

Supplementary Material (ESI) for CrystEngComm
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Influence of N-donor Site in 5-(x-pyridyl)-1H-tetrazole Ligands (x = 2, 4) on Assembly of Polyoxometalates-based Compounds Modified by Multinuclear Metal Clusters and Infinite Chains

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Table. S1. Selected bond lengths (Å) and bond angles (°) for compounds 1–4.

Cu(1)-N(2)	1.940(8)	Cu(1)-N(7)	1.959(8)
Cu(1)-N(1)	2.029(8)	Cu(1)-N(6)	2.036(9)
O(1W)-Cu(3)	1.921(7)	O(1W)-Cu(4)	1.965(7)
O(2W)-Cu(2)	1.995(7)	N(9)-Cu(2)	2.215(9)
N(11)-Cu(2)	2.027(9)	N(12)-Cu(2)	1.994(9)
N(13)-Cu(3)	2.009(8)	N(14)-Cu(4)	2.014(9)
N(16)-Cu(3)	2.052(8)	N(17)-Cu(3)	1.991(8)
N(18)-Cu(2)	1.972(9)	Cu(3)-N(4)#2	2.325(10)
Cu(4)-O(1W)#3	1.965(7)	Cu(4)-N(14)#3	2.014(9)
N(2)-Cu(1)-N(7)	170.8(4)	N(2)-Cu(1)-N(1)	81.3(3)
N(7)-Cu(1)-N(1)	99.0(3)	N(2)-Cu(1)-N(6)	97.6(4)
N(7)-Cu(1)-N(6)	80.9(3)	N(1)-Cu(1)-N(6)	173.0(4)
Cu(3)-O(1W)-Cu(4)	119.5(4)	C(2)-N(1)-Cu(1)	114.3(7)
C(4)-N(1)-Cu(1)	127.1(7)	C(1)-N(2)-Cu(1)	115.4(7)
N(3)-N(2)-Cu(1)	138.8(7)	N(3)-N(4)-Cu(3)#2	114.8(7)
N(5)-N(4)-Cu(3)#2	133.8(7)	C(8)-N(6)-Cu(1)	114.1(7)
C(12)-N(6)-Cu(1)	128.0(8)	C(7)-N(7)-Cu(1)	115.6(7)
N(8)-N(7)-Cu(1)	139.5(7)	N(8)-N(9)-Cu(2)	121.4(6)
C(18)-N(11)-Cu(2)	125.0(8)	C(14)-N(11)-Cu(2)	115.8(7)

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N(13)-N(12)-Cu(2)	138.1(6)	C(13)-N(12)-Cu(2)	114.4(7)
N(12)-N(13)-Cu(3)	128.3(6)	N(14)-N(13)-Cu(3)	122.4(7)
N(13)-N(14)-Cu(4)	118.4(6)	N(15)-N(14)-Cu(4)	129.7(7)
C(24)-N(16)-Cu(3)	128.6(8)	C(20)-N(16)-Cu(3)	113.8(7)
C(19)-N(17)-Cu(3)	113.1(7)	N(18)-N(17)-Cu(3)	140.1(7)
N(19)-N(18)-Cu(2)	124.4(7)	N(17)-N(18)-Cu(2)	126.4(7)
N(18)-Cu(2)-O(2W)	92.6(4)	N(18)-Cu(2)-N(12)	93.7(4)
O(2W)-Cu(2)-N(12)	163.6(3)	N(18)-Cu(2)-N(11)	173.6(4)
O(2W)-Cu(2)-N(11)	93.4(4)	N(12)-Cu(2)-N(11)	79.9(4)
N(18)-Cu(2)-N(9)	90.2(4)	O(2W)-Cu(2)-N(9)	97.1(3)
N(12)-Cu(2)-N(9)	98.0(3)	N(11)-Cu(2)-N(9)	91.2(3)
O(1W)-Cu(3)-N(17)	173.9(3)	O(1W)-Cu(3)-N(13)	88.8(3)
N(17)-Cu(3)-N(13)	91.4(3)	O(1W)-Cu(3)-N(16)	98.3(3)
N(17)-Cu(3)-N(16)	80.5(3)	N(13)-Cu(3)-N(16)	167.0(3)
O(1W)-Cu(3)-N(4)#2	93.5(3)	N(17)-Cu(3)-N(4)#2	92.5(3)
N(13)-Cu(3)-N(4)#2	90.7(3)	N(16)-Cu(3)-N(4)#2	99.7(4)
O(1W)-Cu(4)-O(1W)#3	180.0(1)	O(1W)-Cu(4)-N(14)	89.7(3)
O(1W)#3-Cu(4)-N(14)	90.3(3)	O(1W)-Cu(4)-N(14)#3	90.3(3)
O(1W)#3-Cu(4)-N(14)#3	89.7(3)	N(14)-Cu(4)-N(14)#3	180.0(2)

Symmetry codes for **1**: #1 -x+2,-y-2,-z+2 #2 -x+3,-y-1,-z+1 #3 -x+4,-y-1,-z

Cu(1)-N(3)#1	1.974(7)	Cu(1)-N(1)	1.990(7)
Cu(1)-N(2)	2.001(7)	Cu(1)-O(1W)	2.012(7)
Cu(1)-N(7)	2.253(6)	N(9)-Cu(2)#3	1.984(6)
N(10)-Cu(2)	2.005(6)	N(11)-Cu(3)	2.012(7)
N(12)-Cu(3)	2.037(6)	N(14)-Cu(2)	2.014(7)
Cu(2)-N(9)#3	1.984(6)	Cu(3)-N(11)#4	2.012(7)
Cu(3)-N(12)#4	2.037(6)	N(3)#1-Cu(1)-N(1)	174.3(3)
N(3)#1-Cu(1)-N(2)	93.9(3)	N(1)-Cu(1)-N(2)	80.8(3)
N(3)#1-Cu(1)-O(1W)	90.9(3)	N(1)-Cu(1)-O(1W)	92.0(3)

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N(2)-Cu(1)-O(1W)	142.1(3)	N(3)#1-Cu(1)-N(7)	100.0(2)
N(1)-Cu(1)-N(7)	83.5(2)	N(2)-Cu(1)-N(7)	104.7(2)
O(1W)-Cu(1)-N(7)	111.4(3)	C(2)-N(1)-Cu(1)	115.8(5)
C(6)-N(1)-Cu(1)	125.9(6)	C(1)-N(2)-Cu(1)	112.7(5)
N(3)-N(2)-Cu(1)	141.9(5)	N(4)-N(3)-Cu(1)#1	126.2(5)
N(2)-N(3)-Cu(1)#1	123.7(5)	C(12)-N(6)-Cu(2)	126.7(6)
C(8)-N(6)-Cu(2)	114.7(5)	N(8)-N(7)-Cu(1)	112.1(4)
C(7)-N(7)-Cu(1)	139.5(5)	N(8)-N(9)-Cu(2)#3	126.1(5)
N(10)-N(9)-Cu(2)#3	124.9(4)	C(7)-N(10)-Cu(2)	113.0(5)
N(9)-N(10)-Cu(2)	141.2(5)	C(18)-N(11)-Cu(3)	126.7(6)
C(14)-N(11)-Cu(3)	114.3(6)	C(13)-N(12)-Cu(3)	110.6(5)
N(13)-N(12)-Cu(3)	142.4(6)	N(13)-N(14)-Cu(2)	132.0(6)
N(15)-N(14)-Cu(2)	115.3(5)	N(9)#3-Cu(2)-N(10)	93.7(2)
N(9)#3-Cu(2)-N(14)	92.5(2)	N(10)-Cu(2)-N(14)	173.5(3)
N(9)#3-Cu(2)-N(6)	174.3(2)	N(10)-Cu(2)-N(6)	80.7(3)
N(14)-Cu(2)-N(6)	93.1(3)	N(11)#4-Cu(3)-N(11)	180.0(1)
N(11)#4-Cu(3)-N(12)	98.2(3)	N(11)-Cu(3)-N(12)	81.8(3)
N(11)#4-Cu(3)-N(12)#4	81.8(3)	N(11)-Cu(3)-N(12)#4	98.2(3)
N(12)-Cu(3)-N(12)#4	180.0(1)		

Symmetry codes for **2**: #1 -x-1,-y+1,-z #2 -x-2,-y,-z-1 #3 -x-1,-y,-z #4 -x-2,-y-1,-z

Ag(1)-N(6)	2.162(7)	Ag(1)-N(7)	2.176(7)
Ag(1)-Ag(2)	3.1280(11)	N(2)-Ag(3)	2.287(7)
Ag(2)-N(5)	2.183(7)	Ag(2)-N(8)	2.280(7)
Ag(2)-O(5)	2.474(7)	Ag(3)-N(10)	2.215(7)
Ag(3)-N(9)#5	2.327(7)	N(9)-Ag(3)#7	2.327(7)
N(6)-Ag(1)-N(7)	162.4(3)	N(6)-Ag(1)-Ag(2)	94.9(2)
N(7)-Ag(1)-Ag(2)	67.53(19)	N(3)-N(2)-Ag(3)	121.7(6)
C(1)-N(2)-Ag(3)	132.3(6)	N(5)-Ag(2)-N(8)	157.3(3)

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N(5)-Ag(2)-O(5)	108.5(3)	N(8)-Ag(2)-O(5)	94.0(3)
N(5)-Ag(2)-Ag(1)	93.6(2)	N(8)-Ag(2)-Ag(1)	64.12(18)
O(5)-Ag(2)-Ag(1)	157.63(18)	N(10)-Ag(3)-N(2)	127.0(3)
N(10)-Ag(3)-N(9)#5	130.9(3)	N(2)-Ag(3)-N(9)#5	101.5(3)
N(3)#6-N(5)-Ag(2)	125.0(6)	N(4)-N(5)-Ag(2)	122.8(6)
C(8)-N(6)-Ag(1)	119.5(6)	C(7)-N(6)-Ag(1)	123.8(6)
N(10)-N(7)-Ag(1)	135.4(6)	N(8)-N(7)-Ag(1)	113.4(5)
N(7)-N(8)-Ag(2)	113.2(5)	N(9)-N(8)-Ag(2)	137.8(6)
C(9)-N(9)-Ag(3)#7	120.5(6)	N(8)-N(9)-Ag(3)#7	132.4(5)
N(7)-N(10)-Ag(3)	130.0(5)	C(9)-N(10)-Ag(3)	124.6(6)
Symmetry codes for 3 : #1 -x,-y-1,-z-2 #2 x+1,-y-1/2,z+1/2 #3 x-1,y,z #4 x+1,y,z			
#5 x,-y-1/2,z+1/2	#6 x-1,-y-1/2,z-1/2	#7 x,-y-1/2,z-1/2	
Ag(1)-N(4)	2.319(11)	Ag(1)-N(7)#2	2.357(10)
Ag(1)-N(5)	2.369(10)	Ag(2)-N(2)	2.221(11)
Ag(2)-N(12)#3	2.229(12)	Ag(2)-N(13)#2	2.321(11)
N(7)-Ag(1)#4	2.357(10)	N(12)-Ag(2)#5	2.229(12)
N(13)-Ag(2)#4	2.321(11)	N(4)-Ag(1)-N(7)#2	104.3(4)
N(4)-Ag(1)-N(5)	113.3(4)	N(7)#2-Ag(1)-N(5)	106.2(4)
N(2)-Ag(2)-N(12)#3	136.3(4)	N(2)-Ag(2)-N(13)#2	110.2(4)
N(12)#3-Ag(2)-N(13)#2	111.3(4)	N(4)-N(2)-Ag(2)	124.8(9)
N(1)-N(2)-Ag(2)	123.9(9)	N(2)-N(4)-Ag(1)	121.9(8)
N(3)-N(4)-Ag(1)	128.8(9)	N(13)-N(5)-Ag(1)	117.1(8)
C(1)-N(5)-Ag(1)	136.4(8)	N(13)-N(7)-Ag(1)#4	123.9(8)
N(12)-N(7)-Ag(1)#4	127.1(8)	C(1)-N(12)-Ag(2)#5	133.4(8)
N(7)-N(12)-Ag(2)#5	119.9(8)	N(7)-N(13)-Ag(2)#4	119.8(8)
N(5)-N(13)-Ag(2)#4	129.6(8)		
Symmetry codes for 4 : #1 -x,y,-z+3/2 #2 -x+1/2,y-1/2,-z+3/2 #3 x,y-1,z #4 -			
x+1/2,y+1/2,-z+3/2	#5 x,y+1,z		

Table S2. Hydrogen-bonding geometries (Å, °) of compounds **1** and **4**

	D-H...A	D-H	H...A	D...A	D-H...A
Compound 1	C10-H10A...O19	0.929	2.533	3.400	155.41
Compound 4	C9-H9A...O18	0.930	3.019	3.066	84.12

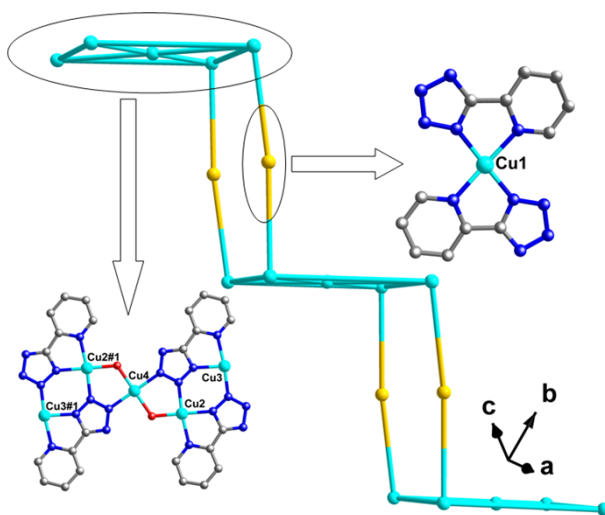


Fig. S1. The penta-nuclear clusters and mono-nuclear Cu^{II} subunits for construction of the 1D stair-like Cu-2-ptz chain in **1**. **Symmetry codes: #1 4-x, -1-y, -z**

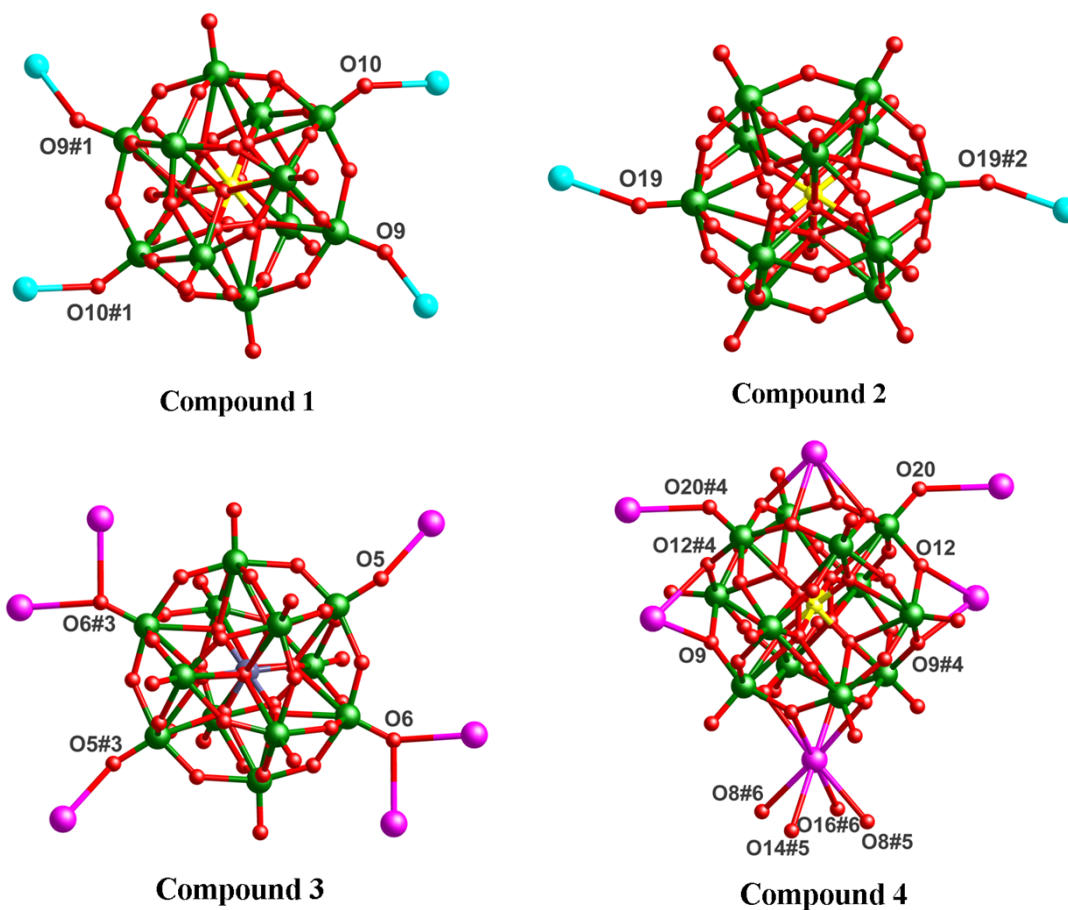


Fig. S2. The coordination modes of Keggin anions in the title compounds. Symmetry codes: #1 2-x, -2-y, 2-z; #2 -2-x, -y, -1-z; #3 -x, -1-y, -2-z; #4 -x, y, -1.5-z; #5 -x, 1+y, 1.5-z; #6 x, 1+y, z.

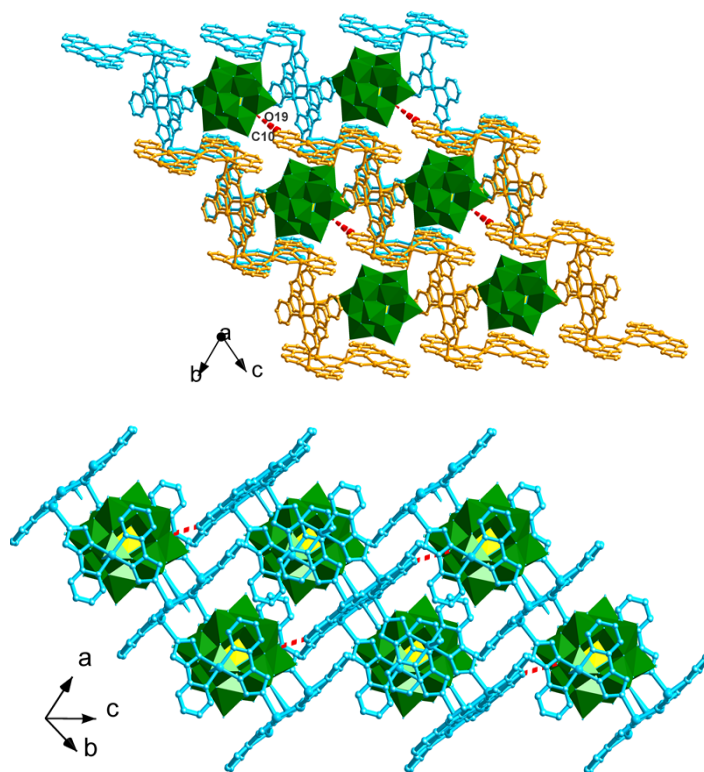


Fig. S3. The 3D supramolecular framework of **1** with C–H···O (C(10)–H(10A)···O(19)) = 3.400 Å hydrogen-bonding interactions.

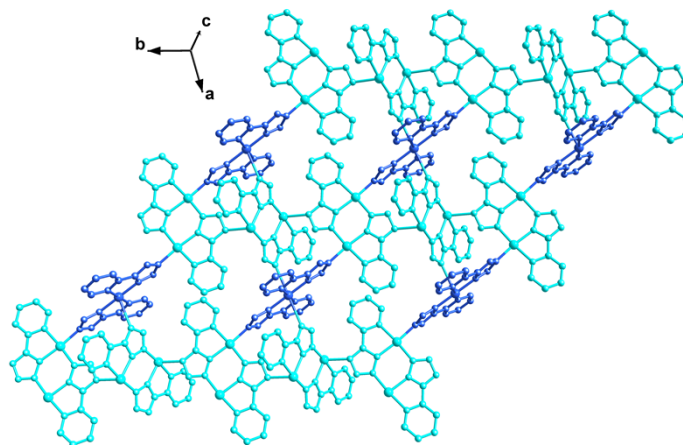


Fig. S4. The 2D layer of compound **2**.

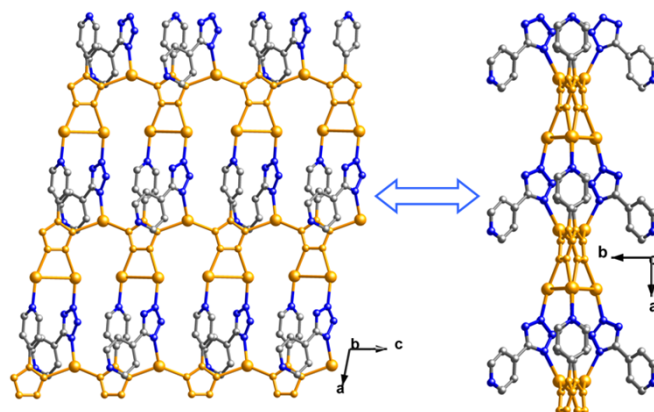


Fig. S5. The 2D layer of compound **3** viewed along *b*- and *c*-axis.

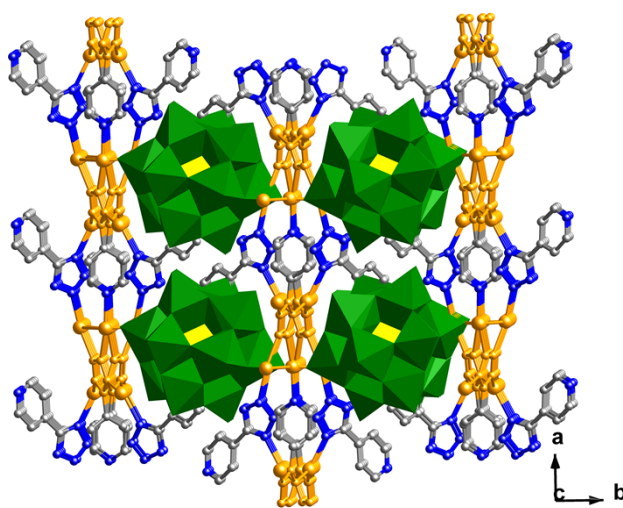


Fig. S6. The 3D structure of compound **3**.

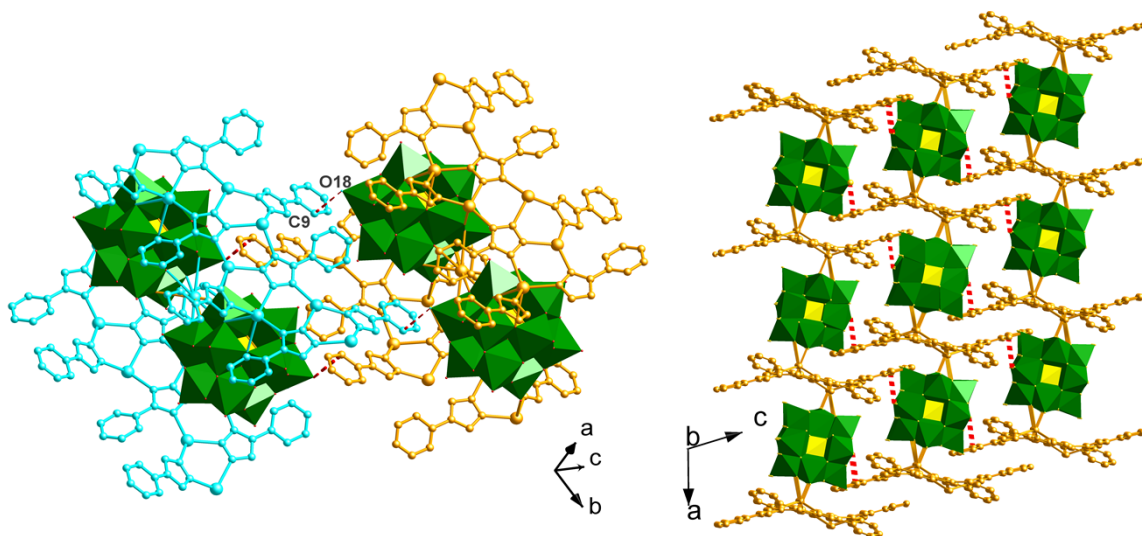


Fig. S7. The 3D supramolecular framework of **4** with C–H···O (C(9)–H(9A)···O(18) = 3.066 Å) hydrogen-bonding interactions.

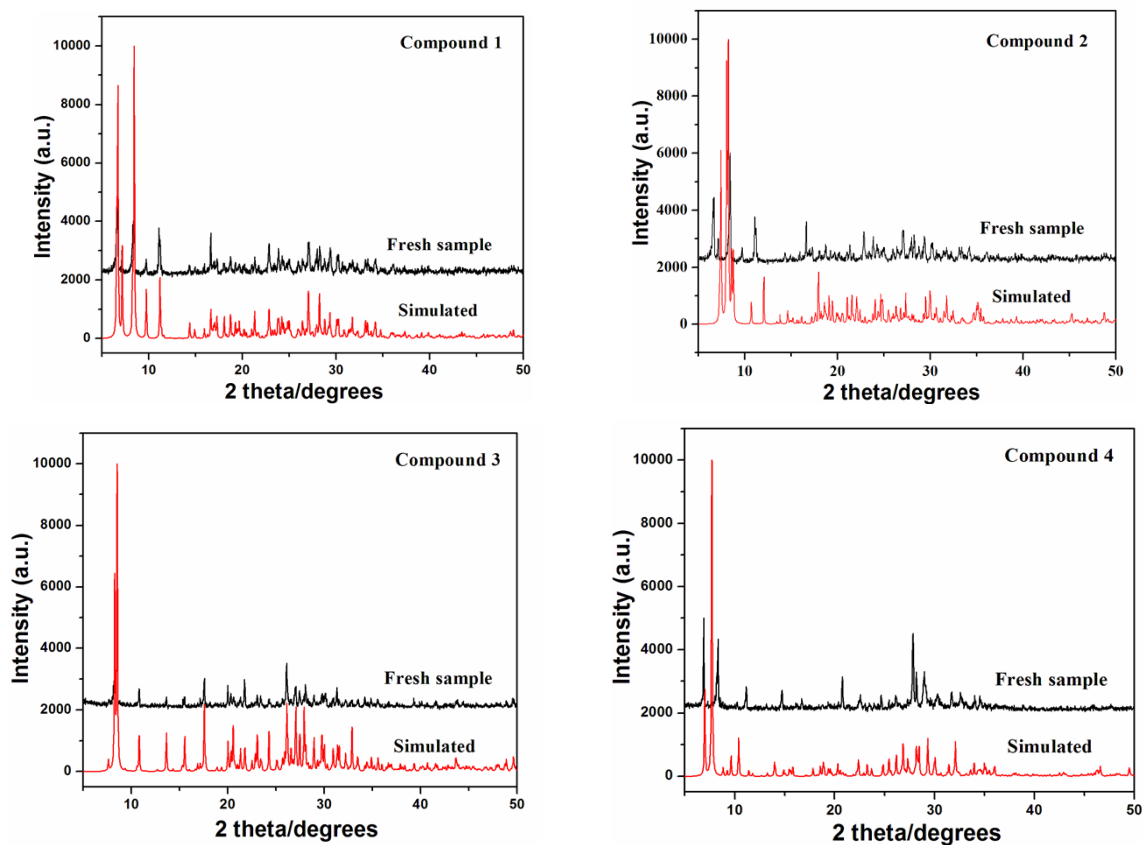


Fig. S8. The XRD patterns of compounds 1–4.

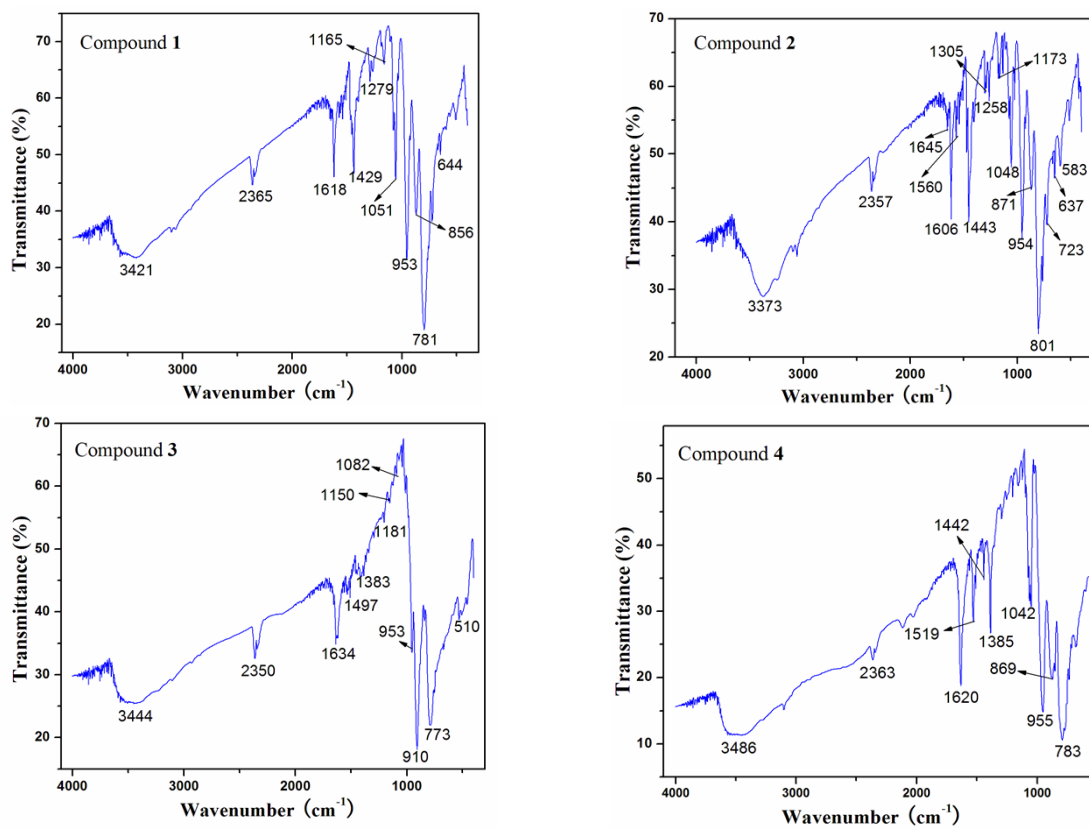


Fig. S9. The IR spectra of compounds 1–4.

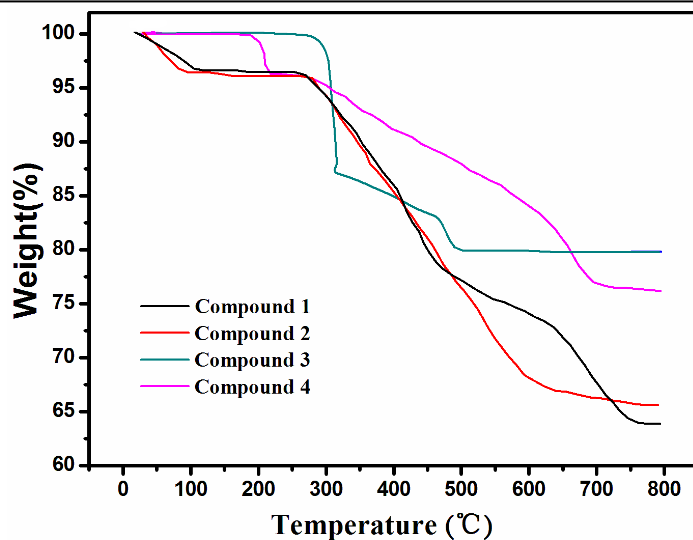


Fig. S10. The TGA curves of the compounds 1–4.

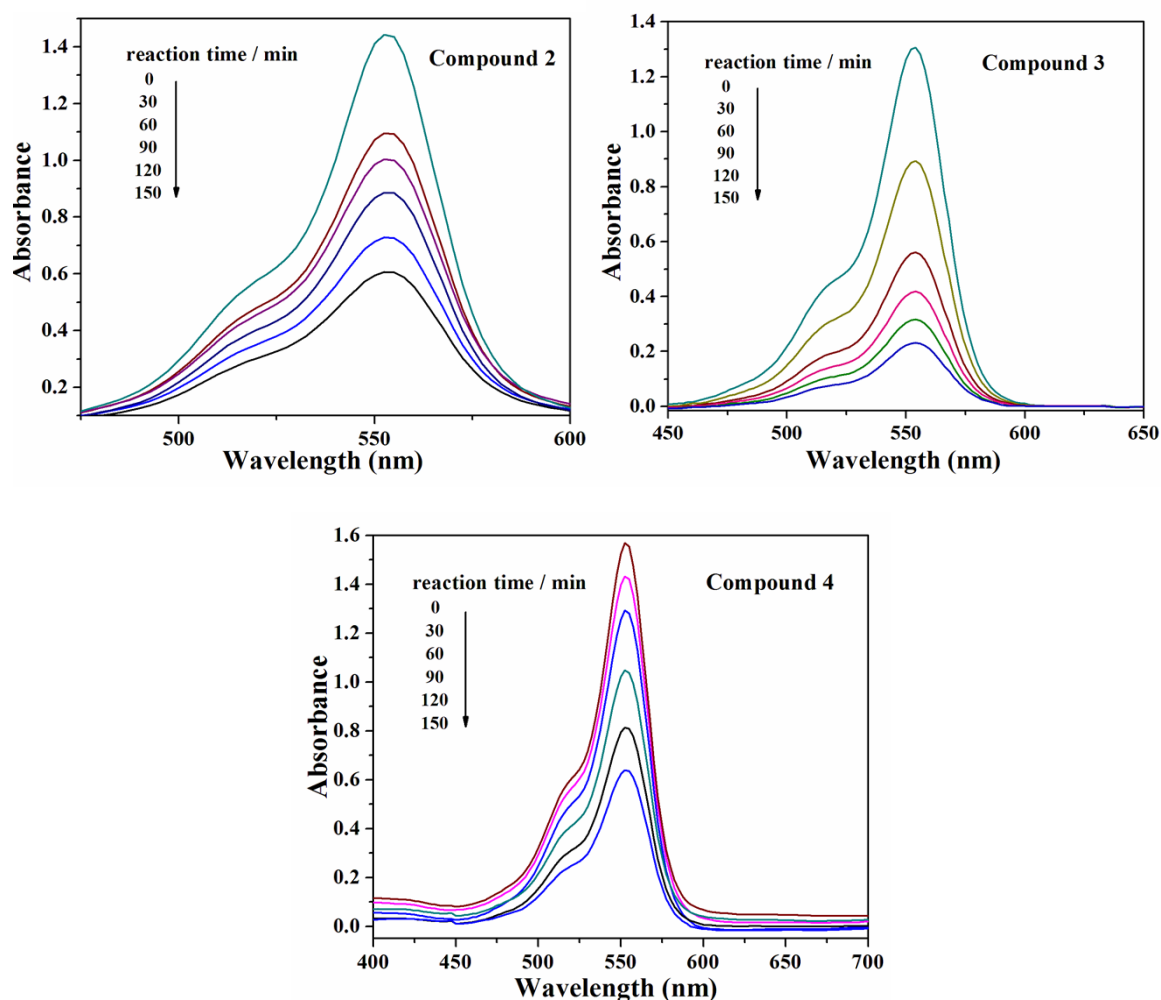


Fig. S11. Absorption spectra of the RhB solution during the decomposition reaction under UV irradiation with the presence of compounds 2–4.

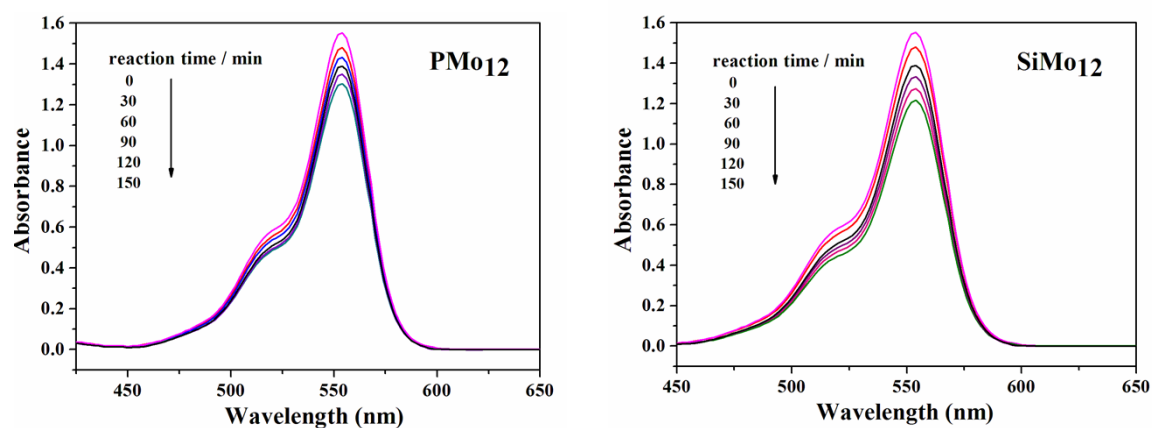


Fig. S12. Absorption spectra of the RhB solution during the decomposition reaction under UV irradiation with the presence of parent POM (PMo₁₂ and SiMo₁₂).

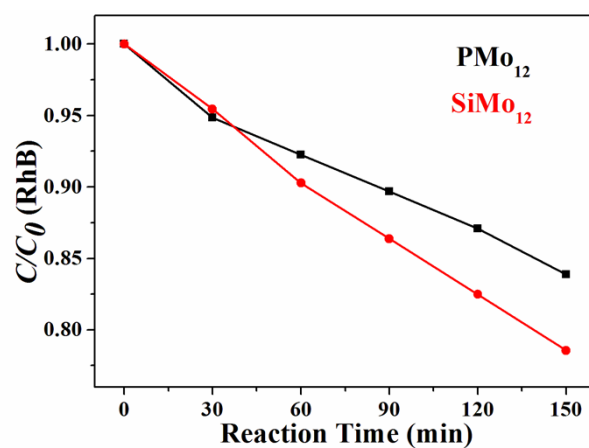


Fig. S13. Photocatalytic decomposition rate of MB solution under UV irradiation with the use of parent POMs. (Black: PMo₁₂ and Red: SiMo₁₂).

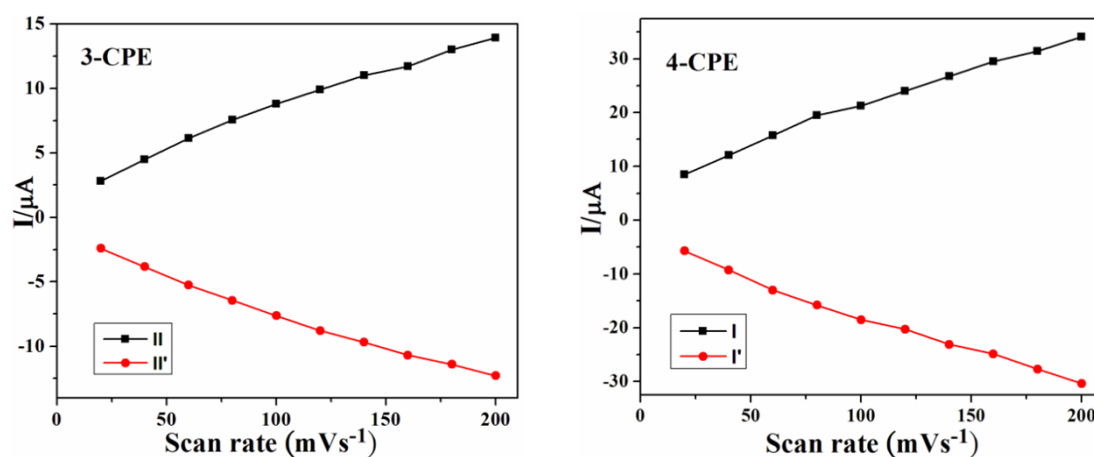


Fig. S14. The dependence of anodic peak (II) and cathodic peak (II') currents for 3-CPE and anodic peak (I) and cathodic peak (I') currents for 4-CPE on scan rates.

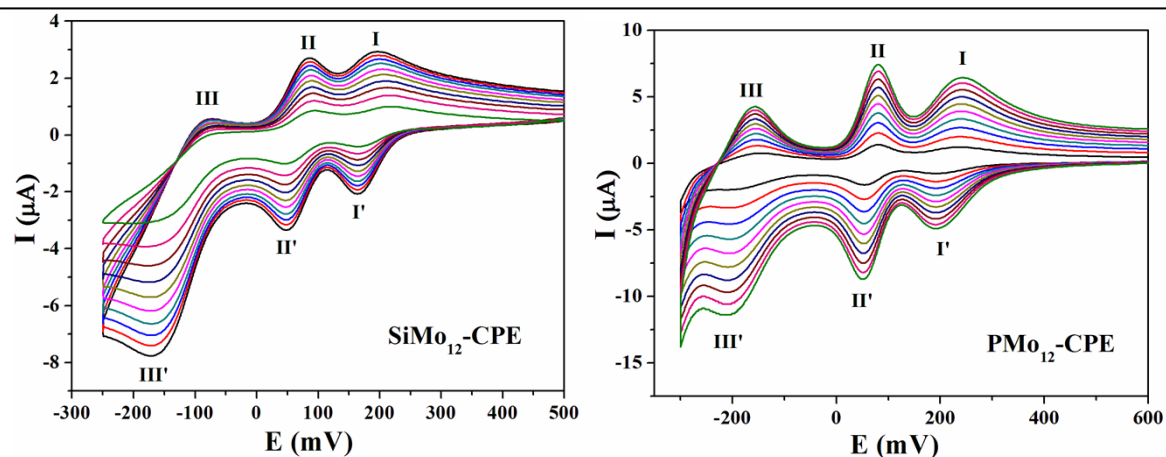


Fig. S15. Cyclic voltammograms of the SiMo₁₂- and PMo₁₂-CPEs in 0.1 M H₂SO₄ + 0.5 M Na₂SO₄ aqueous solution at different scan rates (from inner to outer: 20, 40, 60, 80, 100, 120, 140, 160, 180 and 200 mVs⁻¹, respectively).

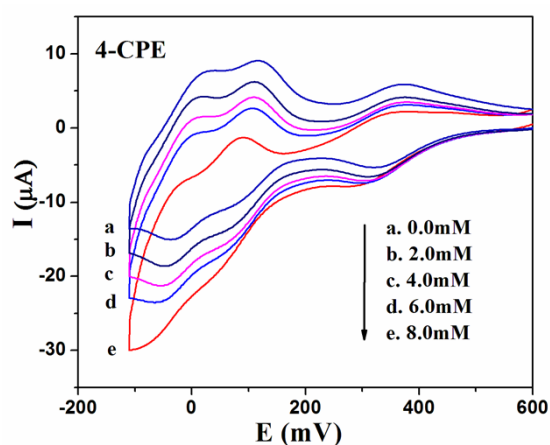


Fig. S16. Cyclic voltammograms of the 4-CPE in 0.1 M H₂SO₄ + 0.5 M Na₂SO₄ aqueous solution containing 0.0–8.0 mM KNO₂. Scan rate: 200mV·s⁻¹.