Supporting Information

## Bright White Upconversion Emission from Yb<sup>3+</sup>, Er<sup>3+</sup>, and Tm<sup>3+</sup>-Codoped Gd<sub>2</sub>O<sub>3</sub> Nanotubes

Kezhi Zheng, Daisheng Zhang, Dan Zhao, Ning Liu, Feng Shi, and Weiping Qin\*

State Key Laboratory on Integrated Optoelectronics, College of Electronic Science

and Engineering, Jilin University, Changchun 130012, China

\* Corresponding author: wpqin@jlu.edu.cn

Fig. S1 show the XRD patterns of  $Gd(OH)_3:10\%Yb^{3+}/1\%Er^{3+}/0.7\%Tm^{3+}$  and  $Gd_2O_3:10\%Yb^{3+}/1\%Er^{3+}/0.7\%Tm^{3+}$  nanocrystals. All diffraction peaks can be readily indexed to pure hexagonal  $Gd(OH)_3$  and cubic  $Gd_2O_3$ , which are in good agreement with the standard values for the  $Gd(OH)_3$  and  $Gd_2O_3$  (JCPDS No.83–2037 and 11–0604), respectively. No other impurity peaks can be detected from the XRD patterns, indicating that the nanocrystals are single-phased and  $Ln^{3+}$  ions have effectively incorporated into the  $Gd(OH)_3$  and  $Gd_2O_3$  host lattices.



Fig. S1. XRD patterns of (a)  $Gd(OH)_3:10\%Yb^{3+}/1\%Er^{3+}/0.7\%Tm^{3+}$  and (b)  $Gd_2O_3:10\%Yb^{3+}/1\%Er^{3+}/0.7\%Tm^{3+}$  nanocrystals. The standard data for  $Gd(OH)_3$  (JCPDS No.83–2037) and  $Gd_2O_3$  (JCPDS No.11–0604) are also presented in the figure for comparison.

Fig. S2 shows the TGA curve of Gd(OH)<sub>3</sub>:10%Yb<sup>3+</sup>/1%Er<sup>3+</sup>/0.7%Tm<sup>3+</sup> nanocrystals. It can be observed that there are two major stages of rapid weight loss at about 311°C and 420°C, indicating the existence of intermediate phase other than Gd(OH)<sub>3</sub> and Gd<sub>2</sub>O<sub>3</sub> during the thermal conversion process. The weight loss for the two stages and the total weight loss are 8.86%, 4.32%, and 13.18%, respectively. The total weight loss is in agreement with the theoretical value of Gd(OH)<sub>3</sub>, calculated from the reaction of its complete dehydration to produce Gd<sub>2</sub>O<sub>3</sub>. Since most of the rare earth compounds can exist in the form of LnOOH, this dehydration process can be supposed to be two steps: Gd(OH)<sub>3</sub>  $\rightarrow$  GdOOH + H<sub>2</sub>O and 2GdOOH  $\rightarrow$  Gd<sub>2</sub>O<sub>3</sub> + H<sub>2</sub>O. The theoretical weight loss for the two processes is 8.64% and 4.73%, respectively, which is quite close to the experimental data.



Fig. S2. TGA curve of  $Gd(OH)_3$ :10%Yb<sup>3+</sup>/1%Er<sup>3+</sup>/0.7%Tm<sup>3+</sup> nanotubes.

Fig. S3 describes the XRD pattern for the product calcinated at 300°C. It fit well with the GdOOH, which has been well addressed by *C Chang et al.* (*Mater. lett.*, 2005, **59**, 1037. and *Nanotechnology*, 2006, **17**, 1981.).



**Fig. S3.** XRD pattern of GdOOH:10%Yb<sup>3+</sup>/1%Er<sup>3+</sup>/0.7%Tm<sup>3+</sup>