

**Supplementary material:**

**Protonation of water clusters in the cavities of acidic zeolites:  $(H_2O)_n \cdot H\text{-CHA}$ ,  $n=1\text{-}4$**

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**Table S1.** Selected harmonic frequencies ( $\text{cm}^{-1}$ ) calculated for the 1:1 complex<sup>a)</sup>

PBE	BLYP	Assignment <sup>a)</sup>	Experiment <sup>b)</sup>
3720	3685	$\nu(O_h H)$	3670
3589	3571	$\nu(O_h H_b)$	3540
2408	2544		$\sim 2710^{\text{c})}$ (2875; 2400)
(1289) <sup>d)</sup>	(1110) <sup>d)</sup>	$\nu(O1H_{bh})$	(~910) <sup>d)</sup>
1569	1574	$\delta(H_2O_h)$	1630
1413	1402	$\delta(O1H_{bh})$	1406, 1380; 1355 <sup>e)</sup>
976	938	$\gamma(O1H_{bh})$	795, 765 <sup>f)</sup>

<sup>a)</sup>  $\nu$ ,  $\delta$ , and  $\gamma$  denote stretching, in-plane and out-of-plane bending modes, respectively. For atomic labels see Fig. 2. <sup>b)</sup> Frequencies obtained for a coverage of  $\sim 1$  water per acid site [20].

<sup>c)</sup> Minimum of AB doublet (maxima in parenthesis) due to FERMI resonance with overtone of  $\delta(O1H_{bh})$  [20]. <sup>d)</sup> Shift with respect to dry zeolite. <sup>e)</sup> Estimated from the minimum of the AB doublet at  $2710 \text{ cm}^{-1}$  [20]. <sup>f)</sup> For  $H_2O$  adsorbed on HSAPO-34 [17], the IR spectrum of  $H_2O$  adsorbed on HSSZ-13 was measured in the  $3750 - 1360 \text{ cm}^{-1}$  range only [20].

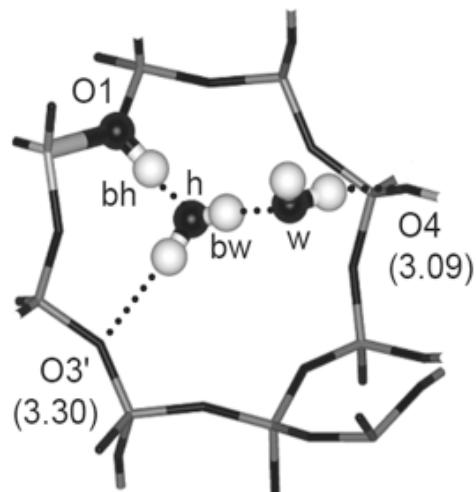
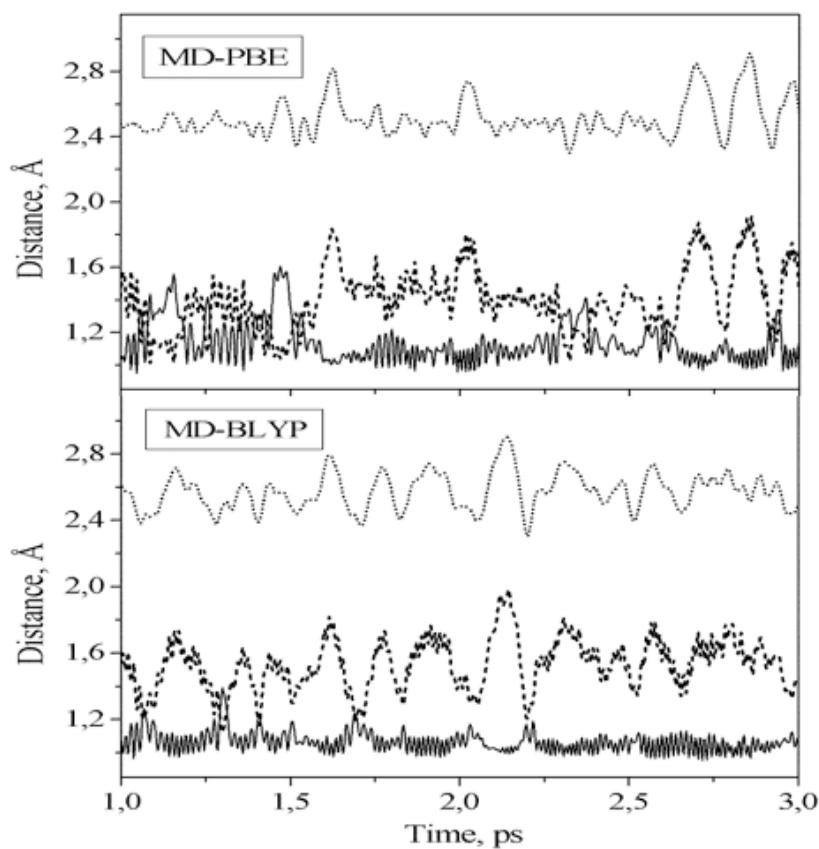
**Table S2.** Relative stabilities,  $\Delta E$  (kJ/mol), and selected interatomic distances ( $\text{\AA}$ ) of different NC and IP species, corresponding to minima on the BLYP potential energy surface of the 2:1 complex.<sup>a</sup>

Species <sup>b</sup>	$\Delta E$	O1...O <sub>h</sub>	H <sub>bh</sub> ...O <sub>h</sub>	O <sub>h</sub> ...O <sub>w</sub>	O <sub>h</sub> ...H <sub>bw</sub>	O <sub>h</sub> ...O3'	O <sub>w</sub> ...O4
NC	0.0	2.49	1.40	2.75	1.00	3.30	3.09
NC	0.3	2.49	1.39	2.76	1.00	3.22	3.12
IP	2.0	2.51	1.11	2.63	1.02	3.12	3.16
IP <sup>c</sup>	6.1	2.54	1.06	2.48	1.08	2.84	3.12; 2.90

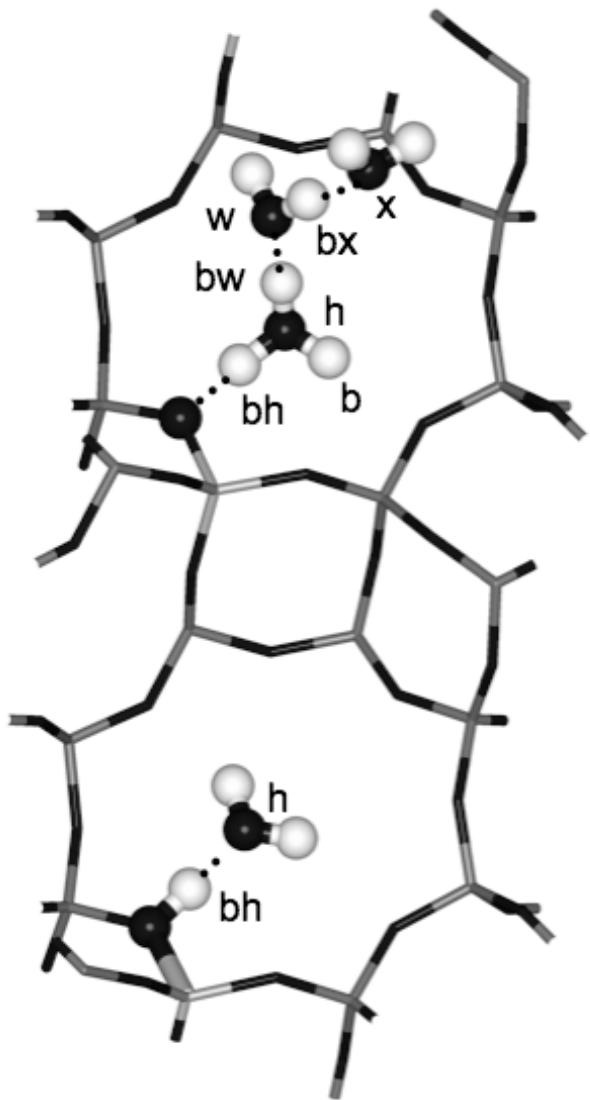
<sup>a</sup> The atomic labels are the same as in Fig. 3.

<sup>b</sup> NC - neutral complex, IP – ion pair.

<sup>c</sup> In the less-stable IP species both OH groups of the second water interact with the framework oxygen atoms by weak H-bonds.



**Figure S1.** Comparison of interatomic distances obtained by molecular dynamics with BLYP and PBE for the 2:1  $\text{H}_2\text{O}:\text{H-CHA}$  complex. Continuous line:  $\text{O1} \dots \text{H}_{\text{bh}}$ ; broken line:  $\text{H}_{\text{bh}} \dots \text{O}_{\text{h}}$ ; dotted line:  $\text{O1} \dots \text{O}_{\text{h}}$ , see structure in the lower part of the figure.



**Figure S2.** The structure of the 1:1 and 3:1 H<sub>2</sub>O-H-CHA complexes in the double cell.

**Table S3.** Selected calculated harmonic frequencies ( $\text{cm}^{-1}$ ) for the 1:1/3:1 loading (2:1 global loading)

BLYP	PBE	Assignment <sup>a)</sup>
3717, 3703, 3625, 3586, 3576, 3450	3760, 3723, 3611, 3600, 3571, 3529	$\nu(\text{OH})$ of OH, free or weakly H-bonded to the framework
3260	3238	$\nu(\text{O}_w\text{H}_{by})$
2507	2288	$\nu^*(\text{O}_1\text{H}_{bh})$
2380	2273	$\nu(\text{O}_h\text{H}_{bw}) + \nu(\text{O}_h\text{H}_{bh})$
2060	1968	$\nu(\text{O}_h\text{H}_{bh}) + \nu(\text{O}_h\text{H}_{bw})$
1674, 1654 1623, 1597, 1549	1679, 1641, 1616, 1588, 1551	$\delta(\text{H}_3\text{O}_h^+), \delta(\text{H}_2\text{O}_h)$ $\delta(\text{H}_2\text{O}_y), \delta(\text{H}_2\text{O}_w), \delta^*(\text{H}_2\text{O}_h)$
1418	1447	$\delta^*(\text{O}_1\text{H}_{bh})$

<sup>a)</sup>  $\nu$ ,  $\delta$ , and  $\gamma$  denote stretching, in-plane and out-of-plane bending modes, respectively. For atomic labels see Figs. 2 and 4a. The asterisk marks the vibrations of the NC structure.