Control of the site and potential of reduction and oxidation processes in **p**-expanded quinoxalinoporphyrins

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Supporting Information Figures S1-S14.



Figure S1. Cyclic voltammograms of dinitroquinoxaline $Q(NO_2)_2$ (19) and analogous dinitroquinoxalioporphyrins [PQ(NO₂)₂]M 12 in pyridine with 0.1 M TBAP.



Figure S2. Cyclic voltammograms of the nitroquinoxaline $Q[(CH_3)_25NO_2]$ (17) and analogous nitroquinoxalinoporphyrins [PQ(5NO₂)]M(9) in pyridine with 0.1 M TBAP.



Figure S3. Thin-layer UV-visible spectral changes obtained before and after the first reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]$ 2H in CH₂Cl₂ solutions containing 0.1 M TBAP.^{ref}



Figure S4. Thin-layer UV-visible spectral changes obtained before and after the first reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]Cu$ in CH_2Cl_2 solutions containing 0.1 M TBAP.



Figure S5. Thin-layer UV-visible spectral changes obtained before and after the first reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]Zn$ in CH₂Cl₂ solutions containing 0.1 M TBAP.



Figure S6. Thin-layer UV-visible spectral changes obtained before and after the first reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]$ 2H in pyridine solutions containing 0.1 M TBAP.

[PQ(5NH₂)]Cu (**4**)



Figure S7. Thin-layer UV-visible spectral changes obtained before and after the first reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]Cu$ in pyridine solutions containing 0.1 M TBAP.

Figure S8. Thin-layer UV-visible spectral changes obtained before and after the first reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]Zn$ in pyridine solutions containing 0.1 M TBAP.

Figure S9. Thin-layer UV-visible spectral changes obtained before and after the second reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]$ 2H in CH₂Cl₂ solutions containing 0.1 M TBAP.

 $[PQ(OCH_3)_2]Cu(5)$

Data not presented due to slow chemical reaction on the spectroelectrochemical time scale

[PQ(5NO₂)]Cu (9)

Data not presented due to slow chemical reaction on the spectroelectrochemical time scale

(PQBr)Cu (7)

Data not presented due to slow chemical reaction on the spectroelectrochemical time scale

Figure S10. Thin-layer UV-visible spectral changes obtained before and after the second reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]Cu$ in CH_2Cl_2 solutions containing 0.1 M TBAP.

[PQ(5NO₂)]Zn (9)

Data not presented due to slow chemical reaction on the spectroelectrochemical time scale

[PQ(NO₂NH₂)]Zn (10)

Data not presented due to slow chemical reaction on the spectroelectrochemical time scale

(PQ)Zn (**6**)

Data not presented due to multiple reduction products

[PQ(6NO₂)]Zn (11)

Data not presented due to slow chemical reaction on the spectroelectrochemical time scale

(PQBr)Zn (7)

[(PQ(NO₂)₂]Zn (**12**)

Data not presented due to slow chemical reaction on the spectroelectrochemical time scale

Data not presented due to slow chemical reaction on the spectroelectrochemical time scale

(PQCl₂)Zn (8)

Figure S11. Thin-layer UV-visible spectral changes obtained before and after the second reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]Zn$ in CH₂Cl₂ solutions containing 0.1 M TBAP.

[PQ(OCH₃)₂]2H (5)

(PQ)2H (6)

Data not presented due to slow chemical reaction on the spectroelectrochemical time scale

(PQBr)2H (7) $(PQCl_2)2H (8)$ $(PQCl_2)2H (8)$

Figure S12. Thin-layer UV-visible spectral changes obtained before and after the second reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]$ 2H in pyridine solutions containing 0.1 M TBAP.

Figure S13 Thin-layer UV-visible spectral changes obtained before and after the second reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]Cu$ in pyridine solutions containing 0.1 M TBAP.

Figure S14. Thin-layer UV-visible spectral changes obtained before and after the second and possibly third reduction of various quinoxalinoporphyrins $[PQ(R_1R_2R_3)]Zn$ in pyridine solutions containing 0.1 M TBAP.