

## **Artificial, Switchable K<sup>+</sup>-gated Ion Channels based on Flow-Through Titania-Nanotube Arrays**

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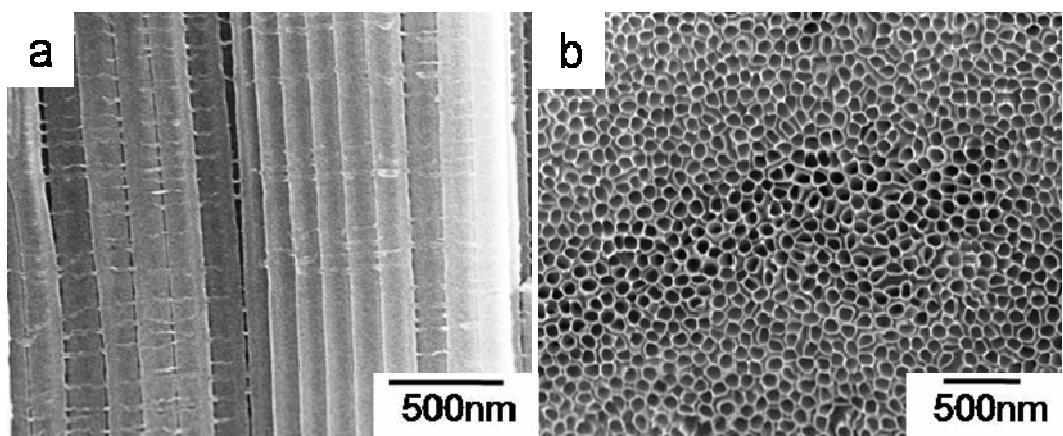


Fig. S1: SEM side view of free-standing TNT arrays anodized in ethylene glycol: (a) cross-sectional view, and (b) top view.

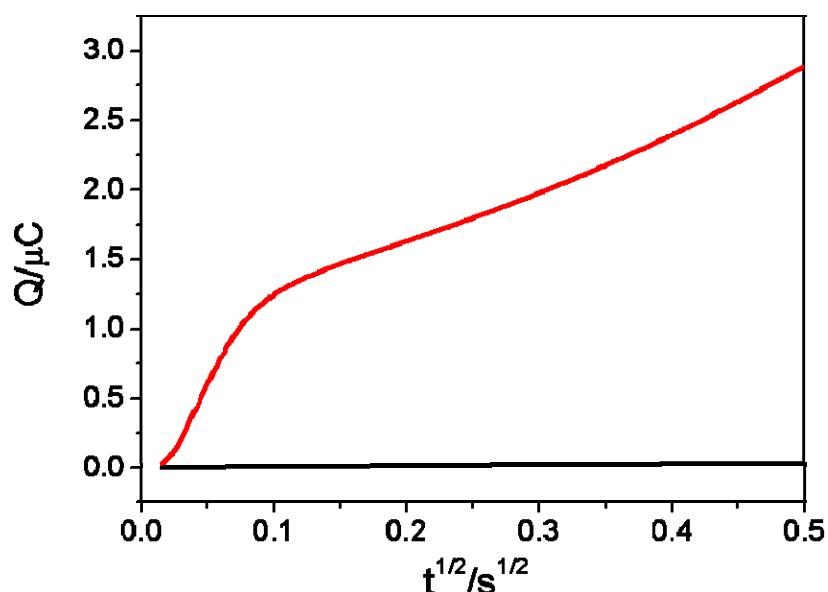


Fig. S2: Chronocoulometry of G-rich DNA modified TNT-Au NPs electrode in Tris-HCl (black) and in 50  $\mu$ M Ru(NH<sub>3</sub>)<sub>6</sub>Cl<sub>3</sub>(red).

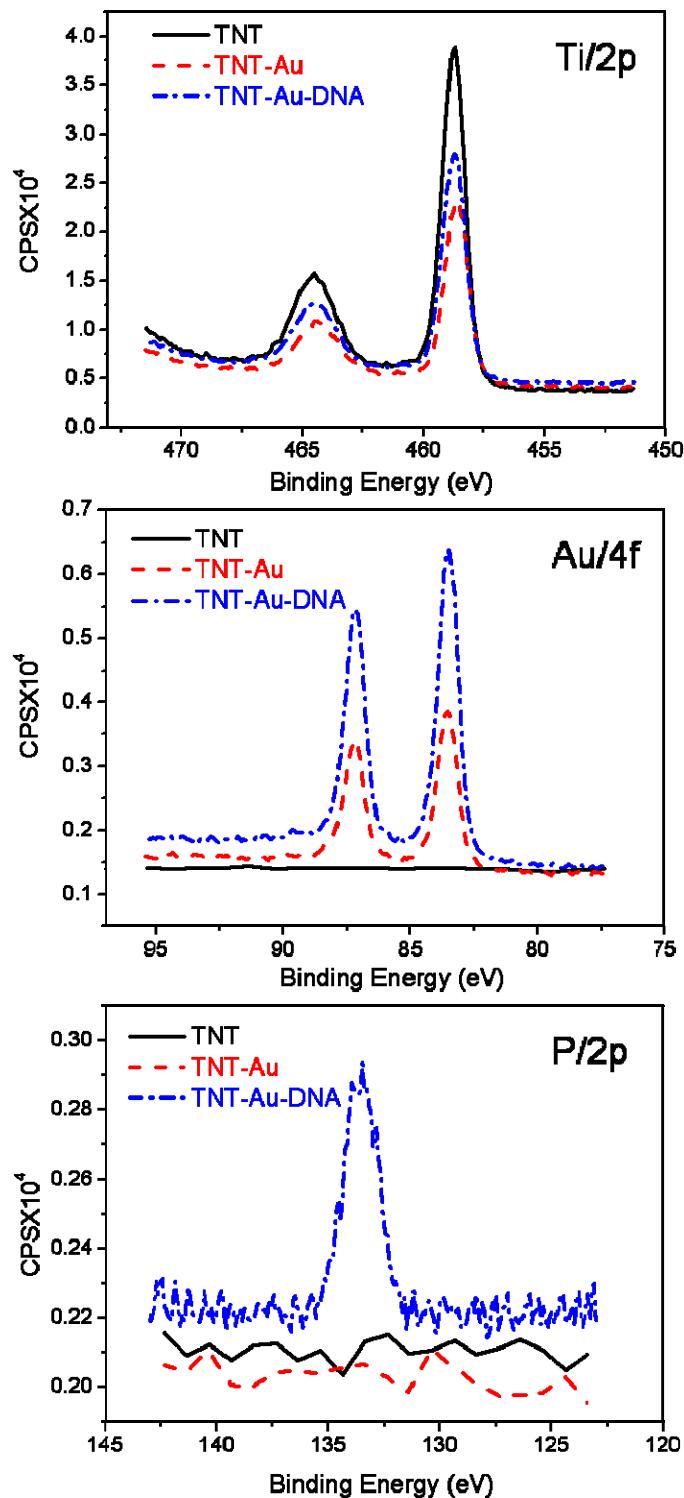


Fig. S3: XPS characterization of Au loading and DNA modification on TNT. These data confirm the success of Au NPs loading ( $Au/4f$ ) in the TNT and DNA modification ( $P/2p$ ) on the TNT-Au electrode.

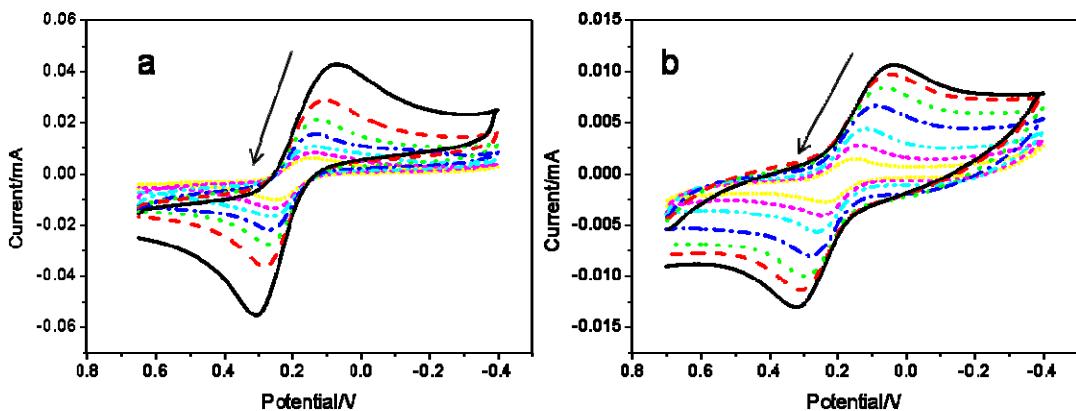


Fig. S4: Cyclic voltammetry of the TNT-Au NPs electrode (a) and the TNT-Au NPs electrode modified with G-rich DNA (b) in 5mM  $\text{Na}_4\text{Fe}(\text{CN})_6$  with 0.2M NaCl as supporting electrolyte. Scan rate: 500mV/s, 200 mV/s, 100 mV/s, 50 mV/s, 20 mV/s, 10 mV/s, 5mV/s.

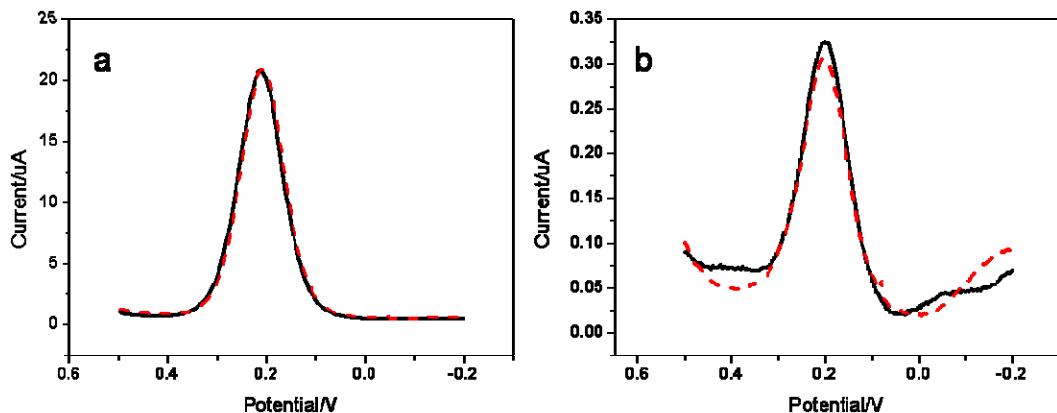


Fig. S5: DPV of the control groups in solution A ( $\text{Na}^+$ , black solid) and solution B ( $\text{K}^+$ , red dash): (a) TNT-Au NPs electrode without DNA modified; (b) TNT-Au NPs eletrode modified with random DNA. Baseline had been deducted.

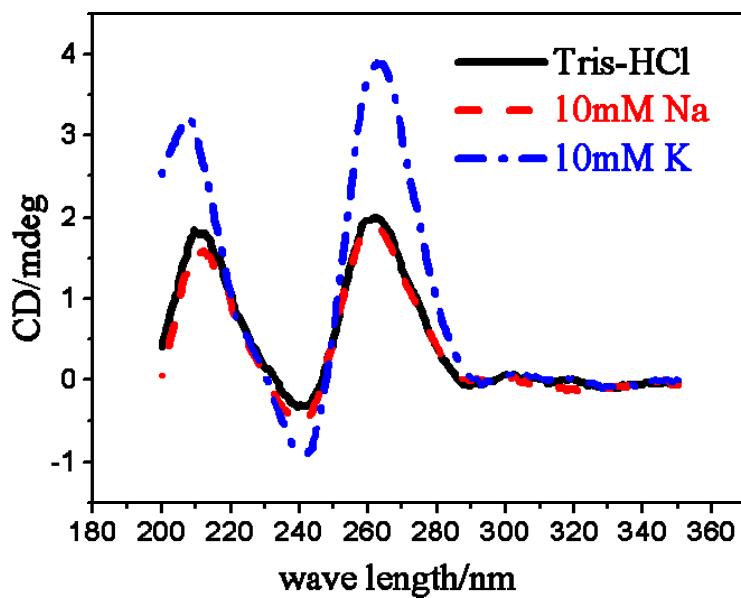


Fig. S6: CD spectra recorded over 200-350 nm on 2  $\mu$ M G-rich DNA in 10 mM Tris-HCl (pH 7.4) (black), 10 mM NaCl in 10 mM Tris-HCl (red) and 10 mM KCl in 10mM Tris-HCl (blue). All these curves had a positive ellipticity maximum at 264 nm and a negative minimum at 240 nm, which belonged to the DNA quadruplex.<sup>1</sup>

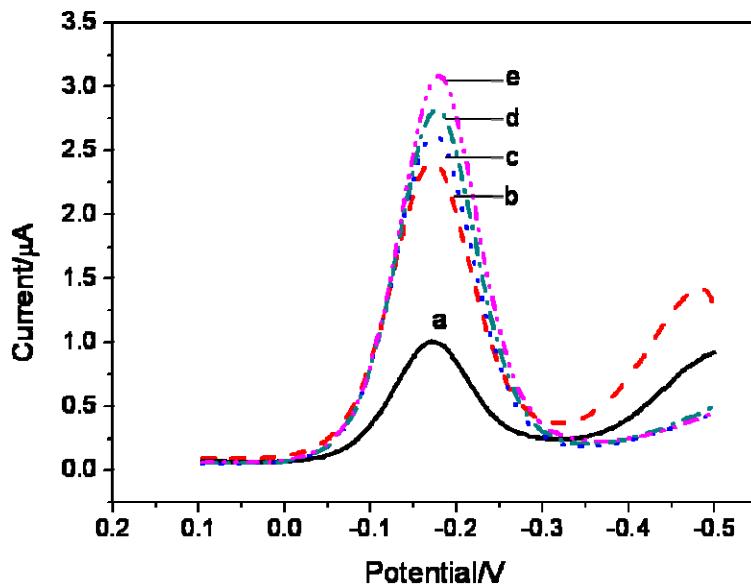


Fig. S7: DPV of the TNT-Au NPs electrode in 50  $\mu$ M  $\text{Ru}(\text{NH}_3)_6\text{Cl}_3$  with 200 mM NaCl (a); 50  $\mu$ M  $\text{Ru}(\text{NH}_3)_6\text{Cl}_3$  with 10mM KCl and different concentration of NaCl (b-e); 200 mM (b), 300 mM (c), 400 mM (d), 500 mM (e).

## References :

1. M. Lu, Q. Guo, N. R. Kallenbach, *Biochemistry*, 1992, **31**, 2455-2459.