

Calculated molecular parameters (B3LYP\6-311+G(2d,p)) and heats of formation (CBS-Q) of silicon-bearing species.

For comparison, experimental values (where available) are shown in brackets.<sup>12,13,37,41,48</sup>

Species	Geometry <sup>a</sup>	Dipole moment <sup>b</sup>	Rotational constants <sup>c</sup>	Vibrational freqs. <sup>d</sup>	$\Delta H_f^\circ(0\text{ K})^e$
SiO ( $^1\Sigma^+$ )	$r(\text{Si-O})=1.515$ [1.510]	3.2320 [3.0982]	21.629 [21.787]	1247.4 [1241.6] ( $\sigma$ )	-109.87 [-104.3±0.8 <sup>13</sup> ]
SiO <sub>2</sub> ( $^1\Sigma^+$ )	$r(\text{Si-O})=1.512$	0	6.912	297.6(×2) [272.5]( $\pi_u$ ) 991.4 ( $\sigma_g$ ) 1437.8 [1416.5] ( $\sigma_u$ )	-295.09 [-305±33 <sup>12</sup> ] [-320.72 <sup>41</sup> ]
SiH ( $^2\Pi_J$ )	$r(\text{Si-H})=1.529$ [1.520]	0.1118	222.2 [224.8]	2021.2 [2041.8] ( $\sigma$ )	366.20 [375±8 <sup>12</sup> ] [366.9 <sup>41</sup> ]
SiH <sub>2</sub> ( $^1A_1$ )	$r(\text{Si-H})=1.523$ [1.514] $\angle(\text{H-Si-H})=91.53$ [92.0]	0.1464	238.144 210.611 111.766	1023.1 [1001.8] ( $a_1$ ) 2043.0 [1981.6] ( $a_1$ ) 2043.5 [1993.4] ( $b_2$ )	264.30 [275±3 <sup>13</sup> ]
SiC ( $^*^3\Pi_J$ ) <sup>f</sup>	$r(\text{Si-C})=1.717$	1.7634	20.417 [18.167]	976.5 [983] ( $\sigma$ )	731.4 [715±33 <sup>12</sup> ]
SiOCO ( $^3A_1$ ) $C_1$	$r(\text{Si-O})=1.706$ $r(\text{O-C})=1.325$ $r(\text{C-O})=1.179$ $\angle(\text{Si-O-C})=137.70$ $\angle(\text{O-C-O})=130.62$ $\angle(\text{SiOC-O})=-83.69$	2.3104	46.878 3.265 3.110	21.9 <sup>g</sup> (a) 173.5 (a) 539.0 (a) 740.5 (a) 1045.9 (a) 1855.2 (a)	63.30

OSiCO ( <sup>3</sup> A'')	$r(\text{O-Si})=1.551$	2.7713	38.510	118.0	(a')	54.79
	$r(\text{Si-C})=1.827$			290.8	(a'')	
	$r(\text{C-O})=1.155$			398.3	(a')	
	$\angle(\text{O-Si -C})=118.44$			567.0	(a')	
	$\angle(\text{Si-C-O})=170.96$			1070.7	(a')	
				2021.8	(a')	
SiOH ( <sup>2</sup> A')	$r(\text{Si-O})=1.662$	1.4871	793.38952	766.1	(a')	-7.80
	$r(\text{O-H})=0.965$			848.8	(a')	
	$\angle(\text{Si-O-H})=119.76$			3809.9	(a')	
HSiO ( <sup>2</sup> A')	$r(\text{H-Si})=1.528$	3.4073	305.85407	613.1	(a')	19.32
	$r(\text{Si-O})=1.532$			1163.7	(a')	
	$\angle(\text{H-Si-O})=118.60$			1910.5	(a')	
HSiOH- <i>cis</i> ( <sup>1</sup> A')	$r(\text{H-Si})=1.543$	1.5970	171.885	634.1	(a'')	-101.01
	$r(\text{Si-O})=1.665$			742.9	(a')	
	$r(\text{O-H})=0.963$			835.9	(a')	
	$\angle(\text{H-Si-O})=97.73$			963.8	(a')	
	$\angle(\text{Si-O-H})=119.50$			1931.1	(a')	
				3841.7	(a')	
HSiOH- <i>trans</i> ( <sup>1</sup> A')	$r(\text{H-Si})= 1.526$	1.2157	174.997	673.0	(a'')	-101.09
	$r(\text{Si-O})= 1.670$			806.7	(a')	
	$r(\text{O-H})= 0.964$			830.7	(a')	
	$\angle(\text{Hi-Si-O})=94.38$			962.5	(a')	
	$\angle(\text{Si-O-H})=114.85$			2013.2	(a')	
				3846.0	(a')	

HSiOH ( <sup>3</sup> A) <i>C<sub>1</sub></i>	$r(\text{H-Si})= 1.497$	1.3790	214.505	368.4	(a)	67.65
	$r(\text{Si-O})= 1.667$			15.339	(a)	
	$r(\text{O-H})= 0.963$			14.857	(a)	
	$\angle(\text{H-Si-O})=115.43$				(a)	
	$\angle(\text{Si-O-H})=117.47$				(a)	
	$\angle(\text{HSiO-H})=-99.82$				(a)	
SiOH <sub>2</sub> ( <sup>1</sup> A') planar <i>C<sub>2v</sub></i>	$r(\text{Si-O})= 2.050$	4.7986	395.264	205.6	(b <sub>1</sub> )	240.40
	$r(\text{O-H})= 0.965$			10.236	(a <sub>1</sub> )	
	$\angle(\text{H-O-H})=111.47$			9.978	(b <sub>2</sub> )	
					(a <sub>1</sub> )	
					(a <sub>1</sub> )	
					(b <sub>2</sub> )	
SiOH <sub>2</sub> ( <sup>3</sup> A) <i>C<sub>s</sub></i>	$r(\text{Si-O})= 2.246$	3.8737	311.891	239.7	(a')	166.82
	$r(\text{O-H})= 0.966$			8.768	(a')	
	$\angle(\text{Si-O-H})=108.74$			8.648	(a'')	
	$\angle(\text{HOH-Si})=117.09$				(a')	
					(a')	
H <sub>2</sub> SiO ( <sup>1</sup> A') planar <i>C<sub>2v</sub></i>	$r(\text{Si-O})=1.523$	3.8195	166.12548	682.4	(b <sub>2</sub> )	-104.35
	$r(\text{Si-H})=1.481$			18.533	(b <sub>1</sub> )	
	$\angle(\text{H-Si-H})=112.12$			16.673	(a <sub>1</sub> )	
					(a <sub>1</sub> )	
					(a <sub>1</sub> )	
					(b <sub>2</sub> )	

---

<sup>a</sup> Geometrical parameters:  $r$  in Å and  $\angle$  in degrees. <sup>b</sup> In Debye ( $= 3.336 \times 10^{-30}$  Cm). <sup>c</sup> In GHz. <sup>d</sup> In  $\text{cm}^{-1}$ . The symmetry of the vibration is shown in parenthesis. <sup>e</sup> In  $\text{kJ mol}^{-1}$ . Evaluated uncertainty =  $\pm 7 \text{ kJ mol}^{-1}$ .<sup>40</sup> The heats of formation at 0 K of the corresponding atomic species are: Si(<sup>3</sup>P),  $446 \pm 8 \text{ kJ mol}^{-1}$ ; O(<sup>3</sup>P),  $246.98 \pm 0.08 \text{ kJ mol}^{-1}$ ; H(<sup>2</sup>S),  $216.164 \pm 0.004 \text{ kJ mol}^{-1}$ ; and C(<sup>3</sup>P),  $711.7 \pm 0.4 \text{ kJ mol}^{-1}$ .<sup>13</sup> <sup>f</sup> The nature of the ground state is uncertain.<sup>49</sup> <sup>g</sup> Hindered rotor.

#### References:

- 12 M. W. Chase, Jr., *NIST-JANAF Thermochemical Tables*, 4th ed.; *J. Phys. Chem. Ref. Data*, Vol. 9, 1998.
- 13 L. A. Curtiss, K. Raghavachari, P. C. Redfern and J. A. Pople, *J. Chem. Phys.*, 1997, **106**, 1063-1079.
- 37 NIST, Computational Chemistry Comparison and Benchmark DataBase Release 14, NIST Standard Reference Database 101, 2006, <http://srdata.nist.gov/cccbdb>
- 40 L. A. Curtiss, K. Raghavachari, P. C. Redfern and B. B. Stefanov, *J. Chem. Phys.*, 1998, **108**, 692-697.
- 41 L. V. Gurvich, I. V. Veyts and C. B. Alcock, *Thermodynamic Properties of Individual Substances*, 4<sup>th</sup> ed.; Hemisphere Pub. Co., New York, 1989.
- 48 K. P. Huber, G. Herzberg, J. W. Gallagher and I. R.D. Johnson. in *NIST Chemistry WebBook*, NIST Standard Reference Database Number 69, E. P. J. Linstrom and W. G. Mallard, ed., National Institute of Standards and Technology, Gaithersburg MD, <http://webbook.nist.gov>, 2005.