

DRY [BMI][PF ₆]						
SOLUTE	BMI ⁺	PF ₆ ⁻	Box Size (Å)	Temp (K)	Time (ns)	
<i>Cis</i> - CMPO	200	200	41.7 x 41.7 x 41.7	300 / 400	2 / 2	
<i>Trans</i> - CMPO	200	200	41.6 x 41.6 x 41.6	300 / 400	2 / 2	
[UO ₂ (NO ₃)(CMPO) _{bi}] ⁺	300	301	47.7 x 47.7 x 47.7	300 / 400	2 / 2	
[UO ₂ (NO ₃)(CMPO) _{mono}] ⁺	300	301	47.7 x 47.7 x 47.7	300 / 400	2 / 2	
[UO ₂ (NO ₃)(CMPO) _{mono}] ⁺ + NO ₃ ⁻ + CMPO	200	200	42.0 x 42.0 x 42.0	300 / 400	2 / 2	
[UO ₂ (NO ₃) ₂ (CMPO) ₂]	200	200	41.9 x 41.9 x 41.9	300 / 400	2 / 2	
[UO ₂ (NO ₃) ₂ (CMPO) ₂] (NO ₃ ⁻ cst) ^{a)}	200	200	41.8 x 41.8 x 41.8	300 / 400	2 / 2	
HUMID [BMI][PF ₆]						
SOLUTE	BMI ⁺	PF ₆ ⁻	H ₂ O	Box Size (Å)	Temp (K)	Time (ns)
<i>Cis</i> - CMPO	200	200	200	42.7 x 42.7 x 42.7	300	2
<i>Trans</i> - CMPO	200	200	200	42.6 x 42.6 x 42.6	300	2
[UO ₂ (NO ₃)(CMPO) _{bi}] ⁺	200	201	200	42.8 x 42.8 x 42.8	300	2
[UO ₂ (NO ₃)(CMPO) _{bi}] ⁺ (NO ₃ ⁻ cst) ^{a)}	200	201	200	42.8 x 42.8 x 42.8	300	2
[UO ₂ (NO ₃)(CMPO) _{mono}] ⁺	200	201	200	42.7 x 42.7 x 42.7	300	2
[UO ₂ (NO ₃)(CMPO) _{mono}] ⁺ (NO ₃ ⁻ cst) ^{a)}	200	201	200	42.7 x 42.7 x 42.7	300	2
[UO ₂ (NO ₃)(CMPO) _{mono}] ⁺ + NO ₃ ⁻ + CMPO	200	200	200	42.9 x 42.9 x 42.9	300	2
[UO ₂ (NO ₃) ₂ (CMPO) ₂]	200	200	200	42.8 x 42.8 x 42.8	300	2
[UO ₂ (NO ₃) ₂ (CMPO) ₂] (NO ₃ ⁻ cst) ^{a)}	200	200	200	42.8 x 42.8 x 42.8	300	2

Table S1: Characteristics of the simulated systems.^{a)} NO₃⁻ constrained to be bidentate

Solute	Temp (K)	BMI ⁺	PF ₆ ⁻	E _{solv}
<i>Cis</i> – CMPO	300	-51 (4)	-48 (3)	-99 (5)
<i>Cis</i> – CMPO	400	-50 (5)	-49 (4)	-99 (6)
<i>Trans</i> – CMPO	300	-57 (4)	-34 (3)	-92 (4)
<i>Trans</i> – CMPO	400	-55 (4)	-37 (4)	-93 (6)
[UO ₂ (NO ₃)(CMPO) _{bi}] ⁺	300	340 (7)	-547 (12)	-207 (12)
[UO ₂ (NO ₃)(CMPO) _{bi}] ⁺	400	342 (8)	-529 (10)	-186 (9)
[UO ₂ (NO ₃)(CMPO) _{mono}] ⁺	300	331 (7)	-583 (9)	-252 (9)
[UO ₂ (NO ₃)(CMPO) _{mono}] ⁺	400	359 (8)	-579 (11)	-221 (9)
[UO ₂ (NO ₃) ₂ (CMPO) ₂]	300	-101 (5)	-42 (5)	-148 (8)
[UO ₂ (NO ₃) ₂ (CMPO) ₂]	400	-114 (7)	-52 (7)	-166 (8)
[UO ₂ (NO ₃) ₂ (CMPO) ₂] (NO ₃ ⁻ cst) ^{a)}	300	-101 (5)	-62 (5)	-163 (6)
[UO ₂ (NO ₃) ₂ (CMPO) ₂] (NO ₃ ⁻ cst) ^{a)}	400	-115 (7)	-69 (6)	-184 (8)

Table S2: Interaction energies (kcal/mol) and fluctuations (between brackets) of the different solutes with the [BMI][PF₆] “dry” ionic liquid and its ionic components.

^{a)} NO₃⁻ constrained to be bidentate

Solute	BMI ⁺	PF ₆ ⁻	H ₂ O	E _{solv}
<i>Cis</i> – CMPO	-40 (4)	-44 (3)	-32 (6)	-115 (7)
<i>Trans</i> – CMPO	-33 (3)	-36 (3)	-39 (4)	-108 (5)
[UO ₂ (NO ₃)(CMPO) _{bi}] ⁺	312 (8)	-488 (10)	-82 (9)	-260 (11)
[UO ₂ (NO ₃)(CMPO) _{bi}] ⁺ (NO ₃ ⁻ cst) ^{a)}	279 (8)	-437 (11)	-53 (8)	-211 (10)
[UO ₂ (NO ₃)(CMPO) _{mono}] ⁺	263 (12)	-402 (21)	-180 (23)	-320 (13)
[UO ₂ (NO ₃)(CMPO) _{mono}] ⁺ (NO ₃ ⁻ cst) ^{a)}	304 (6)	-475 (10)	-92 (7)	-264 (9)
[UO ₂ (NO ₃) ₂ (CMPO) ₂]	-108 (8)	-55 (8)	-65 (7)	-228 (9)
[UO ₂ (NO ₃) ₂ (CMPO) ₂] (NO ₃ ⁻ cst) ^{a)}	-96 (7)	-58 (8)	-40(8)	-195 (10)

Table S3: Interaction energies (kcal/mol) and fluctuations (between brackets) of the different solutes with the [BMI][PF₆][H₂O] “humid” ionic liquid and its different components.

^{a)} NO₃⁻ constrained to be bidentate

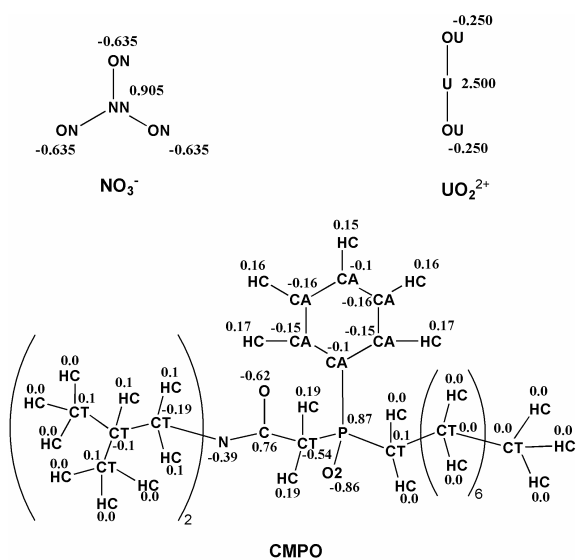


Figure S1: Charges and AMBER atom types of the different solutes.

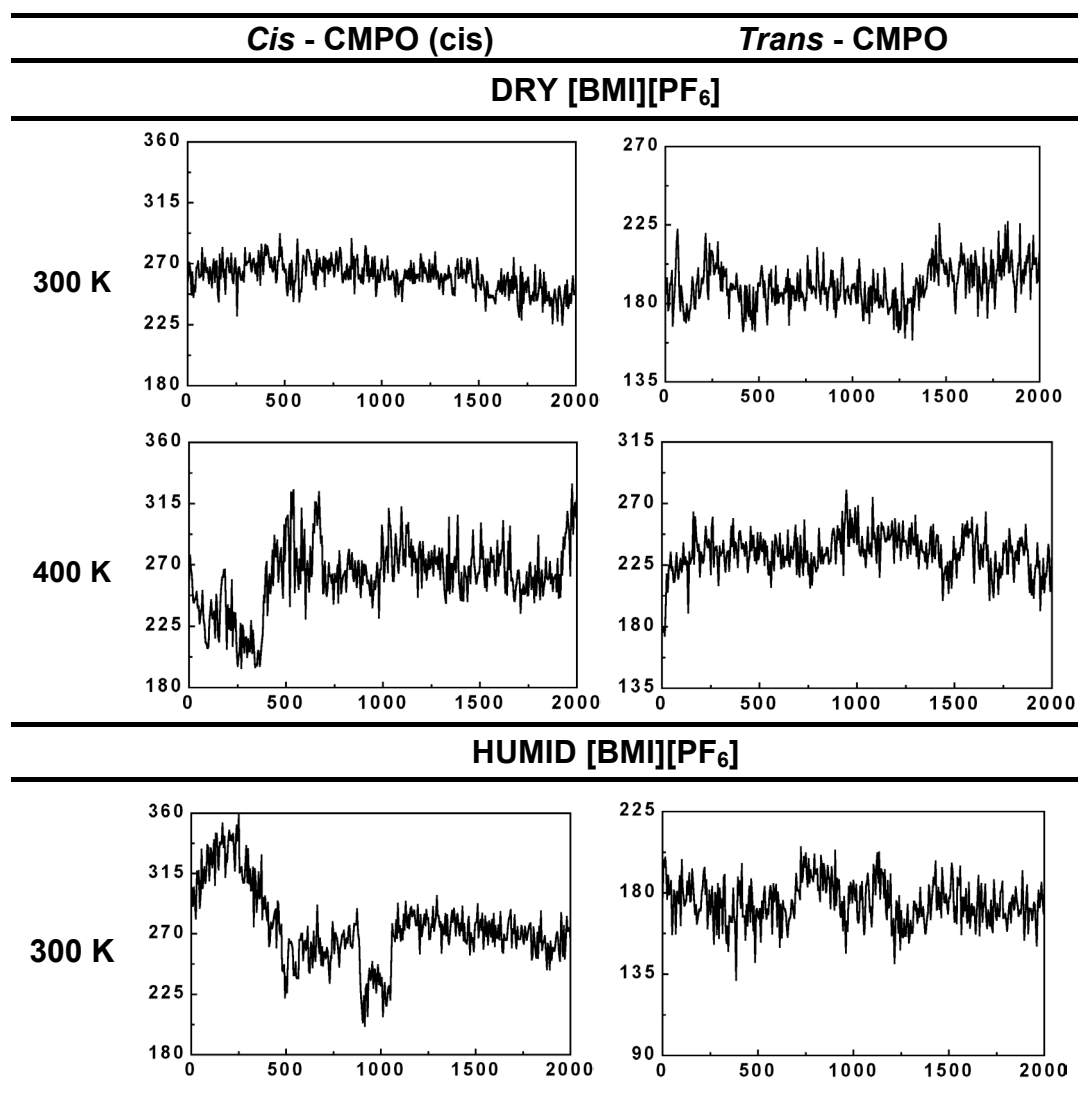


Figure S2: CMPO in “dry” and “humid” [BMI][PF₆]: Variation of the OC-PO dihedral as a function of time (ps). Starting in a *cis* form (left) and a *trans* form (right)

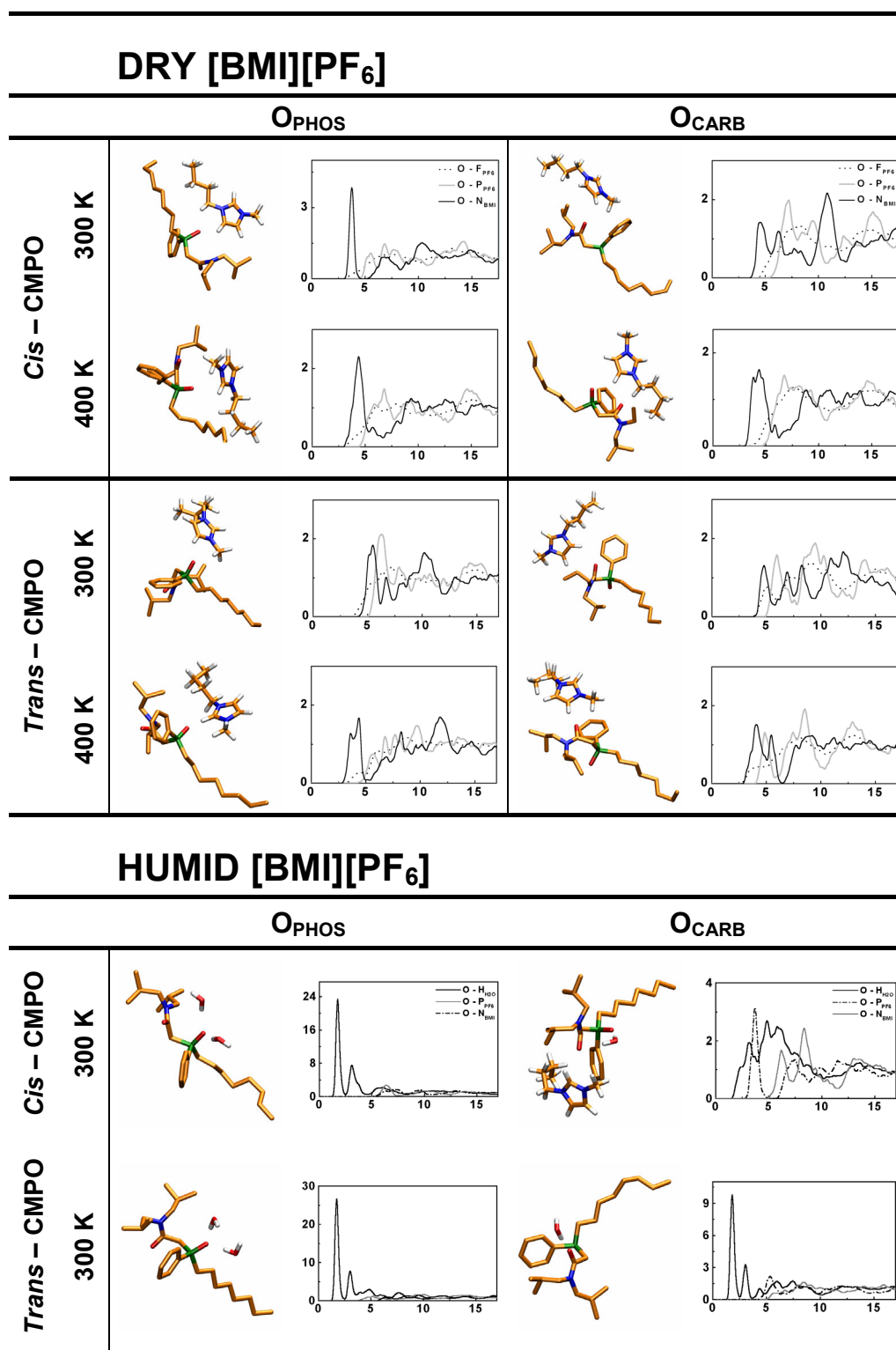


Figure S3: Free CMPO ligands in “dry” and “humid” [BMI][PF₆]: Final snapshot of the first solvation shell around the O_{PHOS} and O_{CARB} and corresponding radial distribution functions.

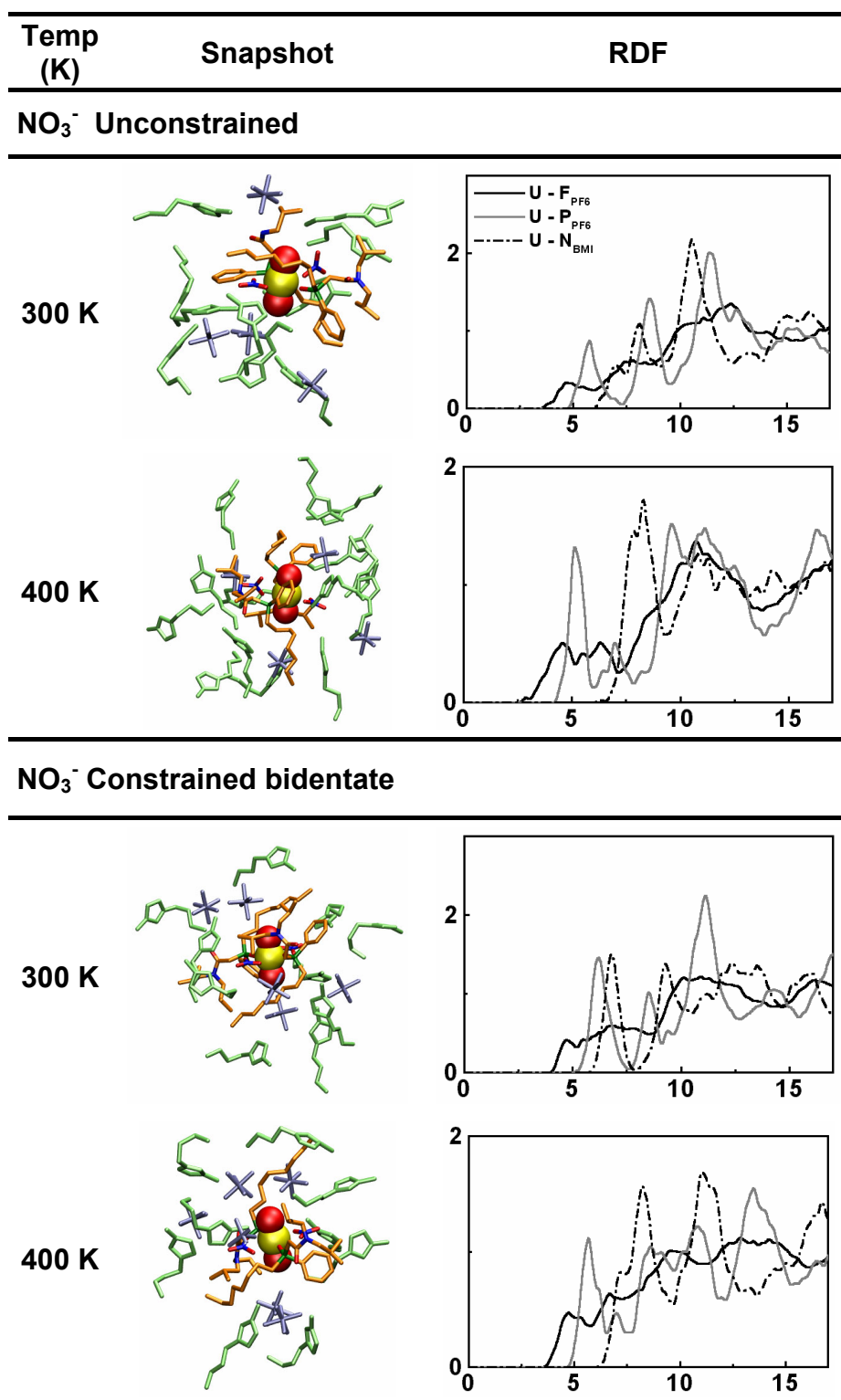


Figure S4: [UO₂(NO₃)₂(CMPO)₂] complexes in “dry” [BMI][PF₆]: Final snapshot of the first solvation shell (up to 8 Å) around the UO₂²⁺ cation and radial distribution function around the U atom.