

# Theoretical Prediction of the Solubility of Fluorinated C<sub>60</sub>

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## Supplementary Material

**Table S1** Results for the calculated solubilities ( $\log x_{2,\text{calc}}$ ) using equation 5. Values are tabulated according to increasing value for  $\log x_{2,\text{exp}}$ , as taken from reference <sup>16</sup>, with adapted values due to the forming of solvated crystals underlined (see text). Values for  $\log x_2$  lower than  $(-4,34 \pm 0,43)$  printed in green indicate solubilities above the “ideal” solubility threshold.

Solvent	$\log x_{2,\text{exp}}$	$\log x_{2,\text{calc}}$
Chlorobenzene	<u>-2.991</u>	-3.267
Benzene	<u>-3.161</u>	-3.176
Toluene	-3.377	-3.301
Tetrachloromethane	<u>-3.420</u>	-3.773
Aniline	-3.875	-3.736
Dichloromethane	-4.699	-4.580
Trichloromethane	-4.745	-4.403
n-Heptane	<u>-4.921</u>	-4.728
1,2-Dichloroethane	-5.046	-5.106
Tetrahydrofuran	-5.155	-5.200
Cyclohexane	-5.268	-5.313
Ethanol	-7.086	-7.651
Methanol	-8.699	-8.320
Dimethylsulfoxide	-	-12.510
Water	-	-16.879

**Table S2** Electrostatic ( $\Delta E_{\text{es}}$ ), cavity ( $\Delta E_{\text{cav}}$ ), dispersion ( $\Delta E_{\text{disp}}$ ), repulsion ( $\Delta E_{\text{rep}}$ ) and total energy difference ( $\Delta E_{\text{tot}}$ ) between the solvent and the gas phase for C<sub>60</sub>F<sub>2</sub> in kcal/mol.

Solvent	$\Delta E_{\text{es}}$	$\Delta E_{\text{cav}}$	$\Delta E_{\text{disp}}$	$\Delta E_{\text{rep}}$	$\Delta E_{\text{tot}}$
Water	-12.54	47.32	-42.44	2.43	-5.23
Dimethylsulfoxide	-7.74	44.65	-44.28	2.52	-4.85
Methanol	-4.29	34.08	-36.74	2.04	-4.91
Ethanol	-3.82	34.70	-37.64	2.06	-4.70
Dichloromethane	-4.26	34.83	-36.71	2.06	-4.08
Tetrahydrofuran	-6.83	40.92	-40.12	2.18	-3.85
Chlorobenzene	-6.52	40.89	-41.31	3.46	-3.48
Toluene	-1.67	37.46	-41.46	3.59	-2.08
Benzene	-3.02	38.12	-40.67	3.58	-1.99

**Table S3** Electrostatic ( $\Delta E_{es}$ ), cavity ( $\Delta E_{cav}$ ), dispersion ( $\Delta E_{disp}$ ), repulsion ( $\Delta E_{rep}$ ) and total energy difference ( $\Delta E_{tot}$ ) between the solvent and the gas phase for  $C_{60}F_4$  in kcal/mol.

Solvent	$\Delta E_{es}$	$\Delta E_{cav}$	$\Delta E_{disp}$	$\Delta E_{rep}$	$\Delta E_{tot}$
Water	-15.51	48.84	-43.38	2.49	-7.56
Dimethylsulfoxide	-10.46	46.14	-45.23	2.57	-6.98
Methanol	-6.74	35.21	-37.56	2.09	-7.00
Ethanol	-6.25	35.86	-38.48	2.11	-6.76
Dichloromethane	-6.50	36.00	-37.43	2.08	-5.85
Tetrahydrofurane	-9.09	42.29	-41.01	2.23	-5.58
Chlorobenzene	-8.64	42.27	-42.16	3.52	-5.01
Toluene	-3.01	38.73	-42.34	3.66	-2.96
Benzene	-4.33	39.41	-41.54	3.65	-2.81

**Table S4** Electrostatic ( $\Delta E_{es}$ ), cavity ( $\Delta E_{cav}$ ), dispersion ( $\Delta E_{disp}$ ), repulsion ( $\Delta E_{rep}$ ) and total energy difference ( $\Delta E_{tot}$ ) between the solvent and the gas phase for  $C_{60}F_6$  in kcal/mol.

Solvent	$\Delta E_{es}$	$\Delta E_{cav}$	$\Delta E_{disp}$	$\Delta E_{rep}$	$\Delta E_{tot}$
Water	-19.01	50.93	-44.09	2.53	-9.64
Dimethylsulfoxide	-13.90	48.16	-45.89	2.59	-9.04
Methanol	-9.82	36.75	-38.14	2.11	-9.10
Ethanol	-9.29	37.44	-39.07	2.14	-8.78
Dichloromethane	-9.25	37.59	-38.02	2.10	-7.58
Tetrahydrofurane	-11.98	44.16	-41.62	2.25	-7.19
Chlorobenzene	-11.36	44.14	-42.87	3.57	-6.52
Toluene	-4.97	40.46	-43.06	3.71	-3.86
Benzene	-6.28	41.16	-42.24	3.70	-3.66

**Table S5** Electrostatic ( $\Delta E_{es}$ ), cavity ( $\Delta E_{cav}$ ), dispersion ( $\Delta E_{disp}$ ), repulsion ( $\Delta E_{rep}$ ) and total energy difference ( $\Delta E_{tot}$ ) between the solvent and the gas phase for  $C_{60}F_8$  in kcal/mol.

Solvent	$\Delta E_{es}$	$\Delta E_{cav}$	$\Delta E_{disp}$	$\Delta E_{rep}$	$\Delta E_{tot}$
Water	-22.50	53.08	-44.93	2.59	-11.76
Dimethylsulfoxide	-17.10	50.25	-46.72	2.63	-10.94
Methanol	-12.67	38.34	-38.86	2.16	-11.03
Ethanol	-12.09	39.07	-39.80	2.18	-10.64
Dichloromethane	-11.91	39.23	-38.67	2.12	-9.23
Tetrahydrofurane	-14.73	46.08	-42.39	2.29	-8.75
Chlorobenzene	-13.95	46.07	-43.65	3.63	-7.90
Toluene	-6.83	42.23	-43.86	3.78	-4.68
Benzene	-8.14	42.96	-43.03	3.77	-4.44

**Table S6** Electrostatic ( $\Delta E_{es}$ ), cavity ( $\Delta E_{cav}$ ), dispersion ( $\Delta E_{disp}$ ), repulsion ( $\Delta E_{rep}$ ) and total energy difference ( $\Delta E_{tot}$ ) between the solvent and the gas phase for  $C_{60}F_{10}$  in kcal/mol.

Solvent	$\Delta E_{es}$	$\Delta E_{cav}$	$\Delta E_{disp}$	$\Delta E_{rep}$	$\Delta E_{tot}$
Water	-25.63	55.23	-45.78	2.64	-13.54
Dimethylsulfoxide	-20.05	52.34	-47.55	2.67	-12.59
Methanol	-15.23	39.93	-39.58	2.20	-12.68
Ethanol	-14.71	40.70	-40.53	2.22	-12.32
Dichloromethane	-14.32	40.87	-39.32	2.15	-10.62
Tetrahydrofuran	-17.21	48.01	-43.16	2.33	-10.03
Chlorobenzene	-16.33	48.01	-44.43	3.69	-9.06
Toluene	-8.48	44.01	-44.67	3.85	-5.29
Benzene	-9.79	44.76	-43.82	3.83	-5.02

**Table S7** Electrostatic ( $\Delta E_{es}$ ), cavity ( $\Delta E_{cav}$ ), dispersion ( $\Delta E_{disp}$ ), repulsion ( $\Delta E_{rep}$ ) and total energy difference ( $\Delta E_{tot}$ ) between the solvent and the gas phase for  $C_{60}F_{12}$  in kcal/mol.

Solvent	$\Delta E_{es}$	$\Delta E_{cav}$	$\Delta E_{disp}$	$\Delta E_{rep}$	$\Delta E_{tot}$
Water	-28.75	57.41	-46.62	2.69	-15.27
Dimethylsulfoxide	-22.87	54.44	-48.53	2.73	-14.23
Methanol	-17.79	41.53	-40.30	2.24	-14.32
Ethanol	-17.23	42.35	-41.26	2.26	-13.88
Dichloromethane	-16.36	42.52	-40.25	2.19	-11.90
Tetrahydrofuran	-19.65	49.95	-43.93	2.38	-11.25
Chlorobenzene	-18.56	49.95	-45.31	3.76	-10.16
Toluene	-10.15	45.80	-45.49	3.92	-5.92
Benzene	-11.46	46.57	-44.62	3.90	-5.61

**Table S8** Electrostatic ( $\Delta E_{es}$ ), cavity ( $\Delta E_{cav}$ ), dispersion ( $\Delta E_{disp}$ ), repulsion ( $\Delta E_{rep}$ ) and total energy difference ( $\Delta E_{tot}$ ) between the solvent and the gas phase for  $C_{60}F_{14}$  in kcal/mol.

Solvent	$\Delta E_{es}$	$\Delta E_{cav}$	$\Delta E_{disp}$	$\Delta E_{rep}$	$\Delta E_{tot}$
Water	-32.10	59.55	-47.44	2.74	-17.25
Dimethylsulfoxide	-26.00	56.52	-49.18	2.75	-15.91
Methanol	-20.46	43.11	-40.99	2.28	-16.06
Ethanol	-19.74	43.96	-41.97	2.30	-15.45
Dichloromethane	-19.03	44.14	-40.61	2.19	-13.31
Tetrahydrofuran	-22.18	51.86	-44.67	2.41	-12.58
Chlorobenzene	-21.05	51.87	-45.98	3.80	-11.36
Toluene	-11.91	47.56	-46.27	3.98	-6.64
Benzene	-13.25	48.36	-45.38	3.96	-6.31

**Table S9** Electrostatic ( $\Delta E_{es}$ ), cavity ( $\Delta E_{cav}$ ), dispersion ( $\Delta E_{disp}$ ), repulsion ( $\Delta E_{rep}$ ) and total energy difference ( $\Delta E_{tot}$ ) between the solvent and the gas phase for  $C_{60}F_{16}$  in kcal/mol.

Solvent	$\Delta E_{es}$	$\Delta E_{cav}$	$\Delta E_{disp}$	$\Delta E_{rep}$	$\Delta E_{tot}$
Water	-35.09	61.82	-48.38	2.81	-18.84
Dimethylsulfoxide	-28.83	58.72	-50.03	2.79	-17.35
Methanol	-22.91	44.78	-41.76	2.33	-17.56
Ethanol	-22.23	45.68	-42.74	2.35	-16.94
Dichloromethane	-21.32	45.87	-41.25	2.22	-14.48
Tetrahydrofurane	-24.62	53.89	-45.46	2.46	-13.73
Chlorobenzene	-23.30	53.91	-46.86	3.88	-12.37
Toluene	-13.51	49.44	-47.19	4.06	-7.20
Benzene	-14.87	50.26	-46.29	4.05	-6.85

**Table S10** Electrostatic ( $\Delta E_{es}$ ), cavity ( $\Delta E_{cav}$ ), dispersion ( $\Delta E_{disp}$ ), repulsion ( $\Delta E_{rep}$ ) and total energy difference ( $\Delta E_{tot}$ ) between the solvent and the gas phase for  $C_{60}F_{18}$  in kcal/mol.

Solvent	$\Delta E_{es}$	$\Delta E_{cav}$	$\Delta E_{disp}$	$\Delta E_{rep}$	$\Delta E_{tot}$
Water	-42.58	68.73	-49.11	2.86	-20.10
Dimethylsulfoxide	-35.71	65.28	-50.80	2.84	-18.39
Methanol	-28.53	49.79	-42.41	2.37	-18.78
Ethanol	-27.72	50.79	-43.42	2.39	-17.96
Dichloromethane	-26.69	51.00	-41.86	2.24	-15.31
Tetrahydrofurane	-30.76	59.91	-46.19	2.50	-14.54
Chlorobenzene	-29.39	59.94	-47.50	3.92	-13.03
Toluene	-18.73	54.97	-47.86	4.11	-7.51
Benzene	-20.15	55.88	-46.93	4.10	-7.10

**Table S11** Electrostatic ( $\Delta E_{es}$ ), cavity ( $\Delta E_{cav}$ ), dispersion ( $\Delta E_{disp}$ ), repulsion ( $\Delta E_{rep}$ ) and total energy difference ( $\Delta E_{tot}$ ) between the solvent and the gas phase for  $C_{60}F_{20}$  in kcal/mol.

Solvent	$\Delta E_{es}$	$\Delta E_{cav}$	$\Delta E_{disp}$	$\Delta E_{rep}$	$\Delta E_{tot}$
Water	-40.70	66.46	-50.20	2.91	-21.53
Dimethylsulfoxide	-33.88	63.20	-51.88	2.89	-19.67
Methanol	-27.32	48.19	-43.40	2.42	-20.11
Ethanol	-26.38	49.18	-44.44	2.44	-19.20
Dichloromethane	-25.33	49.39	-42.66	2.27	-16.33
Tetrahydrofurane	-28.77	58.01	-47.25	2.55	-15.46
Chlorobenzene	-27.59	58.05	-48.37	3.97	-13.94
Toluene	-16.65	53.24	-48.8	4.17	-8.04
Benzene	-18.03	54.12	-47.83	4.15	-7.59

**Table S12** Cavity volume for C<sub>60</sub>F<sub>2n</sub> with n = 1-10, calculated within the PCM formalism.

Solvent	C <sub>60</sub> F <sub>2</sub>	C <sub>60</sub> F <sub>4</sub>	C <sub>60</sub> F <sub>6</sub>	C <sub>60</sub> F <sub>8</sub>	C <sub>60</sub> F <sub>10</sub>	C <sub>60</sub> F <sub>12</sub>	C <sub>60</sub> F <sub>14</sub>	C <sub>60</sub> F <sub>16</sub>	C <sub>60</sub> F <sub>18</sub>	C <sub>60</sub> F <sub>20</sub>
Chlorobenzene	380.44	394.17	407.27	420.76	433.71	447.91	461.53	474.74	490.20	501.09
Benzene	381.98	393.92	406.82	420.30	433.17	447.49	460.72	473.82	489.19	500.69
Toluene	380.49	393.78	407.26	420.82	433.57	447.81	461.49	474.82	490.19	501.14
Dichloromethane	381.25	393.01	405.96	419.23	432.25	445.66	459.74	473.27	487.27	499.78
Tetrahydrofurane	382.88	393.14	406.70	420.11	433.03	447.04	460.93	473.74	488.82	500.50
Ethanol	381.87	392.79	405.73	419.37	431.91	445.34	459.39	472.00	486.31	499.32
Methanol	380.80	392.24	405.20	418.16	431.37	444.38	457.54	470.34	485.30	495.46
Dimethylsulfoxide	379.73	394.49	406.29	419.83	432.92	446.51	459.99	473.45	488.24	500.05
Water	380.37	390.96	403.81	416.31	429.15	442.48	454.79	467.61	481.35	493.06

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