Electronic Supplementary Information

Electron Transport Properties in Fluorinated Copper-Phthalocyanine Films: Importance of Vibrational Reorganization Energy and Molecular Microstructure

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1. Calculated Vibrational Reorganization Energy of $F_{16}H_2Pc$

Table S1. B3LYP/6-31G(d) calculations of
frequencies and vibrational reorganization
energies, λ_v , of A_g normal modes for F ₁₆ H ₂ Pc in
its neutral state.

Calculated	2	
Frequency	λ _ν	
(cm^{-1})	(meV)	(%)
103	2.73	2.25
161	0.02	0.02
271	0.05	0.04
271	0.07	0.06
313	0.46	0.38
326	0.09	0.07
436	0.48	0.39
462	0.24	0.20
568	0.88	0.73
722	8.27	6.84
743	9.01	7.45
947	22.76	18.81
969	4.01	3.31
1190	1.76	1.45
1334	0.05	0.04
1337	3.76	3.10
1381	0.18	0.15
1402	3.05	2.52
1441	21.04	17.38
1520	7.06	5.83
1524	0.33	0.27
1560	0.21	0.18
1617	0.86	0.71
1655	0.46	0.38
1668	33.08	27.34
3620	0.08	0.06
sum (meV) =	121.0	

2. Difference Spectrum between F_{16} CuPc thin-films grown on different substrate temperature



Fig. S1. Raman spectra ($\lambda_{exc} = 633$ nm) of F₁₆CuPc thin-films deposited at different substrate temperature: (a) 30 °C; (b) 120 °C. The spectra were normalized to the intensity of the 1540 cm⁻¹. Trace (a-b) is the difference spectrum, obtained by subtracting the spectra of (a) and (b). The bands discussed in the manuscript are labeled.

3. Polarized Raman spectrum



Fig. S2. Polarized Raman spectra ($\lambda_{exc} = 633$ nm) of the F₁₆CuPc film. The polarizations of the incident and detected light are parallel (*x*,*x*) and perpendicular (*x*,*y*) to each other. The dashed lines serve as guidelines.