

Complementary Supporting Information's

**Electron Transfer Reaction of Light Harvesting Zinc Naphthalocyanine–
Subphthalocyanine Self-Assembled Dyad: Spectroscopic, Electrochemical,
Computational, and Photochemical Studies**

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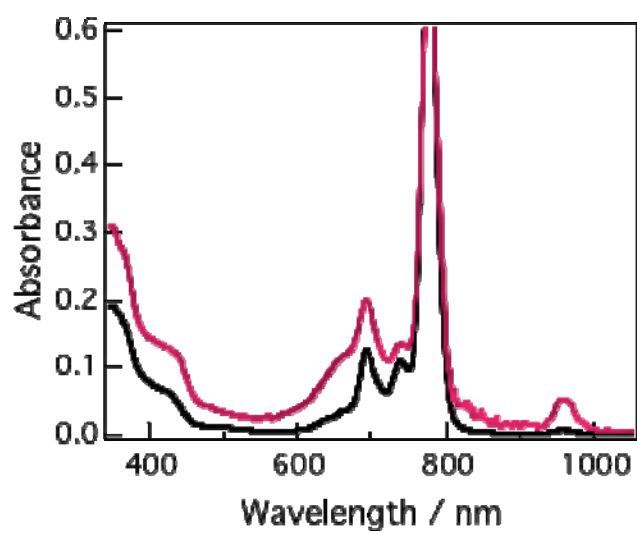


Fig. S1 Absorption profiles of ZnNc in the absence (black) and presence (red) of $[(Ru(bpy)_3)_3]^{3+}$ in dichloromethane.

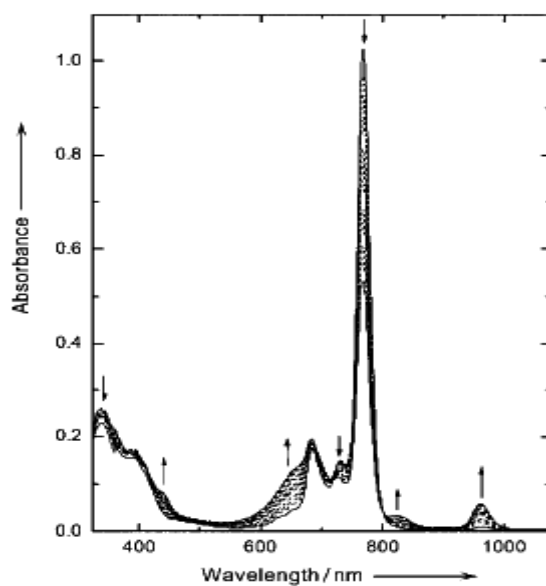


Fig. S2 Spectral changes observed during the first oxidation of ZnNc (3.5 μM) in *o*-dichlorobenzene, 0.1M (TBA)ClO₄. Potential applied 0.8 V vs. Ag/AgNO₃. (Ref. *ChemPhysChem*. 2003, **4**, 474).

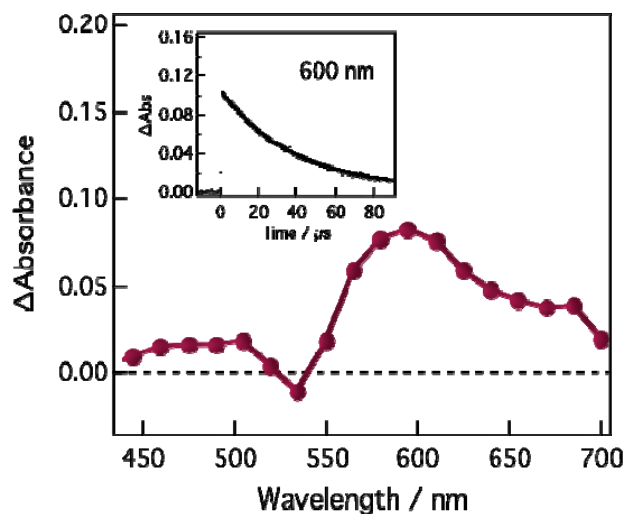


Fig. S3 Nanosecond transient spectra of SubPc(py):ZnNc dyad in *o*-dichlorobenzene; $\lambda_{\text{ex}} = 430 \text{ nm}$.

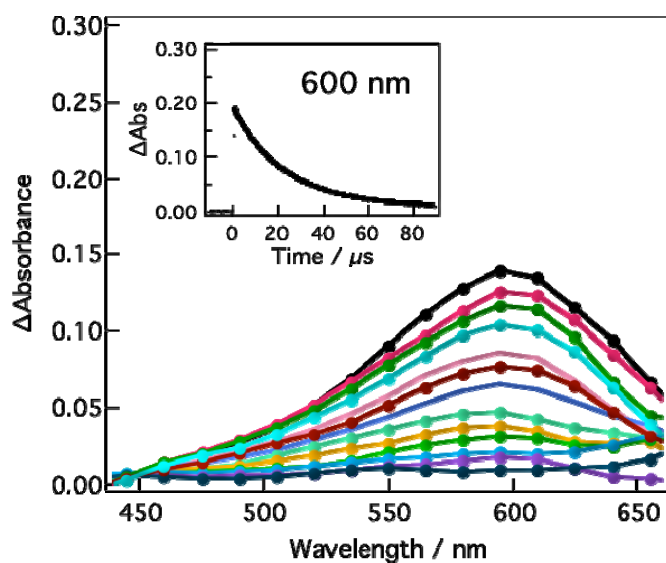


Fig. S4 Nanosecond transient spectra of ZnNc in *o*-dichlorobenzene; $\lambda_{\text{ex}} = 430$ nm.

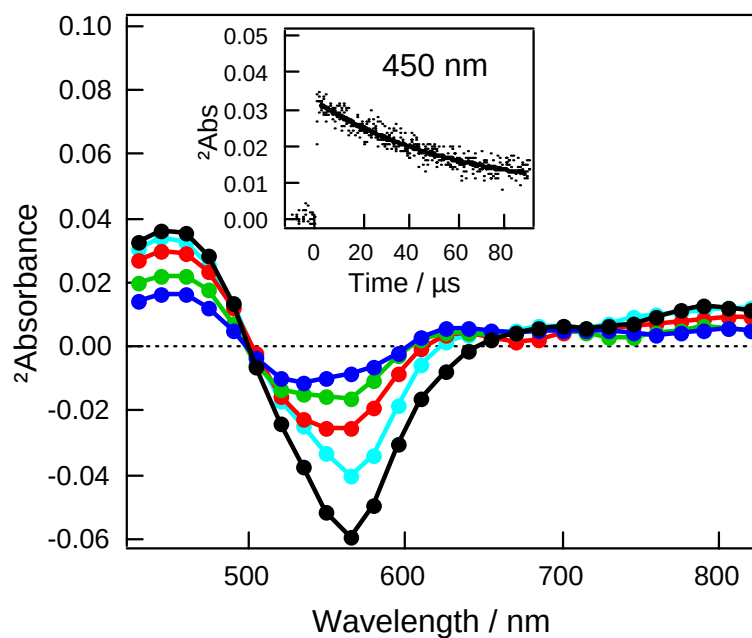


Fig. S5 Nanosecond transient spectra of SubPc(py) in *o*-dichlorobenzene; $\lambda_{\text{ex}} = 550$ nm.

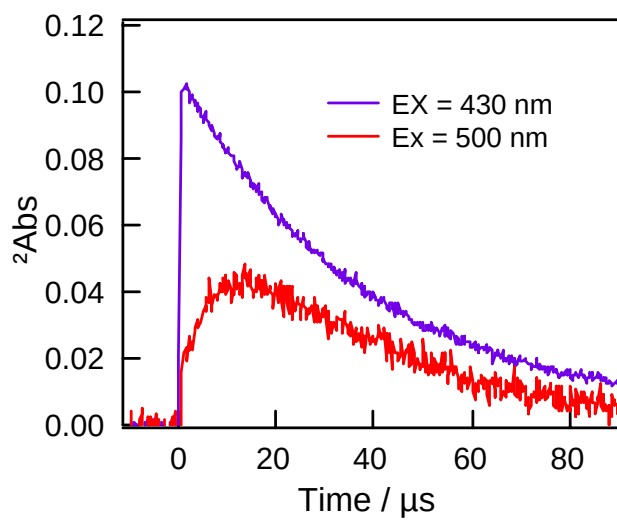


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