## Species with negative electron affinity and standard DFT methods

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## **Supplementary Material**

Part 1. Calculation Details

## Table S1. Detailed results with 6-31+G\* and 6-311+G(2df,p) basis sets\*

		B3LYP/6-31+G*									B3LYP/6-311+G(2df,p) // B3LYP/6-31+G*					
species	type	μ	LVUMO	E(neutral)	ZPE	E(vertica	E(anion)	ZPE	VEA	AEA	E(n)	E(vertica	E(anion)	VEA	AEA	
						anion)						anion)				
11,1-dichloroethylene	V İ	1.51	-0.95	-997.7811	0.0336	-997.7453			-0.975		-997.8702	-997.8382		-0.873		
2 1,3-cyclohexadiene	V	0.49	-0.91	-233.4288	0.1225	-233.3924			-0.992		-233.4950	-233.4620		-0.898		
3 acetaldehyde	V	2.99	-1.08	-153.8397	0.0557	-153.7893			-1.372		-153.8891	-153.8414		-1.298		
4 adenine N	v	2.46	-0.92	-467.3399	0.1120	-467.3100			-0.847		-467.4779	-467.4507		-0.740		
5 bro mobe nzene	V	1.85	-0.79	-2803.3846		-2803.3498			-0.946		-2805.8651	-2805.8331		-0.870		
6 butadiene	v	0.00	-1.10	-156.0011	0.0852	-155.9700			-0.846		-156.0487	-156.0207		-0.763		
7 chlorobenzene	v	1.89	-0.78	691.8549		-691.8149			-1.091		691.9511	-691.9151		-0.980		
8 chloroethylene	v	1.59	-0.59	-538.1910	0.0428	-538.1390			-1.417		-538.2480	-538.1995		-1.320		
9 chloroform	V	1.25	-1.62	-1419.2843	0.0199	-1419.2720			-0.335		-1419.3940	-1419.3818		-0.331		
10 cis-dichloroethylene	V	2.02	-0.76	-997.7843	0.0342	-997.7348			-1.345		-997.8727	-997.8282		-1.209		
11 cyclopentyl radical	v	0.20	5.44	-195.9017	0.1261	-195.8723	-195.8892	0.1233	-0.799	-0.263	-195.9572	195 9315	195.9473	-0.698	-0.192	
12 cytosine	v	6.85	-1.32	394.9498	0.0984	-394.9266	-394.9439	0.0944	-0.631	-0.051	-395.0752	-395.0550	-395.0687	-0.547	-0.068	
13 dimethylaminomethyl radical	v	1.33	-0.86	173 8273	0.1073	173.7962	-173.8096	0.1042	-0.845	-0.399	-173.8807	173 8524	173.8643	-0.768	-0.364	
14 ethviene	v	0.00	-0.22	-78.5933	0.0511	78.5281		•••••	-1.774		-78.6192	-78.5565		1.706		
15 ethyl radical	v	0.34	-1.52	-79.1629	0.0595	-79.1415	-79.1492	0.0578	-0.580	-0.323	-79.1879	-79.1698	-79.1769	-0.495	-0.253	
16 fluorobenzene	v	1.75	-0.75	331 5012	0.0924	331,4576	-331.4640	0.0841	-1.187	-0.788	331,5978	331 5577	331,5622	1.092	-0.743	
17 isopropyl radical	v	0.25	-1 20	118 4837	0.0882	118 4547	118 4686	0.0860	-0 791	-0 352	118 5 199	118 4943	118 5068	-0.698	-0 296	
18 naphtalene	v	0.00	1.33	385,9069	0.1476	385,8930	385.8973	0.1414	-0.378	-0.093	-386.0107	-386.0003	386.0030	-0.283	-0.039	
19 norbornadiene	v	0.04	0 5 6	271 4878	0 1286	-271 4460		•••••	1 1 3 6		271 5622	271 5231		1 0 6 5		
20 nyrazine	v	0.00	1 83	-264 3286	0.0769	264 3186	-264 3256	0 0726	-0 270	0.036	-264 4021	-264 3946	-264 4002	0 203	0.068	
21 nyridazine	v	4 43	1 80	-264 2989	0.0763	-264 2876	-264 2954	0.0713	-0.308	0.039	264 3725	264 3635	264 3700	-0 244	0.067	
22 pyridine	v	2 38	1.00	-248 2958	0.0889	-248 2616	201.2551	0.0715	-0.932	0.055	248 3648	-248 3337	201.5700	-0.846	0.007	
23 pyrimidine	v	2 48	1 57	-264 3351	0.0772	-264 3156	-264 3230	0 0717	-0 531	-0 179	-264 4087	-264 3917	-264 3977	-0.461	-0 149	
24 styrene	v	0.21	1 2 9	-309 6613	0 1335	-309 6453	20	0.07 1.	-0 434	0.1.5	-309 7477	-309 7351	201.5577	-0 342	0.1.0	
25 t-buthv	v	0.27	-4 75	157 8045	0.1355	157 7761	-157 7950	0 1143	-0 772	-0 186	157 8515	-157 8264	-157 8432	-0.684	-0 152	
26 thiophene	v	0.55	-0.64	-553 0114	0.0667	-552 9628	10.0000	0.11.10	-1 322	0.100	-553 0895	-553 0438	197.10.191	-1 242	0.101	
27 thymine	v	4 63	154	-454 1581	0 1149	-454 1444	-454 1587	0 1102	-0 372	0 145	-454 3001	-454 2890	-454 3004	0 301	0 137	
28 trans-dichloroethylene	v	0.00	-0.94	-997 7839	0.0339	-997 7433	10 112001		-1 105	0.2.5	-997 8722	-997 8361		-0.984	0.10.	
29 trichloroethylene	v	0.94	-1 12	-1457 3716	0.0248	-1457 3396			-0.872		-1457 4923	-1457 4652		-0 739		
30 uracil	v	4 67	168	-414 8377	0.0871	-414 8246	-414 8401	0.0827	-0355	0 184	414 9689	-414 9588	-414 9708	-0 274	0 172	
31 1 2 4-trimethylbenzene	ŇV	0 40	0.21	350 2132	0 1836	-350 1715	-350 1717	0 1817	-1 134	-1079	350 3106	-350 2735	350 2734	1 0 0 8	-0.961	
32 acetone	NV	3 1 9	-0.75	-193 1663	0.0838	-193 1200	550127 27	0.101.	-1 2 5 8	1.07.5	-193 2267	-193 1882	550.275	-1 048	0.501	
33 aniline	NV	1.63	0.28	287.6161	0.1171	-287.5757			-1.099		-287.7028	-287.6654		1.018		
34 anisole	NV	1 3 9	-0.37	-346 7855	0 1333	-346 7440			-1 129		-346 8870	346 8512		-0975		
35 cis-butene	NV	0.03	0.33	-157 2299	0 1079	-157 1767			-1 448		-157 2778	-157 2341		1 1 90		
36 cvclohexene	NV	0.35	0.46	234.6562	0.1466	-234.6094			-1.272		234,7231	-234.6826		1 101		
37 furan	NV	0 73	-0.07	-230 0314	0 0700	-229 9794			-1 417		-230 1010	-230 0560		1 2 2 6		
38 m-xvlene	NV	0.36	-0.29	310 8956	0 1554	-310 8554			-1 092		-310 9822	-310 9457		-0 9 92		
39 o-xvlene	NV	0.67	-0.20	310 8950	0 1562	-310 8510			-1 198		-310 9815	310 9431		1 0 4 6		
40 phenol	NV	1 42	-0.48	-307 4803	0 1045	-307 4399			-1 099		-307 5753	-307 5388		-0 9 9 4		
41 propene	NV	0.43	0.10	117 9139	0.0798	117.8548			-1.610		117,9509	117 9020		1.328		
42 pyrrole	NV	1 93	0.65	210 1778	0.0825	-210 1375			-1 096		210 2417	-210 2060		0973		
43 trans-butene	NV	0.00	0.42	157 2336	0.1081	157.1792			-1.480		157.2815	157 2301		-1.397		
44 trimethylethylene	NV	0.31	0.37	-196.5480		196 4992			-1.329		196.6068	196 5657		-1.117		
45 CO2	NV	0.00	-0.56	188.5904	0.0116	188,5565	-188.5805	0.0083	-0.921	-0.182	-188.6569	188.6233	188.6415	0.915	-0.332	
46 guanine	NV	6.80	-0.58	-542 5764	0.1168	-542 5606	542 5621	0.1139	0.431	0.311	-542 7433	-542 7286	-542 7297	-0.400	-0.291	
* Diple moment in Debves, LVUMO	enera	y in e	V, Eneraie	es in Hartrees	, VEA an	d AEA in eV				0.0.2.2						

 $^{\ast}$  Dipole moment in Debyes, LVUMO energy in eV, Energies in Hartrees, VEA and AEA in eV



Figure S1. Correlation between the experimental and B3LYP/6-31+G\* data. See text and Figure 1.

Table S2. Detailed results with AUG-CC-PVDZ and AUG-CC-PVTZ basis sets*															
		B3LYP/AUG-cc-pVDZ B3LYP/AUG-CC-PVTZ // B3LYP/AUG-CC-PVDZ													
				E(neutral)	ZPE	E(vertic a	E(anion)	ZPE	VEA	AEA	E(n)	E(vertical	E(anion)	VEA	AEA
species	type	μ	LVUMO			anion)						anion)			
11,3-cyclohexadiene	v	0.50	-1.01	-233.4449	0.1216	-233.4261			-0.511		-233.5067	-233.4916		-0.411	
9 chloroform	v	1.11	-1.55	-1419.3600	0.0196	-1419.3492			-0.293		-1419.4135	-1419.4046		-0.243	
11 cyclopentyl radical	v	0.19		-195.9147	0.1249	-195.8921	-195.9085	0.1224	-0.614	-0.098	-195.9678	-195.9469	-195.9609	-0.567	-0.117
13 dimethylaminomethyl radical	v	1.30		-173.8398	0.1059	-173.8189	-173.8276	0.1030	-0.568	-0.252	-173.8902	-173.8711	-173.8776	-0.519	-0.265
15 ethyl radical	v	0.32		-79.1668	0.0588	-79.1495	-79.1576	0.0568	-0.470	-0.194	-79.1929	-79.1772	-79.1837	-0.428	-0.194
17 isopropy   radica	v	0.17		-118.4897	0.0870	-118.4684	-118.4803	0.0850	-0.578	-0.200	-118.5271	-118.5074	-118.5172	-0.535	-0.213
23 pyrimidine	v	2.38	-1.62	-264.3566	0.0767	-264.3393			-0.469		264,4194	-264.4071		-0.335	
24 styrene	v	0.20	-1.38	-309.6830	0.1330	-309.6705			-0.339		-309.7624	-309,7509		-0.311	
25 t-buthy	v	0.19		-157.8125	0.1154	-157.7953	-157.8085	0.1127	-0.467	-0.037	-157.8608	-157.8449	-157.8560	-0.431	-0.055
3 acetaldehyde	NV	2.91	-1.15	-153.8532	0.0550	-153.8342			-0.517		-153.8962	-153.8812		-0.408	
32 acetone	NV	3.11	-0.81	-193.1816	0.0827	-193.1664	-193.1665	0.0814	-0.414	-0.375	-193.2359	-193.2238	-193.2236	-0.328	-0.297
33 aniliine	NV	1.57	-0.40	287.6438	0.1164	-287.6289			-0.405		-287.7167	-287.7052		-0.313	
34 anisole	NV	1.31	0.49	-346.8126	0.1323	-346.7962			-0.445		-346.9028	-346.8900		-0.347	
5 bromoben zene	NV	1.86	-0.91	-2805.8307	0.0900	-2805.8160	-2805.8438	0.0865	-0.398	0.453	-2805.9609	-2805.9494	-2805.9715	-0.311	0.385
6 butadiene	NV	0.00	-1.19	-156.0121	0.0849	-155.9875			-0.669		-156.0569	-156.0380		-0.517	
7 chlorobenzene	NV	1.89	-0.87	-691.8948	0.0906	-691.8786	691.8788	0.0892	-0.442	-0.398	-691.9672	-691.9544	691.9542	-0.349	-0.315
35 cis-butene	NV	0.31	0.19	-157.2389	0.1068	-157.2188			-0.546		-157.2868	-157.2704		-0.444	
36 cyclohexene	NV	0.37	0.38	-234.6720	0.1454	-234.6535			-0.504		-234.7354	-234.7205		-0.405	
14 ethylene	NV	0.00	-0.34	-78.5984	0.0508	-78.5719	-78.5719	0.0503	-0.720		-78.6237	-78.6029	78.6028	-0.568	-0.555
16 fluorobenzene	NV	1.62	-0.85	-331.5256	0.0918	-331.5078	-331.5080	0.0908	-0.484	-0.454	-331.6124	-331.5982	331.5981	-0.386	-0.364
37 furan	NV	0.67	-0.20	230.0517	0.0695	-230.0304			-0.581		230,1114	-230.0942		-0.467	
19 norbornadiene	NV	0.09	-0.66	-271.5072	0.1276	-271.4881			-0.520		-271.5751	-271.5597		-0.420	
40 phenol	NV	1.27	-0.60	-307.5119	0.1043	-307,4969			-0.408		-307.5897	-307.5780		-0.317	
41 propene	NV	0.43	-0.03	-117.9209	0.0791	-117.8981	-117.8982	0.0785	-0.620	-0.600	-117.9576	-117.9390	-117.9389	-0.507	-0.491
22 pyridine	NV	2.28	-1.14	-248.3142	0.0883	248.2980			-0.440		-248.3756	-248.3630		-0.343	
42 pyrrole	NV	1.84	0.52	-210.1982	0.0822	-210.1820			-0.441		-210.2518	-210.2392		-0.345	
43 trans-butene	NV	0.00	0.28	-157.2425	0.1070	-157.2201			-0.610		-157.2903	-157.2720		0.497	
44 trimethylethylene	NV	0.29	0.28	-196.5592	0,1341	-196.5401			-0.519		-196.6178	-196.6023		-0.420	
30 uraci	NV	4.54	-1.74	-414.8843	0.0868	-414.8805	-414.8813	0.0856	-0.102	-0.046	-414.9845	-414.9831	414.9834	-0.038	0.004
45 CO2	NV	0.00	-0.53	-188.6142	0.0116	-188.5812			-0.898		-188.6632	-188.6355		-0.754	
* Diple moment in Debyes, LVUMC	Diple moment in Debyes, LVUMC energy in eV. Energies in Harrees, VEA and AEA in eV														



**Figure S2**. Correlation between experimental and calculated electron affinities (B3LYP/AUG-ccpVTZ basis). Same as figure S1.



**Figure S3**. Koopman's Theorem correlation at the B3LYP/6-31+G\* level.



Figure S4. Koopman's Theorem correlation at the B3LYP/AUG-cc-pVDZ level.



Part 2. Structure, Dipole, SOMO diagrmas and spin density of some species under study

**Figure S5**. Acetone B3LYP/6-31+G\*, Non -Valence RA. Top: LUMO of the neutral. Lower left corner: SOMO of the vertical RA. Lower right corner: spin density. The cyan arrow show the dipole orientation.



**Figure S6.** Neutral and non valence RA obtained at the B3LYP/6-31+G\* level for furan presented as in figure S5.



Figure S7. Non valence RA found for phenol (organized as Fig. S5).



Figure S8. Spin density in the trimethylethylene RA.



**Figure S9**. Ethyl radical and its anions. From top to bottom, SOMO of the neutral, HOMO of the vertical anion and HOMO of the optimized anion.

## Part 3. Geometries of all species under study.

All the XYZ coordinates in Angstorms for all the optimized species under study have been gathered in a single ASCII file named Structures-xyz.txt