

Supplementary Material for PCCP
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MOLECULAR DYNAMICS STUDY OF DICARBOLLIDE ANIONS IN
NITROBENZENE SOLUTION AND AT ITS AQUEOUS INTERFACE.
SYNERGISTIC EFFECT IN THE Eu(III) ASSISTED EXTRACTION.

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Supplementary Materials

	$\text{Cs}^+ \cdots \text{N}_{\text{NBZ}}$	$\text{Cs}^+ \cdots \text{O}_{\text{NBZ}}$	$\text{Cs}^+ \cdots \text{Cl}_{\text{CCD}}$	$\text{Cs}^+ \cdots \text{Co}_{\text{CCD}}$	
30 Cs⁺, CCD⁻ in nitrobenzene					
	6.1	12.7	2.4	1.7	
	<i>4.0 ; 5.8</i>	<i>3.2 ; 5.9</i>	<i>3.7 ; 5.6</i>	<i>6.8 ; 10.1</i>	
	$\text{Cs}^+ \cdots \text{N}_{\text{NBZ}}$	$\text{Cs}^+ \cdots \text{O}_{\text{NBZ}}$	$\text{Cs}^+ \cdots \text{Cl}_{\text{CCD}}$	$\text{Cs}^+ \cdots \text{Co}_{\text{CCD}}$	$\text{Cs}^+ \cdots \text{O}_{\text{H}_2\text{O}}$
30 Cs⁺, CCD⁻ in the 50:50 mixture c)					
a)	0.7	0.8	0.5	0.5	7.8
	<i>4.2 ; 5.2</i>	<i>3.3 ; 4.2</i>	<i>3.8 ; 5.2</i>	<i>7.4 ; 10.0</i>	<i>3.2 ; 4.2</i>
b)	2.1	2.5	1.4	1.1	5.3
	<i>4.2 ; 5.2</i>	<i>3.3 ; 4.2</i>	<i>3.8 ; 5.5</i>	<i>7.4 ; 10.1</i>	<i>3.1 ; 4.3</i>
a)	1.3	1.5	0.5	0.6	7.0
	<i>4.2 ; 5.2</i>	<i>3.3 ; 4.2</i>	<i>3.8 ; 5.2</i>	<i>6.9 ; 9.9</i>	<i>3.1 ; 4.2</i>
b)	2.7	3.1	0.9	1.1	4.9
	<i>4.2 ; 5.2</i>	<i>3.3 ; 4.2</i>	<i>3.8 ; 5.2</i>	<i>6.9 ; 9.8</i>	<i>3.1 ; 4.3</i>
a)	0.6	0.7	0.2	<i>id)</i>	7.8
	<i>4.2 ; 5.1</i>	<i>3.3 ; 4.2</i>	<i>4.0 ; 5.1</i>		<i>3.2 ; 4.1</i>
b)	2.7	3.2	0.3	<i>id)</i>	5.8
	<i>4.2 ; 5.1</i>	<i>3.3 ; 4.2</i>	<i>4.0 ; 5.1</i>		<i>3.1 ; 4.4</i>
30 Cs⁺, CCD⁻ in the 90:10 mixture					
a)	1.5	2.0	0.3	<i>id)</i>	7.0
	<i>4.2 ; 5.1</i>	<i>3.3 ; 4.3</i>	<i>3.9 ; 5.5</i>		<i>3.1 ; 4.2</i>
10 Eu³⁺, 30 CCD⁻ at the nitrobenzene – water interface					
a)	3.7	5.5	<i>id)</i>	<i>id)</i>	9.0
	<i>5.8 ; 6.8</i>	<i>5.2 ; 6.0</i>			<i>2.5 ; 3.2</i>
b)	8.7	12.4	<i>id)</i>	<i>id)</i>	9.0
	<i>5.7 ; 6.9</i>	<i>5.2 ; 6.0</i>			<i>2.5 ; 3.2</i>

Table S1: Average coordination numbers of $\text{M}^{\text{n}+}$ cations and CCD^- anions in neat NBZ and in the NBZ – water binary mixtures. They are obtained by integration of the first peaks of the RDF's whose maximum and integration distance are given in italics (*second line*). a) Averages over all Cs^+ or Eu^{3+} ions. b) Averages over Cs^+ or Eu^{3+} ions in the nitrobenzene phase. *id)* value not reported because the peak is ill-defined.

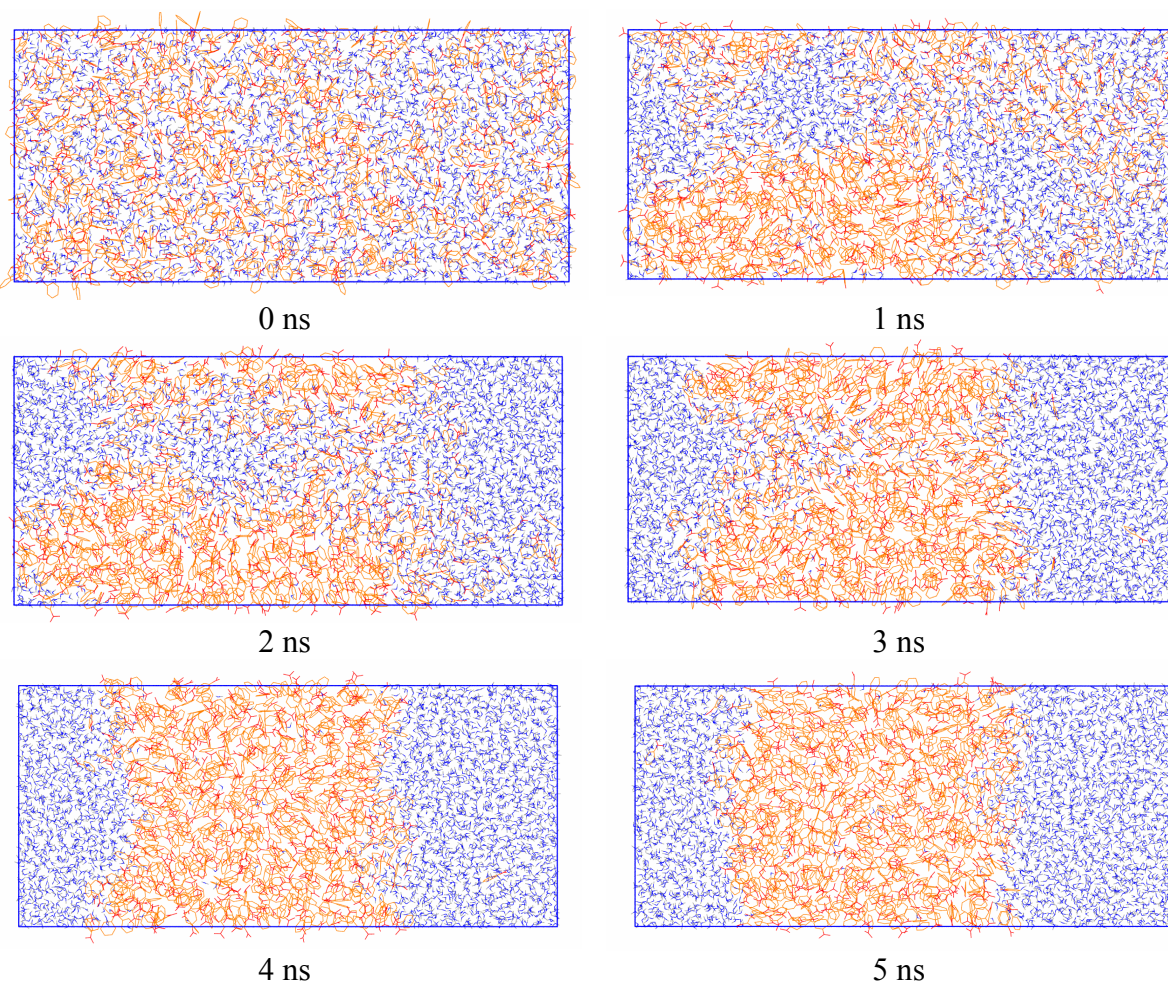


Figure S1: Phase separation of a randomly mixed 50:50 water / NBZ liquids. Snapshots along the dynamics.

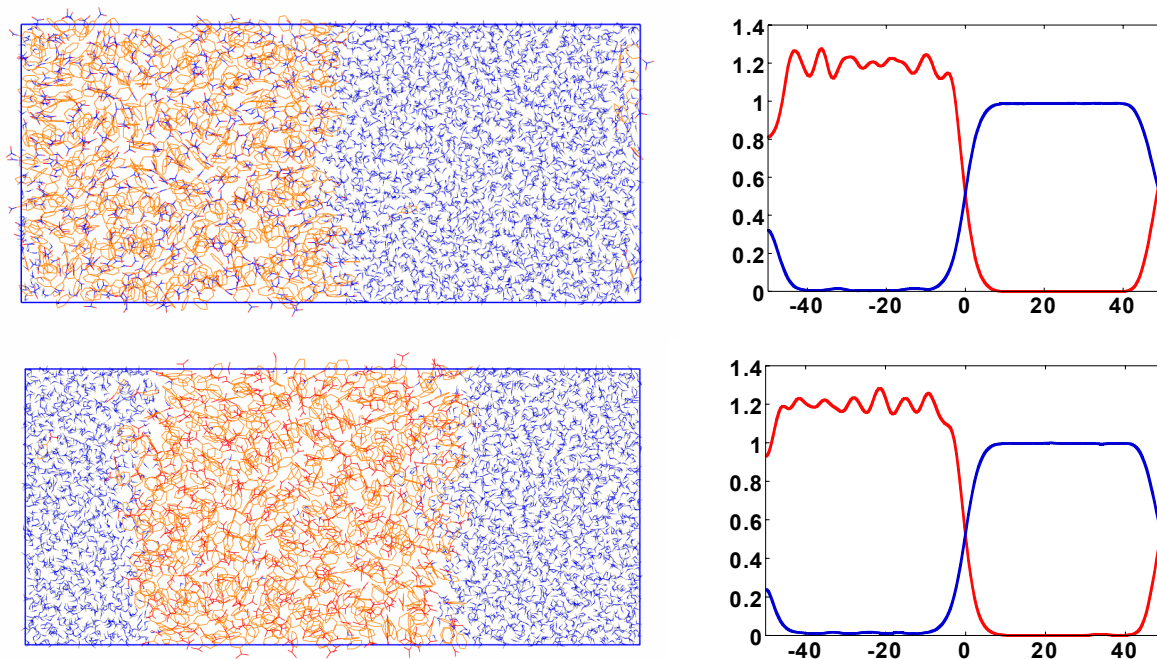


Figure S2 : Final snapshots and density curves of the dynamics that started with juxtaped (top) and randomly mixed (bottom) 50:50 Water / NBZ liquids.

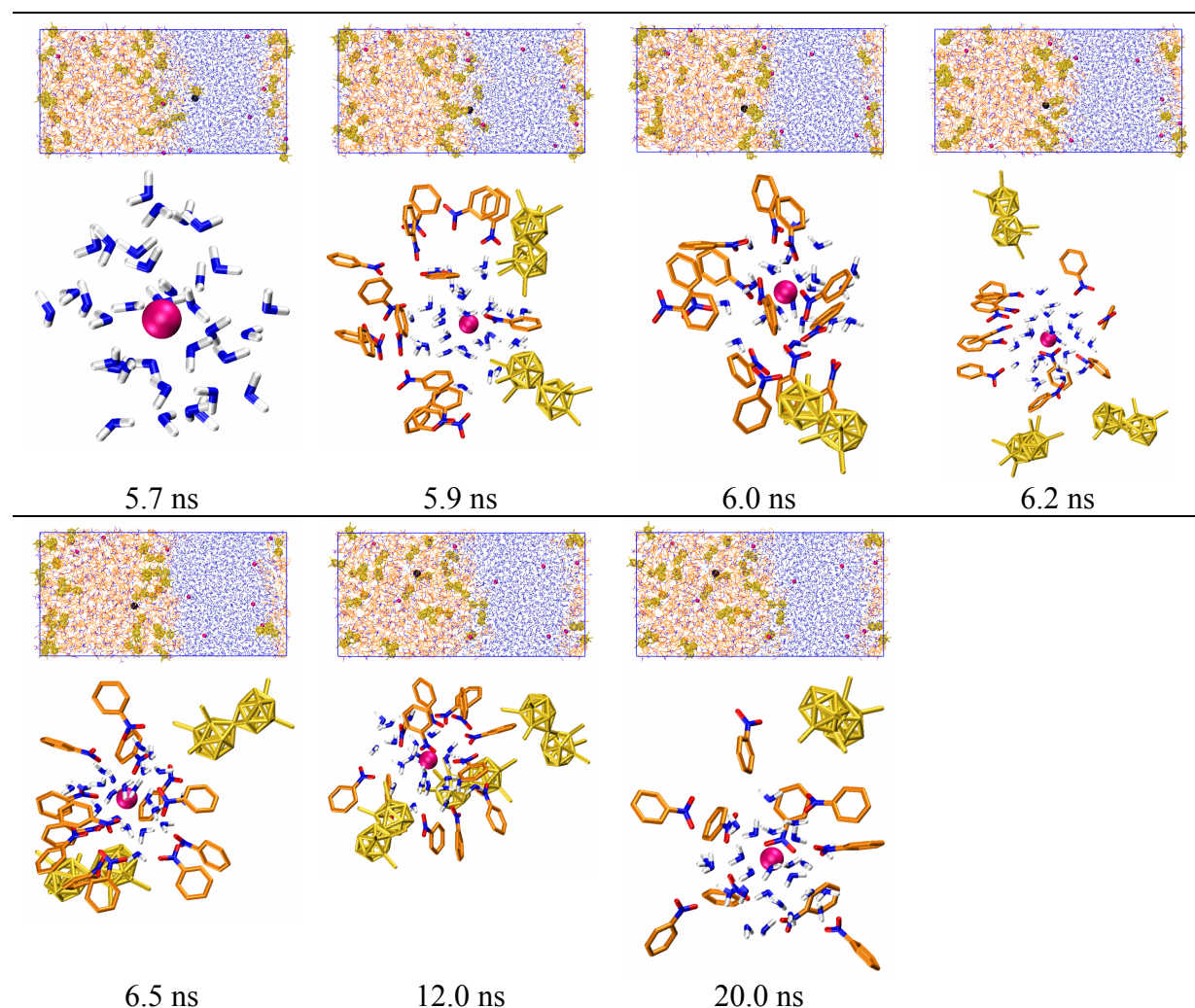


Figure S3 : 30 CCD^- , 10 Eu^{3+} (initially in water) at the NBZ - water interface. Snapshots of an Eu^{3+} cation during his migration from water to nitrobenzene.

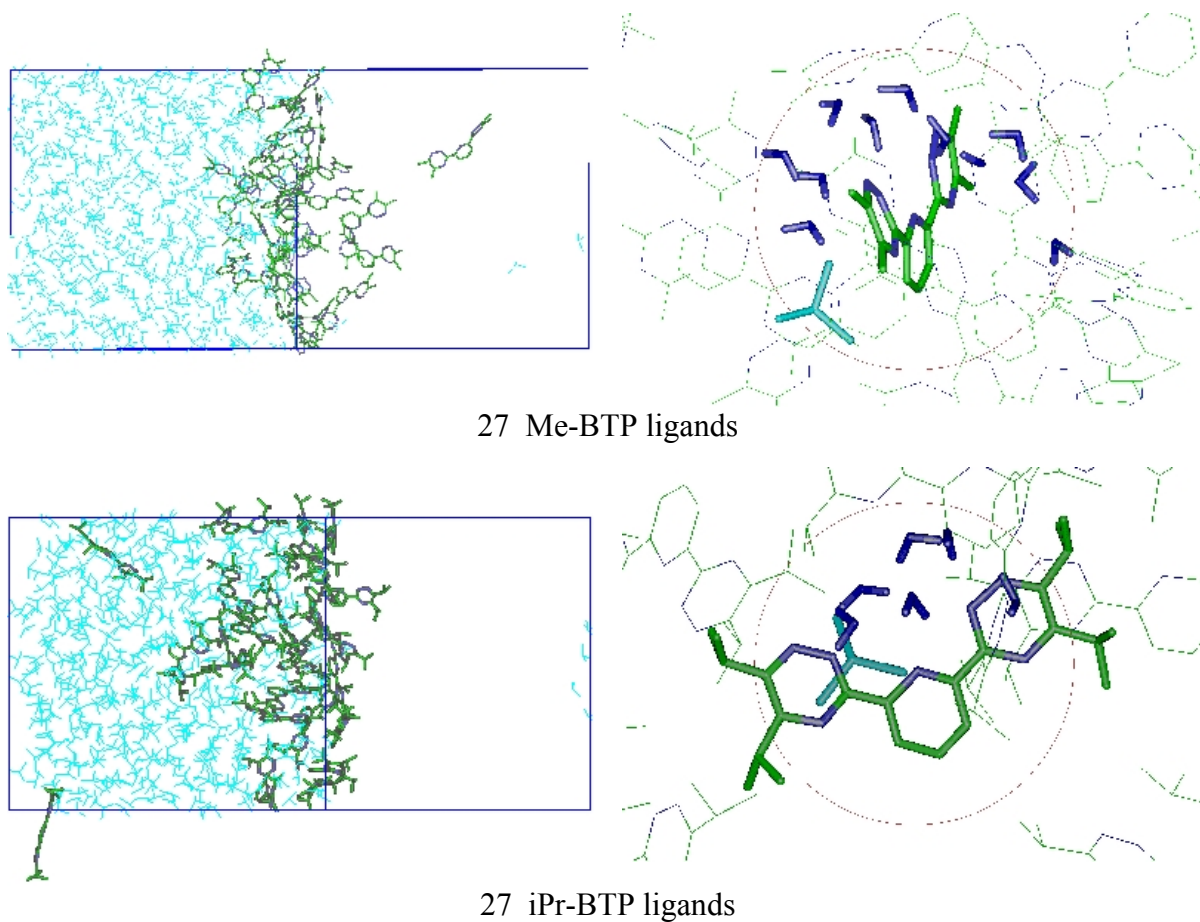
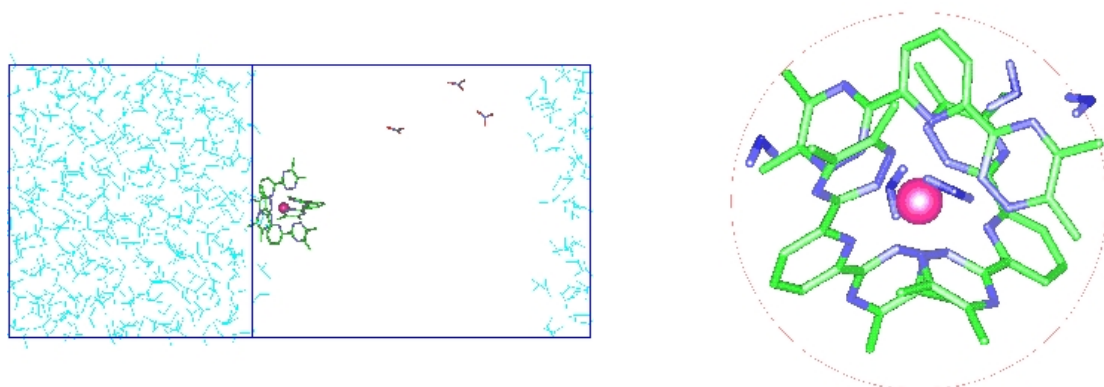
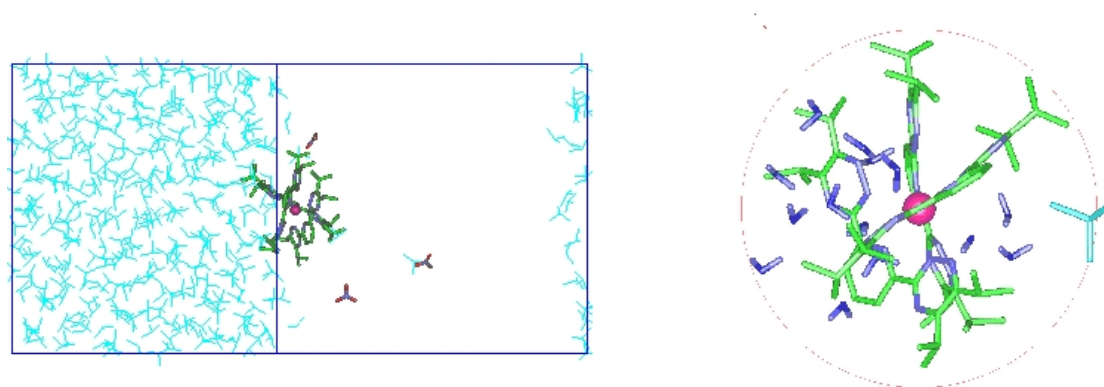


Figure S4 : MD simulation of BTP ligands at the chloroform – water interface. Comparison of Me-BTP and iPr-BTP. G. Wipff, EUROPART EEC meeting (contract F16W-CT-2003-508-854) in Madrid, June 2004.



$\text{Eu}(\text{Me-BTP})_3^{3+} \text{ 3 NO}_3^-$ complex



$\text{Eu}(\text{iPr-BTP})_3^{3+} \text{ 3 NO}_3^-$ complex

Figure S5 : MD simulation of $\text{Eu}(\text{BTP})_3^{3+}$ complex at the chloroform – water interface. Comparison of Me-BTP and iPr-BTP. G. Wipff, EUROPART EEC meeting in Madrid, June 2004.

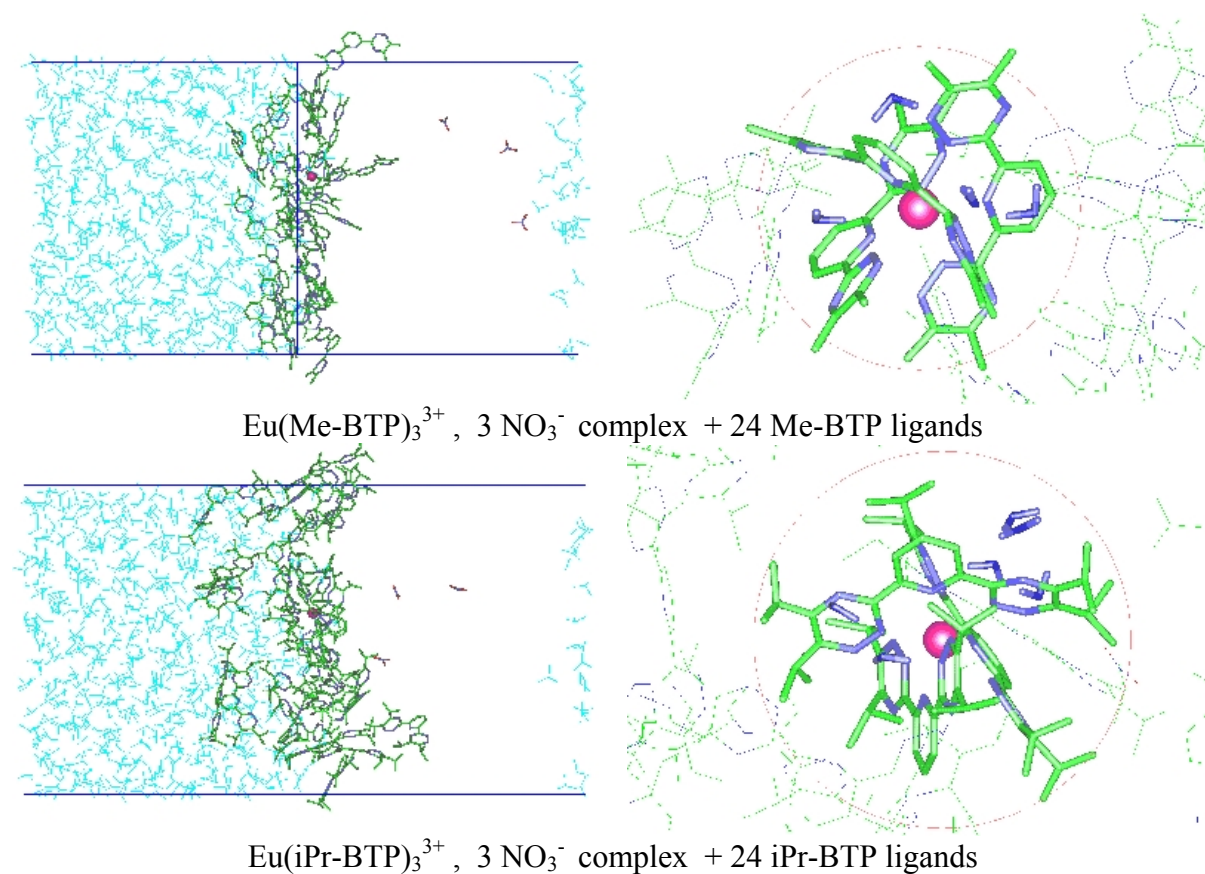


Figure S6 : MD simulation of $\text{Eu}(\text{BTP})_3^{3+}$ complexes + 24 BTP ligands at the chloroform – water interface. Comparison of Me-BTP and iPr-BTP. G. Wipff, EUROPART EEC meeting in Madrid, June 2004.

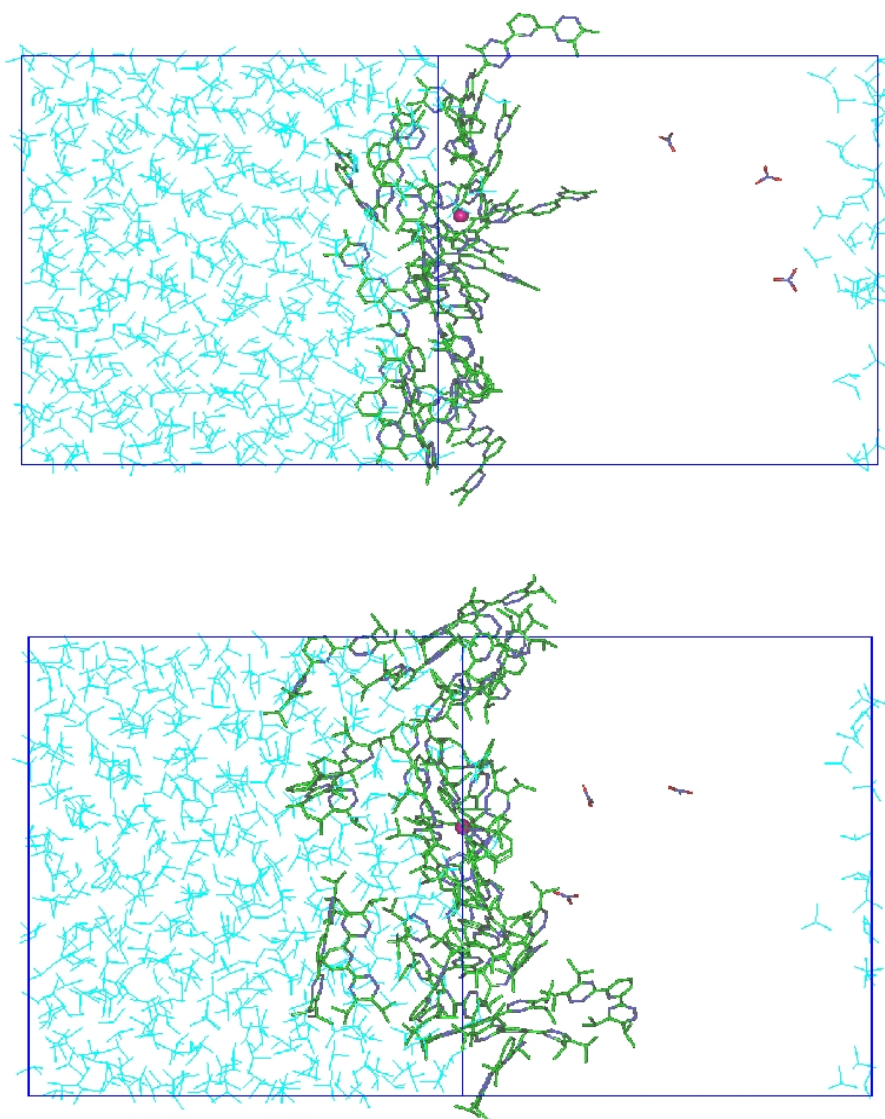


Figure S7 : Zoom on the $\text{Eu}(\text{Me-BTP})_3^{3+}$, 3NO_3^- complex + 24 Me-BTP ligands (*top*) and $\text{Eu}(\text{iPr-BTP})_3^{3+}$, 3NO_3^- complex + 24 iPr-BTP ligands (*bottom*) at the chloroform – water interface (water hidden for clarity). G. Wipff, EUROPART EEC meeting in Madrid, June 2004.