

# Theoretical investigation of the conformation, acidity, basicity and hydrogen bond ability of halogenated ethers

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TABLE 1S Selected NBO parameters in isolated halogenated ethers ( $\sigma^*$  occupancies and s-orbital character at C carbon in CH bond). Results of MP2/6-311++G(d,p) calculations.

	<b>I</b>	<b>II</b>	<b>III</b>	<b>IV</b>	<b>V</b>
$\sigma^*(C_1H_6)$	0.008	0.021	0.020	0.037	0.035
$\sigma^*(C_3H_7)$		0.014	-	0.012	-
$\sigma^*(C_3H_9)$	0.008	0.007	0.020	0.007	0.036
%s C <sub>1</sub> (H <sub>4</sub> )	25.92	27.38	27.88	-	-
%s C <sub>1</sub> (H <sub>6</sub> )	25.64	27.02	27.08	27.76	28.89
%s C <sub>3</sub> (H <sub>7</sub> )	25.92	25.99	-	26.66	-
%s C <sub>3</sub> (H <sub>9</sub> )	25.64	25.81	27.08	25.79	30.08

TABLE 2S E<sup>(2)</sup> energies (kcal mol<sup>-1</sup>) from lone pair of O to the  $\sigma^*$  orbitals in the protonated ethers. The numbers in parentheses indicate the differences between the protonated and isolated molecules. Results of MP2/6-311++G(d,p) calculations.

	<b>I(H<sup>+</sup>)</b>	<b>II(H<sup>+</sup>)</b>	<b>III(H<sup>+</sup>)</b>	<b>IV(H<sup>+</sup>)</b>	<b>V(H<sup>+</sup>)</b>
LP(1)O →					
$\sigma^*(C_1H_4)$	1.45(+0.47)	0.96(-2.66)	2.64(+1.03)	-	-
$\sigma^*(C_1H_6)$	0.91(-2.40)	1.41(-2.20)	0.59(-3.33)	1.24(-3.97)	0.00
$\sigma^*(C_3H_7)$	1.46(+0.48)	1.28(-0.90)	-	1.71(+0.95)	-
$\sigma^*(C_3H_8)$	3.27(+2.28)	3.01(+3.01)	0.99(-0.62)	2.89(+2.13)	-
$\sigma^*(C_3H_9)$	0.91(-2.40)	0.91(-2.08)	1.23(-2.58)	-	0.00
$\sigma^*(C_1X)^a$	-	6.38(+5.72)	1.48(-0.80)	5.85(+1.42))	1.30 (-2.17)
$\sigma^*(C_3Cl_7)$	-	-	6.13(+3.85)	-	-
$\sigma^*(C_1F_5)$	-	-	-	-	5.63(+2.95)

<sup>a</sup> X=Cl<sub>5</sub> for II, III and IV, X=Cl<sub>4</sub> for V

Table 3S Selected NBO parameters in O-protonated halogenated ethers ( $\sigma^*$  occupancies and % s-character at C carbon in CH bond). Results of MP2/6-311++G(d,p) calculations.

	<b>I(H<sup>+</sup>)</b>	<b>II(H<sup>+</sup>)</b>	<b>III(H<sup>+</sup>)</b>	<b>IV(H<sup>+</sup>)</b>	<b>V(H<sup>+</sup>)</b>
$\sigma^*(C_1H_4)$	0.004	0.021	0.022	-	-
$\sigma^*(C_1Cl_4)$	-	-	-	0.035	0.031
$\sigma^*(C_1Cl_5)$	-	0.015	0.008	0.043	0.051 <sup>a</sup>
$\sigma^*(C_1H_6)$	0.003	0.019	0.019	0.040	0.036
$\sigma^*(C_3H_7)$	0.004	0.004	-	0.003	0.039 <sup>b</sup>
$\sigma^*(C_3H_9)$	0.003	0.003	0.019	0.004	0.034
%sC <sub>1</sub> (H <sub>4</sub> )	27.43	28.32	28.32	-	-
%sC <sub>1</sub> (H <sub>6</sub> )	27.41	28.41	28.96	28.92	30.46
%sC <sub>3</sub> (H <sub>7</sub> )	27.44	27.42	-	28.13	-
%sC <sub>3</sub> (H <sub>9</sub> )	27.41	27.42	28.40	27.93	32.43

<sup>a</sup> Corresponds to  $\sigma^*(C_1F_5)$ . <sup>b</sup> Corresponds to  $\sigma^*(C_3F_8)$ .

TABLE 4S The NBO charges (e) on all the atoms in the isolated ethers.  
 Results of MP2/6-311++G(d,p) calculations.

<b>I</b>		<b>II</b>		<b>III</b>		<b>IV</b>		<b>V</b>	
atom	charge	atom	charge	atom	charge	atom	charge	atom	charge
C <sub>1</sub>	-0.095	C <sub>1</sub>	0.103	C <sub>1</sub>	0.078	C <sub>1</sub>	0.206	C <sub>1</sub>	0.614
O <sub>2</sub>	-0.649	O <sub>2</sub>	-0.636	O <sub>2</sub>	-0.634	O <sub>2</sub>	-0.640	O <sub>2</sub>	-0.632
C <sub>3</sub>	-0.095	C <sub>3</sub>	-0.108	C <sub>3</sub>	0.078	C <sub>3</sub>	-0.120	C <sub>3</sub>	0.995
H <sub>4</sub>	0.131	H <sub>4</sub>	0.145	H <sub>4</sub>	0.170	Cl <sub>4</sub>	-0.060	Cl <sub>4</sub>	-0.064
H <sub>5</sub>	0.131	Cl <sub>5</sub>	-0.134	Cl <sub>5</sub>	-0.110	Cl <sub>5</sub>	-0.060	F <sub>5</sub>	-0.408
H <sub>6</sub>	0.158	H <sub>6</sub>	0.172	H <sub>6</sub>	0.179	H <sub>6</sub>	0.183	H <sub>6</sub>	0.163
H <sub>7</sub>	0.131	H <sub>7</sub>	0.137	Cl <sub>7</sub>	-0.110	H <sub>7</sub>	0.161	F <sub>7</sub>	-0.393
H <sub>8</sub>	0.131	H <sub>8</sub>	0.157	H <sub>8</sub>	0.170	H <sub>8</sub>	0.161	F <sub>8</sub>	-0.397
H <sub>9</sub>	0.158	H <sub>9</sub>	0.165	H <sub>9</sub>	0.179	H <sub>9</sub>	0.170	H <sub>9</sub>	0.122

TABLE 5S E<sup>(2)</sup> energies (kcal mol<sup>-1</sup>) from two lone pairs of O to the σ\* orbitals in the isolated molecules. Results of B3LYP/6-311++G(d,p) calculations.

	<b>I</b>	<b>II</b>	<b>III</b>	<b>IV</b>	<b>V</b>
LP(1)O →					
σ*(C <sub>1</sub> H <sub>4</sub> )	0.78	3.14	1.50	-	-
σ*(C <sub>1</sub> H <sub>6</sub> )	2.49	2.75	2.98	3.81	3.21
σ*(C <sub>3</sub> H <sub>7</sub> )	0.78	1.65	-	0.57	-
σ*(C <sub>3</sub> H <sub>8</sub> )	0.78	0.00	1.50	0.57	-
σ*(C <sub>3</sub> H <sub>9</sub> )	2.49	2.31	2.98	2.50	2.92
σ*(C <sub>1</sub> Cl <sub>5</sub> )	-	0.58	1.87	3.99	2.83 <sup>a</sup>
σ*(C <sub>1</sub> F <sub>5</sub> )	-	-	-	-	2.52
σ*(C <sub>3</sub> F <sub>7</sub> )	-	-	-	-	2.97
LP(2)O →					
σ*(C <sub>1</sub> H <sub>4</sub> )	6.51	3.88	4.72	-	-
σ*(C <sub>1</sub> H <sub>6</sub> )	0.00	0.64	0.00	0.00	0.00
σ*(C <sub>3</sub> H <sub>8</sub> )	6.51	6.07	4.72	4.79	-
σ*(C <sub>1</sub> Cl <sub>5</sub> )	-	21.95	15.50	16.49	13.72 <sup>b</sup>
σ*(C <sub>1</sub> F <sub>5</sub> )	-	-	-	-	12.61
σ*(C <sub>3</sub> F <sub>7</sub> )	-	-	-	-	6.72

<sup>a</sup> E<sup>(2)</sup> from LP(1)O to σ\*(C<sub>1</sub>Cl<sub>4</sub>). <sup>b</sup> E<sup>(2)</sup> from LP(2)O to σ\*(C<sub>1</sub>Cl<sub>4</sub>).