

Supplementary information for Kaltsoyannis and Plane

Table S1: B3LYP geometric data for NO₂, NO₃ and INO₃ (planar). Bond lengths r in Å, bond and dihedral angles \angle in degrees. Atom numbers for INO₃ are given in Figure 1.

	$r(\text{O1-I1})$	$r(\text{O1-N1})$	$r(\text{N1-O2})$	$r(\text{N1-O3})$	$\angle\text{I1O1N1}$	$\angle\text{O1N1O2}$	$\angle\text{O2N1O3}$
NO ₂		1.192	1.192			134.5	
NO ₃		1.232				120.0	
INO ₃	2.023	1.452	1.198	1.195	117.8	118.0	131.5

Table S2: B3LYP geometric data for I₂O₃. Bond lengths r in Å, bond and dihedral angles \angle in degrees. Atom numbers are given in Figure 2.

$r(\text{I1-O1})$	$r(\text{O1-I2})$	$r(\text{I2-O2})$	$r(\text{I2-O3})$	$\angle\text{I1O1I2}$	$\angle\text{O1I2O2}$	$\angle\text{O1I2O3}$	$\angle\text{O2I2O3}$	$\angle\text{I1O1I2O2}$	$\angle\text{I1O1I2O3}$
2.019	1.971	1.784	1.784	121.1	102.5	102.5	108.1	56.0	56.0

Table S3: B3LYP geometric data for I₂O₄. Bond lengths r in Å, bond and dihedral angles \angle in degrees. Atom numbers are given in Figure 3.

$r(\text{O1-I1})$	$r(\text{I1-O2})$	$r(\text{O2-I2})$	$r(\text{I2-O3})$	$r(\text{I2-O4})$	$\angle\text{O1I1O2}$	$\angle\text{I1O2I2}$	$\angle\text{O2I2O3}$	$\angle\text{O2I2O4}$	$\angle\text{O3I2O4}$	$\angle\text{O1I1O2I2}$	$\angle\text{I1O2I2O3}$	$\angle\text{I2O2I2O4}$
1.824	2.010	1.997	1.783	1.787	103.9	123.1	100.1	99.3	109.0	-93.1	-108.9	2.5

Table S4: B3LYP geometric data for I₂O₅. Bond lengths r in Å, bond and dihedral angles \angle in degrees. Atom numbers are given in Figure 4.

Variable		Variable		Variable	
$r(\text{O1-I1})$	1.779	$\angle\text{O1I1O2}$	108.9	$\angle\text{O1I1O3I2}$	173.5
$r(\text{I1-O2})$	1.781	$\angle\text{O1I1O3}$	96.2	$\angle\text{O2I1O3I2}$	-76.0
$r(\text{I1-O3})$	2.019	$\angle\text{O2I1O3}$	99.5	$\angle\text{I1O3I2O4}$	-120.4
$r(\text{O3-I2})$	1.969	$\angle\text{I1O3I2}$	123.7	$\angle\text{I1O3I2O5}$	-9.4
$r(\text{I2-O4})$	1.779	$\angle\text{O3I2O4}$	100.6		
$r(\text{I2-O5})$	1.790	$\angle\text{O3I2O5}$	99.6		

Supplementary Material for PCCP
This journal is © The Owner Societies 2008

Table S5: B3LYP geometric data for I₂, IO, OIO and three isomers of I₂O₂; IOIO, IOOI and OI(I)O. Bond lengths r in Å, bond and dihedral angles \angle in degrees. Experimental data (in italics) for I₂ and IO from *Thermodynamic Properties of Individual Substances*, Hemisphere, New York, 1989, and for O₃ from N. N. Greenwood and A. Earnshaw, *Chemistry of the Elements*, Pergamon Press, Oxford, 2nd edition, 1997.

	$r(\text{I-I})$	$r(\text{I-O})$	$r(\text{O-O})$	$r(\text{O-I})$	$r(\text{I-O})$	$\angle\text{OOO}$	$\angle\text{IOI}$	$\angle\text{OIO}$	$\angle\text{IOO}$	$\angle\text{OOI}$	$\angle\text{OII}$	$\angle\text{IOIO}$	$\angle\text{IOOI}$	$\angle\text{IIOO}$
I ₂	2.708 <i>2.665</i>													
IO		1.892 <i>1.868</i>												
OIO				1.819				110.5						
IO ₃		1.804						110.8						
O ₃			1.255 <i>1.278</i>			118.3 <i>116.8</i>								
IOIO (chain)		2.029		1.994	1.828		122.0	105.9				73.7		
IOOI (chain)		2.065	1.384	2.065					113.6	113.6			86.3	
OI(I)O	2.823	1.793			1.793						102.8			108.5

Table S6: B3LYP vibrational wavenumbers and point groups for all target systems.

NO₂ (C_{2v})		I₂O₅ (C₁)		IOIO (C₁)		INO (C_s)	
A ₁	766.2	A	36.2	A	64.3	A'	233.2
A ₁	1385.6	A	48.1	A	224.5	A'	512.4
B ₂	1688.0	A	59.9	A	328.6	A'	1873.0
		A	161.8	A	509.3		
NO₃ (D_{3h})		A	174.5	A	534.9	ICN (C_{∞v})	
E'	218.5	A	249.3	A	806.2	Π	327.1
A ₂ "	806.0	A	261.5			Σ	506.2
E'	1096.5	A	297.6	OI(I)O (C₁)		Σ	2277.3
A ₁ '	1126.0	A	313.5	A	138.6		
		A	459.4	A	146.3	I₂ (D_{∞h})	
INO₃ (C_s)		A	583.4	A	215.6	Σ _g	214.6
A"	97.7	A	866.1	A	285.9		
A'	181.0	A	881.3	A	847.6	IO (C_{∞v})	
A'	368.2	A	905.3	A	882.4	Σ	685.3
A'	579.8	A	911.2				
A'	729.4			IOIO (C₁)		OIO (C_{2v})	

Supplementary Material for PCCP
This journal is © The Owner Societies 2008

A''	752.2	I₂O₄ (C₁)		A	64.3	A ₁	264.4
A'	824.3	A	34.5	A	125.8	A ₁	799.6
A'	1311.4	A	66.3	A	220.5	B ₂	821.1
A'	1716.2	A	92.6	A	433.0		
		A	159.7	A	568.3	IO₃ (C₁)	
I₂O₃ (C₁)		A	239.2	A	804.0	A	258.7
A	43.4	A	255.5			A	259.1
A	88.2	A	305.2	HI (C_{∞v})		A	268.2
A	236.8	A	431.3	Σ	2318.7	A	765.0
A	254.6	A	530.8			A	765.2
A	297.5	A	811.2	HOI (C_s)		A	795.3
A	442.5	A	868.2	A'	587.4		
A	625.7	A	900.7	A'	1097.6	O₃ (C_{2v})	
A	872.7			A'	3791.3	A ₁	746.0
A	901.2	IOI (C_{2v})				B ₂	1189.4
		A ₁	127.6			A ₁	1248.9
		A ₁	443.0				
		B ₂	619.6				