

## **Supplementary Information**

### **Deep-space glycine formation via Strecker-type reactions activated by ice water dust mantles. A computational approach.**

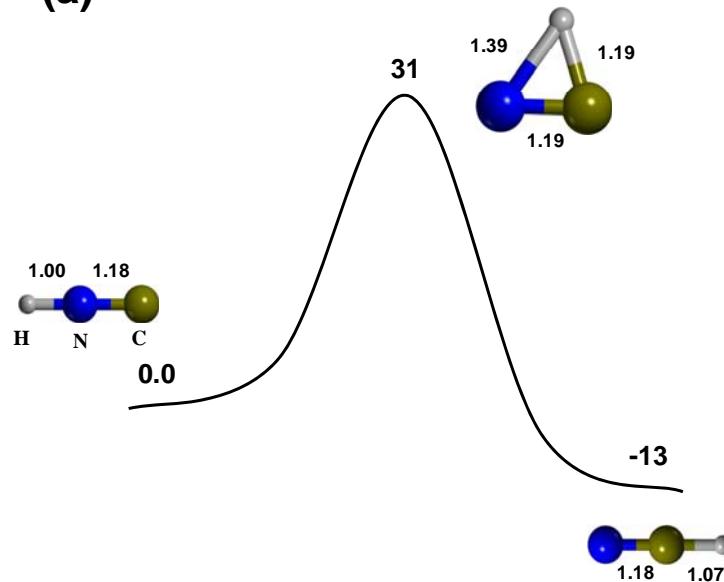
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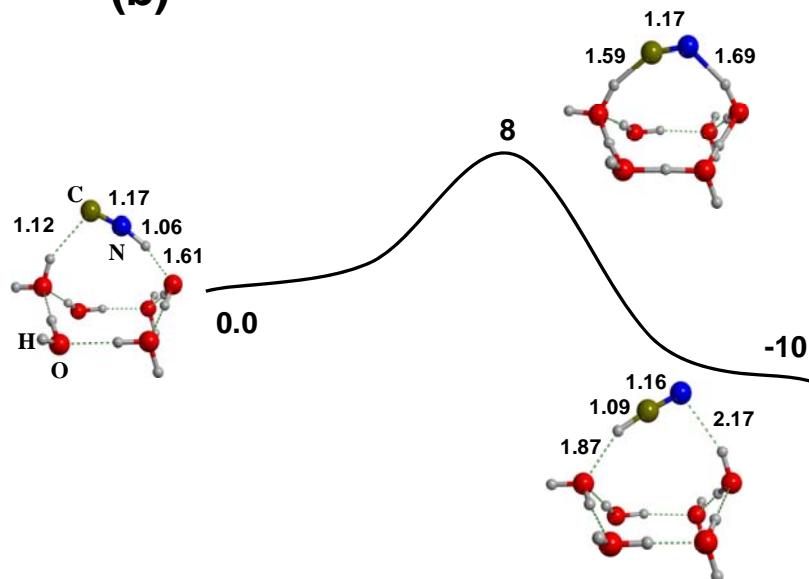
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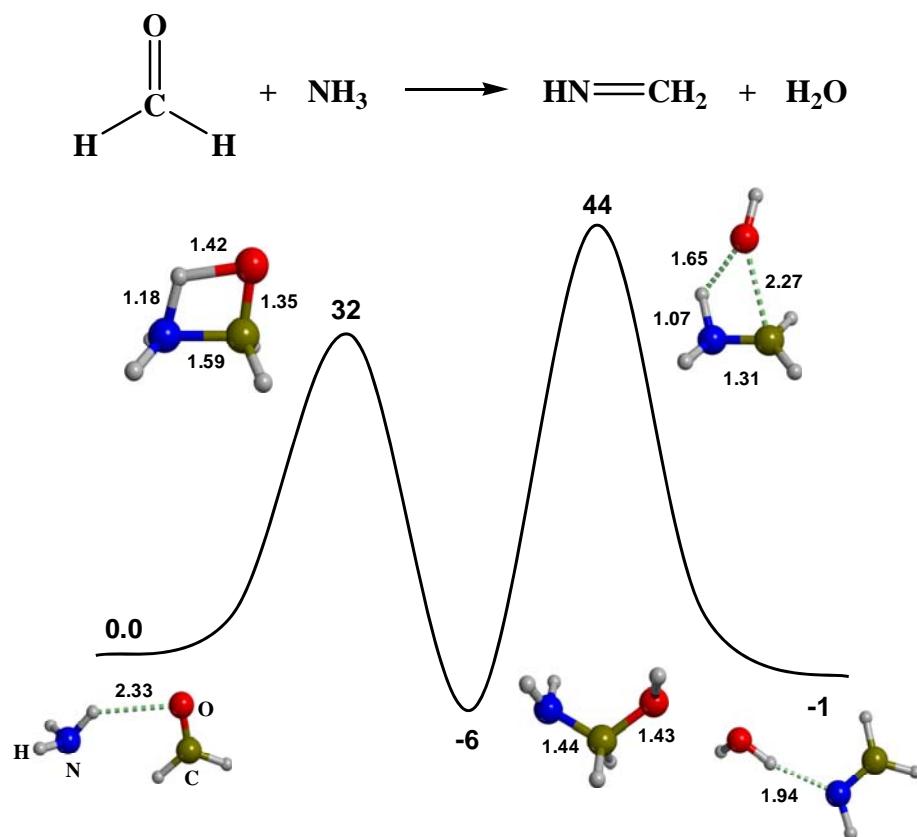
(a)



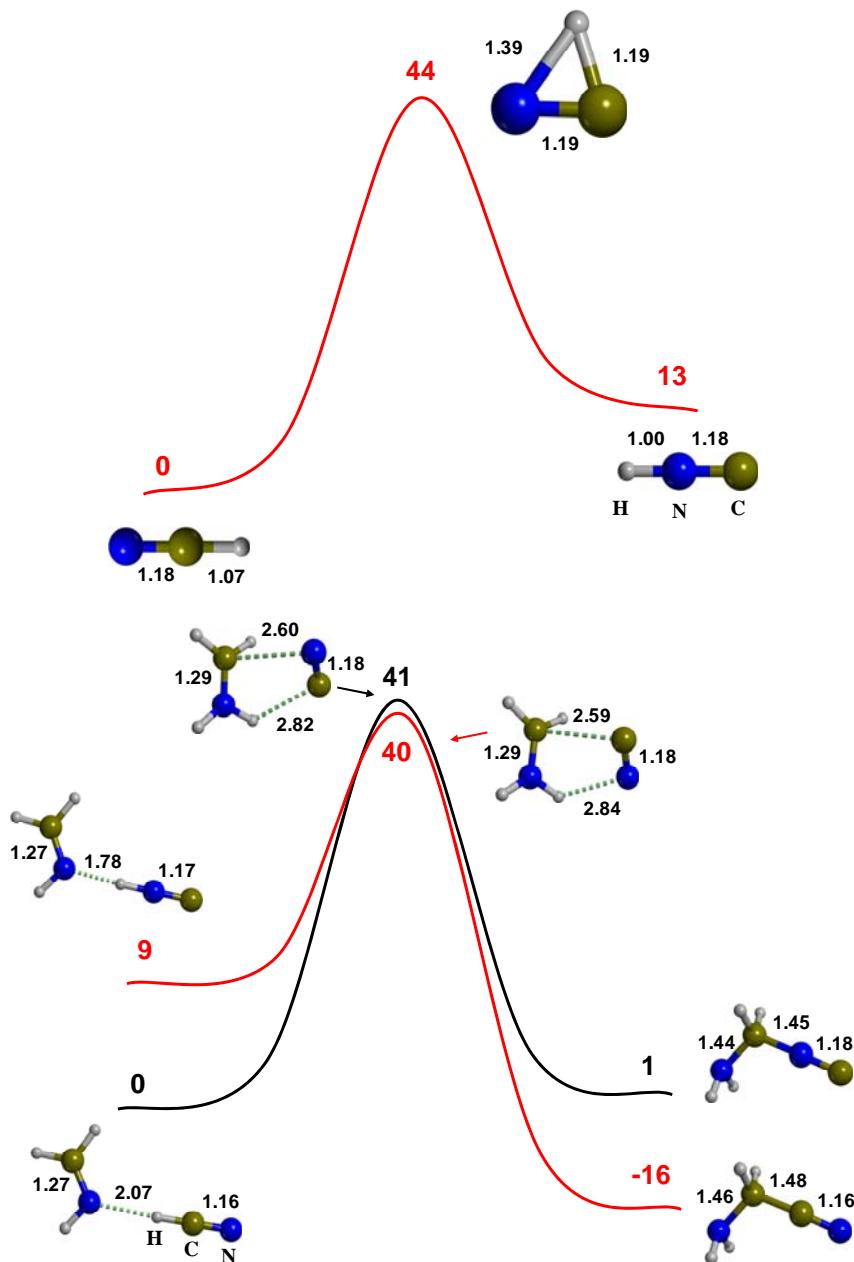
(b)



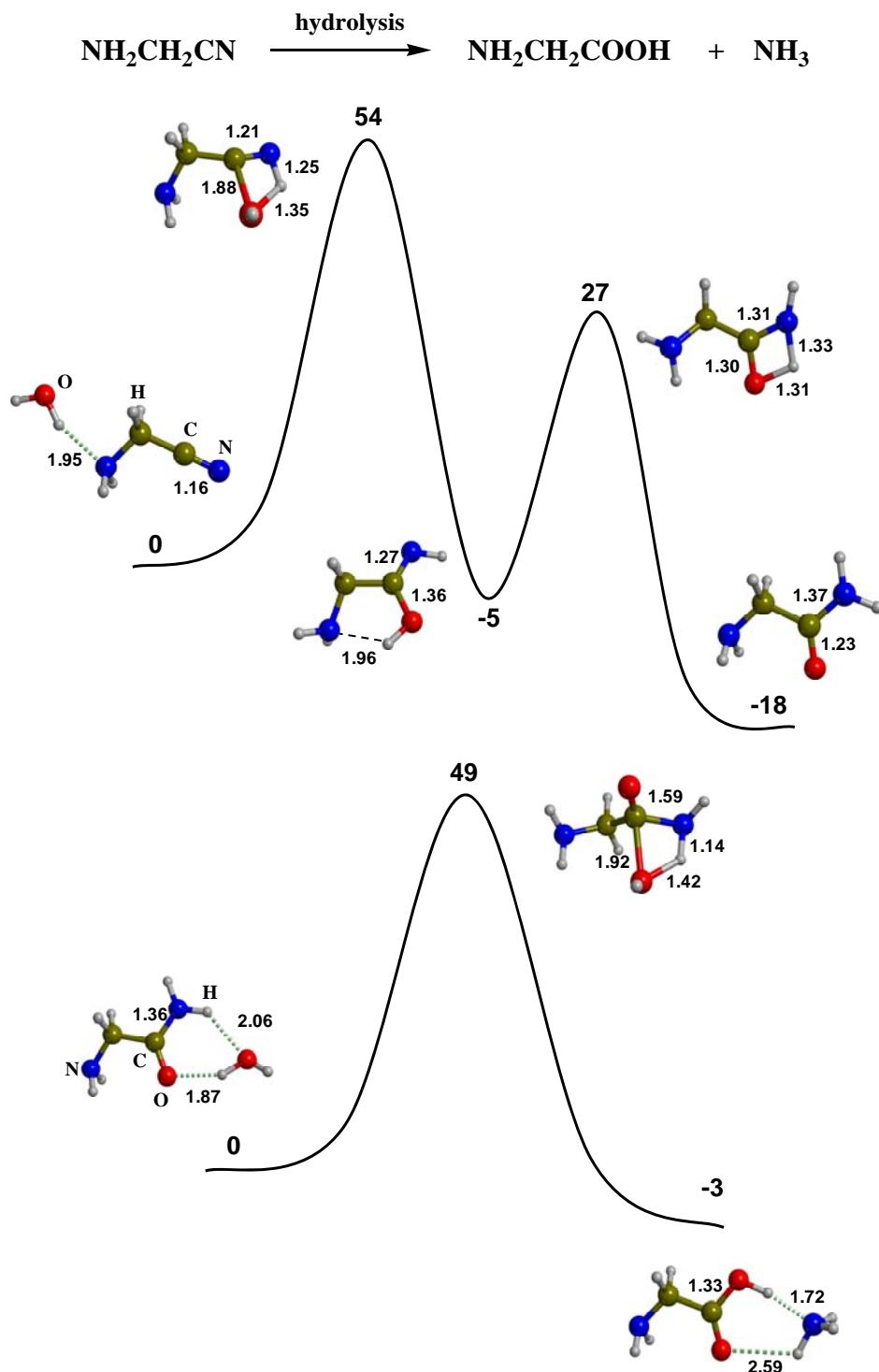
**Figure S1.** B3LYP/6-31+G(d,p) isomerization reaction of CNH → HCN: (a) in gas-phase, (b) at the H<sub>2</sub>O-ice surface. Relative ZPE-corrected energies in kcal mol<sup>-1</sup>, bond distances in Å.



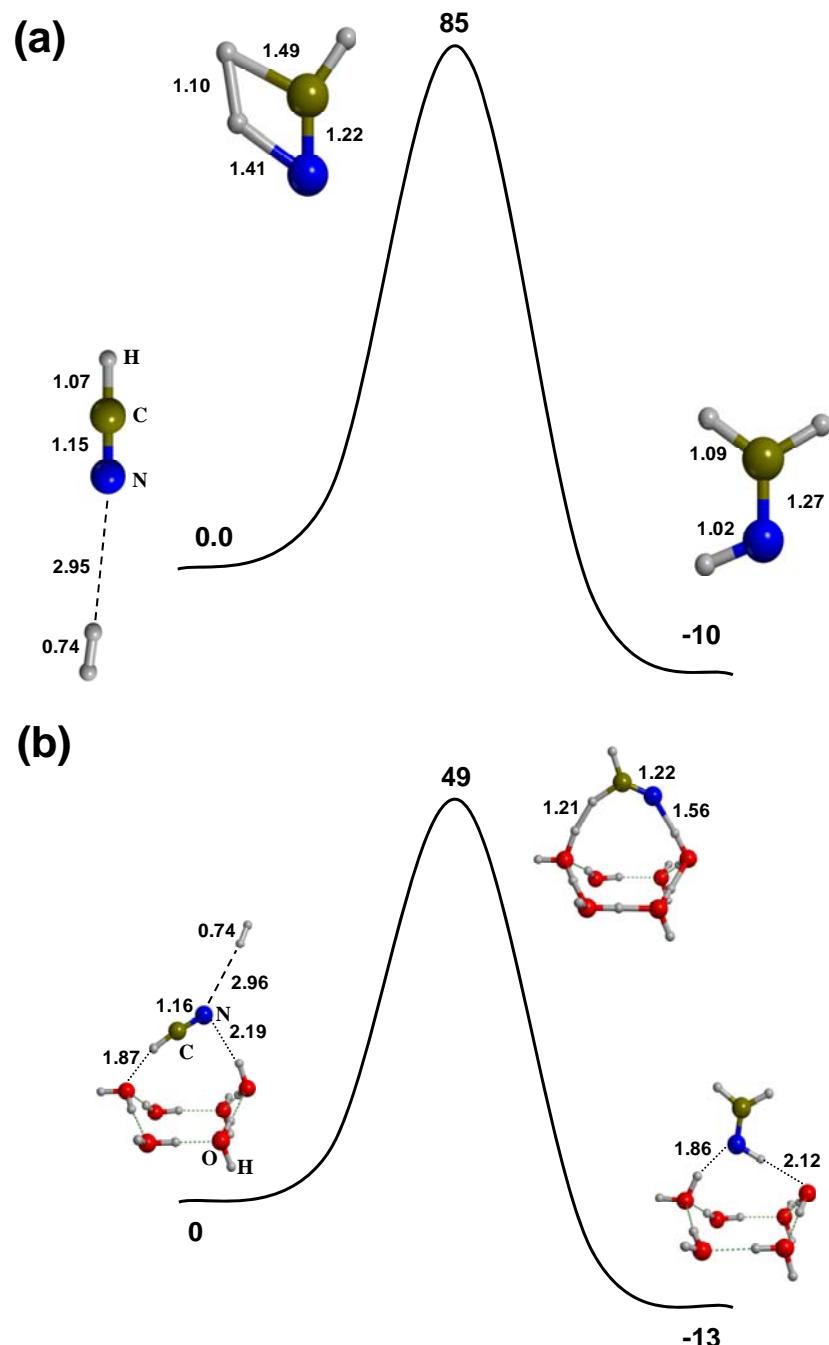
**Figure S2.** B3LYP/6-31+G(d,p) gas-phase ZPE-corrected profile of  $\text{H}_2\text{C=O} + \text{NH}_3 \rightarrow \text{HN}=\text{CH}_2 + \text{H}_2\text{O}$  following the Strecker mechanism. Relative ZPE-corrected energies in kcal mol<sup>-1</sup>, bond distances in Å.



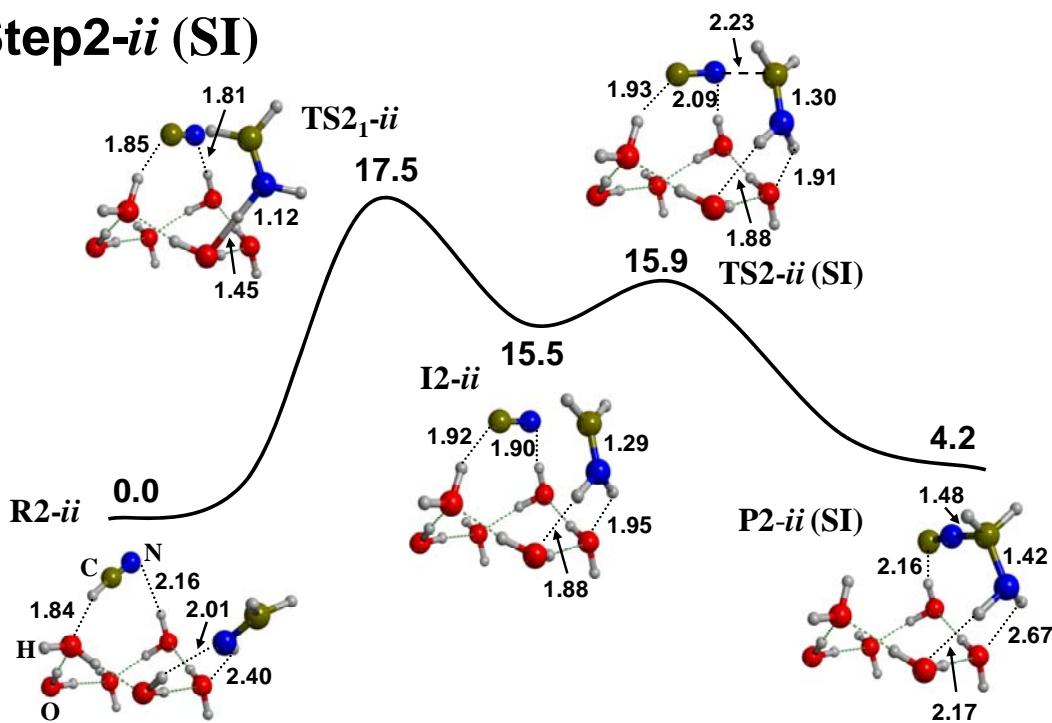
**Figure S3.** B3LYP/6-31+G(d,p) gas-phase ZPE-corrected profile of  $\text{HCN} + \text{NH}=\text{CH}_2 \rightarrow \text{NH}_2\text{CH}_2\text{CN}$  following the Strecker mechanism. Relative ZPE-corrected energies in kcal mol<sup>-1</sup>, bond distances in Å.



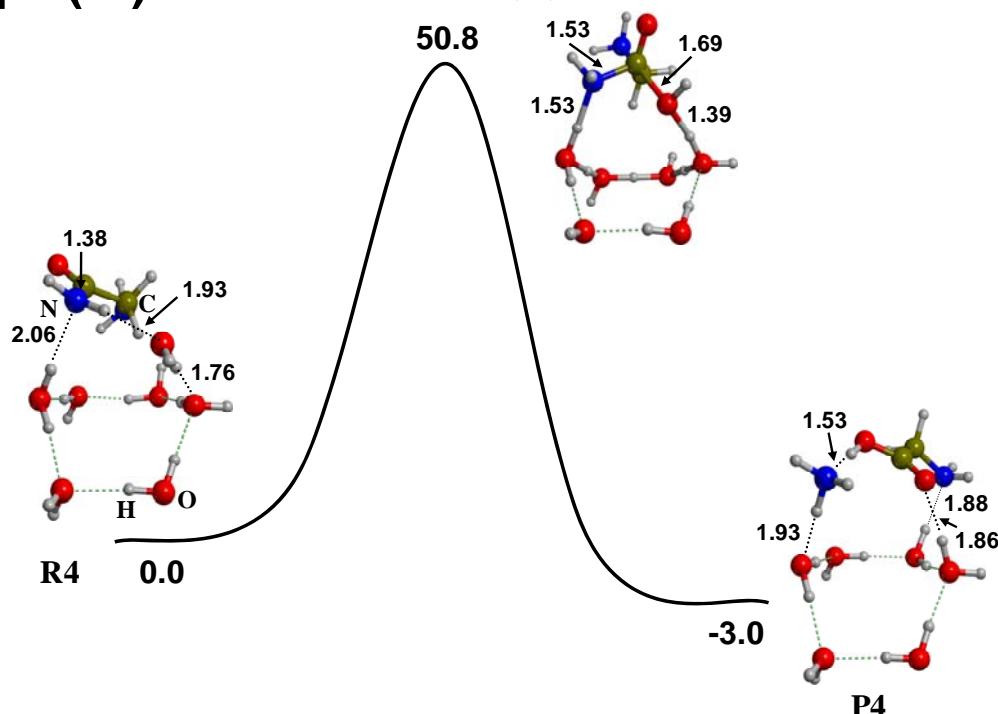
**Figure S4.** B3LYP/6-31+G(d,p) gas-phase ZPE-corrected profile of the hydrolysis of  $\text{NH}_2\text{CH}_2\text{CN}$  to form glycine following the Strecker mechanism. Relative ZPE-corrected energies in  $\text{kcal mol}^{-1}$ , bond distances in  $\text{\AA}$ .



**Figure S5.** B3LYP/6-31+G(d,p) ZPE-corrected profile for the  $\text{H}_2 + \text{HCN} \rightarrow \text{NH}=\text{CH}_2$  reaction: (a) in gas-phase, (b) at the  $\text{H}_2\text{O}$ -ice surface. Relative ZPE-corrected energies in  $\text{kcal mol}^{-1}$ , bond distances in  $\text{\AA}$ .

**Step2-ii (SI)**

**Figure S6.** B3LYP/6-31+G(d,p) ZPE-corrected profile for the  $\text{NH}=\text{CH}_2 + \text{HCN} \rightarrow \text{NHCH}_2\text{NC}$  reaction at the  $\text{H}_2\text{O}$ -ice surface. Relative ZPE-corrected energies in kcal mol<sup>-1</sup>, bond distances in Å.

**Step4 (SI)**

**Figure S7.** B3LYP/6-31+G(d,p) ZPE-corrected profile for the  $\text{NH}_2\text{CH}_2\text{CONH}_2 + \text{H}_2\text{O} \rightarrow \text{NH}_2\text{CH}_2\text{COOH} + \text{NH}_3$  assisted by the  $\text{H}_2\text{O}$ -ice surface. Relative ZPE-corrected energies in kcal mol<sup>-1</sup>, bond distances in Å.