

	Water Diluted	Water Concentrated	Methanol Concentrated
NBu ₄ ⁺	-230	-201	-150
Cs ⁺	-391	-373	-225
H ₃ O ⁺	***	-346	***
H ₅ O ₂ ⁺	***	-336	***
UO ₂ ²⁺	-365	-330	-183
Eu ³⁺	-375	-340	-150
Eu ³⁺ (H ₂ O SPCE)	***	-338	***

TABLE S1: Average interaction energy of one PW³⁻ with the solvent (Averages taken over the last 2 ns of dynamics and over all PW³⁻ anions).

	Water Diluted	Water Concentrated	Methanol Concentrated
NBu ₄ ⁺	4802 (171)	28920 (152)	26263 (138)
Cs ⁺	3678 (131)	24017 (126)	24782 (130)
H ₃ O ⁺	***	26094 (137)	***
H ₅ O ₂ ⁺	***	26275 (138)	***
UO ₂ ²⁺	3924 (140)	26454 (139)	25405 (133)
Eu ³⁺	3798 (136)	26794 (141)	27090 (143)
Eu ³⁺ (H ₂ O SPCE)	***	26736 (140)	***

TABLE S2: PW³⁻ - PW³⁻ interaction energies (kcal/mol). Averages over the last 2 ns of dynamics. In brackets average per PW³⁻...PW³⁻ pair.

	Water Diluted	Water Concentrated	Methanol Concentrated
NBu ₄ ⁺	-1592 (-66)	-3294 (-54)	-3019 (-50)
Cs ⁺	-1157 (-48)	-2873 (-48)	-2789 (-46)
H ₅ O ₂ ⁺	***	-2897 (-48)	***
H ₃ O ⁺	***	-2879 (-48)	***
UO ₂ ²⁺	-1244 (-103)	-2928 (-97)	-2898 (-97)
Eu ³⁺	-1202 (-150)	-2956 (-147)	-3112 (-155)
Eu ³⁺ (H ₂ O SPCE)	***	-2969 (-148)	***

TABLE S3: Average interaction energy between one PW³⁻ and all Mⁿ⁺ counterions (kcal/mol). Averages over the last 2 ns of dynamics and over all PW³⁻. In brackets: average interaction energy per PW³⁻ - Mⁿ⁺ pair.

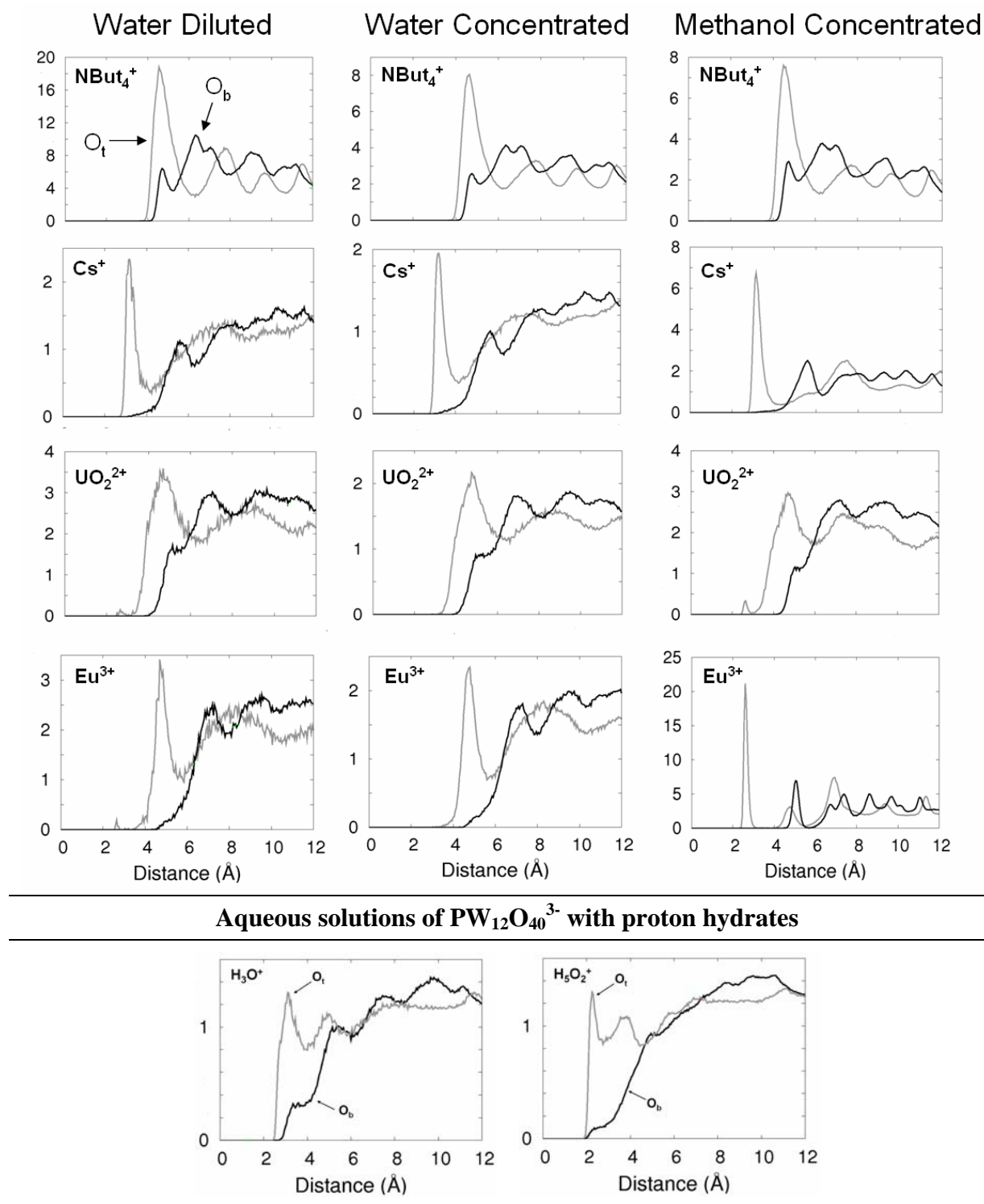


FIGURE S1: $\text{O}_t \dots \text{M}^{n+}$ (grey) and $\text{O}_b \dots \text{M}^{n+}$ (black) RDFs in diluted and concentrated aqueous solutions and in methanol.

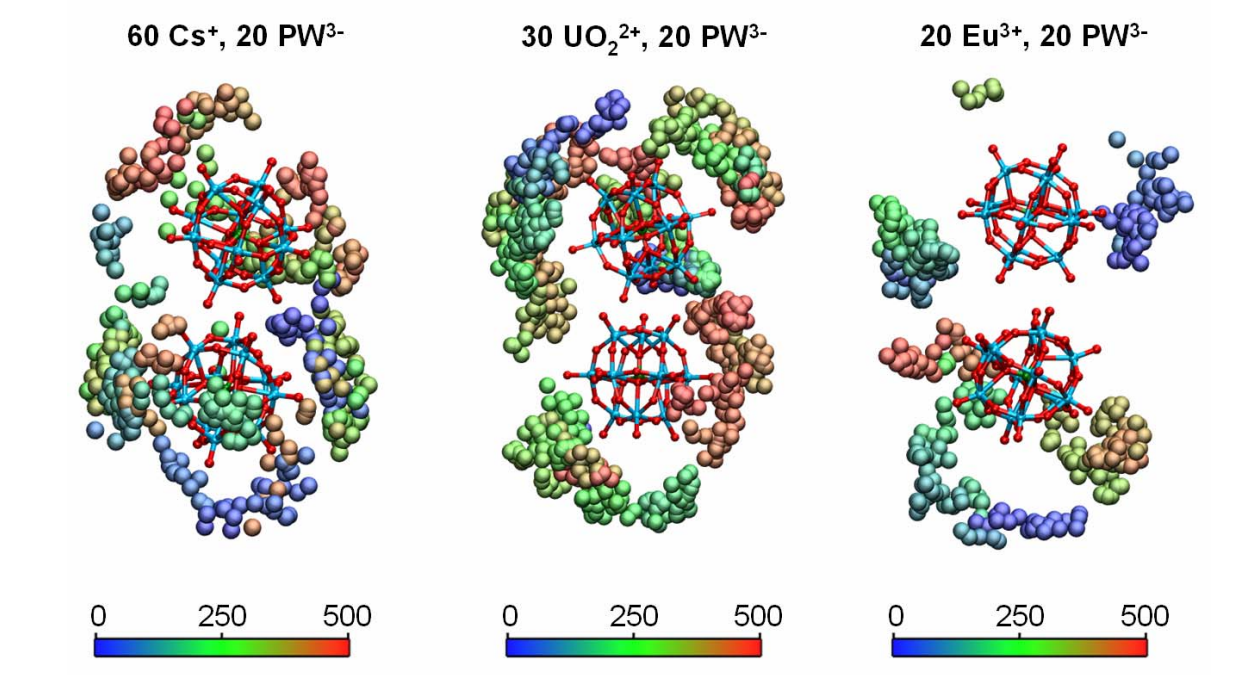


FIGURE S2: (PW³⁻)₂ dimers with Cs⁺, UO₂²⁺ or Eu³⁺ counterions within 3 Å: cumulated views over 1 ns. Different regions of same color correspond to different cations at a given time.

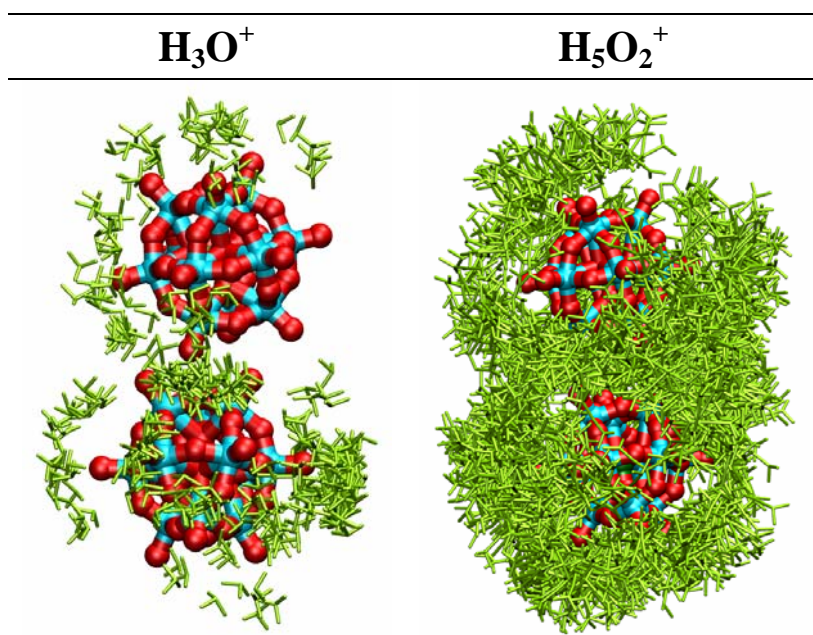


FIGURE S3: $(\text{PW}^{3-})_2$ dimer with H_3O^+ vs H_5O_2^+ counterions in water: cumulated views of first shell M^+ counterions.

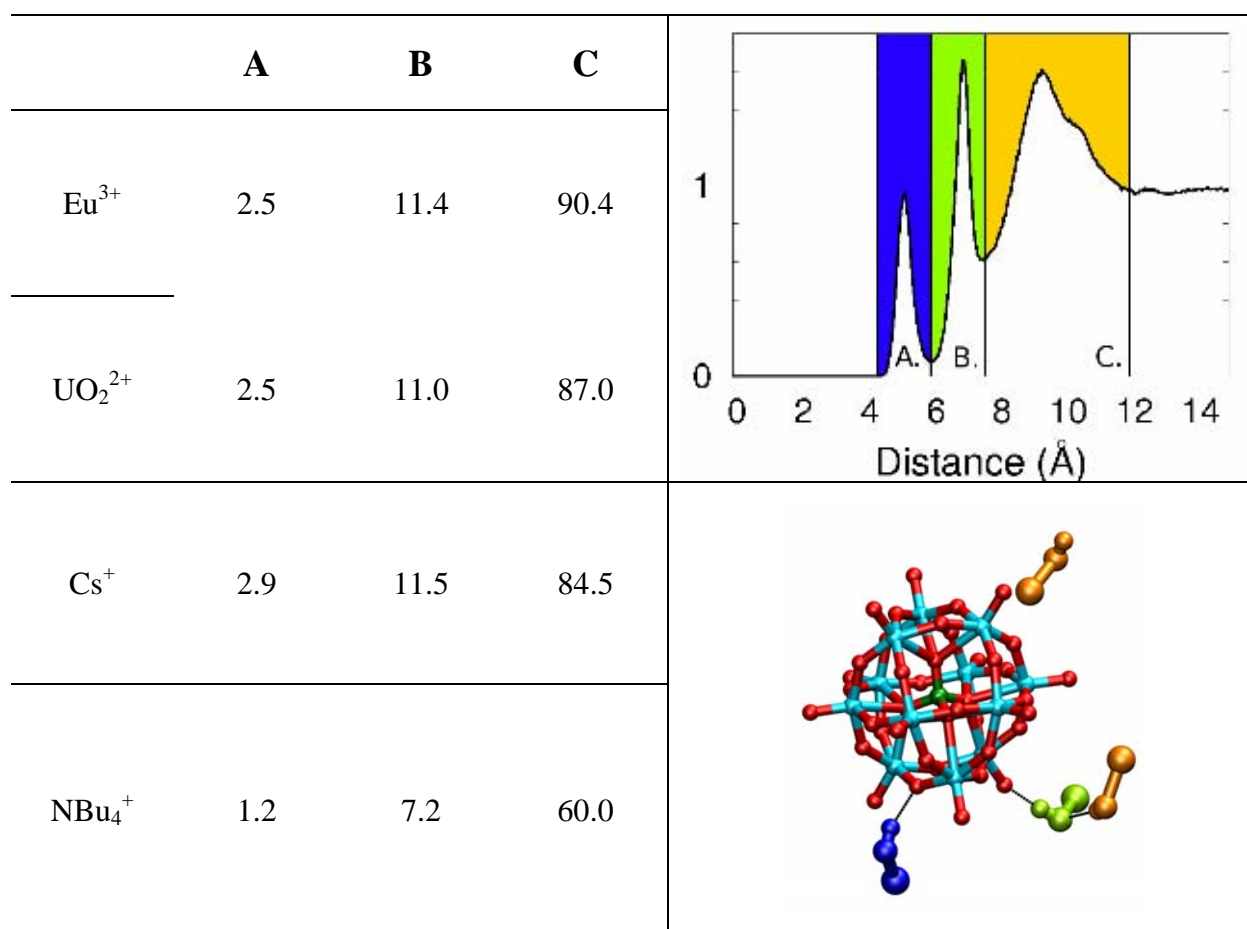


Figure S4: PW³⁻ with NBu₄⁺, Eu³⁺, UO₂²⁺ and Cs⁺ counterions in methanol. *Right:* P...H_{MeOH} RDFs, and snapshots of typical solvation patterns. *Table left:* Integration of the A – C peaks of the RDFs for the different salts.

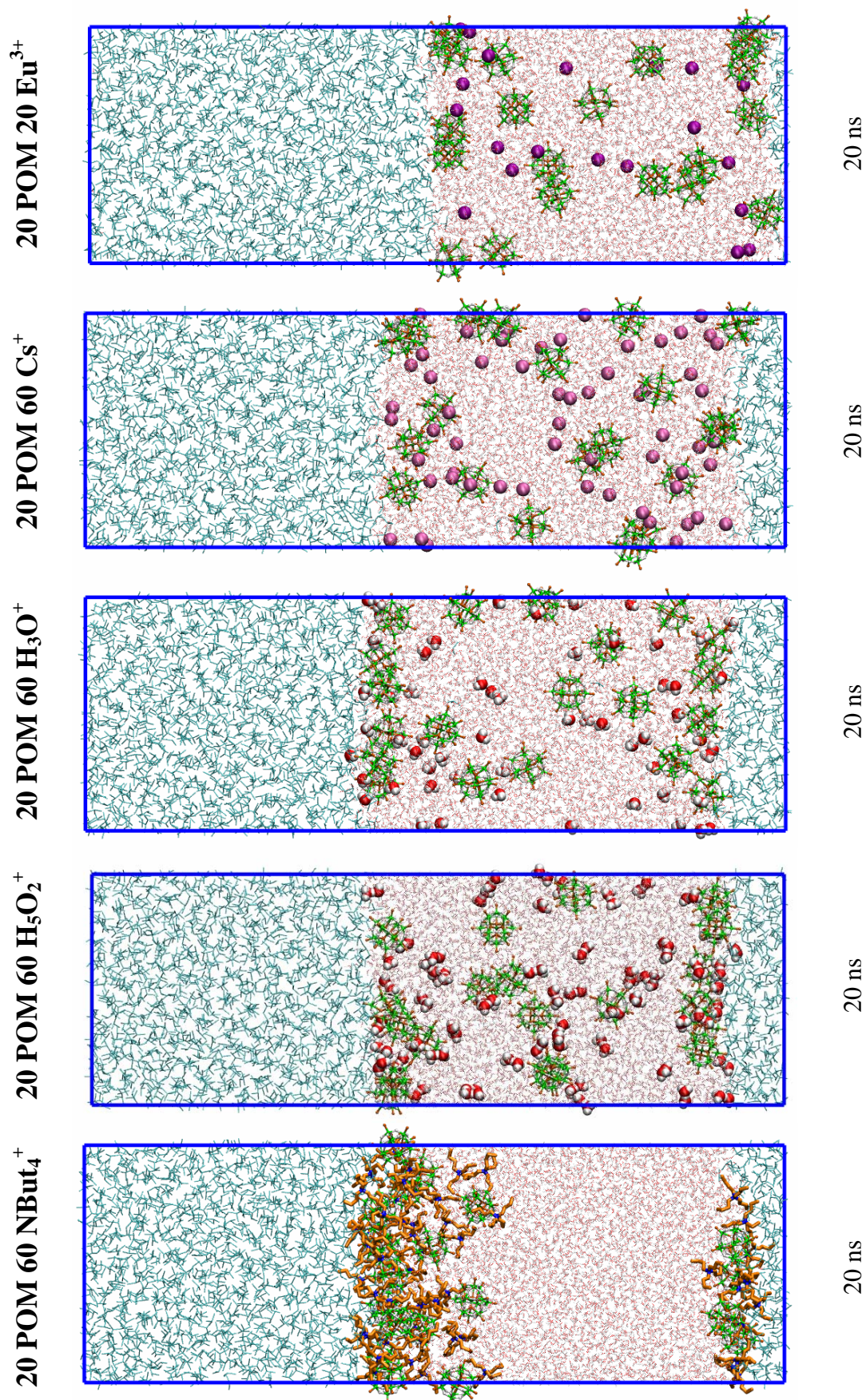


Figure S5: 20 PW³⁻ with either Eu³⁺, Cs⁺, H₃O⁺, H₅O₂⁺ or NBut₄⁺ as counterions at the chloroform / water interface.

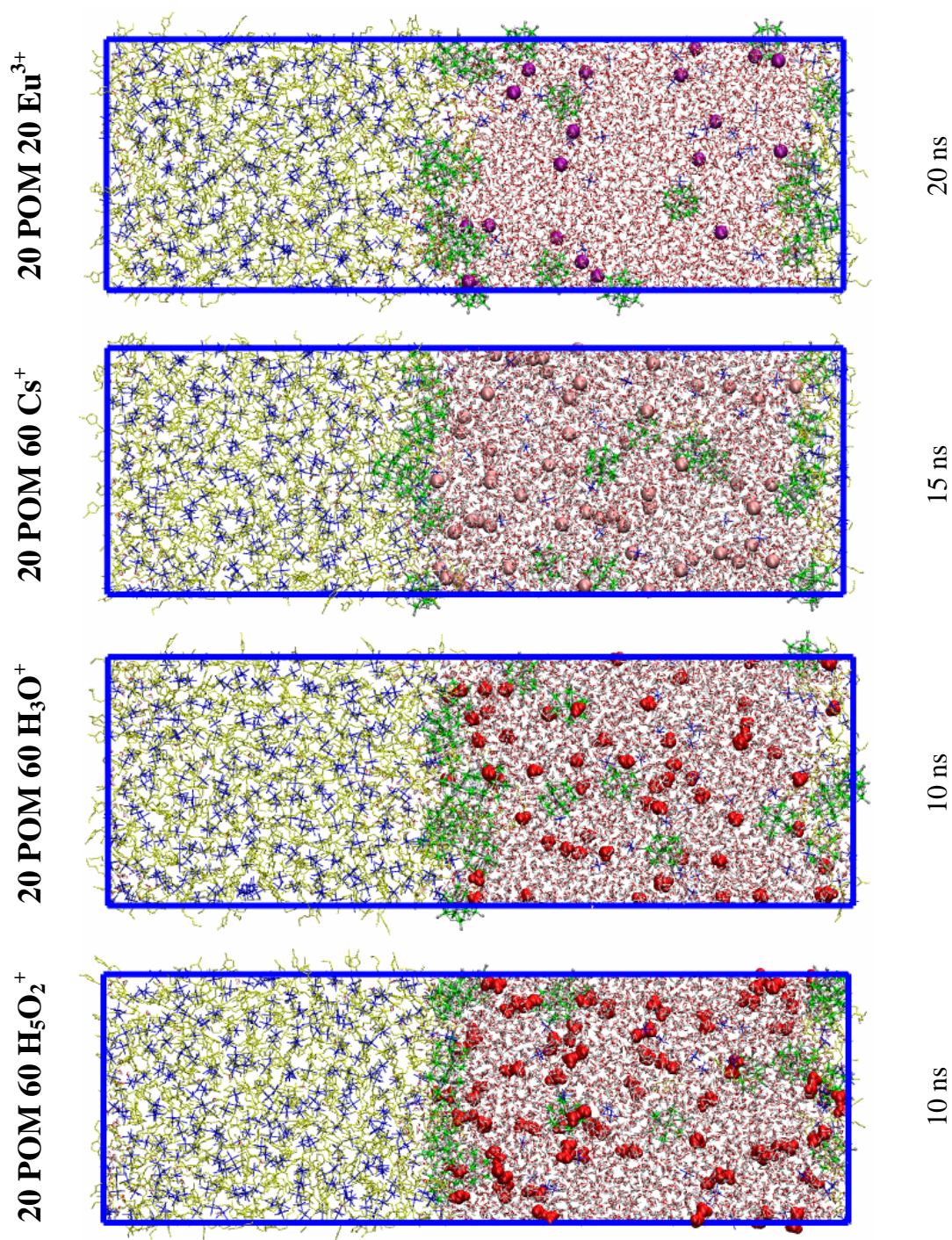


Figure S6: 20 PW^{3-} with either Eu^{3+} , Cs^+ , H_3O^+ or H_5O_2^+ as counterions at the aqueous interface with the $[\text{OMI}][\text{PF}_6]$ room temperature ionic liquid (based on the octyl-methyl imidazolium OMI^+ cation and the PF_6^- anion).