	–155.9(3)	C(27)–C(29)–O(6)–C(30)	–173.7(4)
C(37)–C(36)–O(7)–C(33)	–179.5(4)		
Compound 2			
C(19)–C(20)–O(5)–C(21)	–175.71(17)		
Compound 3			
C(46)–C(45)–O(8)–C(43)	173.5(2)	C(35)–C(38)–O(7)–C(39)	–178.2(3)
C(17)–C(20)–O(5)–C(21)	–171.0(3)	C(28)–C(27)–O(6)–C(25)	176.6(3)
Compound 4			
C(4)–C(6)–O(1)–C(7)	159.0(3)	C(14)–C(13)–O(2)–C(11)	–170.6(3)
C(36)–C(38)–O(7)–C(39)	179.1(3)	C(46)–C(45)–O(8)–C(43)	–172.9(3)
C(54)–C(56)–O(9)–C(57)	178.1(3)	C(64)–C(63)–O(10)–C(61)	–172.6(3)
Compound 5			
C(11)–C(13)–O(3)–C(14)	–167.26(19)	C(21)–C(20)–O(4)–C(19)	–177.6(2)
