

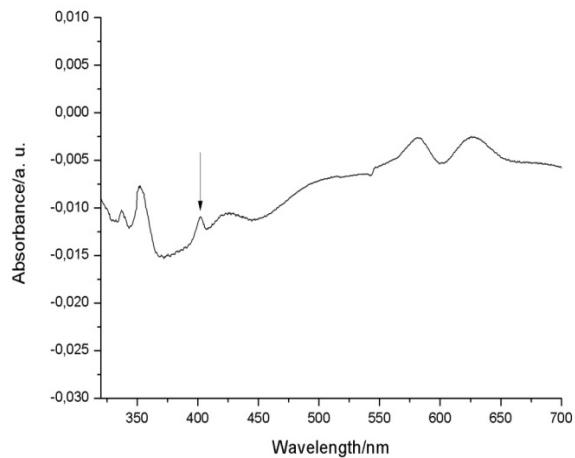
Supporting Information

Optical basicity of ionic liquids

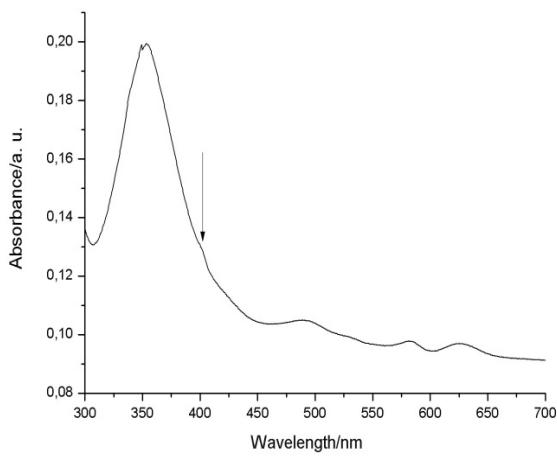
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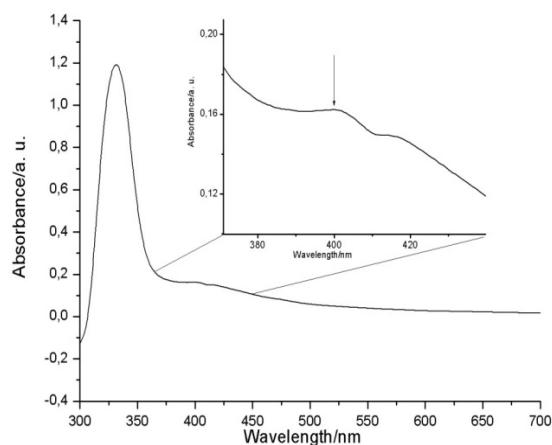
All shown spectra are difference spectra between the neat and Mn(NTf₂) doped ionic liquid.



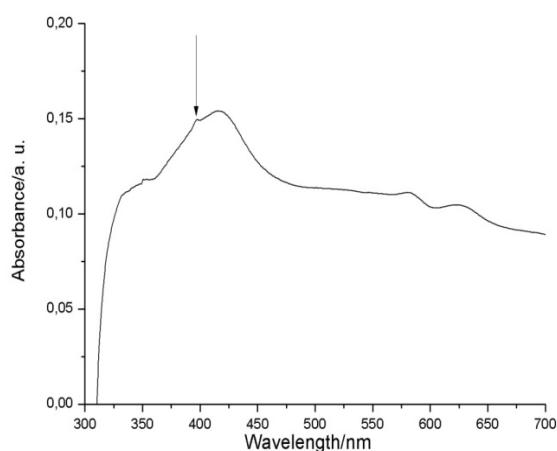
ESI-Fig. 1 UV/Vis spectrum of C₄C₁mimNTf₂ doped with Mn(NTf₂)₂.



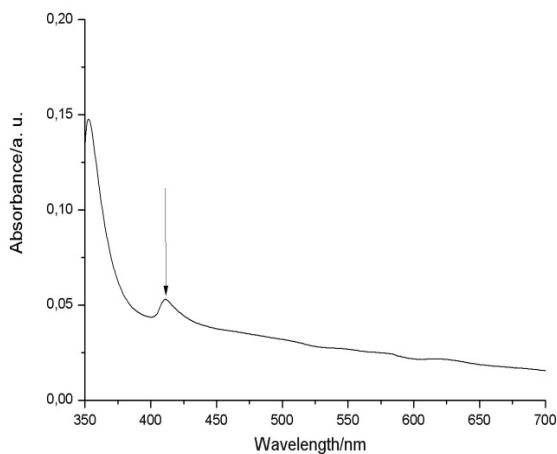
ESI-Fig. 2 UV/Vis spectrum of $\text{C}_4\text{mimNTf}_2$ doped with $\text{Mn}(\text{NTf}_2)_2$.



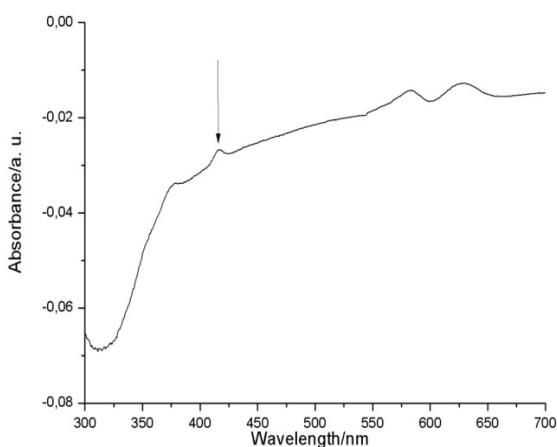
ESI-Fig. 3 UV/Vis spectrum of $\text{C}_4\text{mimClO}_4$ doped with $\text{Mn}(\text{NTf}_2)_2$.



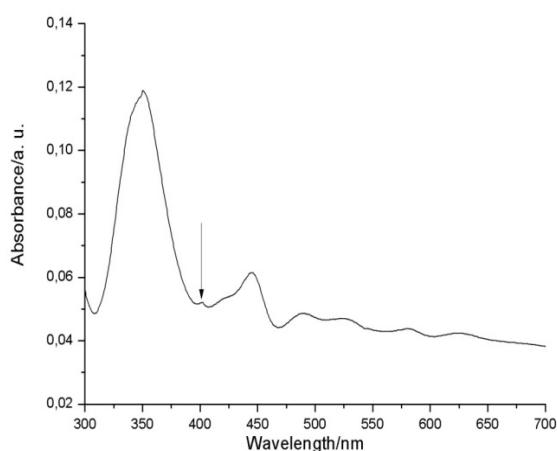
ESI-Fig. 4 UV/Vis spectrum of $\text{C}_6\text{mpyridBF}_4$ doped with $\text{Mn}(\text{NTf}_2)_2$.



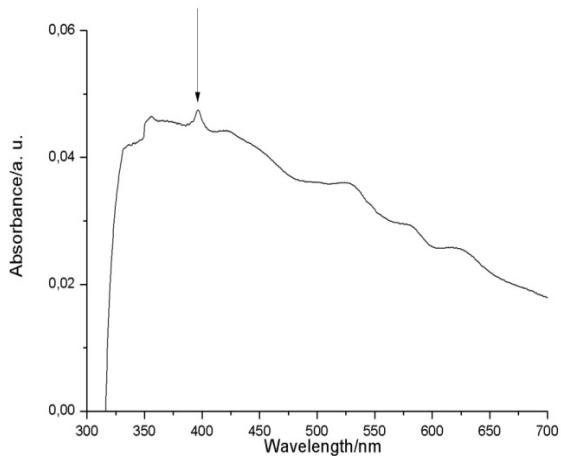
ESI-Fig. 5 UV/Vis spectrum of C_4mimNO_3 doped with $\text{Mn}(\text{NTf}_2)_2$.



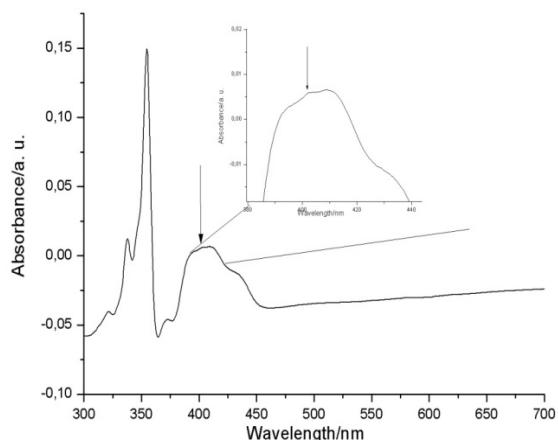
ESI-Fig. 6 UV/Vis spectrum of C_4mimTFA doped with $\text{Mn}(\text{NTf}_2)_2$.



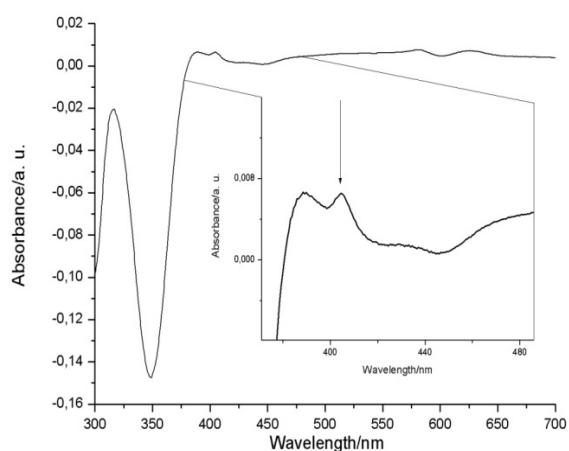
ESI-Fig. 7 UV/Vis spectrum of $\text{C}_4\text{mpyrNTf}_2$ doped with $\text{Mn}(\text{NTf}_2)_2$.



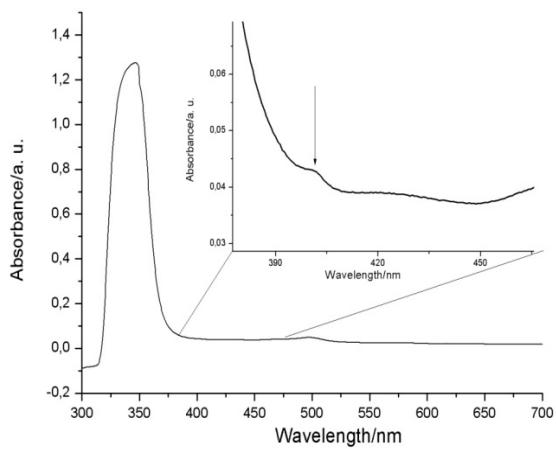
ESI-Fig. 8 UV/Vis spectrum of $\text{C}_4\text{pyrid}\text{BF}_4$ doped with $\text{Mn}(\text{NTf}_2)_2$.



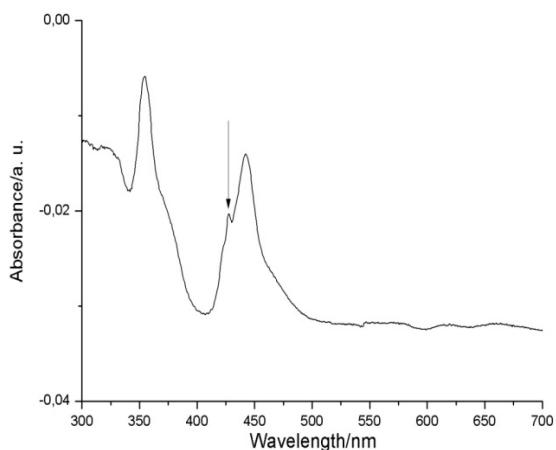
ESI-Fig. 9 UV/Vis spectrum of $\text{C}_{10}\text{mim}\text{NTf}_2$ doped with $\text{Mn}(\text{NTf}_2)_2$.



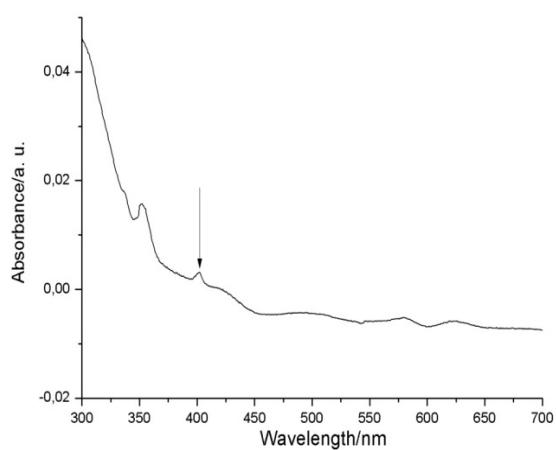
ESI-Fig. 10 UV/Vis spectrum of $\text{CNC}_2\text{emim}\text{NTf}_2$ doped with $\text{Mn}(\text{NTf}_2)_2$.



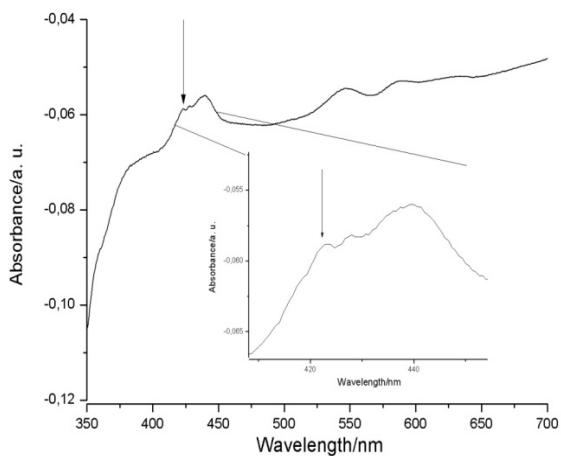
ESI-Fig. 11 UV/Vis spectrum of C₂mimNTf₂ doped with Mn(NTf₂)₂.



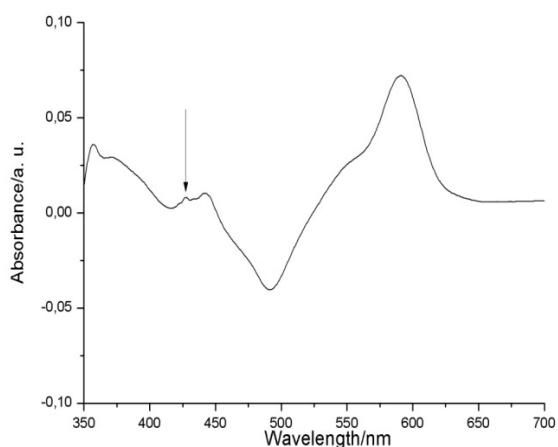
ESI-Fig. 12 UV/Vis spectrum of C₂mimEtSO₄ doped with Mn(NTf₂)₂.



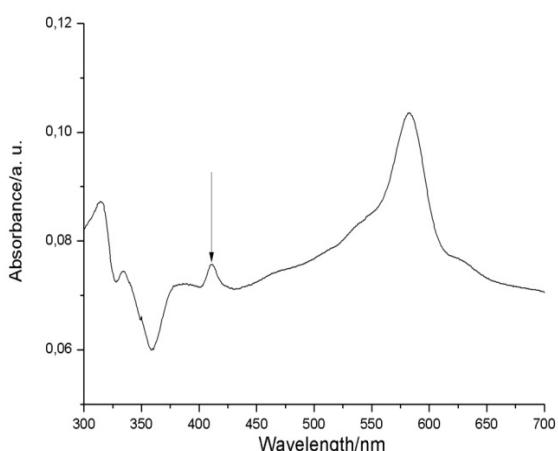
ESI-Fig. 13 UV/Vis spectrum of C₂mimMeSO₃ doped with Mn(NTf₂)₂.



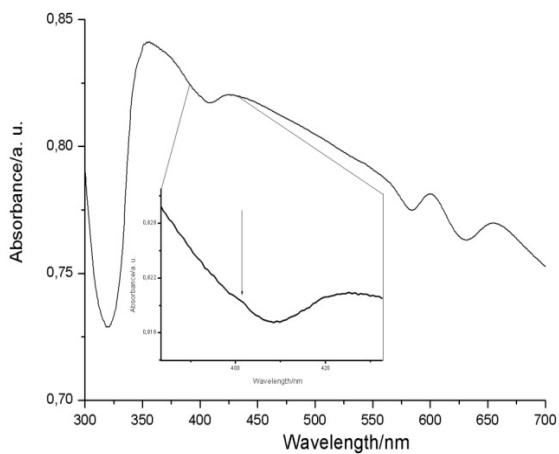
ESI-Fig. 14 UV/Vis spectrum of C₂mimOTos doped with Mn(NTf₂)₂.



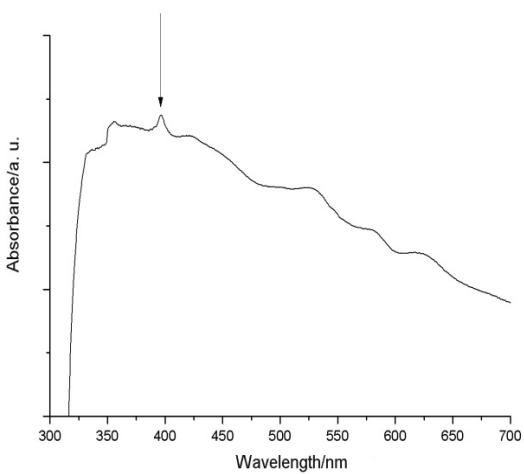
ESI-Fig. 15 UV/Vis spectrum of C₄mpyridEtSO₄ doped with Mn(NTf₂)₂.



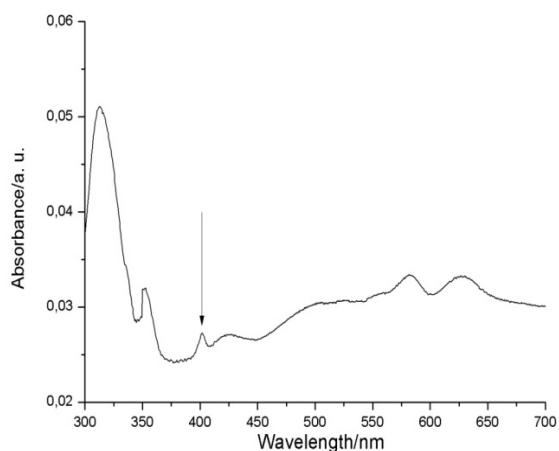
ESI-Fig. 16 UV/Vis spectrum of Et₃NvalCN NTf₂ doped with Mn(NTf₂)₂.



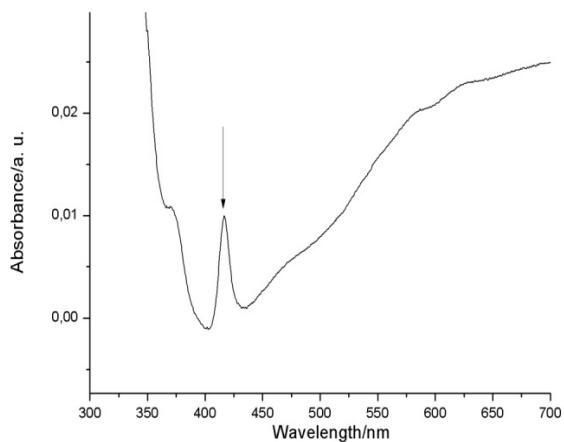
ESI-Fig. 17 UV/Vis spectrum of $\text{C}_6\text{mimNTf}_2$ doped with $\text{Mn}(\text{NTf}_2)_2$.



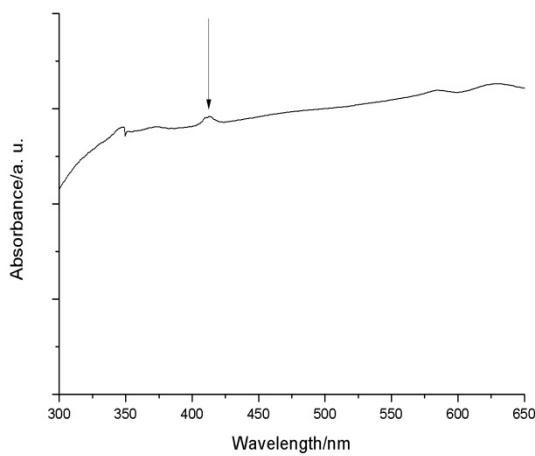
ESI-Fig. 18 UV/Vis spectrum of $\text{C}_4\text{mpyridBF}_4$ doped with $\text{Mn}(\text{NTf}_2)_2$.



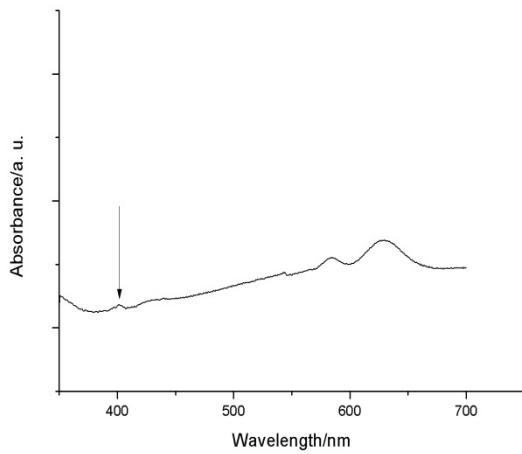
ESI-Fig. 19 UV/Vis spectrum of $\text{C}_6\text{mpyrNTf}_2$ doped with $\text{Mn}(\text{NTf}_2)_2$.



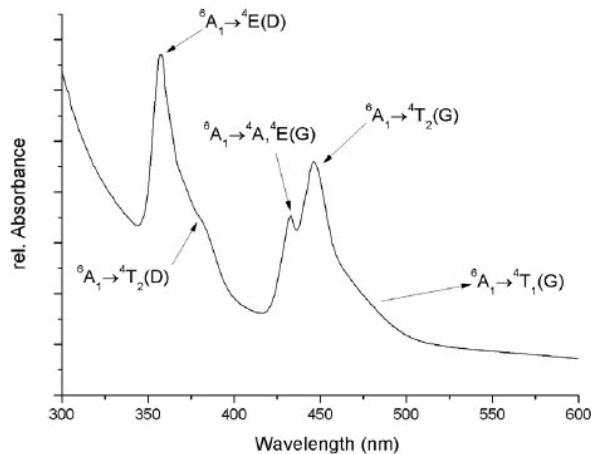
ESI-Fig. 20 UV/Vis spectrum of C₄mimDCA doped with Mn(NTf₂)₂.



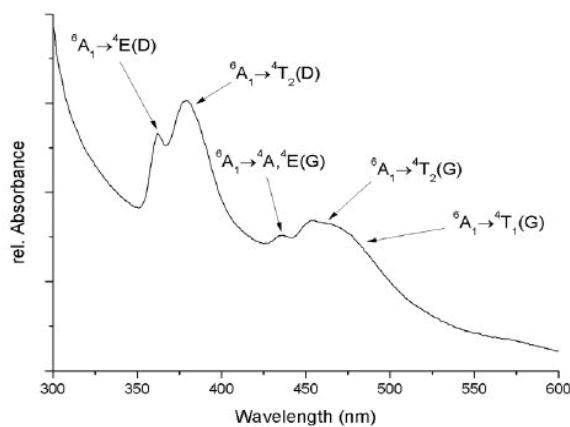
ESI-Fig. 21 UV/Vis spectrum of Acetonitrile doped with Mn(NTf₂)₂.



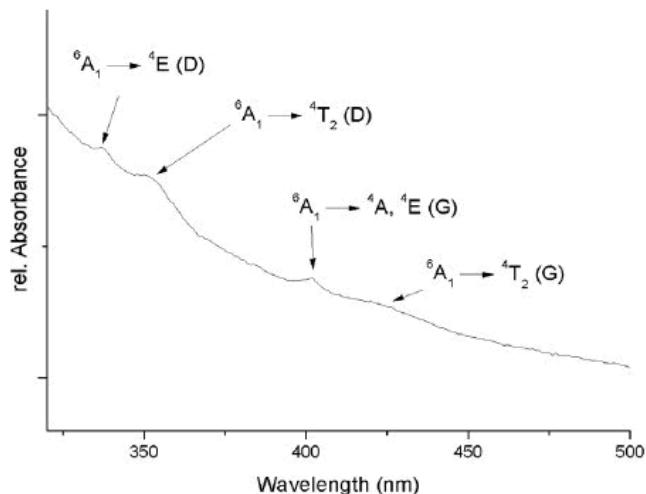
ESI-Fig. 22 UV/Vis spectrum of deionized water doped with Mn(NTf₂)₂.



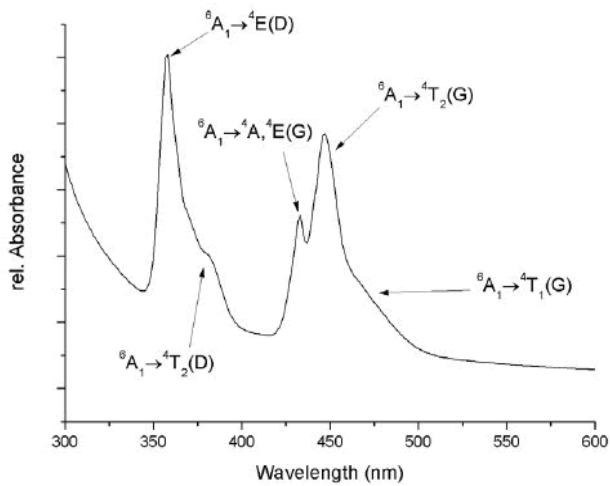
ESI-Fig. 23 UV/Vis spectrum of $[C_2mim]_2[MnCl_4]$.



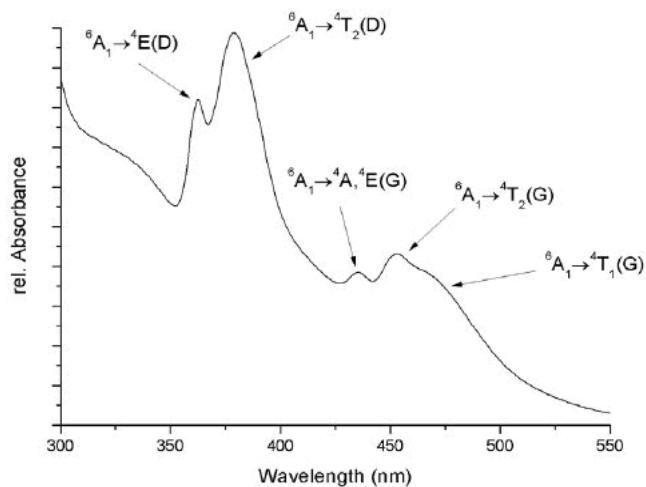
ESI-Fig. 24 UV/Vis spectrum of $[C_2mim]_2[MnBr_4]$.



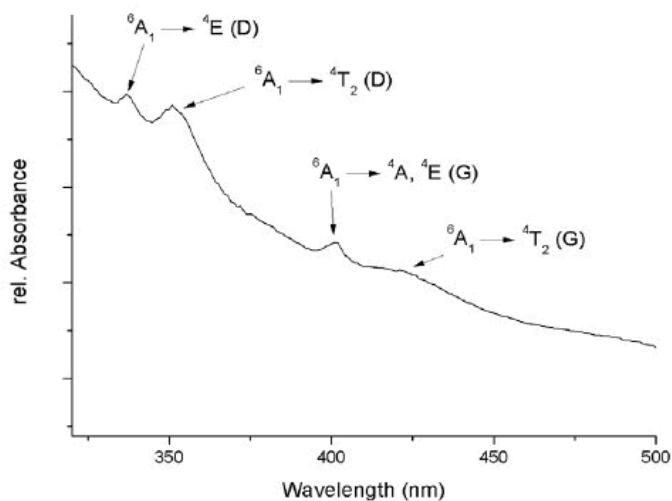
ESI-Fig. 25 UV/Vis spectrum of $[C_2mim][Mn(Tf_2N)_3]$.



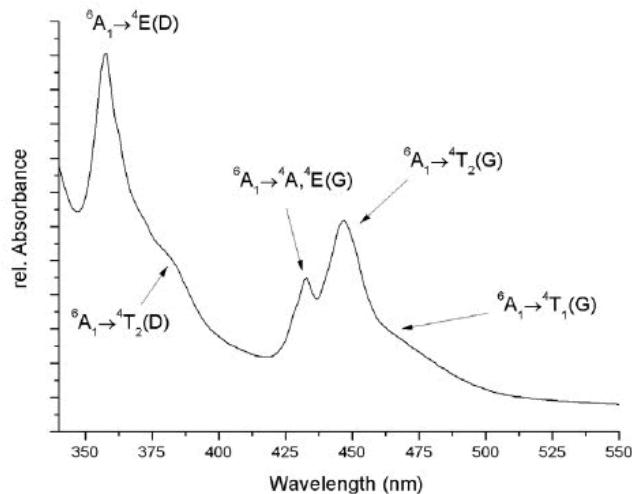
ESI-Fig. 26 UV/Vis spectrum of $[C_3\text{mim}]_2[\text{MnCl}_4]$.



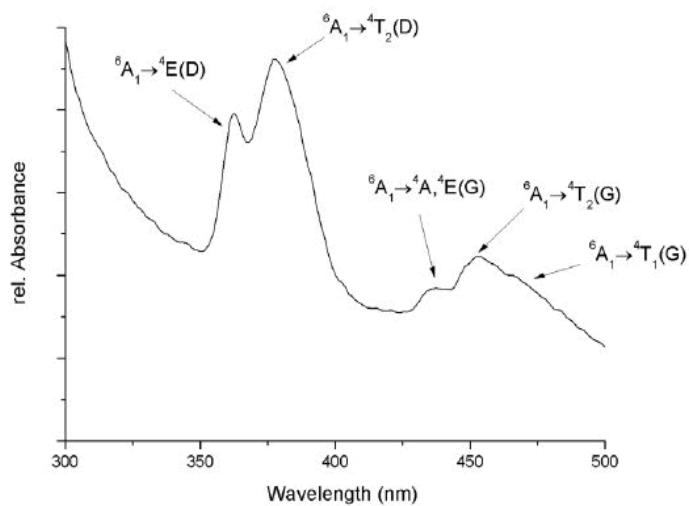
ESI-Fig. 27 UV/Vis spectrum of $[C_3\text{mim}]_2[\text{MnBr}_4]$.



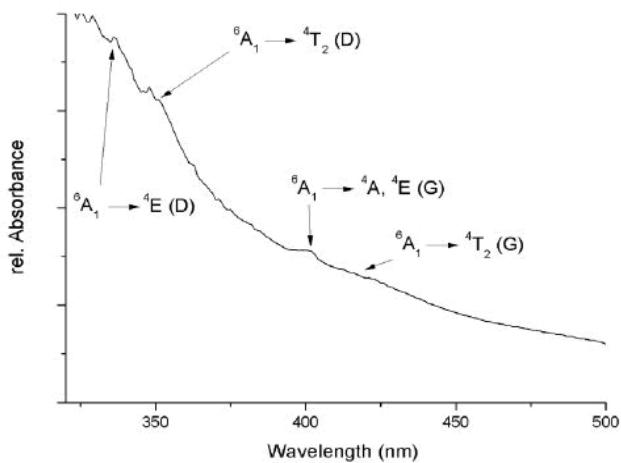
ESI-Fig. 28 UV/Vis spectrum of $[C_3\text{mim}][\text{Mn}(\text{Tf}_2\text{N})_3]$.



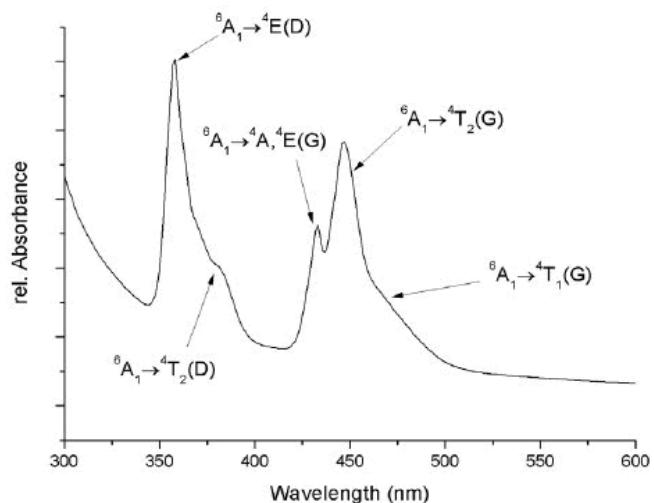
ESI-Fig. 29 UV/Vis spectrum of $[C_4mim]_2[MnCl_4]$.



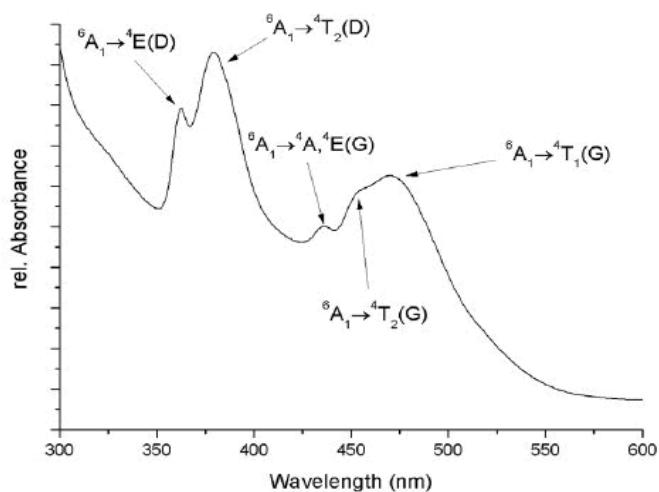
ESI-Fig. 30 UV/Vis spectrum of $[C_4mim]_2[MnBr_4]$.



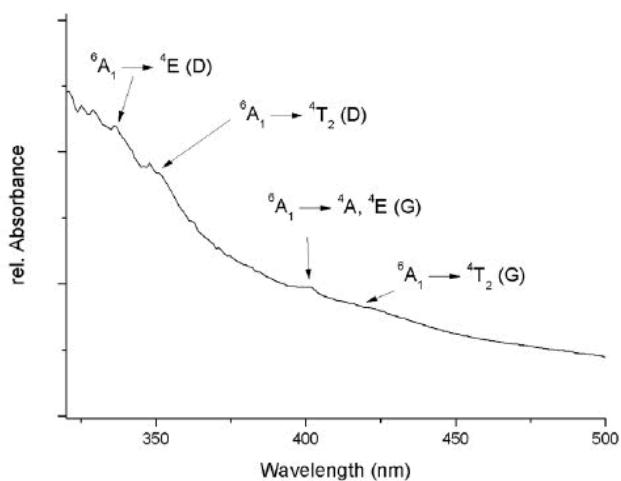
ESI-Fig. 31 UV/Vis spectrum of $[C_4mim][Mn(Tf_2N)_3]$.



ESI-Fig. 32 UV/Vis spectrum of $[C_6\text{mim}]_2[\text{MnCl}_4]$.



ESI-Fig. 33 UV/Vis spectrum of $[C_6\text{mim}]_2[\text{MnBr}_4]$.



ESI-Fig. 34 UV/Vis spectrum of $[C_6\text{mim}][\text{Mn}(\text{Tf}_2\text{N})_3]$.

ESI-Table 1. List of the $^6A_1 \rightarrow ^4A, ^4E(G)$ transition bands and the nephelauxetic parameters h calculated from literature data. The Mn²⁺ site symmetry is assigned to tet (tetrahedral) and oct (octahedral).

compound	$^6A_1 \rightarrow ^4A, ^4E(G)$ transition band/cm ⁻¹	coordination site	nephelauxetic parameter h
MnF ₂ ¹	24937	oct	-0.04
MnF ₂ @ 4 K ¹	25340	oct	-0.27
MnCl ₂ ²	23590	oct	0.74
MnBr ₂ ²	23084	oct	1.03
MnBr ₂ in propanediol-1,2-carbonate ³	23041	tet	1.05
MnBr ₂ in acetonitrile ³	23095	tet	1.02
MnBr ₂ in Bu ₄ NBr ⁴	22992	tet	1.08
[Bu ₄ N] ₂ [MnBr ₄] in hexachlorobutadiene ⁵	23810	tet	0.61
[Bu ₄ N] ₂ [MnCl ₄] in Nujol ⁵	23310	tet	0.90
[Et ₄ N] ₂ [MnCl ₄] @ 77 K ⁶	23400	tet	0.85
[Et ₄ N] ₂ [MnBr ₄] @ 77 K ⁶	23050	tet	1.05
[Me ₄ N] ₂ [MnCl ₄] @ 77 K ⁶	23290	tet	0.91
[Me ₄ N] ₂ [MnBr ₄] @ 77 K ⁶	23000	tet	1.08
[Bu ₄ N] ₂ [MnCl ₄] @ 77 K ⁶	23470	tet	0.81
[Bu ₄ N] ₂ [MnBr ₄] @ 77 K ⁶	23080	tet	1.03
5% MnCl ₂ in KCl ⁷	23880	oct	0.57
Mn(H ₂ O) ₆ ^{2+ 8}	25150	oct	-0.16
Mn(CH ₃ CN) ₆ ^{2+ 8}	24450	oct	0.24

¹ Gmelin *Handbook, Mn*, Vol. C 10, 6, **1982**

² V. Gutmann, K. Fenkart, *Mh. Chem.* **1967**, *Bd. 1*, 277

³ R. Pappalardo, *J. Chem. Phys.* **1960**, 613

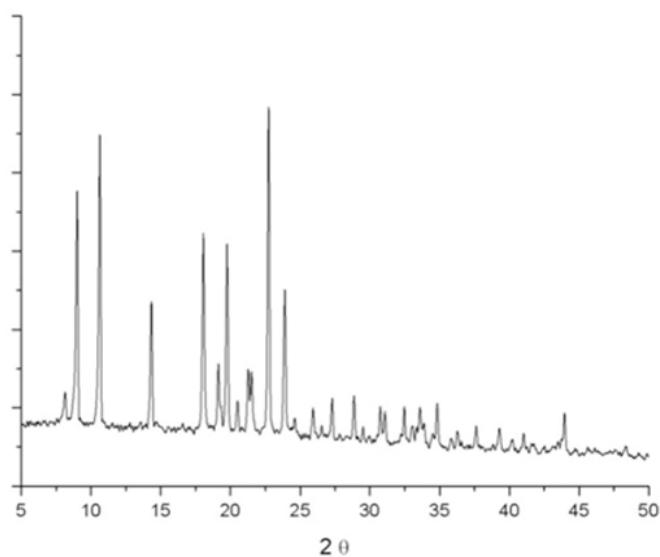
⁴ N. Islam, *App. Spectr.* **1975**, 29, 266

⁵ A. Cotton, D. Goodgame, M. Goodgame, *JACS* **1962**, 84, 167

⁶ D. Oelkrug, A. Wölpl, *Ber. d. Bunsengesellschaft* **1972**, *Bd. 76*, 680

⁷ A. Mehra, P. Venkateswarlu, *J. Chem. Phys.* **1966**, 45, 338

⁸ S. Chan, B. Fung, H. Lütje, *J. Chem. Phys.* **1967**, 47, 2121



ESI-Fig. 33 PXRD of as-synthesized $\text{Mn}(\text{NTf}_2)_2$.