

**Self-assembly and heterogeneous electron
transfer properties of metallo-
octacarboxyphthalocyanine complexes on
gold electrode**

Bolade O. Agboola and Kenneth I. Ozoemena*

Department of Chemistry, University of Pretoria, Pretoria 0002, South Africa

Revised manuscript to PCCP (Paper Ref.: B800611C)

* Author to whom correspondence should be addressed: Tel.: +27-12-4202515;
Fax: +27-12-420 4687. E-mail: kenneth.ozoemena@up.ac.za.

SUPPORTING INFORMATION

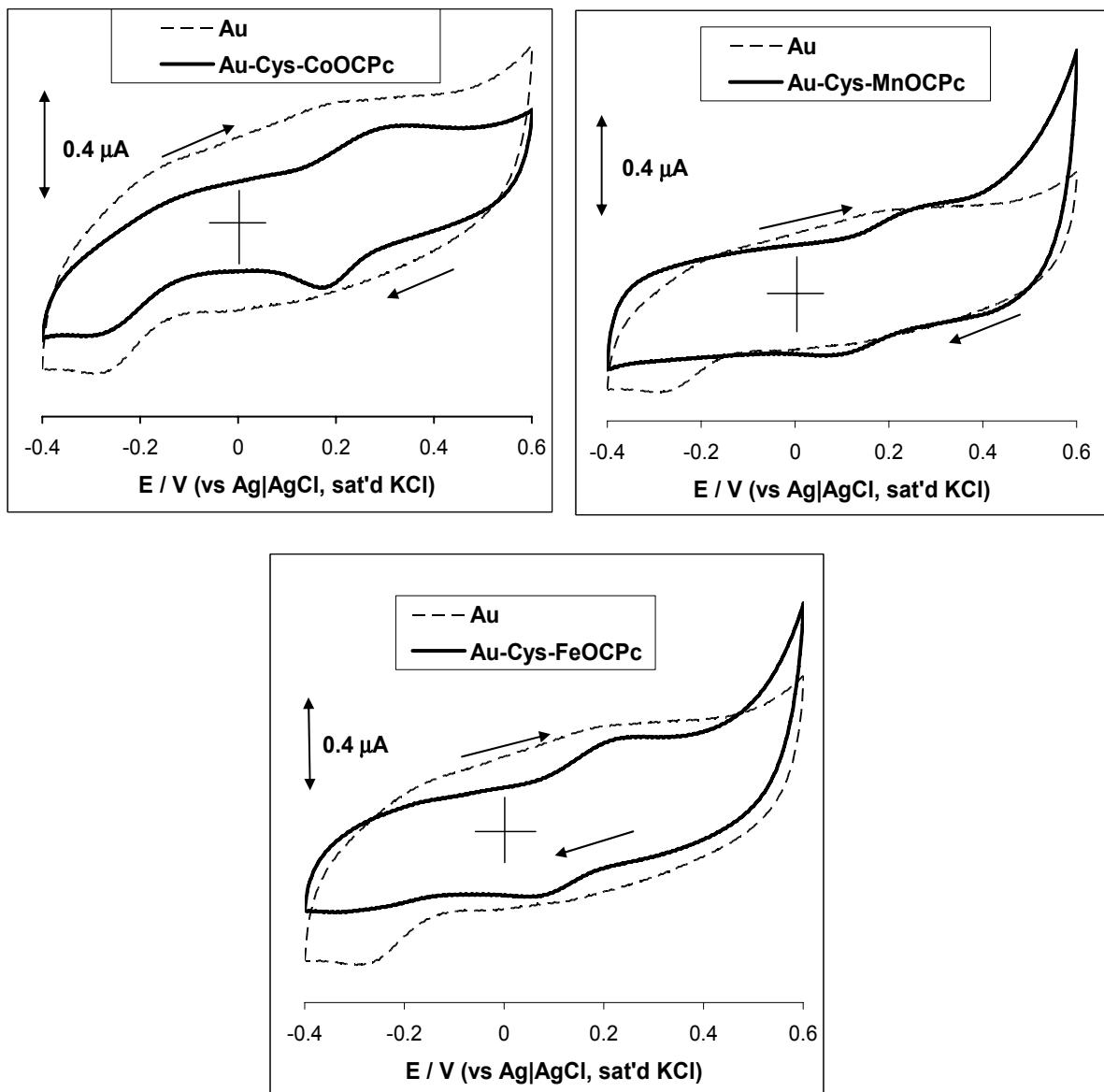


Figure 1 SI. Comparative cyclic voltammetric profiles of the bare Au and Au-Cys-CoOCPc, Au-Cys-FeOCPc and Au-Cys-MnOCPc obtained in 0.05 M phosphate buffer pH 7.4. Scan rate = 50 mV s⁻¹.

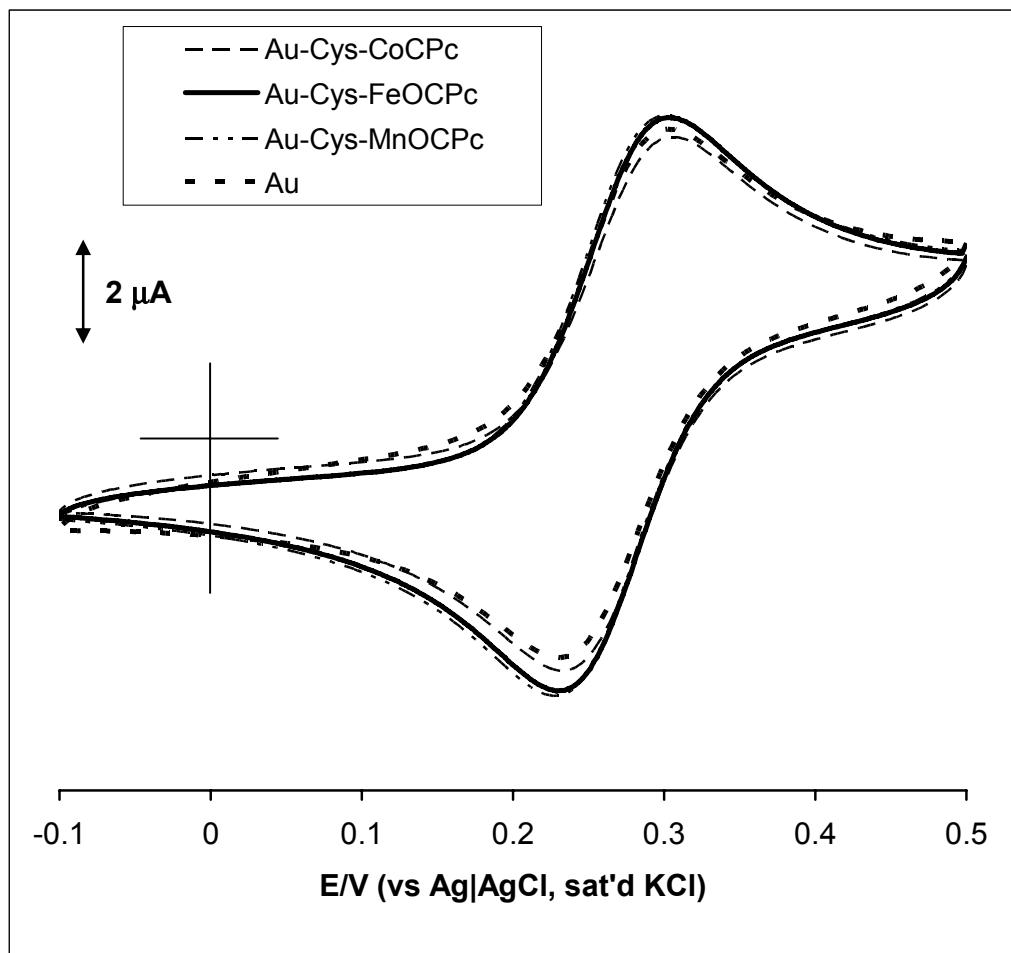


Figure 2 SI. Cyclic voltammetric evolutions of $[\text{Fe}(\text{CN})_6]^{4-}$ / $[\text{Fe}(\text{CN})_6]^{3-}$ in 0.1 M KCl at different electrodes. Scan rate = 50 mV s⁻¹.

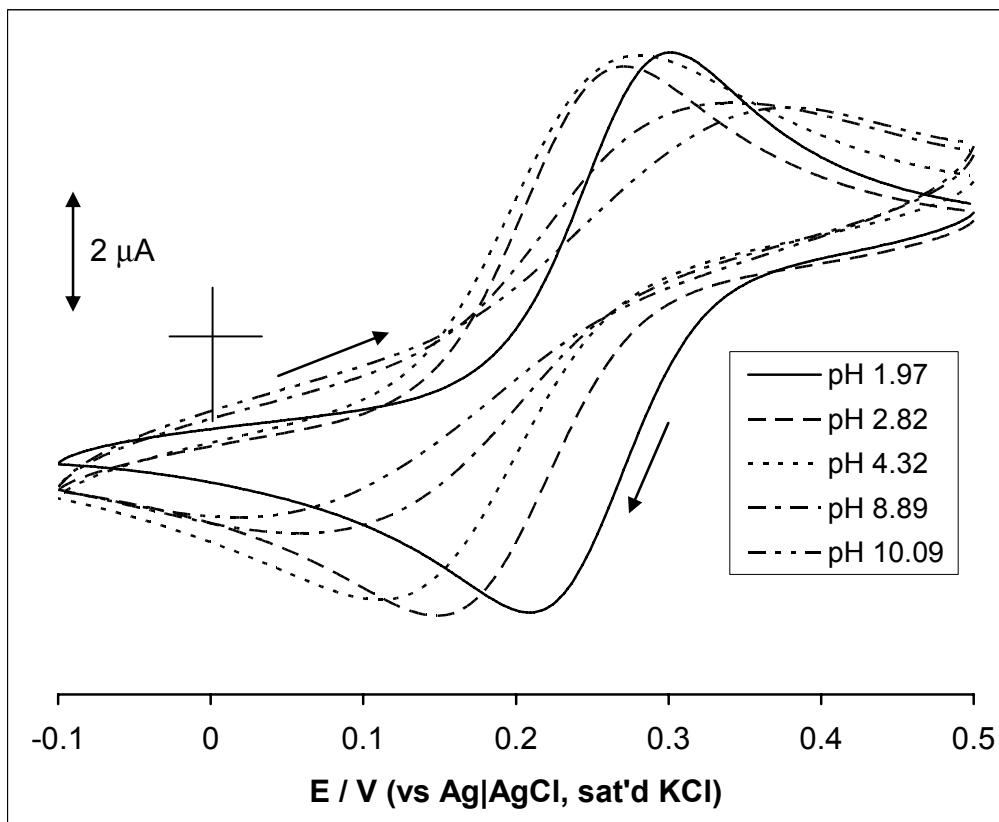


Figure 3 SI: Typical voltammetric responses of the Au-Cys-FeOCPc electrode at different solution pH of the redox probe, $[Fe(CN)_6]^{4-} / [Fe(CN)_6]^{3-}$. Scan rate = 50 mV s⁻¹.