

Electronic Supplementary Information

Reaction Controlled Assemblies and Structural Diversities of Seven Co(II)/Cu(II) Complexes Based on a Bipyridine-dicarboxylate N-oxide Ligand

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Table S1 Crystallographic data and structure refinement parameters for complexes **1-5**

complex	1	2	3	4	5
chemical formula	C ₁₄ H ₁₁ CoN ₃ O ₈	C ₁₂ H ₈ CuN ₂ O ₇	C ₄₈ H ₂₈ Cu ₂ N ₈ O ₁₂	C ₄₈ H ₄₈ Cu ₂ N ₈ O ₂₂	C ₄₄ H ₄₀ Cu ₂ N ₈ O ₁₈
formula weight	408.17	355.74	1035.86	1216.02	1095.92
crystal shape	columnar	block	columnar	block	block
crystal color	red	bright green	blue	blue	blue
temperature (K)	296(2)	296(2)	296(2)	296(2)	296(2)
crystal system	Monoclinic	Monoclinic	Monoclinic	Triclinic	Triclinic
space group	<i>C2/m</i>	<i>P2(1)/n</i>	<i>P2(1)/n</i>	<i>P-1</i>	<i>P-1</i>
<i>a</i> /Å	14.462(2)	5.8488(13)	9.3276(10)	9.213(2)	9.357(4)
<i>b</i> /Å	15.486(3)	16.227(4)	11.9664(13)	12.190(3)	11.325(5)
<i>c</i> /Å	7.4899(12)	13.164(3)	19.099(2)	12.635(3)	12.540(5)
α (°)	90	90	90	109.362(4)	64.338(6)
β (°)	96.971(3)	91.507(3)	92.947(2)	109.077(4)	69.757(6)
γ (°)	90	90	90	91.212(4)	75.003(6)
<i>V</i> /Å ³	1665.0(5)	1249.0(5)	2128.9(4)	1251.8(5)	1114.4(8)
<i>Z</i>	4	4	2	1	1
density (mg/m ³)	1.620	1.892	1.616	1.613	1.633
μ (mm ⁻¹)	1.080	1.790	1.078	0.944	1.044
<i>F</i> (000)	820	716.0	1052	626	562
reflections collected	4251	6642	10555	6190	5612
<i>R</i> _{int}	0.0504	0.0335	0.0463	0.0212	0.0247
number of parameters	129	199	316	361	325
Goodness-of-fit on <i>F</i> ²	1.075	1.075	1.002	1.052	1.028
<i>R</i> ₁ ^a , <i>wR</i> ₂ ^b [<i>I</i> > 2σ(<i>I</i>)]	0.0642, 0.1755	0.0408, 0.1228	0.0410, 0.0877	0.0450, 0.1330	0.0559, 0.1634
<i>R</i> ₁ , <i>wR</i> ₂ (all data)	0.0897, 0.1940	0.0656, 0.1756	0.0659, 0.0978	0.0538, 0.1567	0.0741, 0.2373

$$^a R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|; \quad ^b wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)]^{1/2}$$

Table S2 The selected bond lengths (Å) and angles (°) for complexes **1-5**.

1			
Co(1)-O(2)#1	2.080(4)	Co(1)-O(2)#2	2.080(4)
Co(1)-O(2)#3	2.080(4)	Co(1)-O(2)	2.080(4)
Co(1)-O(4)	2.138(5)	Co(1)-O(4)#1	2.138(5)
Co(2)-O(3)	2.075(4)	Co(2)-O(3)#4	2.075(4)
Co(2)-O(3)#5	2.075(4)	Co(2)-O(3)#6	2.075(4)
Co(2)-N(1)	2.115(8)	Co(2)-N(1)#4	2.115(8)
O(2)#1-Co(1)-O(2)#2	90.6(3)	O(2)#1-Co(1)-O(2)#3	89.4(3)
O(2)#2-Co(1)-O(2)#3	180	O(2)#1-Co(1)-O(2)	180
O(2)#2-Co(1)-O(2)	89.4(3)	O(2)#3-Co(1)-O(2)	90.6(3)
O(2)#1-Co(1)-O(4)	92.48(2)	O(2)#2-Co(1)-O(4)	87.52(2)
O(2)#3-Co(1)-O(4)	92.48(2)	O(2)-Co(1)-O(4)	87.52(2)
O(2)#1-Co(1)-O(4)#1	87.52(2)	O(2)#2-Co(1)-O(4)#1	92.48(2)
O(2)#3-Co(1)-O(4)#1	87.52(2)	O(2)-Co(1)-O(4)#1	92.48(2)
O(4)-Co(1)-O(4)#1	180	O(3)-Co(2)-O(3)#4	180
O(3)-Co(2)-O(3)#5	93.6(2)	O(3)#4-Co(2)-O(3)#5	86.4(2)
O(3)-Co(2)-O(3)#6	86.4(2)	O(3)#4-Co(2)-O(3)#6	93.6(2)
O(3)#5-Co(2)-O(3)#6	180	O(3)-Co(2)-N(1)	95.31(2)
O(3)#4-Co(2)-N(1)	84.69(2)	O(3)#5-Co(2)-N(1)	95.31(2)
O(3)#6-Co(2)-N(1)	84.69(2)	O(3)-Co(2)-N(1)#4	84.69(2)
O(3)#4-Co(2)-N(1)#4	95.31(2)	O(3)#5-Co(2)-N(1)#4	84.69(2)
O(3)#6-Co(2)-N(1)#4	95.31(2)	N(1)-Co(2)-N(1)#4	180
2			
Cu(1)-O(3)#1	1.929(4)	Cu(1)-O(6)#2	1.973(3)
Cu(1)-O(5)#2	1.974(3)	Cu(1)-O(1)	1.977(3)
Cu(1)-O(7)	2.363(5)		
O(3)#1-Cu(1)-O(6)#2	90.86(2)	O(3)#1-Cu(1)-O(5)#2	158.55(2)
O(6)#2-Cu(1)-O(5)#2	90.57(2)	O(3)#1-Cu(1)-O(1)	88.75(2)
O(6)#2-Cu(1)-O(1)	177.36(2)	O(5)#2-Cu(1)-O(1)	90.75(2)
O(3)#1-Cu(1)-O(7)	108.09(2)	O(6)#2-Cu(1)-O(7)	87.17(2)
O(5)#2-Cu(1)-O(7)	93.36(2)	O(1)-Cu(1)-O(7)	90.47(2)
3			
Cu(1)-O(1)	1.933(2)	Cu(1)-O(2)	2.645(3)
Cu(1)-O(3)#1	1.954(2)	Cu(1)-O(4)#1	2.585(2)
Cu(1)-N(3)	2.000(3)	Cu(1)-N(4)	2.005(3)
O(1)-Cu(1)-O(3)#1	91.72(9)	O(1)-Cu(1)-N(3)	172.85(1)
O(3)#1-Cu(1)-N(3)	93.88(1)	O(1)-Cu(1)-N(4)	93.61(1)

O(3)#1-Cu(1)-N(4)	167.25(1)	N(3)-Cu(1)-N(4)	81.83(1)
4			
Cu(1)-O(1)	1.929(3)	Cu(1)-O(2)	2.856(3)
Cu(1)-O(4)#1	1.961(2)	Cu(1)-N(4)	2.005(3)
Cu(1)-N(3)	2.027(3)	Cu(1)-O(1W)	2.439(3)
O(1)-Cu(1)-O(4)#1	91.21(1)	O(1)-Cu(1)-N(4)	175.37(1)
O(4)#1-Cu(1)-N(4)	92.71(1)	O(1)-Cu(1)-N(3)	94.60(1)
O(4)#1-Cu(1)-N(3)	165.02(1)	N(4)-Cu(1)-N(3)	82.20(1)
O(1)-Cu(1)-O(1W)	90.19(1)	O(4)#1-Cu(1)-O(1W)	94.16(1)
N(4)-Cu(1)-O(1W)	87.06(1)	N(3)-Cu(1)-O(1W)	99.61(1)
5			
Cu(1)-O(1)	1.945(4)	Cu(1)-O(3)#1	1.961(4)
Cu(1)-N(4)	1.997(4)	Cu(1)-N(3)	2.022(4)
Cu(1)-O(1W)	2.420(4)	Cu(1)-O(2)	2.797(6)
O(1)-Cu(1)-O(3)#1	90.94(2)	O(1)-Cu(1)-N(4)	170.91(2)
O(3)#1-Cu(1)-N(4)	93.65(2)	O(1)-Cu(1)-N(3)	94.34(2)
O(3)#1-Cu(1)-N(3)	173.58(2)	N(4)-Cu(1)-N(3)	80.63(2)
O(1)-Cu(1)-O(1W)	92.06(2)	O(3)#1-Cu(1)-O(1W)	93.31(2)
N(4)-Cu(1)-O(1W)	95.50(2)	N(3)-Cu(1)-O(1W)	90.15(2)

Symmetry codes: For **1**, #1 $-x, -y+1, -z+1$; #2 $x, -y+1, z$; #3 $-x, y, -z+1$; #4 $-x, -y, -z$; #5 $x, -y, z$; #6 $-x, y, -z$. For **2**, #1 $x+1, y, z$; #2 $x+1/2, -y+1/2, z+1/2$. For **3**, #1 $-x+2, -y+1, -z+2$; For **4**, #1 $-x+1, -y, -z$; For **5**, #1 $-x+1, -y+1, -z+1$.

Table S3 Hydrogen bond geometries in the crystal structure of **1-5**.

D-H...A	D-H	H...A	D...A(Å)	D-H...A(°)
1				
C(4)-H(4)···O(1)#1	0.93	2.42	3.146(7)	135
C(8)-H(8A)···O(4)#2	0.96	2.54	3.356(17)	142
2				
O(7)-H(7A)···O(2)#1	0.86	2.39	3.160(7)	150
O(7)-H(7B)···O(1)	0.86	2.47	3.093(6)	130
O(7)-H(7B)···O(2)	0.86	1.90	2.704(8)	156
3				
C(1)-H(1)···O(6)#1	0.93	2.19	2.927(4)	136
C(2)-H(2)···O(4)#2	0.93	2.57	3.376(3)	145
C(9)-H(9)···O(5)#3	0.93	2.25	3.095(4)	150

C(18)–H(18)···O(2)#4	0.93	2.40	2.995(5)	121
4				
O(1W)–H(1A)···O(5)#1	0.86	1.89	2.711(6)	160
O(1W)–H(1B)···O(3W)	0.85	2.13	2.983(6)	179
O(2W)–H(2A)···O(2)	0.86	2.03	2.888(5)	179
O(2W)–H(2B)···O(4W)#2	0.86	2.05	2.908(7)	178
O(3W)–H(3A)···O(3)#3	0.86	1.94	2.801(5)	179
O(3W)–H(3B)···O(5)#4	0.85	1.85	2.698(6)	179
O(4W)–H(4A)···O(6)#3	0.87	1.90	2.772(6)	179
O(4W)–H(4B)···O(3W)	0.85	2.08	2.925(6)	178
O(5W)–H(5A)···O(3)#5	0.85	1.91	2.760(7)	180
O(5W)–H(5B)···O(1W)#6	0.85	2.15	2.995(9)	180
C(8)–H(8)···O(4W)#6	0.93	2.39	3.312(7)	170
C(22)–H(22)···O(1W)#6	0.93	2.52	3.376(6)	154
5				
O(1W)–H(1WA)···O(4)#1	0.86	2.14	2.717(7)	124
O(1W)–H(1WB)···O(5)#2	0.86	2.18	2.828(7)	133
O(2W)–H(2A)···O(2)	0.86	2.00	2.852(8)	178
O(2W)–H(2B)···O(3W)	0.86	1.97	2.829(8)	179
O(3W)–H(3A)···O(6)#3	0.85	1.97	2.818(8)	180
O(3W)–H(3B)···O(4)	0.85	2.08	2.930(8)	180
C(8)–H(8)···O(6)#4	0.93	2.25	3.172(8)	169
C(19)–H(19)···O(2W)#5	0.93	2.31	3.175(7)	155

Symmetry codes: For **1**, #1 $1/2+x, 1/2-y, z$; #2 $1/2-x, -1/2+y, 1-z$. For **2**, #1 $-x+2, -y+1, -z+2$. For **3**, #1 $3/2-x, 1/2+y, 3/2-z$; #2 $-1+x, y, z$; #3 $5/2-x, -1/2+y, 3/2-z$; #4 $2-x, -y, 2-z$; For **4**, #1 $-x+1, -y, -z$; #2 $x, y, z-1$; #3 $x+1, y, z+1$; #4 $x, y, z+1$; #5 $x+1, y+1, z+1$; #6 $-x+1, -y+1, -z+1$. For **5**, #1 $-x+1, -y+1, -z+1$; #2 $x, y+1, z$; #3 $x-1, y, z$; #4 $2-x, 1-y, -z$; #5 $1-x, 2-y, -z$.

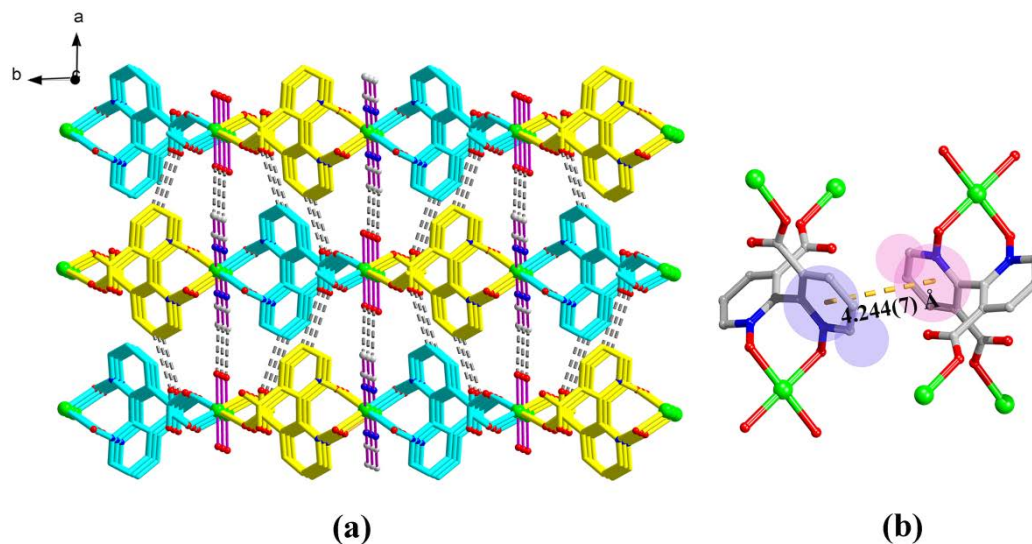


Fig. S1 (a) A view of the 3D supramolecular network of complex **1** constructed through hydrogen bonds (C-H \cdots O) and (b) the weak π - π stacking interactions between the adjacent 2D layers. The broken lines are on behalf of the hydrogen bonds and π \cdots π interactions, and the hydrogen atoms are omitted for clarity.

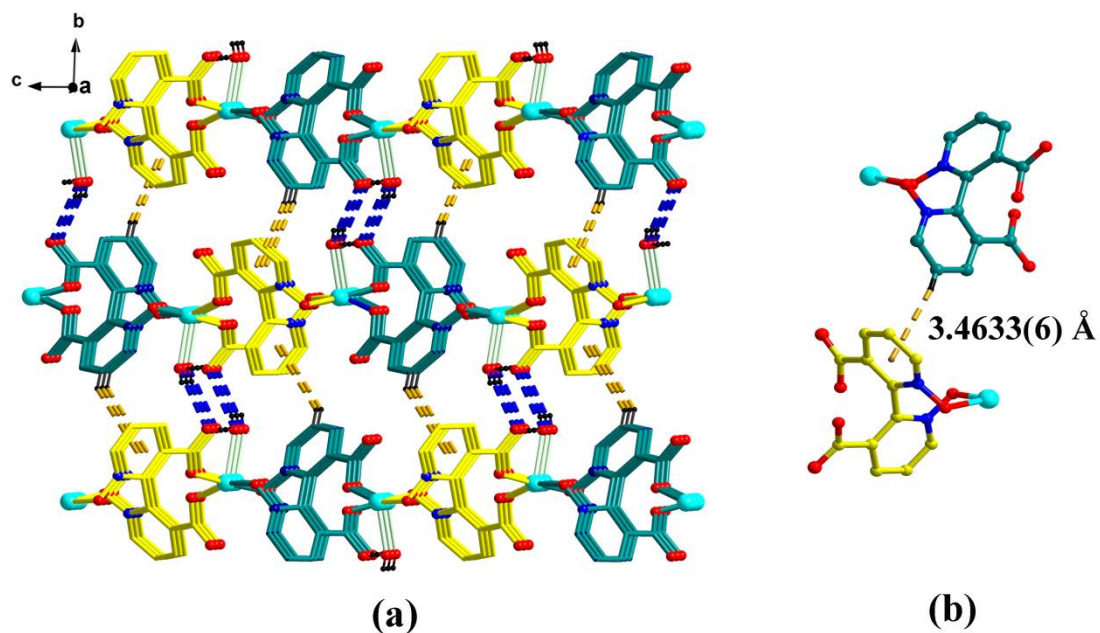


Fig. S2 (a) A view of the 3D supramolecular network of complex **2** constructed through hydrogen bonds (O-H \cdots O) and (b) C-H \cdots π interactions between the adjacent 2D layers. The broken lines are on behalf of the hydrogen bonds and C-H \cdots π interactions, and the hydrogen atoms are omitted for clarity.

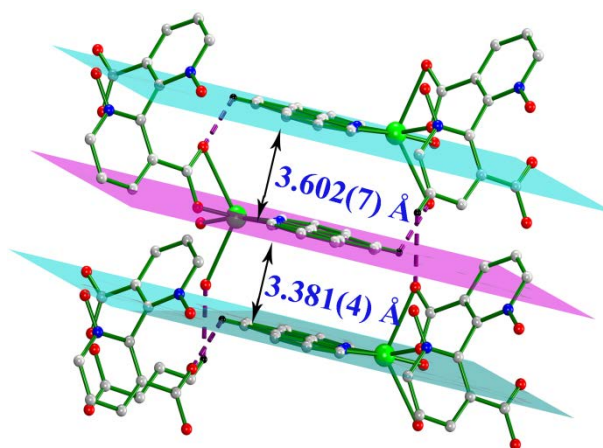


Fig. S3 The π - π stacking interactions can be observed in the 2D layers of complex **3**. The broken lines are on behalf of the hydrogen bonds, and the hydrogen atoms are omitted for clarity.

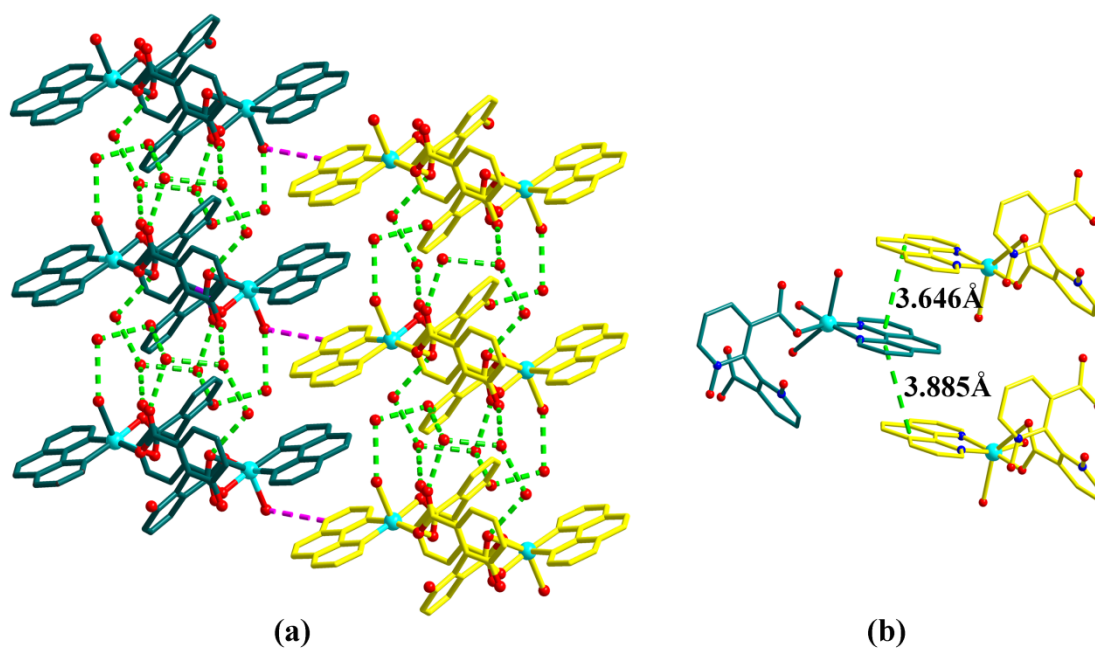


Fig. S4 A view of 2D layer of complex **4** constructed by hydrogen bonds (O-H \cdots O and C-H \cdots O) and π \cdots π stacking interactions between two neighboring phen ligands with the centroids distance of 3.646 Å and 3.885 Å. The broken lines are on behalf of the hydrogen bonds and π \cdots π interactions, and the hydrogen atoms are omitted for clarity.

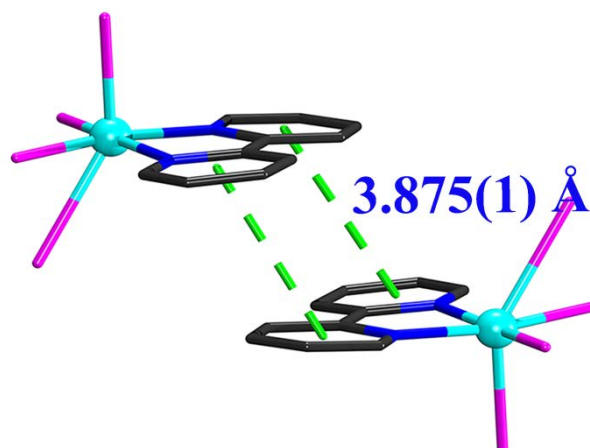
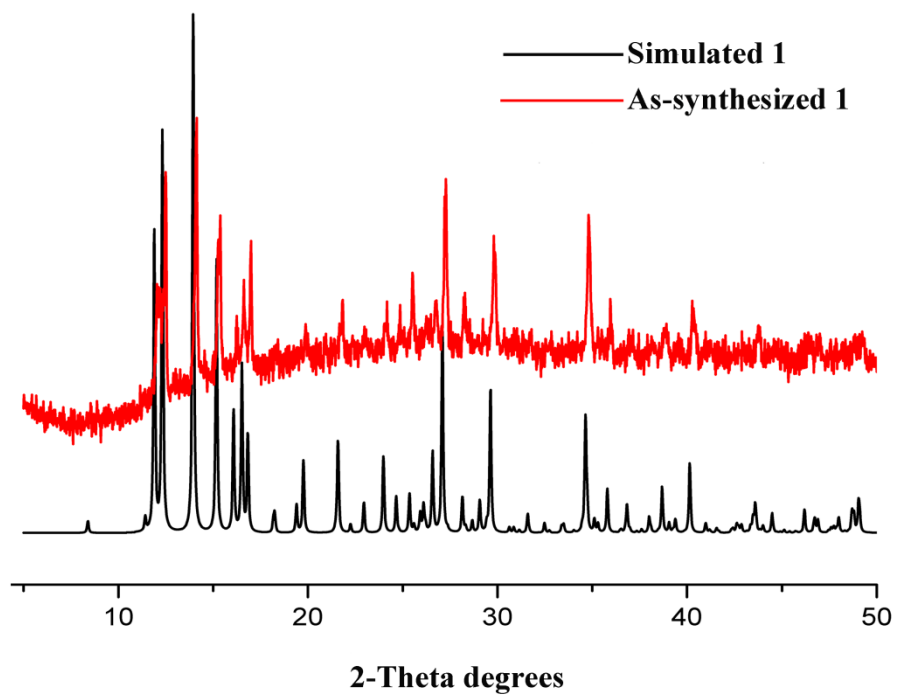
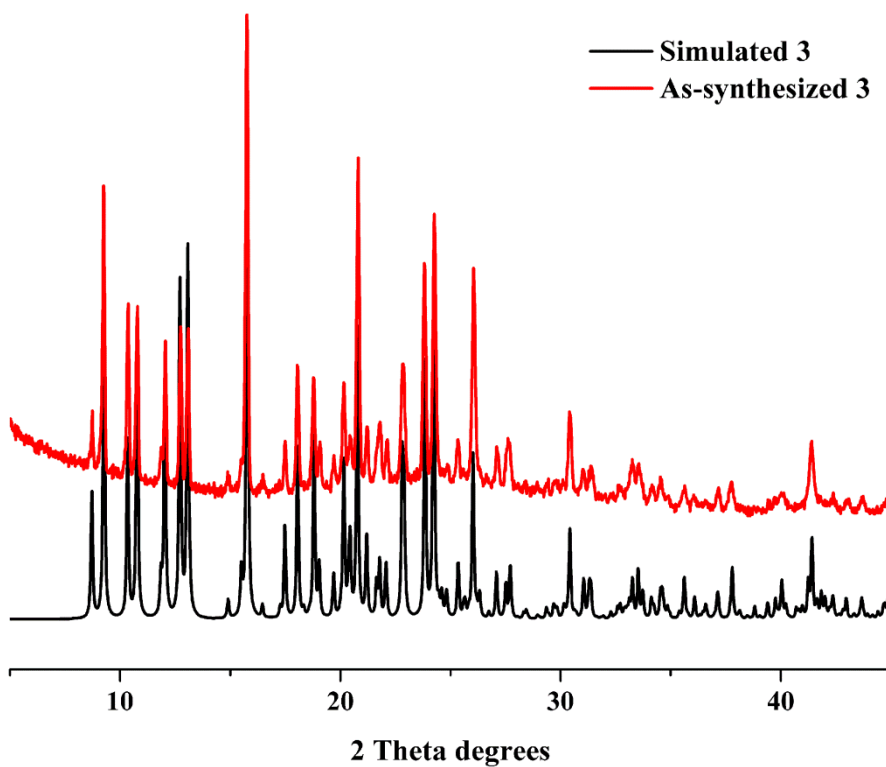
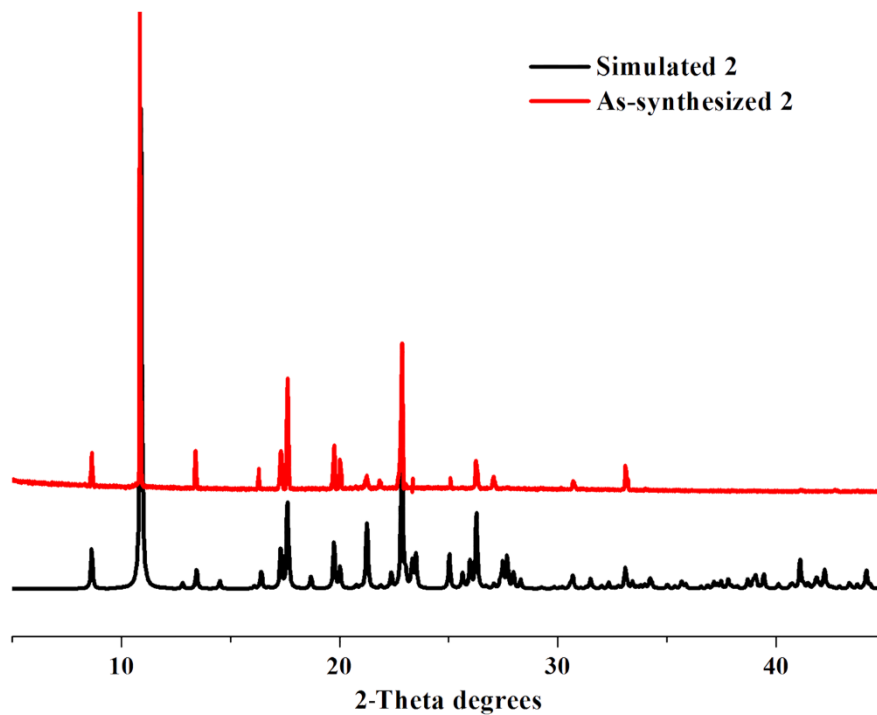


Fig. S5 The π - π stacking interactions can be seen between the 2D layers of complex **5**. The broken lines are on behalf of the π $\cdots\pi$ interactions, and the hydrogen atoms and (*R, S*)-bpdado ligands are omitted for clarity.





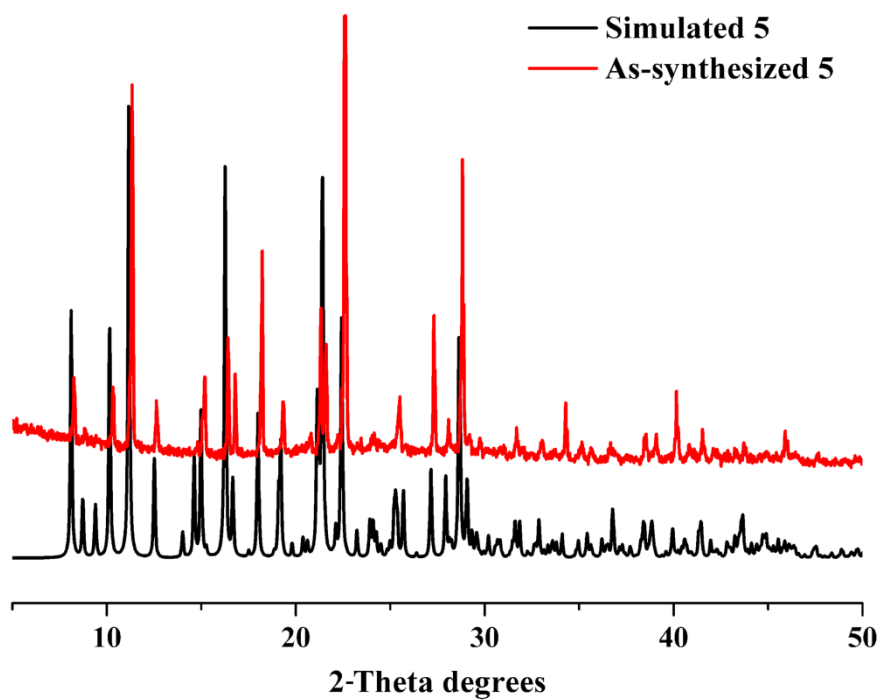
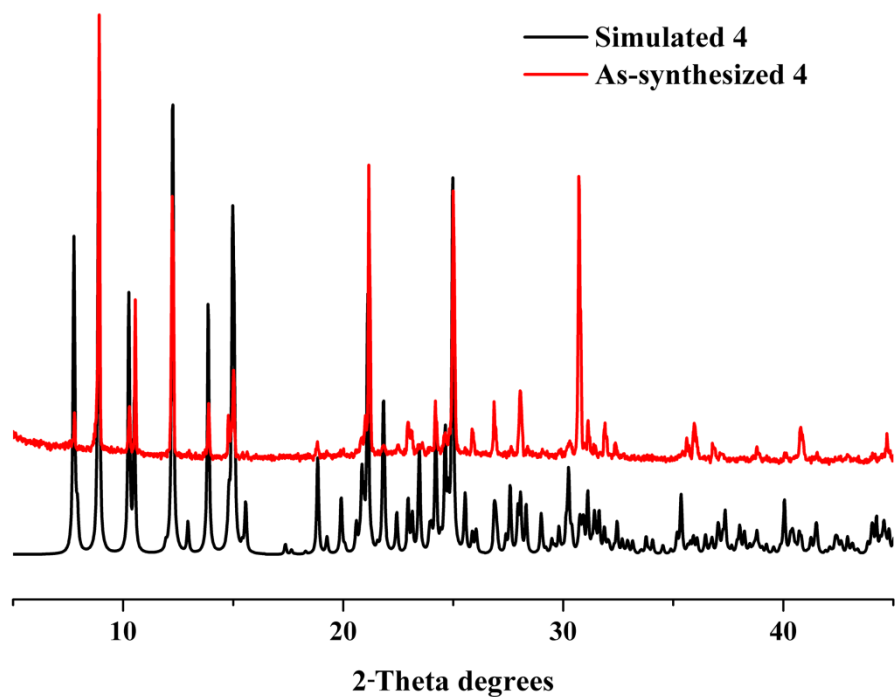
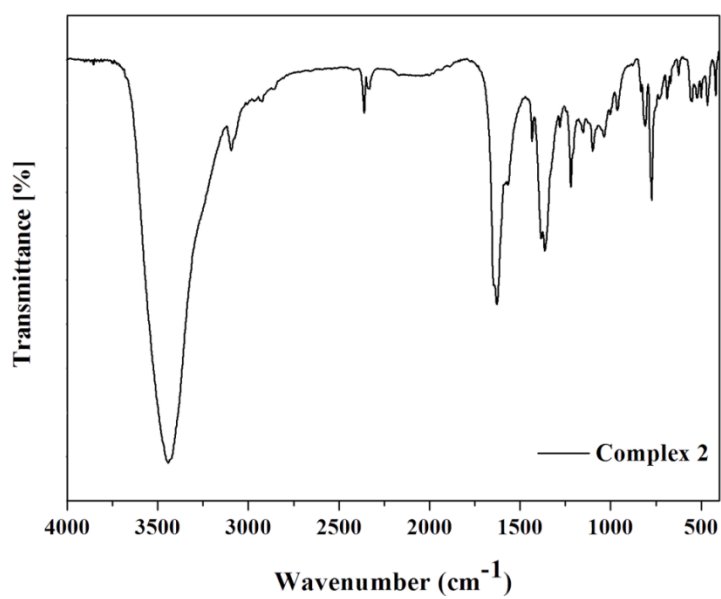
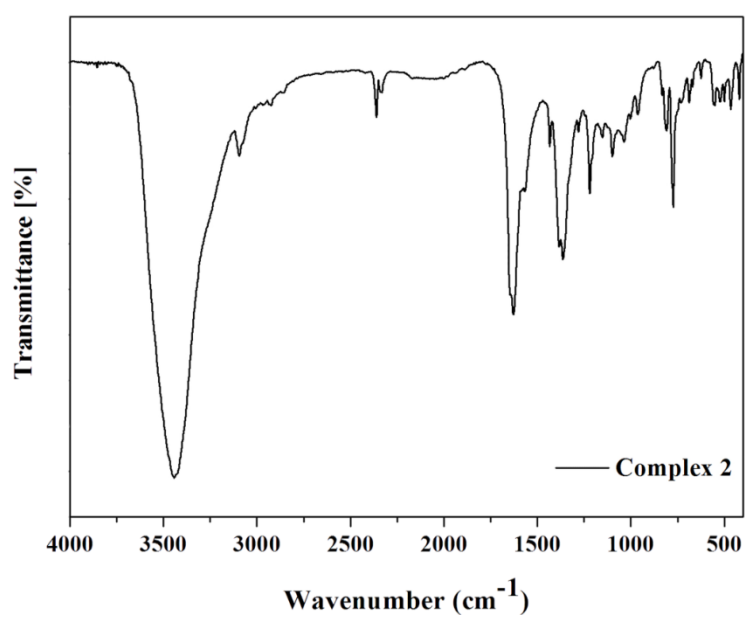
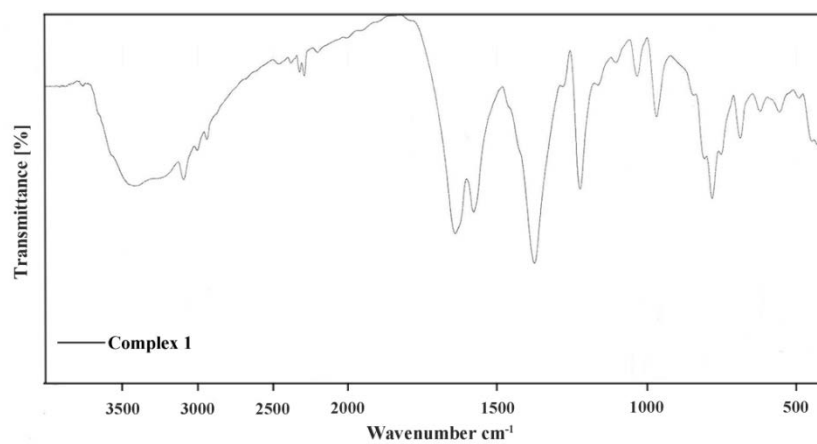


Fig.S6 Comparisons of XRD patterns of the simulated pattern from the single-crystal structure determination and the as-synthesized products in complexes **1-5**.



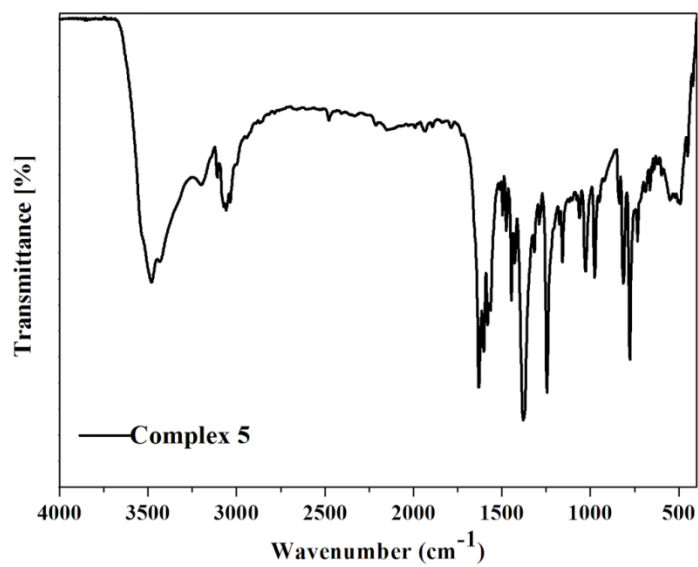
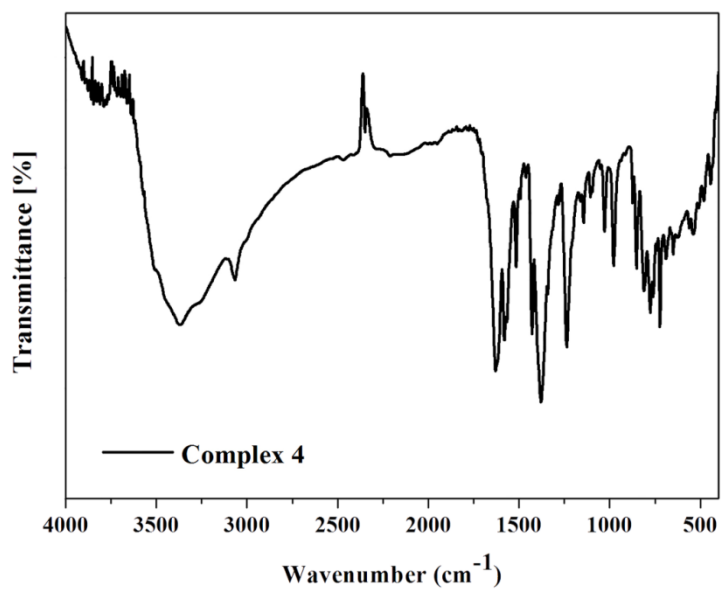
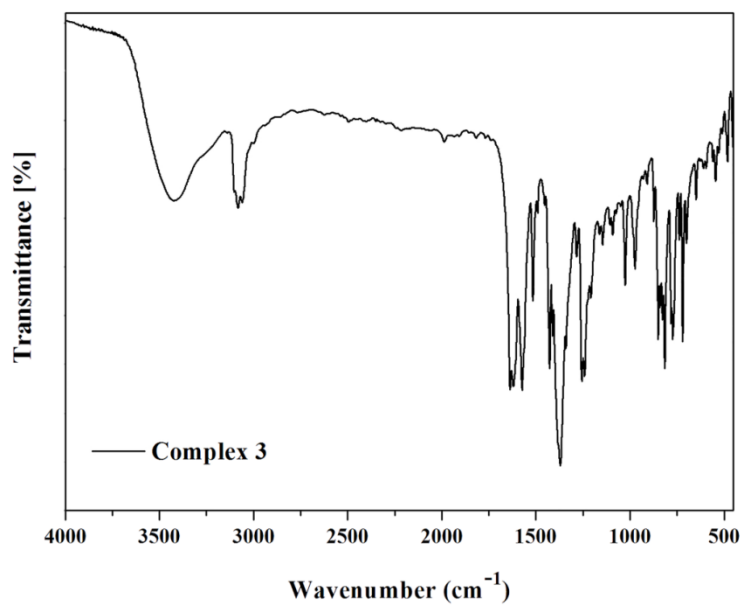


Fig.S7 IR for complexes 1-5.