

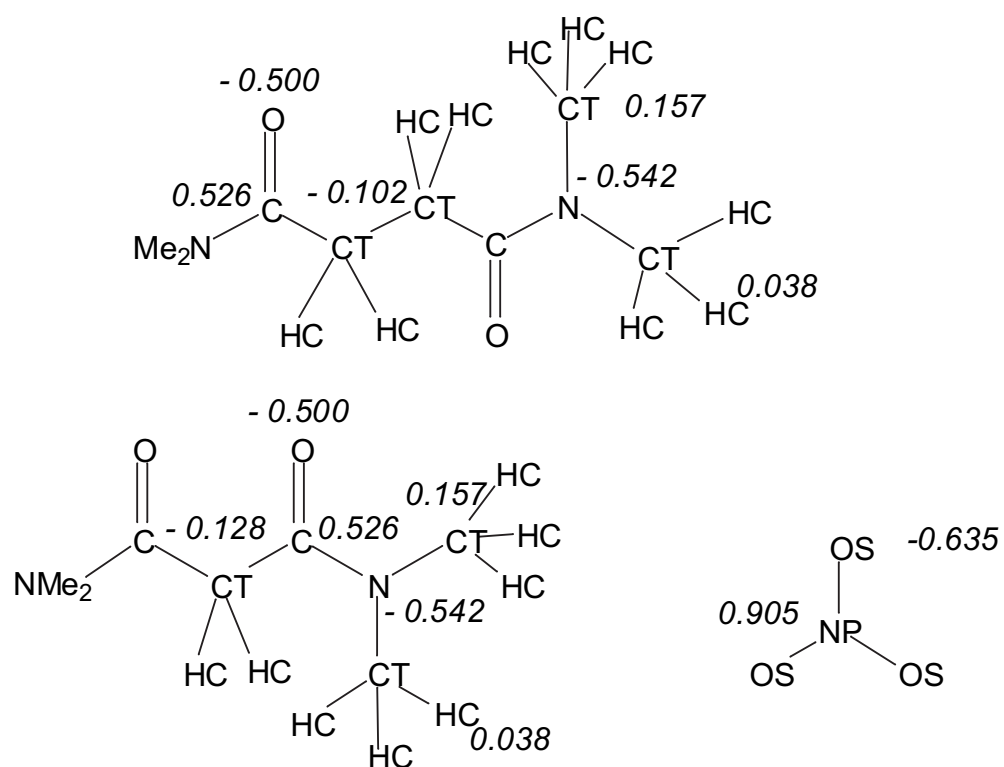


Figure S1. Atomic charges (in cursive script) and AMBER atom types used to simulate the AcA and AccA complexes and NO_3^- .

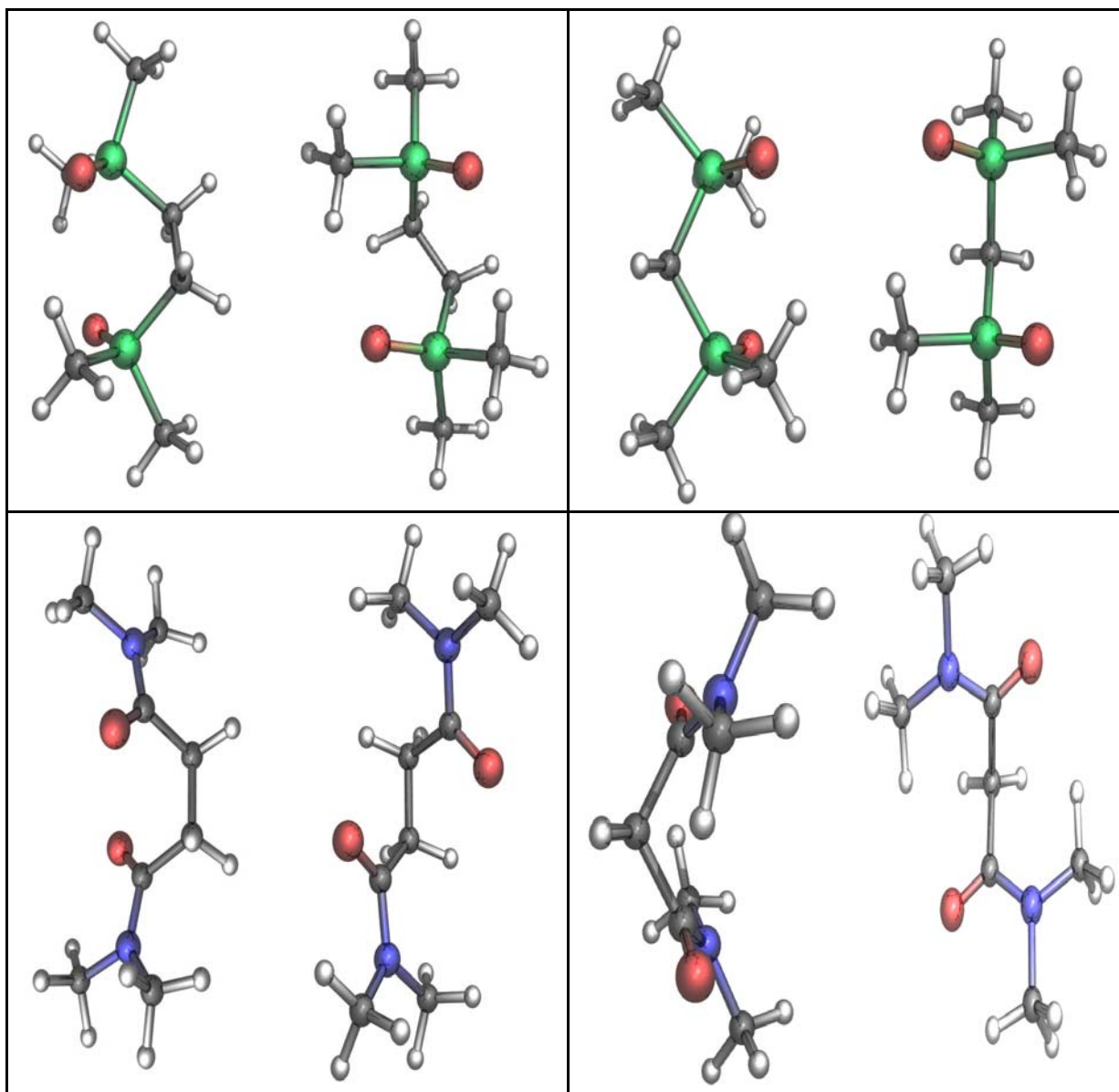


Figure S2. Optimized **L** ligands (orthogonal views). From top left to bottom right: **L** = PccP, PcP, AccA and AcA.

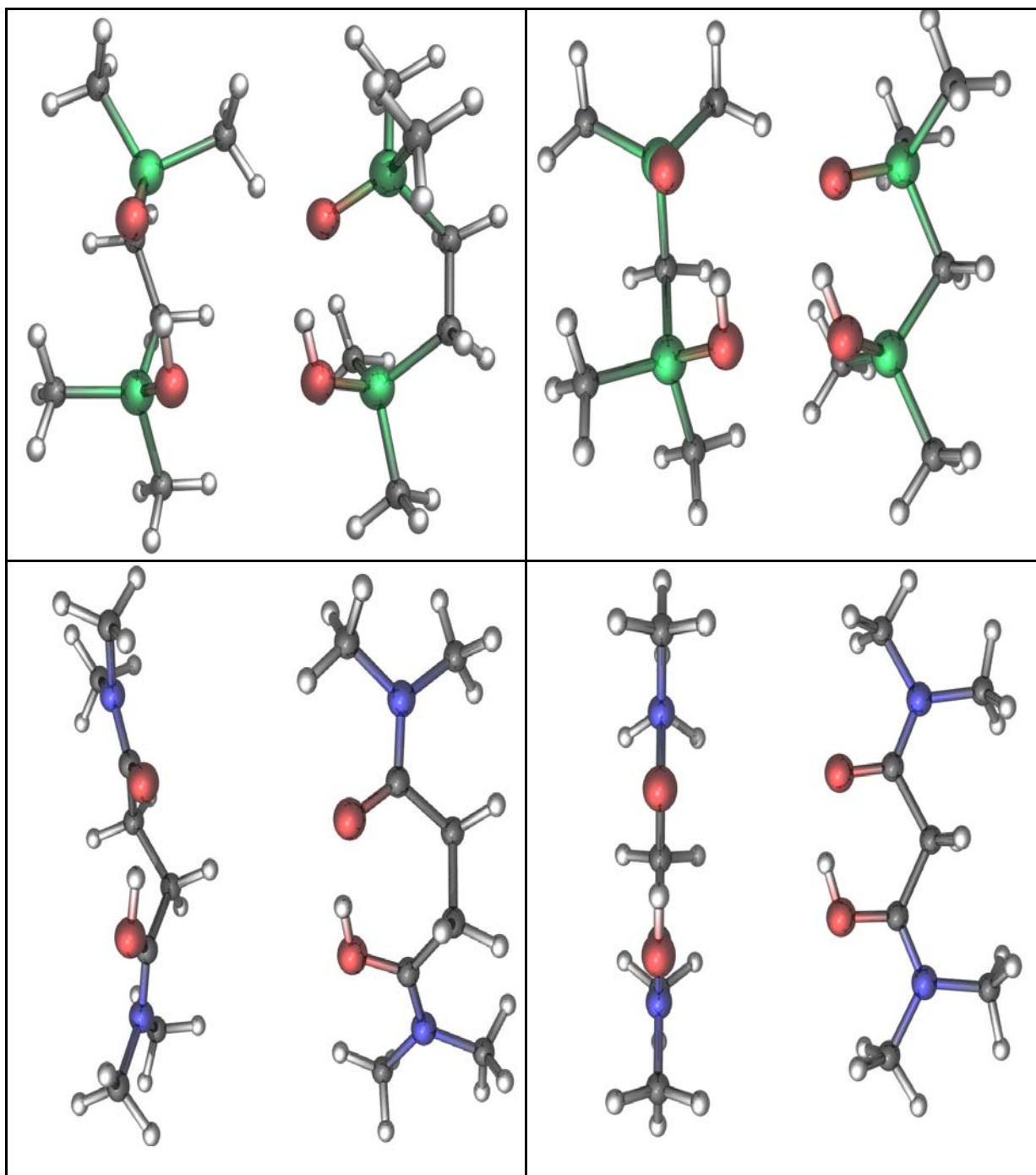


Figure S3. Optimized protonated $L \cdot H^+$ ligands (orthogonal views). From top left to bottom right: L = PccP, PcP, AccA and AcA.

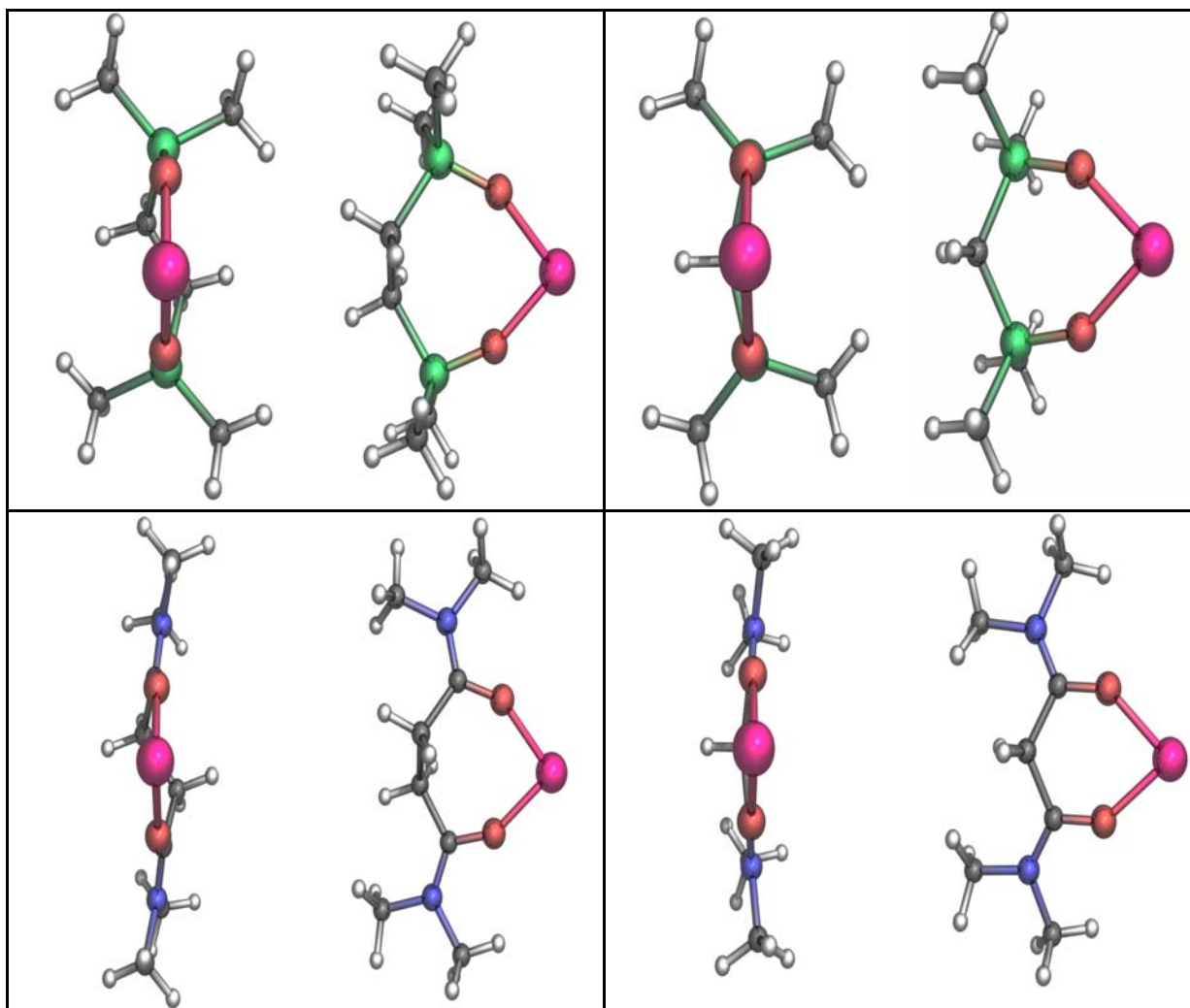


Figure S4. Optimized $\text{Eu} \cdot \text{L}^{3+}$ complexes (orthogonal views). From top left to bottom right: **L** = PccP, PcP, AccA and AcA.

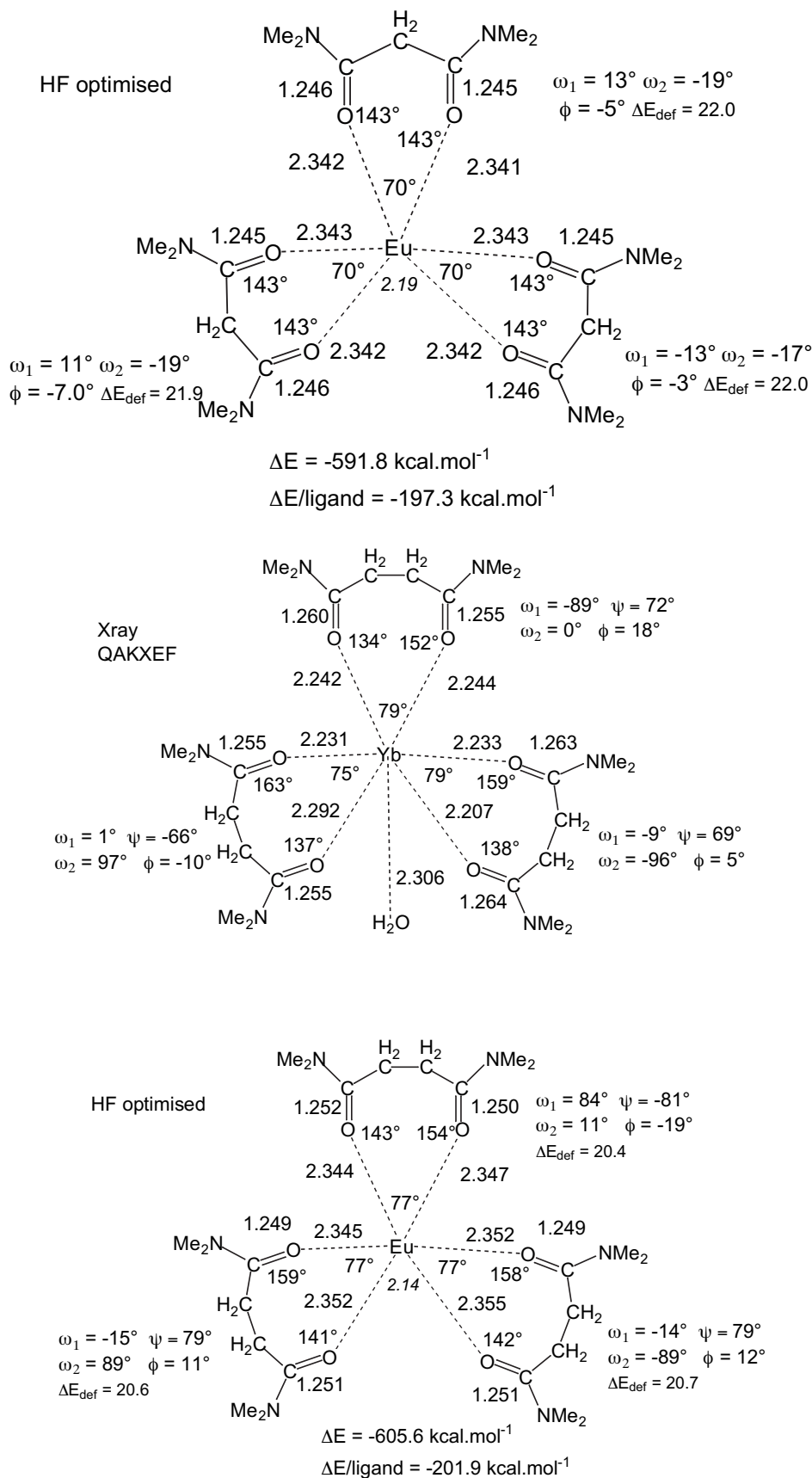


Figure S5. Optimized parameters of $\text{Eu} \cdot (\text{AcA})_3^{3+}$ (top) and $\text{Eu} \cdot (\text{AccA})_3^{3+}$ complexes and Xray structure of the QAKXEF $\text{Yb}(\text{H}_2\text{O})(\text{AccA})_3$ complex. ΔE_{def} is the deformation energy of **L** (kcal/mol).

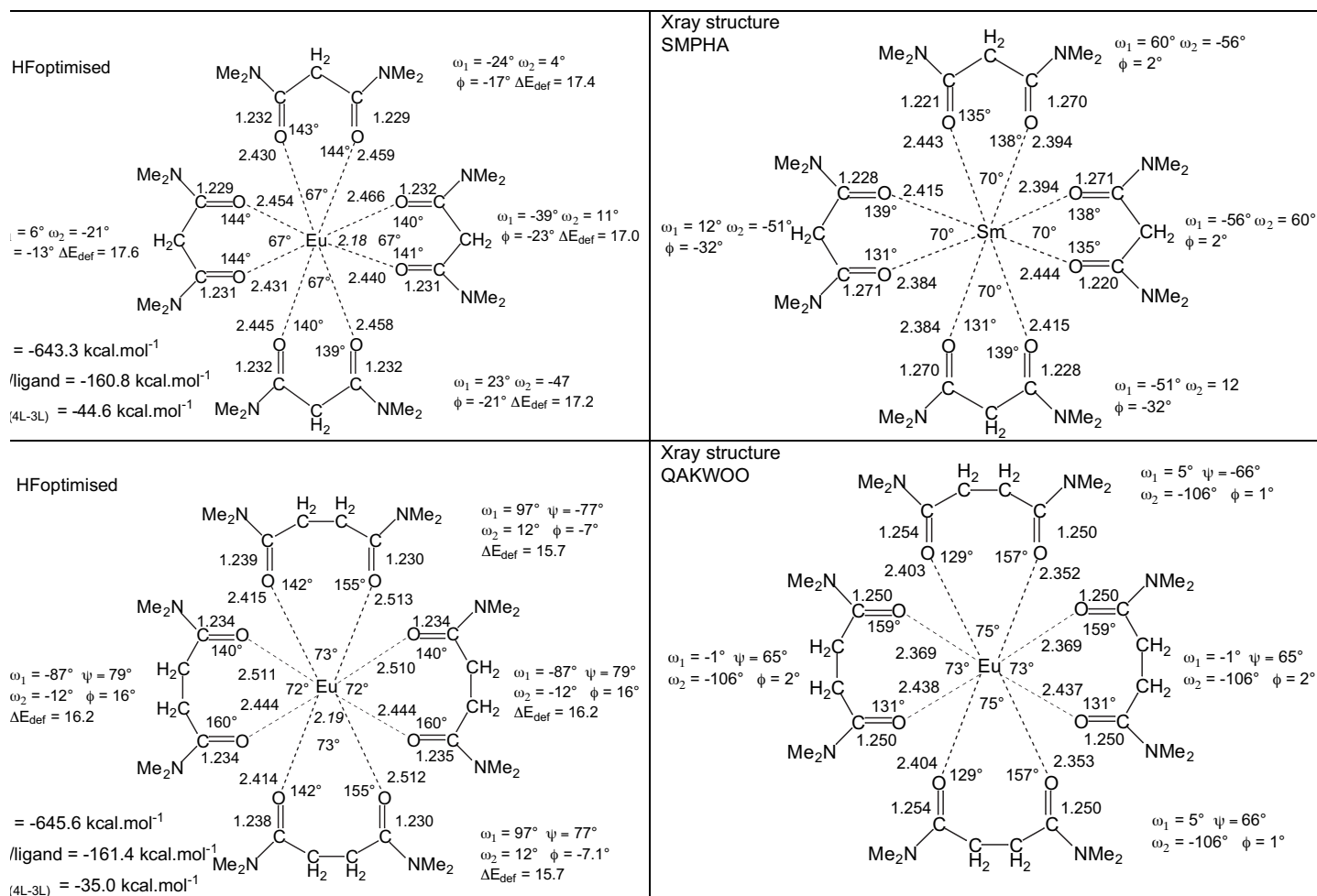


Figure S6. Optimized parameters of $\text{Eu} \cdot (\text{AcA})_4^{3+}$ (top) and $\text{Eu} \cdot (\text{AccA})_4^{3+}$ complexes and Xray structure of the SMPHA $\text{Sm} \cdot (\text{AcA})_4^{3+}$ and QAKWOO $\text{Eu} \cdot (\text{AccA})_4^{3+}$ complexes. ΔE_{def} is the deformation energy of **L** (kcal/mol).

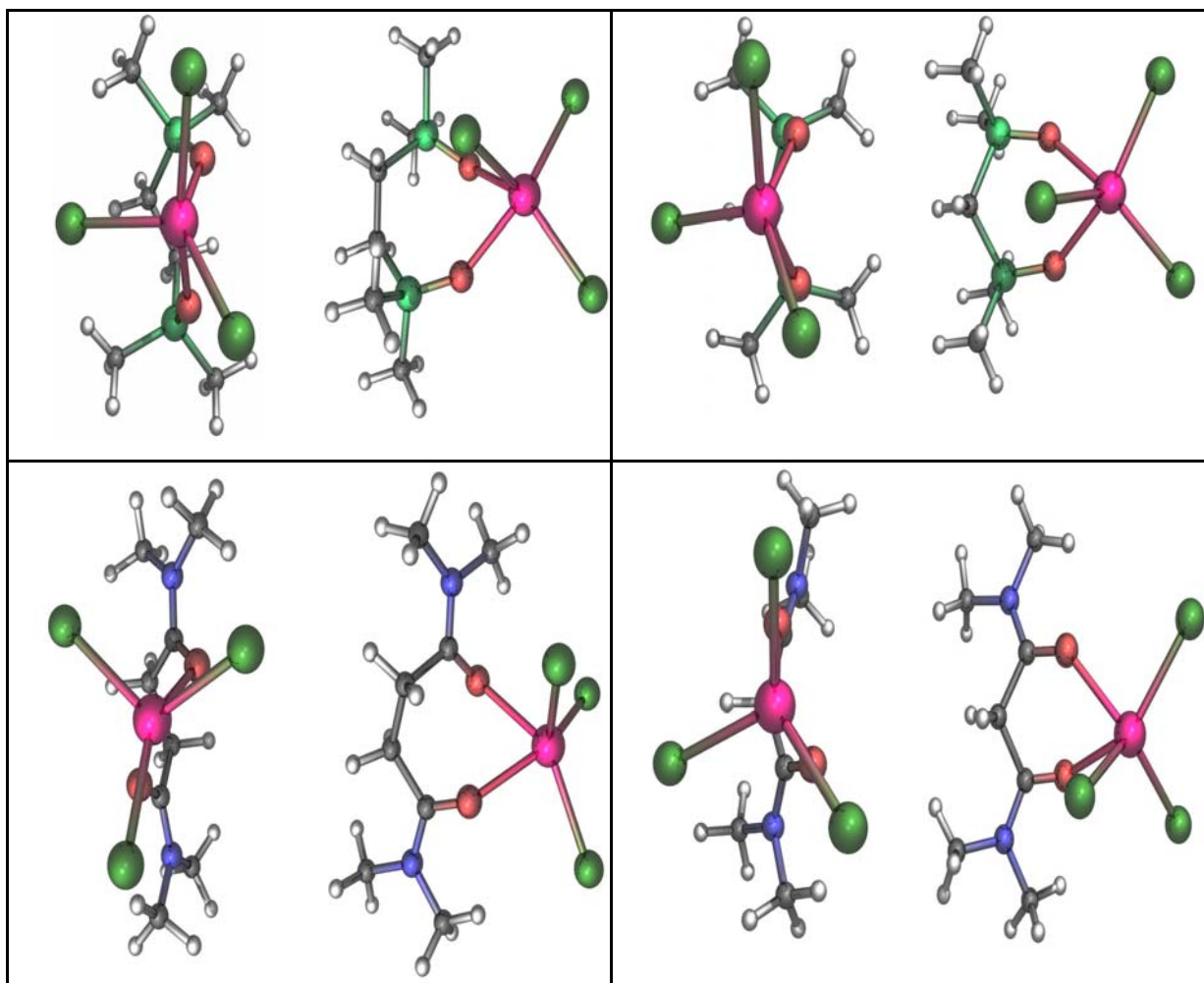


Figure S7. Optimized $\text{EuCl}_3 \cdot \text{L}^{3+}$ bidentate complexes (orthogonal views). From top left to bottom right: **L** = PccP, PcP, AccA and AcA.

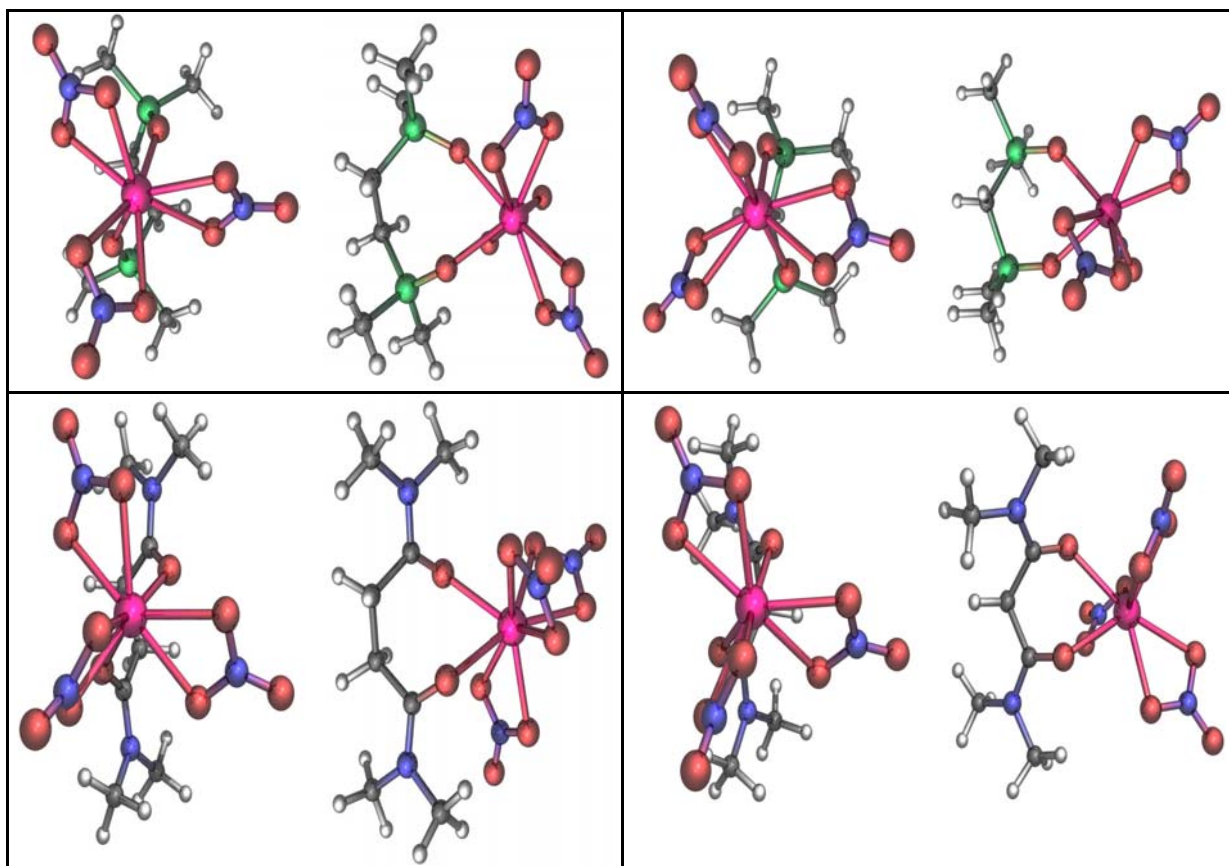


Figure S8. Optimized $\text{Eu}(\text{NO}_3)_3 \cdot \text{L}$ bidentate complexes (orthogonal views). From top left to bottom right: **L** = PccP, PcP, AccA and AcA.

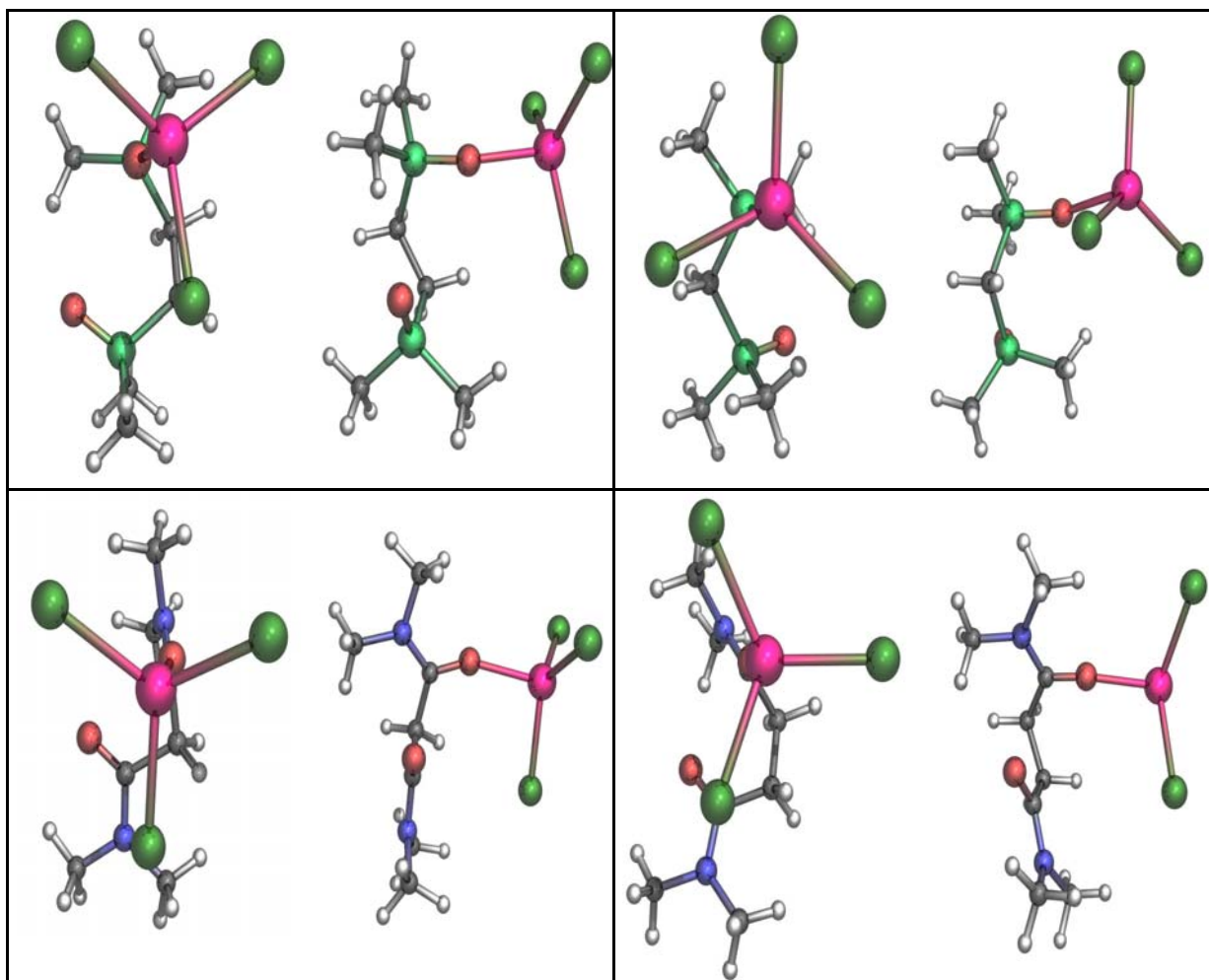


Figure S9. Optimized $\text{EuCl}_3 \cdot \text{L}^{3+}$ monodentate complexes (orthogonal views). From top left to bottom right:
L = PccP, PcP, AccA and AcA.

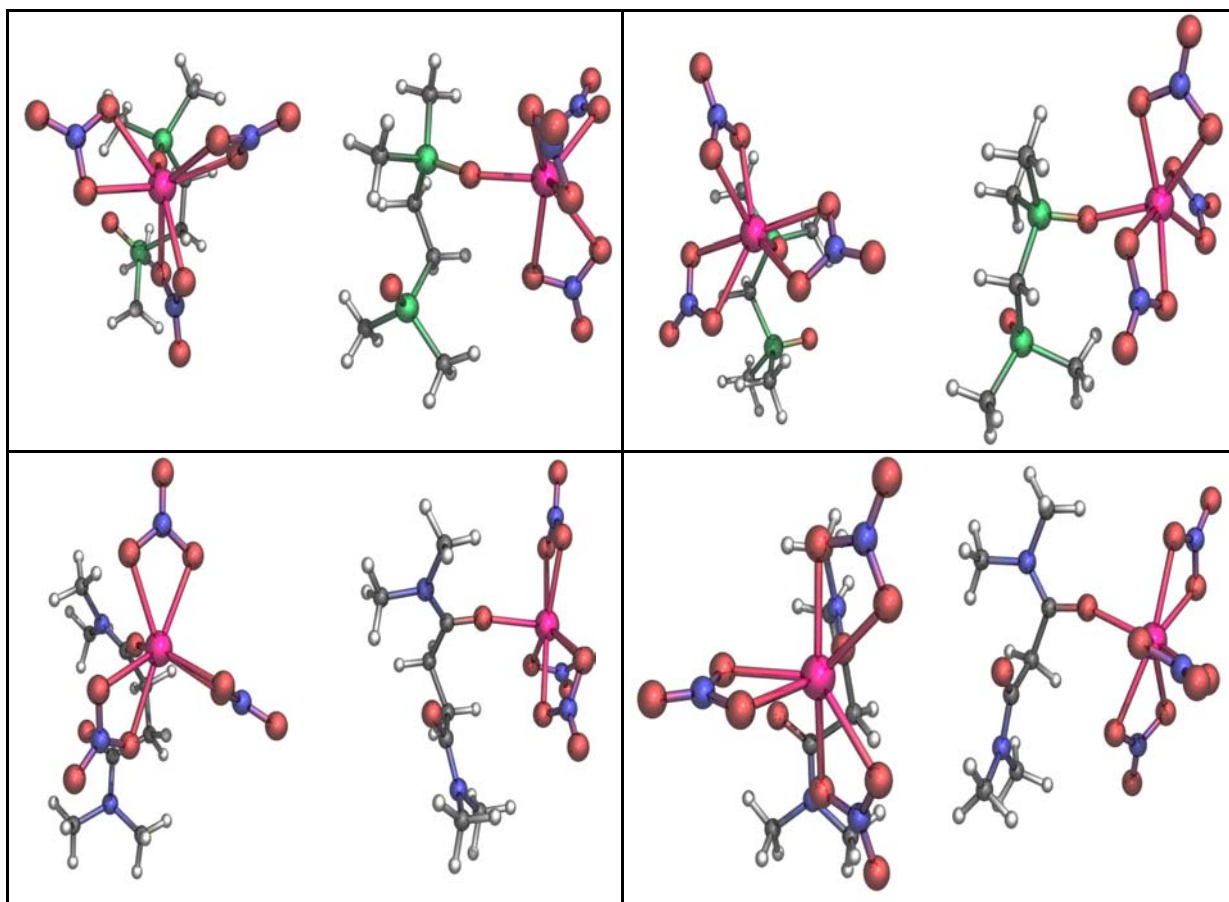


Figure S10. Optimized $\text{Eu}(\text{NO}_3)_3 \cdot \text{L}$ monodentate complexes (orthogonal views). From top left to bottom right: **L** = PccP, PcP, AccA and AcA.