## Theoretical Study of the Temperature Dependence of Dynamic Effects in Thymidylate Synthase

## **Electronic Supplementary Information**

Natalia Kanaan, Maite Roca, Iñaki Tuñón, Sergio Martí,<sup>\*</sup> Vicent Moliner.<sup>\*</sup>

Supplementary Material (ESI) for PCCP This journal is © the Owner Societies 2010



Figure S1. Time evolution of key distances and angle defining the chemical reaction, for reactive trajectories at 278 K. Distance C8-C7 (purple line), C8-Ht (blue line), Ht-C7 (magenta line), C6-S (green line) and the C8-Ht-C7 angle (brown line).



Figure S2. Time evolution of key distances established with residues of the active site for reactive trajectories at 278 K: S-Wat40 distance (dark blue line), S-Arg166 distance (magenta line), O3P-Arg166 distance (red line), O2-Asg169 distance (light blue line) and C6-S distance (green line).



Figure S3. Time evolution of Mulliken charges of key atoms averaged for reactive trajectories at 278 K: C8 (dark blue line), Ht (magenta line), C7 (brown line), C6 (light blue line), S (black line), O4 (green line) and C4 (red line).



Figure S4. Time evolution of key distances and angle defining the chemical reaction, for reactive trajectories at 303 K. Distance C8-C7 (purple line), C8-Ht (blue line), Ht-C7 (magenta line), C6-S (green line) and the C8-Ht-C7 angle (brown line).



Figure S5. Time evolution of key distances established with residues of the active site for reactive trajectories at 303 K: S-Wat40 distance (dark blue line), S-Arg166 distance (magenta line), O3P-Arg166 distance (red line), O2-Asg169 distance (light blue line) and C6-S distance (green line).



Figure S6. Time evolution of Mulliken charges of key atoms averaged for reactive trajectories at 303 K: C8 (dark blue line), Ht (magenta line), C7 (brown line), C6 (light blue line), S (black line), O4 (green line) and C4 (red line).



Figure S7. Time evolution of key distances and angle defining the chemical reaction, for reactive trajectories at 313 K. Distance C8-C7 (purple line), C8-Ht (blue line), Ht-C7 (magenta line), C6-S (green line) and the C8-Ht-C7 angle (brown line).



Figure S8. Time evolution of key distances established with residues of the active site for reactive trajectories at 313 K: S-Wat40 distance (dark blue line), S-Arg166 distance (magenta line), O3P-Arg166 distance (red line), O2-Asg169 distance (light blue line) and C6-S distance (green line).



Figure S9. Time evolution of Mulliken charges of key atoms averaged for reactive trajectories at 313 K: C8 (dark blue line), Ht (magenta line), C7 (brown line), C6 (light blue line), S (black line), O4 (green line) and C4 (red line).



Figure S10. Contributions of individual amino acid residues to interaction electrostatic energies (in kcal/mol) computed at AM1/MM level from averaged representative structures of the transition state. The most important interactions are assigned.



 $Figure \ S11. \ Contributions \ of \ individual \ amino \ acid \ residues \ to \ interaction \ electrostatic \ energies \ (in \ kcal/mol) \ computed \ at \ B3LYP/6-31g(d,p)/MM \ level \ from \ averaged \ representative \ structures \ of \ the \ transition \ state. \ The \ most \ important \ interactions \ are \ assigned.$ 

Table S1. Weights of the coordinates (together with their standard deviations) of the key atoms in the transition vectors obtained using two different sizes of the Hessians (see text). Values have been averaged over 12 different transition structures. Cys146, Folate and dUMP accounts for the rest of the atoms of these residues. Results are reported for 278 K.

	Small Hessian	Large Hessian
C8	$0.218 \pm 0.017$	$0.258 \pm 0.016$
Ht	$0.436 \pm 0.005$	$0.440 \pm 0.005$
C7	$0.246 \pm 0.018$	$0.214 \pm 0.016$
N9	$0.009 \pm 0.005$	$0.007 \pm 0.03$
C6	$0.0134 \pm 0.0014$	$0.0079 \pm 0.0010$
S	$0.0089 \pm 0.0011$	$0.0037 \pm 0.0007$
C5	$0.0168 \pm 0.0012$	$0.0147 \pm 0.0013$
H71	$0.0078 \pm 0.0008$	$0.0091 \pm 0.0010$
H72	$0.0133 \pm 0.0018$	$0.0135 \pm 0.0019$
Cys146	$0.0017 \pm 0.0021$	$0.0011 \pm 0.0013$
Folate	$0.029 \pm 0.007$	$0.027 \pm 0.006$
dUMP	$0.0095 \pm 0.0010$	$0.0111 \pm 0.0009$
Arg166	-	$0.00053 \pm 0.00014$
WAT40	-	$0.00054 \pm 0.00007$

Table S2. Weights of the coordinates (together with their standard deviations) of the key atoms in the transition vectors obtained using two different sizes of the Hessians (see text). Values have been averaged over 12 different transition structures. Cys146, Folate and dUMP accounts for the rest of the atoms of these residues. Results are reported for 303 K.

	Small Hessian	Large Hessian
C8	$0.217 \pm 0.017$	$0.260 \pm 0.017$
Ht	$0.434 \pm 0.006$	$0.440 \pm 0.004$
C7	$0.245 \pm 0.015$	$0.210 \pm 0.014$
N9	$0.012 \pm 0.007$	$0.008 \pm 0.04$
C6	$0.0140 \pm 0.0013$	$0.0085 \pm 0.0014$
S	$0.0095 \pm 0.0012$	$0.0040 \pm 0.0010$
C5	$0.0163 \pm 0.0010$	$0.0138 \pm 0.0010$
H71	$0.0072 \pm 0.0010$	$0.0089 \pm 0.0011$
H72	$0.0133 \pm 0.0014$	$0.0130 \pm 0.0012$
Cys146	$0.002 \pm 0.003$	$0.0014 \pm 0.0018$
Folate	$0.033 \pm 0.009$	$0.029 \pm 0.006$
dUMP	$0.0091 \pm 0.0011$	$0.0108 \pm 0.0011$
Arg166	-	$0.00056 \pm 0.00023$
WAT40	-	$0.00051 \pm 0.00006$

Table S3. Weights of the coordinates (together with their standard deviations) of the key atoms in the transition vectors obtained using two different sizes of the Hessians (see text). Values have been averaged over 12 different transition structures. Cys146, Folate and dUMP accounts for the rest of the atoms of these residues. Results are reported for 313 K.

	Small Hessian	Large Hessian
C8	$0.240 \pm 0.019$	$0.276 \pm 0.018$
Ht	$0.442 \pm 0.010$	$0.444 \pm 0.009$
C7	$0.220 \pm 0.016$	$0.192 \pm 0.014$
N9	$0.010 \pm 0.005$	$0.008 \pm 0.004$
C6	$0.0118 \pm 0.0008$	$0.0072 \pm 0.0007$
S	$0.0076 \pm 0.0008$	$0.0030 \pm 0.0005$
C5	$0.01424 \pm 0.0013$	$0.0127 \pm 0.0013$
H71	$0.0071 \pm 0.0018$	$0.0086 \pm 0.0019$
H72	$0.0148 \pm 0.0020$	$0.0147 \pm 0.0012$
Cys146	$0.0021 \pm 0.0015$	$0.0016 \pm 0.0017$
Folate	$0.031 \pm 0.009$	$0.029 \pm 0.008$
dUMP	$0.0097 \pm 0.0009$	$0.0113 \pm 0.0009$
Arg166	-	$0.00045 \pm 0.00014$
WAT40	-	$0.00052 \pm 0.00008$