

Supporting Information

Intermolecular Photoinduced Electron-Transfer of 1,8-Naphthalimides in Protic Polar Solvents

Dae Won Cho,^{a,b} Mamoru Fujitsuka,^a Ung Chan Yoon,^c and Tetsuro Majima^{*a}

^a *The Institute of Scientific and Industrial Research (SANKEN), Osaka University, Mihogaoka 8-1, Ibaraki, Osaka 567-0047, Japan. Fax: +81- 6-6879-8499; Tel:+81-6-6879-8495; E-mail:*

majima@saken.osaka-u.ac.jp

^b *Department of Chemistry, Chosun University, Gwangju 501-759, Korea. Fax: +82- 62-230-7979; Tel: +82- 62-230-6677; E-mail: dwcho@chosun.ac.kr*

^c *Department of Chemistry and the Chemistry Institute for Functional Materials, Pusan National University, Pusan 609-735, Korea. E-mail: ucyoon@pusan.ac.kr*

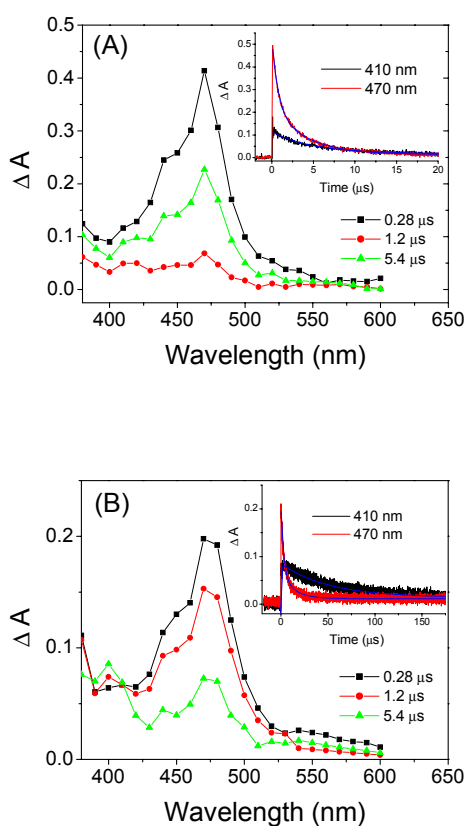


Fig. S1 Transient absorption spectra observed at various delay times during the laser flash photolysis of NI-O (2.9×10^{-5} M) in (A) CH₃CN and (B) H₂O/CH₃CN (v/v = 1:1) with a 355 nm excitation pulse. The inset shows kinetic traces of ΔA at 410 and 470 nm: (A) the decay times are 4.2 and 4.7 μ s monitored at 410 and 470 nm, respectively; (B) the decay times are 59.4 and 5.5 μ s monitored at 410 and 470 nm, respectively.

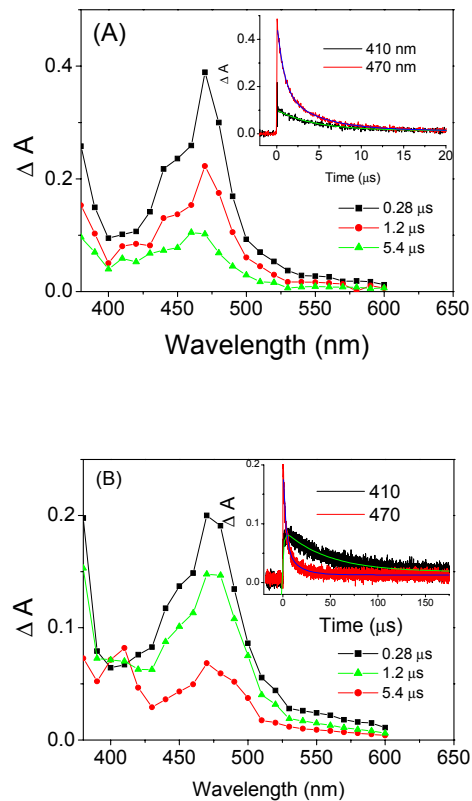


Fig. S2 Transient absorption spectra observed at various delay times during the laser flash photolysis of NI-O3 (2.9×10^{-5} M) in (A) CH₃CN and (B) H₂O/CH₃CN (v/v = 1:1) with a 355 nm excitation pulse. The inset shows kinetic traces of ΔA at 410 and 470 nm: (A) the decay times are 4.5 and 4.5 μ s monitored at 410 and 470 nm, respectively; (B) the decay times are 50.5 and 5.5 μ s monitored at 410 and 470 nm, respectively.

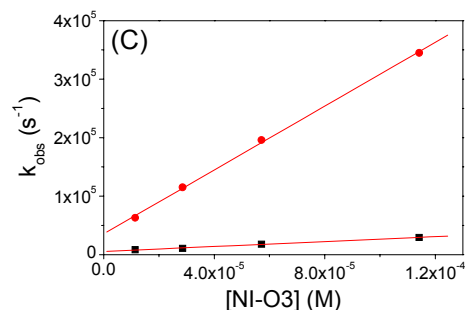
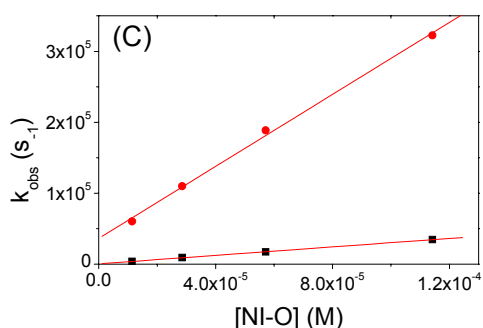
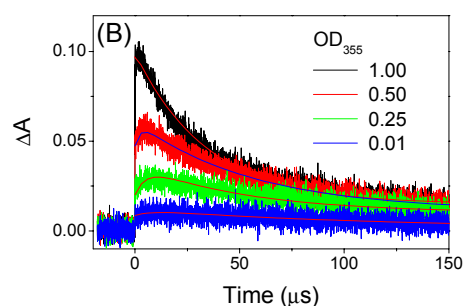
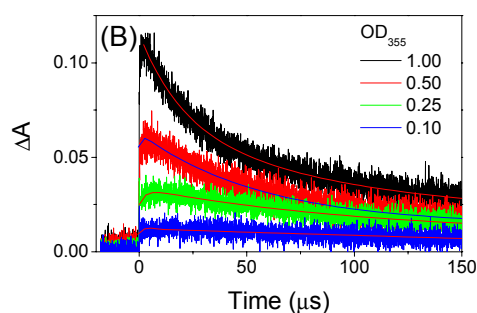
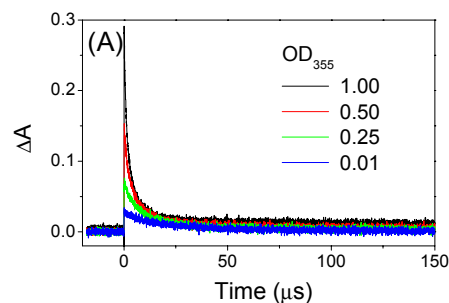
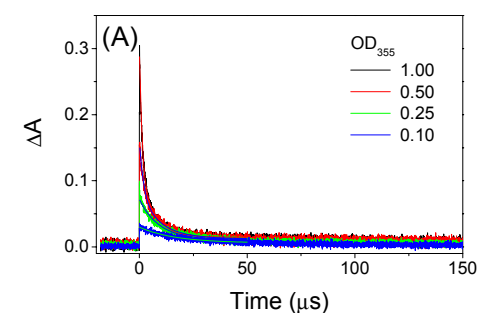


Fig. S3 Triplet decay of (A) NI-O3 at 470 nm and (B) concomitant growth of transient at 410 nm in H₂O/CH₃CN according to change of concentration of NI [OD_{355 nm} = 1.00 (1.1×10⁻⁴ M), 0.50 (5.7×10⁻⁵ M), 0.25 (2.9×10⁻⁵ M), and 0.10 (1.4×10⁻⁵ M)]. (C) Plots of *k*_{obs} vs. [NI-O] for self-quenching of NI derivatives monitored at 470 nm (circles) and at 410 nm (squares) in H₂O/CH₃CN.

Fig. S4 Triplet decay of (A) NI-O3 at 470 nm and (B) concomitant growth of transient at 410 nm in H₂O/CH₃CN according to change of concentration of NI [OD_{355 nm} = 1.00 (1.1×10⁻⁴ M), 0.50 (5.7×10⁻⁵ M), 0.25 (2.9×10⁻⁵ M), and 0.10 (1.4×10⁻⁵ M)]. (C) Plots of *k*_{obs} vs. [NI-O3] for self-quenching of NI derivatives monitored at 470 nm (circles) and at 410 nm (squares) in H₂O/CH₃CN.

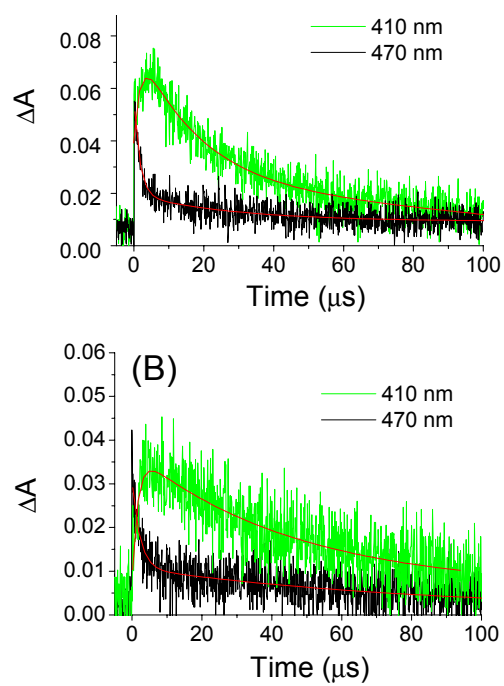


Fig. S5 Triplet decay of (A) NI-C8-PTZ (2.9×10^{-5} M) and (B) NI-O3-PTZ (2.9×10^{-5} M) monitored at 470 nm and concomitant growth of transient at 410 nm (B) in H_2O/CH_3CN .