

**Predictive Elucidation of Conformational
Characteristics and Configurational Properties of
Poly(1-methylphosphirane) and
Poly(1-phenylphosphirane) as a Molecular Design**

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Supplementary Information
(Tables S1 – S4)

Table S1 Geometrical parameters for *ll* form of poly(1-methylphosphirane), used in the refined RIS calculations ^a

Conformation					
Bond					
<i>j</i> - 1	<i>j</i>	<i>j</i> + 1	<i>l_j</i> ^b	$\angle j \wedge (j + 1)$ ^c	ϕ_j ^d
<i>j</i> - 1 = <i>c</i> , <i>j</i> = <i>a</i> , and <i>j</i> + 1 = <i>b</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.880	113.1	2.9
	<i>g</i> ⁺		1.885	119.3	125.1
	<i>g</i> ⁻		1.881	113.2	-101.2
<i>g</i> ⁺	<i>t</i>		1.881	112.9	3.3
	<i>g</i> ⁺		1.885	119.2	130.5
	<i>g</i> ⁻		1.883	114.5	-99.4
<i>g</i> ⁻	<i>t</i>		1.881	112.2	-1.0
	<i>g</i> ⁺		1.891	120.2	101.0
	<i>g</i> ⁻		1.880	113.0	-109.4
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.880	115.7	-1.8
	<i>g</i> ⁺		1.883	120.5	121.5
	<i>g</i> ⁻		1.886	116.3	-80.7
<i>g</i> ⁺	<i>t</i>		1.880	115.6	2.7
	<i>g</i> ⁺		1.884	120.2	120.9
	<i>g</i> ⁻		-	-	-
<i>g</i> ⁻	<i>t</i>		1.880	114.9	-3.4
	<i>g</i> ⁺		1.892	121.6	97.8
	<i>g</i> ⁻		1.894	116.2	-87.1
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.885	115.4	-15.0
	<i>g</i> ⁺		1.884	121.0	133.6
	<i>g</i> ⁻		1.884	116.2	-114.6
<i>g</i> ⁺	<i>t</i>		1.886	115.2	-16.6
	<i>g</i> ⁺		1.886	121.7	125.7
	<i>g</i> ⁻		1.885	117.3	-107.2
<i>g</i> ⁻	<i>t</i>		1.886	114.6	-19.6
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.883	115.8	-117.1
<i>j</i> - 1 = <i>a</i> , <i>j</i> = <i>b</i> , and <i>j</i> + 1 = <i>c</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.537	113.1	0.0
	<i>g</i> ⁺		1.541	115.4	108.9
	<i>g</i> ⁻		1.541	115.8	-108.9
<i>g</i> ⁺	<i>t</i>		1.536	113.0	-5.8
	<i>g</i> ⁺		1.547	114.0	83.1
	<i>g</i> ⁻		1.549	117.3	-104.8
<i>g</i> ⁻	<i>t</i>		1.536	113.2	-4.6
	<i>g</i> ⁺		1.541	116.3	111.4
	<i>g</i> ⁻		1.539	116.1	-129.8
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.536	113.2	4.6
	<i>g</i> ⁺		1.539	116.1	129.8
	<i>g</i> ⁻		1.541	116.3	-111.4
<i>g</i> ⁺	<i>t</i>		1.536	112.9	3.6
	<i>g</i> ⁺		1.547	114.0	83.1
	<i>g</i> ⁻		-	-	-

<i>g</i> ⁻	<i>t</i>	<i>g</i> ⁺	1.537	113.0	-1.5
	<i>g</i> ⁺		1.541	116.0	109.7
	<i>g</i> ⁻		1.541	115.5	-109.7
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.536	119.3	5.8
	<i>g</i> ⁺		1.549	123.0	104.8
	<i>g</i> ⁻		1.547	120.5	-83.1
<i>g</i> ⁺	<i>t</i>		1.535	119.2	0.0
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		-	-	-
<i>g</i> ⁻	<i>t</i>		1.536	119.4	-3.6
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.547	120.5	-83.1
<i>j</i> - 1 = <i>b</i> , <i>j</i> = <i>c</i> , and <i>j</i> + 1 = <i>a</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.880	98.6	-2.9
	<i>g</i> ⁺		1.881	99.8	101.2
	<i>g</i> ⁻		1.885	101.1	-125.1
<i>g</i> ⁺	<i>t</i>		1.885	97.8	15.0
	<i>g</i> ⁺		1.884	99.8	114.6
	<i>g</i> ⁻		1.888	103.3	-133.6
<i>g</i> ⁻	<i>t</i>		1.880	98.1	1.8
	<i>g</i> ⁺		1.886	100.4	80.7
	<i>g</i> ⁻		1.883	100.6	-121.5
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.881	101.2	1.0
	<i>g</i> ⁺		1.880	102.4	109.4
	<i>g</i> ⁻		1.891	103.9	-101.0
<i>g</i> ⁺	<i>t</i>		1.886	100.2	19.6
	<i>g</i> ⁺		1.883	102.2	117.1
	<i>g</i> ⁻		-	-	-
<i>g</i> ⁻	<i>t</i>		1.880	101.0	3.4
	<i>g</i> ⁺		1.884	104.0	87.1
	<i>g</i> ⁻		1.892	102.2	-97.8
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.881	99.7	-3.3
	<i>g</i> ⁺		1.883	102.2	99.4
	<i>g</i> ⁻		1.885	102.4	-130.5
<i>g</i> ⁺	<i>t</i>		1.886	99.0	16.6
	<i>g</i> ⁺		1.885	101.8	107.2
	<i>g</i> ⁻		1.886	104.7	-125.7
<i>g</i> ⁻	<i>t</i>		1.880	99.3	-2.7
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.884	101.5	-120.9

^a Obtained from the geometrical optimization for 1,2-bis(ethylmethylphosphino)ethane at the B3LYP/6-31G(d) level. *j* denotes the current bond. Geometrical parameters of the *dd* form can be derived from the above ones. ^b Length of bond *j*. ^c Angle formed between bonds *j* and *j*+1. ^d Dihedral angle of bond *j*. For the bond designations, see Figure 2.

Table S2 Geometrical parameters for *dl* form of poly(1-methylphosphirane), used in the refined RIS calculations ^a

Conformation					
Bond					
<i>j</i> - 1	<i>j</i>	<i>j</i> + 1	<i>l_j</i> ^b	$\angle j \wedge (j + 1)$ ^c	ϕ_j ^d
<i>j</i> - 1 = <i>c</i> , <i>j</i> = <i>a</i> , and <i>j</i> + 1 = <i>b</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.880	113.1	2.1
	<i>g</i> ⁺		1.884	119.4	130.1
	<i>g</i> ⁻		1.881	113.2	-102.6
<i>g</i> ⁺	<i>t</i>		1.880	113.0	2.6
	<i>g</i> ⁺		1.885	119.2	130.5
	<i>g</i> ⁻		1.884	114.4	-95.3
<i>g</i> ⁻	<i>t</i>		1.880	112.4	-0.2
	<i>g</i> ⁺		1.891	120.5	101.4
	<i>g</i> ⁻		1.880	112.8	-107.9
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.883	115.9	14.0
	<i>g</i> ⁺		1.887	122.8	144.1
	<i>g</i> ⁻		1.886	115.4	-84.3
<i>g</i> ⁺	<i>t</i>		1.884	115.7	14.0
	<i>g</i> ⁺		1.886	122.8	140.5
	<i>g</i> ⁻		1.884	116.4	-93.5
<i>g</i> ⁻	<i>t</i>		1.882	115.3	12.1
	<i>g</i> ⁺		1.892	121.5	99.4
	<i>g</i> ⁻		1.886	114.9	-86.8
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.886	116.2	-19.6
	<i>g</i> ⁺		1.883	120.6	134.1
	<i>g</i> ⁻		1.880	115.8	-102.3
<i>g</i> ⁺	<i>t</i>		1.887	116.3	-19.5
	<i>g</i> ⁺		1.885	122.8	115.5
	<i>g</i> ⁻		1.885	116.7	-88.9
<i>g</i> ⁻	<i>t</i>		1.886	115.6	-20.5
	<i>g</i> ⁺		1.884	122.9	127.6
	<i>g</i> ⁻		1.880	115.8	-108.6
<i>j</i> - 1 = <i>a</i> , <i>j</i> = <i>b</i> , and <i>j</i> + 1 = <i>c</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.537	113.1	5.3
	<i>g</i> ⁺		1.539	115.9	128.8
	<i>g</i> ⁻		1.541	116.2	-111.9
<i>g</i> ⁺	<i>t</i>		1.536	112.8	5.6
	<i>g</i> ⁺		1.544	117.1	104.2
	<i>g</i> ⁻		1.547	114.0	-83.5
<i>g</i> ⁻	<i>t</i>		1.536	113.1	0.2
	<i>g</i> ⁺		1.541	115.7	109.6
	<i>g</i> ⁻		1.541	115.4	-108.4
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.536	119.4	5.6
	<i>g</i> ⁺		1.544	122.8	104.2
	<i>g</i> ⁻		1.547	120.6	-83.5
<i>g</i> ⁺	<i>t</i>		1.535	119.1	3.1
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		-	-	-

<i>g</i> ⁻	<i>t</i>	<i>g</i> ⁺	1.536	119.4	-2.7
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.543	122.8	-105.2
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.536	113.2	0.2
	<i>g</i> ⁺		1.541	115.4	109.6
	<i>g</i> ⁻		1.541	115.8	-108.4
<i>g</i> ⁺	<i>t</i>		1.536	112.9	-2.7
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.543	117.4	-105.2
<i>g</i> ⁻	<i>t</i>		1.536	113.2	-6.4
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.539	116.1	-129.1
<i>j</i> - 1 = <i>b</i> , <i>j</i> = <i>c</i> , and <i>j</i> + 1 = <i>a</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.880	98.6	2.1
	<i>g</i> ⁺		1.884	101.5	130.1
	<i>g</i> ⁻		1.881	99.7	-102.6
<i>g</i> ⁺	<i>t</i>		1.883	97.6	14.0
	<i>g</i> ⁺		1.887	101.6	144.1
	<i>g</i> ⁻		1.886	101.3	-84.3
<i>g</i> ⁻	<i>t</i>		1.886	98.2	-19.6
	<i>g</i> ⁺		1.883	102.9	134.1
	<i>g</i> ⁻		1.880	99.5	-102.3
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.880	99.8	2.6
	<i>g</i> ⁺		1.885	102.4	130.5
	<i>g</i> ⁻		1.884	102.2	-95.3
<i>g</i> ⁺	<i>t</i>		1.884	98.8	14.0
	<i>g</i> ⁺		1.886	102.7	140.5
	<i>g</i> ⁻		1.884	104.3	-93.5
<i>g</i> ⁻	<i>t</i>		1.887	99.3	-19.5
	<i>g</i> ⁺		1.885	104.7	115.5
	<i>g</i> ⁻		1.885	101.6	-88.9
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.880	101.2	-0.2
	<i>g</i> ⁺		1.891	104.1	101.4
	<i>g</i> ⁻		1.880	102.3	-107.9
<i>g</i> ⁺	<i>t</i>		1.882	100.5	12.1
	<i>g</i> ⁺		1.892	102.4	99.4
	<i>g</i> ⁻		1.886	104.4	-86.8
<i>g</i> ⁻	<i>t</i>		1.886	100.4	-20.5
	<i>g</i> ⁺		1.884	106.0	127.6
	<i>g</i> ⁻		1.880	102.2	-108.6

^a Obtained from the geometrical optimization for 1,2-bis(ethylmethylphosphino)ethane at the B3LYP/6-31G(d) level. *j* denotes the current bond. Geometrical parameters of the *dl* form can be derived from the above ones. ^b Length of bond *j*. ^c Angle formed between bonds *j* and *j*+1. ^d Dihedral angle of bond *j*. For the bond designations, see Figure 2.

Table S3 Geometrical parameters for *ll* form of poly(1-phenylphosphirane), used in the refined RIS calculations ^a

Conformation					
Bond					
<i>j</i> - 1	<i>j</i>	<i>j</i> + 1	<i>l_j</i> ^b	$\angle j \wedge (j + 1)$ ^c	ϕ_j ^d
<i>j</i> - 1 = <i>c</i> , <i>j</i> = <i>a</i> , and <i>j</i> + 1 = <i>b</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.880	112.6	4.2
	<i>g</i> ⁺		1.887	118.7	128.6
	<i>g</i> ⁻		1.883	112.4	-100.9
<i>g</i> ⁺	<i>t</i>		1.879	112.4	5.9
	<i>g</i> ⁺		1.887	118.1	129.5
	<i>g</i> ⁻		1.885	113.6	-105.5
<i>g</i> ⁻	<i>t</i>		1.875	112.3	1.1
	<i>g</i> ⁺		1.888	121.1	111.0
	<i>g</i> ⁻		1.876	111.6	-105.7
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.881	115.2	4.4
	<i>g</i> ⁺		1.887	122.9	144.8
	<i>g</i> ⁻		1.889	115.6	-82.8
<i>g</i> ⁺	<i>t</i>		1.881	115.0	3.1
	<i>g</i> ⁺		1.886	122.7	141.8
	<i>g</i> ⁻		-	-	-
<i>g</i> ⁻	<i>t</i>		1.876	114.6	2.5
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.882	115.3	-87.0
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.888	114.7	-20.4
	<i>g</i> ⁺		1.891	122.6	108.8
	<i>g</i> ⁻		1.886	115.7	-115.0
<i>g</i> ⁺	<i>t</i>		1.887	114.6	-19.5
	<i>g</i> ⁺		1.890	121.3	118.7
	<i>g</i> ⁻		1.888	116.4	-111.5
<i>g</i> ⁻	<i>t</i>		1.883	113.9	-23.3
	<i>g</i> ⁺		1.909	120.6	70.6
	<i>g</i> ⁻		1.878	114.6	-116.4
<i>j</i> - 1 = <i>a</i> , <i>j</i> = <i>b</i> , and <i>j</i> + 1 = <i>c</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.535	112.6	0.2
	<i>g</i> ⁺		1.539	114.7	113.6
	<i>g</i> ⁻		1.539	115.2	-113.6
<i>g</i> ⁺	<i>t</i>		1.533	112.5	0.3
	<i>g</i> ⁺		1.543	115.9	99.1
	<i>g</i> ⁻		1.540	116.8	-105.4
<i>g</i> ⁻	<i>t</i>		1.536	112.4	-4.5
	<i>g</i> ⁺		1.542	116.4	101.9
	<i>g</i> ⁻		1.538	115.4	-127.1
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.536	112.4	4.5
	<i>g</i> ⁺		1.538	115.7	127.1
	<i>g</i> ⁻		1.542	115.6	-101.9
<i>g</i> ⁺	<i>t</i>		1.534	112.5	7.1
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		-	-	-

<i>g</i> ⁻	<i>t</i>	<i>g</i> ⁺	1.537	112.4	0.1
	<i>g</i> ⁺		1.542	115.4	108.3
	<i>g</i> ⁻		1.542	114.8	-108.3
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.533	118.7	-0.3
	<i>g</i> ⁺		1.540	122.6	105.4
	<i>g</i> ⁻		1.543	122.5	-99.1
<i>g</i> ⁺	<i>t</i>		1.531	119.0	1.1
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		-	-	-
<i>g</i> ⁻	<i>t</i>		1.534	118.8	-7.1
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		-	-	-
<i>j</i> - 1 = <i>b</i> , <i>j</i> = <i>c</i> , and <i>j</i> + 1 = <i>a</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.880	98.9	-4.2
	<i>g</i> ⁺		1.883	99.9	100.9
	<i>g</i> ⁻		1.887	101.5	-128.6
<i>g</i> ⁺	<i>t</i>		1.888	97.9	20.4
	<i>g</i> ⁺		1.886	100.2	115.0
	<i>g</i> ⁻		1.891	103.4	-108.8
<i>g</i> ⁻	<i>t</i>		1.881	98.2	-4.4
	<i>g</i> ⁺		1.889	100.9	82.8
	<i>g</i> ⁻		1.887	102.0	-144.8
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.875	101.3	-1.1
	<i>g</i> ⁺		1.879	102.2	105.7
	<i>g</i> ⁻		1.888	105.3	-111.0
<i>g</i> ⁺	<i>t</i>		1.883	100.2	23.3
	<i>g</i> ⁺		1.878	101.8	116.4
	<i>g</i> ⁻		1.909	104.5	-70.6
<i>g</i> ⁻	<i>t</i>		1.876	100.7	-2.5
	<i>g</i> ⁺		1.882	104.5	87.0
	<i>g</i> ⁻		-	-	-
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.879	100.0	-5.9
	<i>g</i> ⁺		1.885	101.6	105.5
	<i>g</i> ⁻		1.887	102.2	-129.5
<i>g</i> ⁺	<i>t</i>		1.887	99.1	19.5
	<i>g</i> ⁺		1.888	101.2	111.5
	<i>g</i> ⁻		1.890	104.5	-118.7
<i>g</i> ⁻	<i>t</i>		1.881	99.2	-3.1
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.886	102.6	-141.8

^a Obtained from the geometrical optimization for 1,2-bis(ethylphenylphosphino)ethane at the B3LYP/6-31G(d) level. *j* denotes the current bond. Geometrical parameters of the *dd* form can be derived from the above ones. ^b Length of bond *j*. ^c Angle formed between bonds *j* and *j*+1. ^d Dihedral angle of bond *j*. For the bond designations, see Figure 2.

Table S4 Geometrical parameters for *dl* form of poly(1-phenylphosphirane), used in the refined RIS calculations ^a

Conformation					
Bond					
<i>j</i> - 1	<i>j</i>	<i>j</i> + 1	<i>l_j</i> ^b	$\angle j \wedge (j + 1)$ ^c	ϕ_j ^d
<i>j</i> - 1 = <i>c</i> , <i>j</i> = <i>a</i> , and <i>j</i> + 1 = <i>b</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.880	112.5	4.2
	<i>g</i> ⁺		1.887	119.1	126.2
	<i>g</i> ⁻		1.883	112.1	-100.3
<i>g</i> ⁺	<i>t</i>		1.881	112.5	6.9
	<i>g</i> ⁺		1.887	118.3	128.3
	<i>g</i> ⁻		1.885	113.7	-103.0
<i>g</i> ⁻	<i>t</i>		1.874	112.4	-1.7
	<i>g</i> ⁺		1.893	120.1	102.3
	<i>g</i> ⁻		1.875	111.5	-105.5
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.884	115.3	11.5
	<i>g</i> ⁺		1.894	122.4	154.2
	<i>g</i> ⁻		-	-	-
<i>g</i> ⁺	<i>t</i>		1.885	115.2	13.8
	<i>g</i> ⁺		1.888	122.5	142.6
	<i>g</i> ⁻		1.890	116.2	-90.5
<i>g</i> ⁻	<i>t</i>		1.878	114.9	5.2
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.884	113.8	-86.4
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.884	116.3	-16.9
	<i>g</i> ⁺		1.887	121.5	125.8
	<i>g</i> ⁻		1.884	115.7	-108.0
<i>g</i> ⁺	<i>t</i>		1.883	116.2	-15.9
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.887	116.1	-91.9
<i>g</i> ⁻	<i>t</i>		1.879	115.7	-17.0
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.876	114.3	-109.6
<i>j</i> - 1 = <i>a</i> , <i>j</i> = <i>b</i> , and <i>j</i> + 1 = <i>c</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.535	112.5	3.9
	<i>g</i> ⁺		1.537	115.3	123.5
	<i>g</i> ⁻		1.540	116.3	-100.2
<i>g</i> ⁺	<i>t</i>		1.532	112.2	3.7
	<i>g</i> ⁺		1.542	115.9	97.5
	<i>g</i> ⁻		1.543	114.6	-92.7
<i>g</i> ⁻	<i>t</i>		1.536	112.8	3.2
	<i>g</i> ⁺		1.540	115.2	108.1
	<i>g</i> ⁻		1.539	114.9	-118.4
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.532	119.1	3.7
	<i>g</i> ⁺		1.542	122.4	97.5
	<i>g</i> ⁻		1.543	121.5	-92.7
<i>g</i> ⁺	<i>t</i>		1.531	118.7	3.7
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		-	-	-

<i>g</i> ⁻	<i>t</i>	<i>g</i> ⁺	1.534	118.9	-1.9
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.542	122.9	-102.6
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.536	112.1	3.2
	<i>g</i> ⁺		1.540	114.3	108.1
	<i>g</i> ⁻		1.539	115.7	-118.4
<i>g</i> ⁺	<i>t</i>		1.534	112.2	-1.9
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.542	116.6	-102.6
<i>g</i> ⁻	<i>t</i>		1.537	112.6	-10.2
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.540	115.8	-129.5
<i>j</i> - 1 = <i>b</i> , <i>j</i> = <i>c</i> , and <i>j</i> + 1 = <i>a</i>					
<i>t</i>	<i>t</i>	<i>t</i>	1.880	98.9	4.2
	<i>g</i> ⁺		1.887	101.5	126.2
	<i>g</i> ⁻		1.883	99.7	-100.3
<i>g</i> ⁺	<i>t</i>		1.884	98.0	11.5
	<i>g</i> ⁺		1.894	102.7	154.2
	<i>g</i> ⁻		-	-	-
<i>g</i> ⁻	<i>t</i>		1.884	97.5	-16.9
	<i>g</i> ⁺		1.887	103.6	125.8
	<i>g</i> ⁻		1.884	100.1	-108.0
<i>t</i>	<i>t</i>	<i>g</i> ⁺	1.881	99.9	6.9
	<i>g</i> ⁺		1.887	102.1	128.3
	<i>g</i> ⁻		1.885	101.9	-103.0
<i>g</i> ⁺	<i>t</i>		1.885	99.1	13.8
	<i>g</i> ⁺		1.888	102.7	142.6
	<i>g</i> ⁻		1.890	104.2	-90.5
<i>g</i> ⁻	<i>t</i>		1.883	98.6	-15.9
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.887	101.9	-91.9
<i>t</i>	<i>t</i>	<i>g</i> ⁻	1.874	101.1	-1.7
	<i>g</i> ⁺		1.893	103.8	102.3
	<i>g</i> ⁻		1.875	102.2	-105.5
<i>g</i> ⁺	<i>t</i>		1.878	100.6	5.2
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.884	104.6	-86.4
<i>g</i> ⁻	<i>t</i>		1.879	99.8	-17.0
	<i>g</i> ⁺		-	-	-
	<i>g</i> ⁻		1.876	102.1	-109.6

^a Obtained from the geometrical optimization for 1,2-bis(ethylphenylphosphino)ethane at the B3LYP/6-31G(d) level. *j* denotes the current bond. Geometrical parameters of the *dl* form can be derived from the above ones. ^b Length of bond *j*. ^c Angle formed between bonds *j* and *j*+1. ^d Dihedral angle of bond *j*. For the bond designations, see Figure 2.