

Electronic Supplementary Information

Efficient photocatalytic hydrogen evolution without an electron mediator using a simple electron donor-acceptor dyad

Hiroaki Kotani, Toshiya Ono, Kei Ohkubo and Shunichi Fukuzumi*

*Department of Material and Life Science, Graduate School of Engineering, Osaka University,
SORST, Japan Science and Technology Agency (JST), Suita, Osaka 565-0871, Japan*

* To whom correspondence should be addressed.

E-mail: fukuzumi@chem.eng.osaka-u.ac.jp.

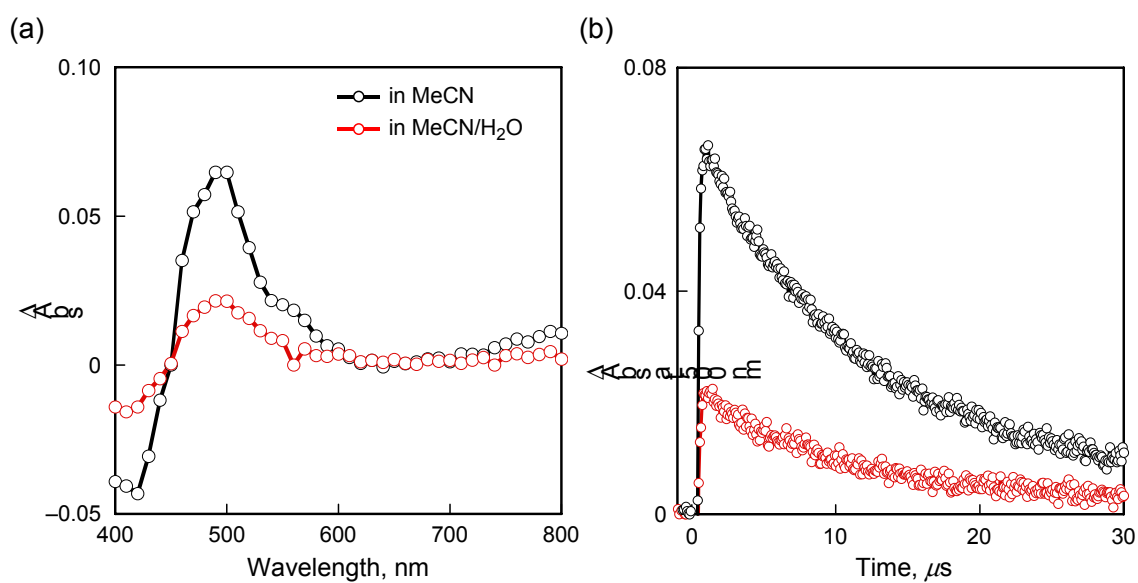


Fig. S1 (a) Transient absorption spectra of Acr⁺-Mes (1.0 × 10⁻⁴ mol dm⁻³) in deaerated MeCN (black) and MeCN/H₂O [1:1 (v/v)] (red) at 298 K taken at 1 μs after nanosecond laser excitation at 430 nm; (b) Decay time profiles of Acr⁺-Mes at 500 nm in deaerated MeCN (black) and MeCN/H₂O [1:1 (v/v)] (red).

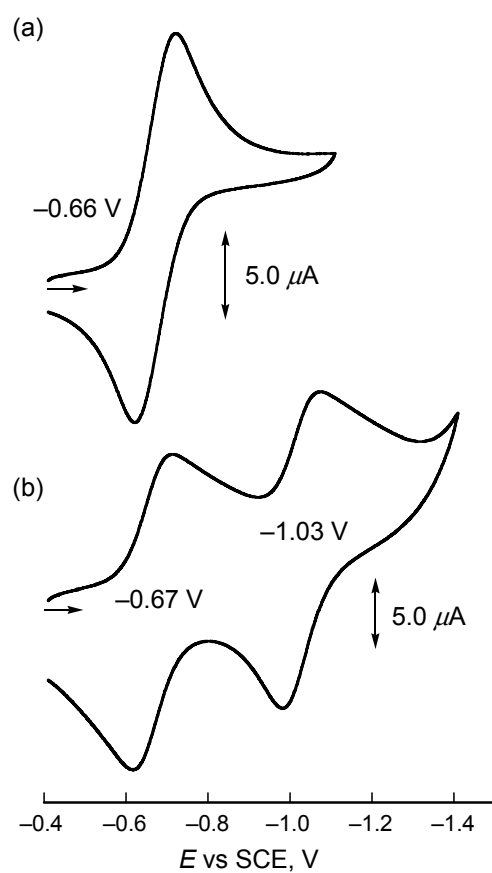


Fig. S2 Cyclic voltammograms of (a) Acr^+-Mes ($2.0 \times 10^{-3} \text{ mol dm}^{-3}$) and (b) MV^{2+} ($2.0 \times 10^{-3} \text{ mol dm}^{-3}$) in a deaerated H_2O and MeCN [1:1 (v/v)] mixed solution (5.0 cm^3) containing 0.10 mol dm^{-3} Bu_4NClO_4 (TBAP) at 298 K; sweep rate (100 mV s^{-1}).

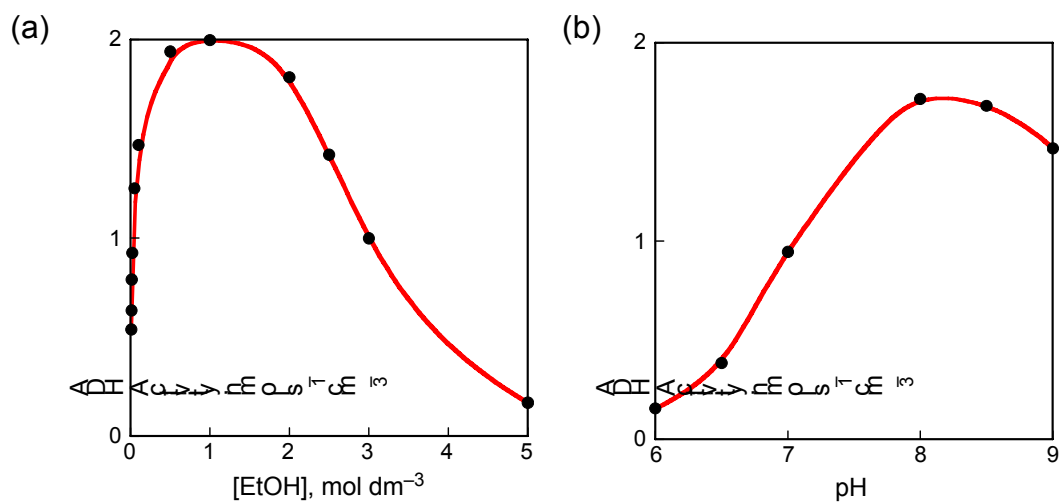


Fig. S3 ADH activity of NADH production vs. (a) [EtOH], (b) pH in a potassium phosphate buffer (pH 7.0, 50 mmol dm⁻³) solution (2.0 cm³) containing NAD⁺ (0.30 mmol dm⁻³), ADH (0.60 units).