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Figure S1 ESI. CV (A) and CVD (B) of MP-Pt in 1 M HClO₄. v = 0.10 Vs⁻¹.

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Figure S2 ESI. CV (upper) and CVD (lower) of a MP-Pt electrode in 1 M H_2SO_4 . $v = 0.015 Vs^{-1}$.



Figure S3 ESI. CV (A) and CVD (B) of MP-Pt in 1 M H_2SO_4 using an extended potential range. v = 0.015 Vs⁻¹.



Figure S4 ESI

Comparison of chronodeflectometric data (red circles) for NPtC in 1 M H₂SO₄ with simulation according to eq. 2 (full black line). dn/dC = 0.011; $Do = 2.6 \times 10^{-5} \text{ cm}^2/\text{s}$; $x = 96 \,\mu\text{m}$; $Cs = 1.62 \times 10^{-3} \text{ mol/l}$. For comparison a simulated curve (full blue line) is calculated assuming that the process is not discontinuous (fast double layer charging) but continuous (diffusion controlled process) with the same constants but $x = 55 \,\mu\text{m}$ and $Cs = 1.8 \times 10^{-3} \text{ mol/l}$. As it can be seen, only the discontinuous model fits the CD profile.



Figure S5 ESI

Comparison of chronodeflectometric data (red circles) for MP-Pt in 1 M H₂SO₄ with simulation according to eq. 2 (full black line). dn/dC = 0.011; $Do = 2.6 \times 10^{-5} \text{ cm}^2/\text{s}$; $x = 90 \text{ }\mu\text{m}$; $Cs = 4.26 \times 10^{-3} \text{ mol/l}$. For comparison a simulated curve (full blue line) is calculated assuming that the process is not discontinuous (fast double layer charging) but continuous (diffusion controlled process) with the same constants but $x = 50 \text{ }\mu\text{m}$ and $Cs = 5.2 \times 10^{-3} \text{ mol/l}$. As it can be seen, only the discontinuous model fits the CD profile.



Figure S6 ESI.

CV (A) and CVD (B) for CO_{ad} oxidation in 1 M H₂SO₄ at MP-Pt. v = 0.015 V s⁻¹. The black lines are measured with full CO coverage, while the red lines correspond to a lower coverage.