

Organic Polyanionic High-Spin Molecular Clusters in Solution:  
Topological-Symmetry Controlled Models for Organic Ferromagnetic  
Metals with *meta*-Benzoylbenzene Linkers

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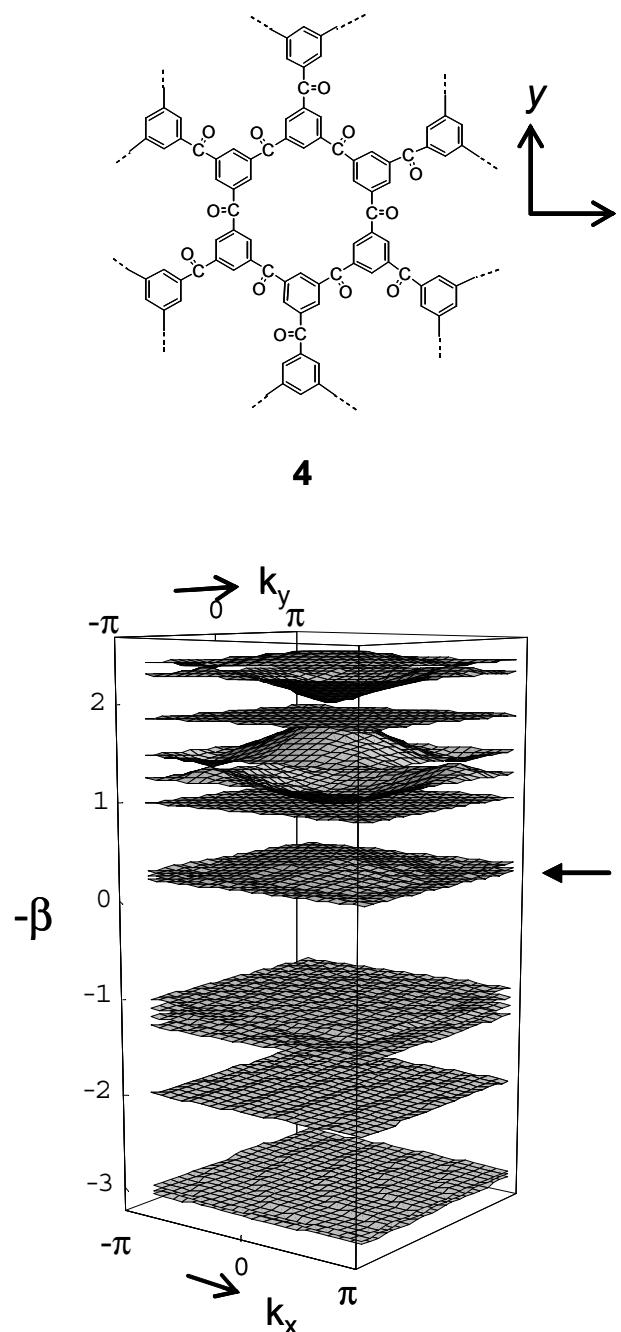
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## Contents of the Supplementary Information

- 1. The two-dimensional  $\pi$ -crystal orbital band structure of a two-dimensional system 4 (Fig. S1)**
- 2. Optimized molecular structures for *m*-benzoylbenzene in its doublet and triplet states**
  - 2.1 The optimized molecular structures for  $1^-K^+$  in the spin-doublet state (Fig. S2) and  $1^{2-}(K^+)_2$  in the triplet state (Fig. S3) obtained by the DFT quantum chemical calculation (UB3LYP/6-311G\*)**
- 3. Optimized molecular structures for *m*-benzoylbenzene-based polyanionic high-spin clusters**
  - 3.1 The optimized structure for  $1^-(K^+)_21^-$  in the triplet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G) (Fig. S4).**
  - 3.2 The optimized structure for  $1^{2-}(K^+)_31^-$  in the quartet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G) (Fig. S5).**
  - 3.3 The optimized structure for  $1^{2-}(K^+)_41^{2-}$  in the quintet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G) (Fig. S6).**
- 4. The clustering structure of the dimer, [Fluorenone $^{-\bullet}$  {Na $^+$ (dimethoxyethan) $_2$ }] $_2$  (Fig. S7), and the comparison of the lengths, Na-O, Na-Na, and O-O and angles, O-Na-O, and Na-O-Na between the X-ray single-crystal analysis and the structural optimization (Fig. S7).**
- 5. Mulliken atomic charges for [Fluorenone $^{-\bullet}$  {Na $^+$ (dimethoxyethan) $_2$ }] $_2$  (Tables S1 and S2)**
- 6. The calculation of  $D_i$  of the intramolecular interaction in terms of point-dipole approximation**
  - 6.1 The calculation of the  $D_{12}$  tensors for the doublet, quartet, and quintet clusters in terms of point-dipole approximation**
  - 6.2 The  $D$  and  $E$  values for the quartet and quintet molecular clusters**
- 7. The charge density and spin density for  $1^-K^+$  in the doublet state,  $1^{2(-\bullet)}K_2$  in the triplet ground state,  $1^-(K^+)_21^-$  in the triplet state,  $1^{2(-\bullet)}(K^+)_31^{(-\bullet)}$  in the quartet state,  $1^{2(-\bullet)}(K^+)_41^{2(-\bullet)}$  in the quintet state, as calculated by the DFT quantum chemical calculations (UB3LYP/6-311G\*/UB3LYP/3-21G)**
- 8. The charge density distribution and the spin density distribution for  $1^{2(-\bullet)}(K^+)_2$  in the triplet state calculated by DFT method (UB3LYP/6-311++G\*\* //UB3LYP/6-311G\*)**

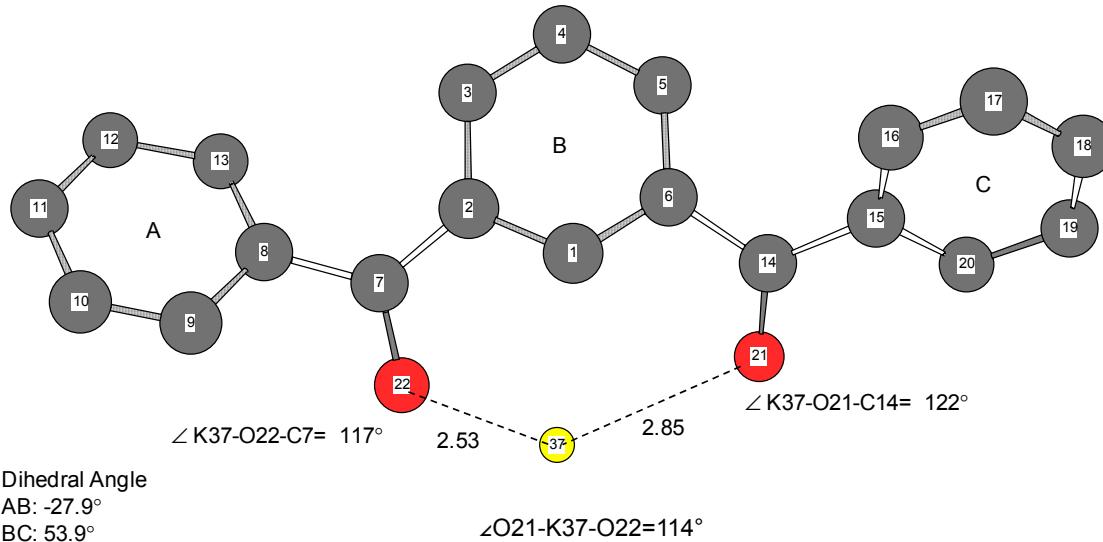
**1. The two-dimensional  $\pi$ -crystal orbital band structure of a two-dimensional system **4** (Fig. S1)**



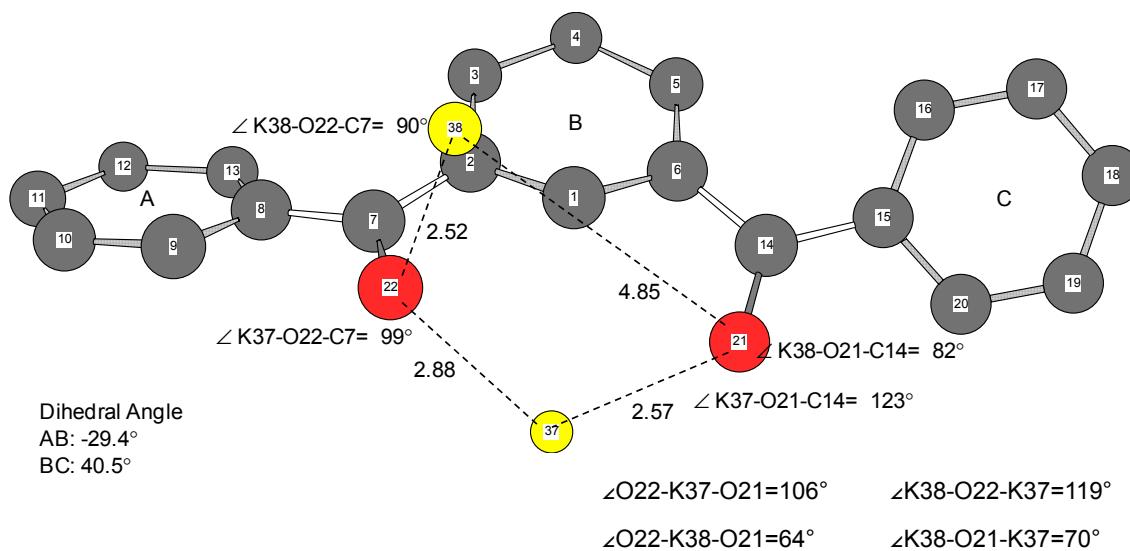
**Fig. S1** The two-dimensional  $\pi$ -Crystal Orbital Band structure of a two-dimensional system **4**. The arrow denotes the Lowest Unoccupied Crystal Orbitals (LUCOs), which are pseudo-flat bands. The flat bands originate in the 1,3,5-connectivity of oligoketones, not in the group-theoretical symmetry. The 2D polymeric system **4** incorporates extra electrons into the flat band in a ferromagnetically exchange-coupled manner in the ground state. The next LUCO is closely located above the LUCOs.

## 2. The optimized molecular structures for *m*-benzoylbenzene in its doublet and triplet states

2.1 The optimized molecular structures for  $\mathbf{1}^{\bullet}\mathbf{K}^+$  in the spin-doublet state (Fig. S2) and  $\mathbf{1}^{2\bullet}(\mathbf{K}^+)_2$  in the triplet state (Fig. S3) obtained by the DFT quantum chemical calculation (UB3LYP/6-311G<sup>\*</sup>)



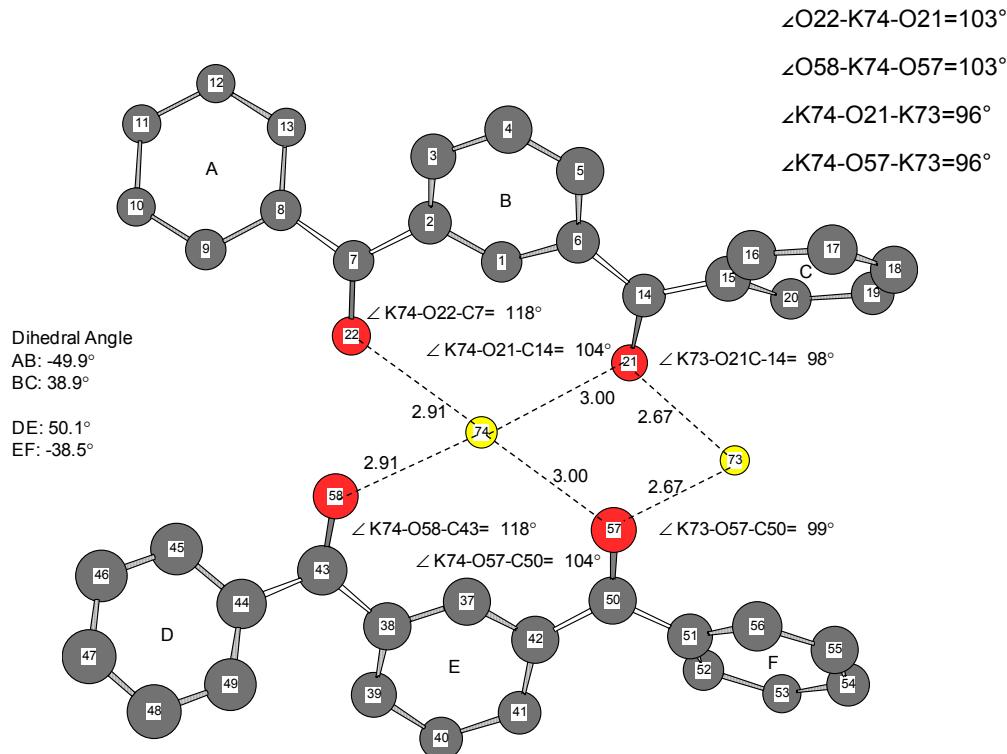
**Fig. S2** The optimized molecular structure for  $\mathbf{1}^{\bullet}\mathbf{K}^+$  in the spin-doublet state obtained by the DFT quantum chemical calculation (UB3LYP/6-311G<sup>\*</sup>).



**Fig. S3** The optimized structure for  $\mathbf{1}^{2\bullet}(\mathbf{K}^+)_2$  in the triplet state obtained by the DFT quantum chemical calculation (UB3LYP/6-311G<sup>\*</sup>).

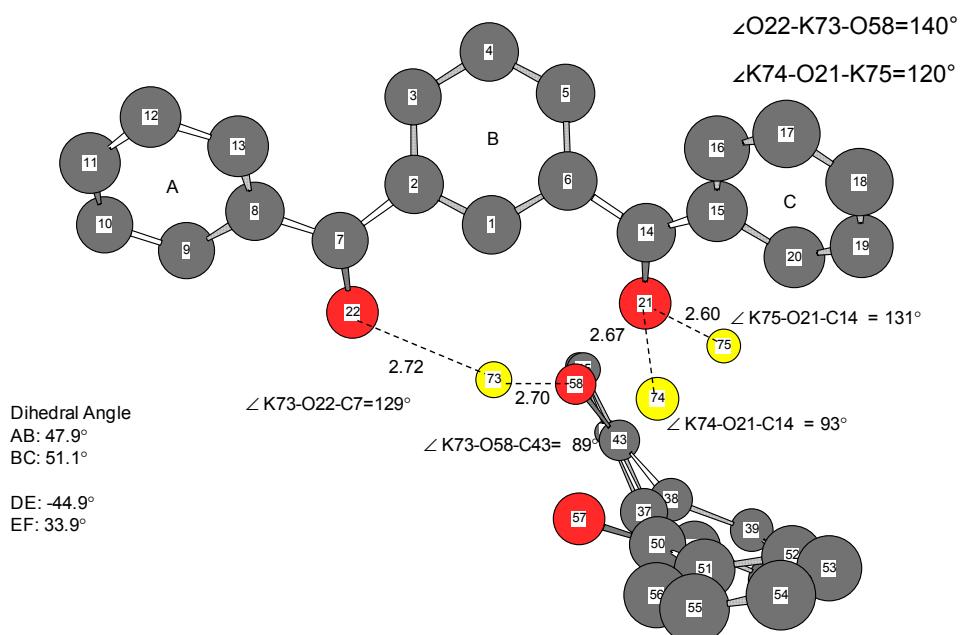
### 3. The optimized molecular structures for *m*-benzoylbenzene-based polyanionic high-spin clusters

#### 3.1 The optimized structure for $1^{-\bullet}(K^+)_21^{-\bullet}$ in the triplet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G) (Fig. S4).



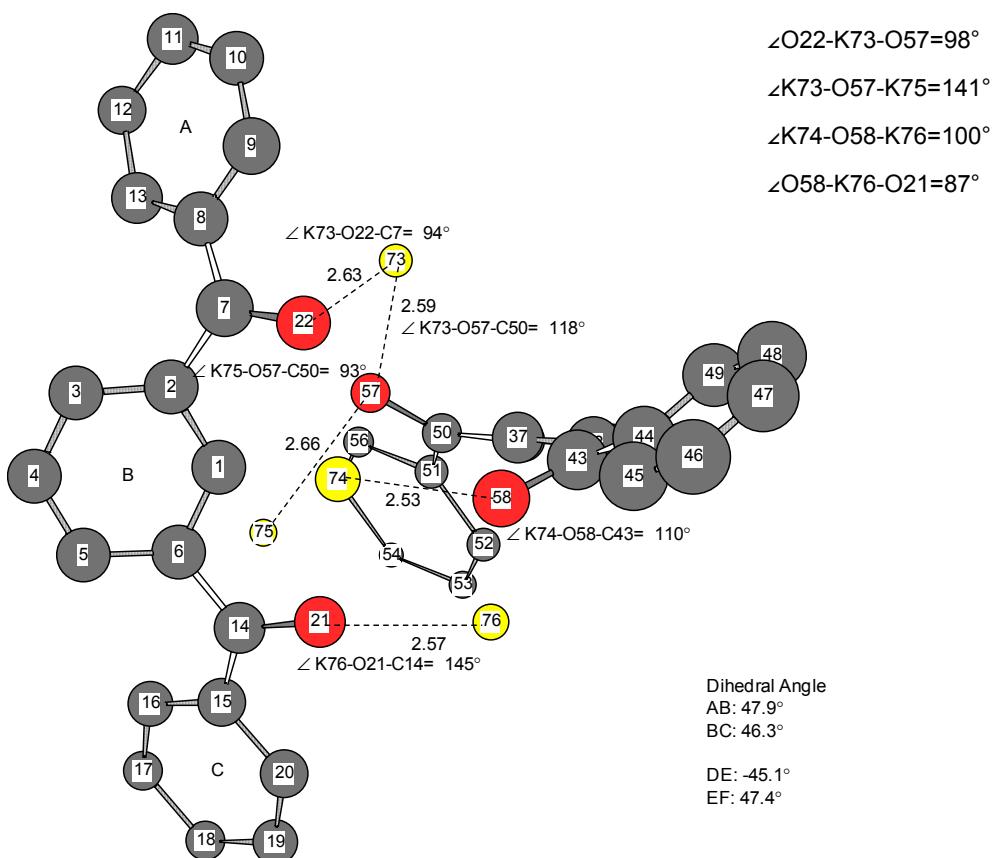
**Fig. S4** The optimized structure for  $1^{-\bullet}(K^+)_21^{-\bullet}$  in the triplet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G).

#### 3.2 The optimized structure for $1^{2-\bullet}(K^+)_31^{-\bullet}$ in the quartet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G) (Fig. S5).



**Fig. S5** The optimized structure for  $\mathbf{1}^{2-\bullet}(\text{K}^+)_3\mathbf{1}^{-\bullet}$  in the quartet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G).

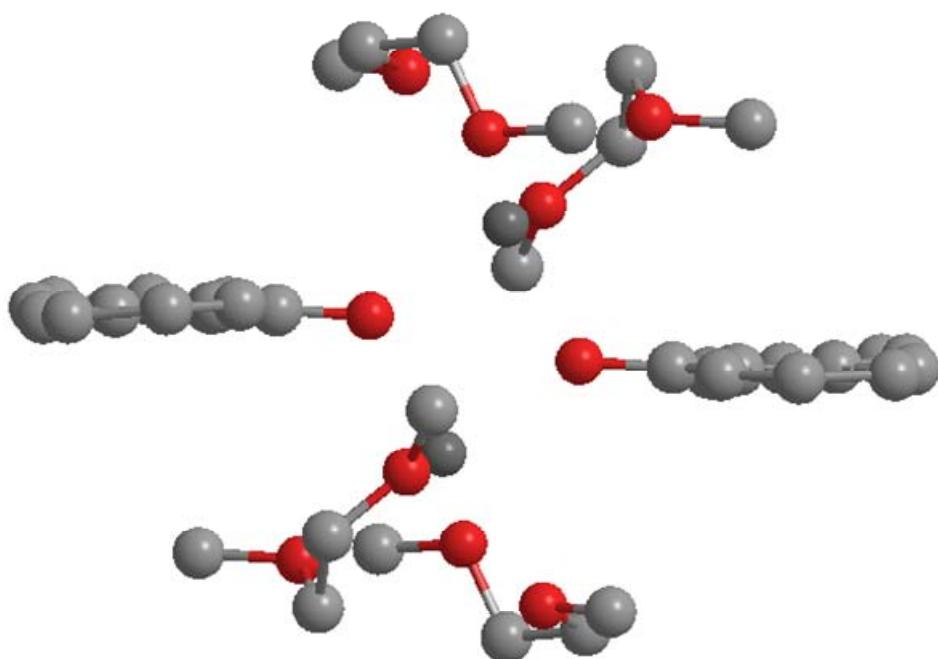
### 3.3 The optimized structure for $1^{\text{--}}(\text{K}^+)_4 1^{\text{--}}$ in the quintet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G) (Fig. S6).



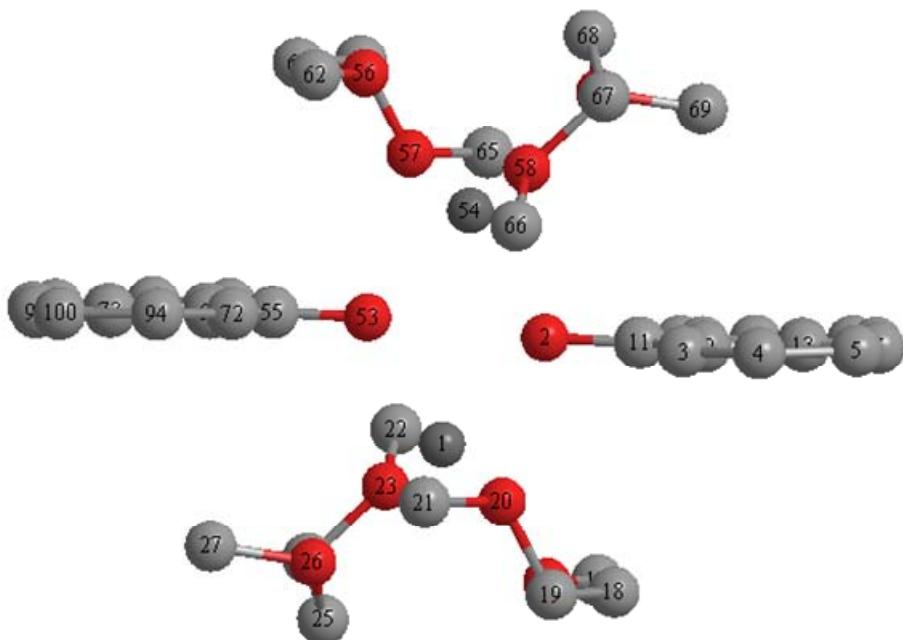
**Fig. S6** The optimized structure for  $\mathbf{1}^{2-\bullet}(\text{K}^+)_4\mathbf{1}^{2-\bullet}$  in the quintet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G).

**4. The clustering structure of the dimer, [Fluorenone<sup>•</sup> {Na<sup>+</sup>(dimethoxyethan)<sub>2</sub>}]<sub>2</sub> (Fig. S7), and the comparison of the lengths, Na-O, Na-Na, and O-O and angles, O-Na-O, and Na-O-Na between the X-ray single-crystal analysis and the structural optimization (Fig. S7).**

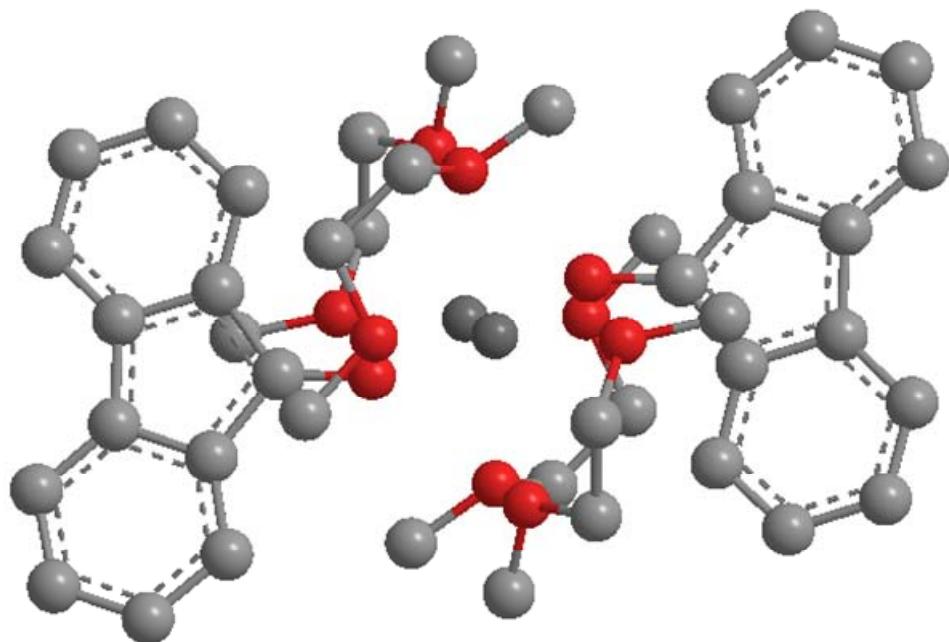
(a)



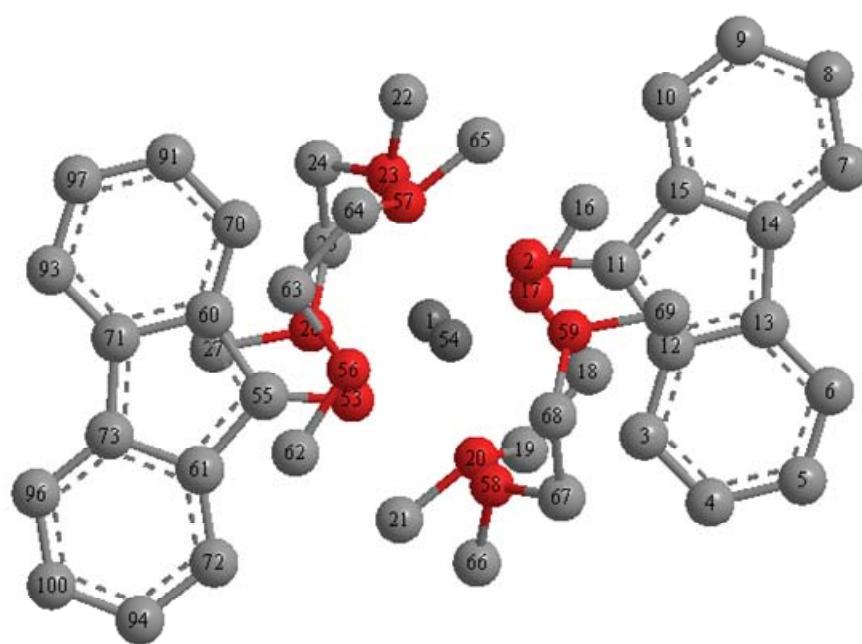
(b)



(c)



(d)



(e)

|                      | X-ray Analysis | Optimization |
|----------------------|----------------|--------------|
| <b>Length/Å</b>      |                |              |
| <b>Na1-O53</b>       | <b>2.398</b>   | <b>2.344</b> |
| <b>Na1-O2</b>        | <b>2.282</b>   | <b>2.325</b> |
| <b>Na54-O53</b>      | <b>2.283</b>   | <b>2.324</b> |
| <b>Na54-O2</b>       | <b>2.398</b>   | <b>2.344</b> |
| <b>Na1-Na54</b>      | <b>3.350</b>   | <b>3.270</b> |
| <b>O53-O2</b>        | <b>3.270</b>   | <b>3.333</b> |
| <b>Angle/Degrees</b> |                |              |
| <b>O53-Na1-O2</b>    | <b>88.6</b>    | <b>91.1</b>  |
| <b>O53-Na54-O2</b>   | <b>88.6</b>    | <b>91.1</b>  |
| <b>Na1-O53-Na54</b>  | <b>91.4</b>    | <b>88.9</b>  |
| <b>Na1-O2-Na54</b>   | <b>91.4</b>    | <b>88.9</b>  |

**Fig. S7** The clustering structure of the dimer, [Fluorenone<sup>-•</sup> {Na<sup>+</sup>(dimethoxyethan)<sub>2</sub>}]<sub>2</sub>. **a, c:** The structures of the dimeric cluster, [Fluorenone<sup>-•</sup> {Na<sup>+</sup>(dimethoxyethan)<sub>2</sub>}]<sub>2</sub> determined by X-ray analysis. **b, d:** The optimized structures for [Fluorenone<sup>-•</sup> {Na<sup>+</sup>(dme)<sub>2</sub>}]<sub>2</sub> in the triplet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G). **e:** The comparison of the lengths, Na-O, Na-Na, and O-O and angles, O-Na-O, and Na-O-Na between the X-ray single-crystal analysis and the structural optimization.

**5. Mulliken atomic charges for [Fluorenone<sup>-•</sup> {Na<sup>+</sup>(dimethoxyethan)<sub>2</sub>}]<sub>2</sub> (Tables S1 and S2)**

**Table S1 Mulliken atomic charges for [Fluorenone<sup>-•</sup> {Na<sup>+</sup>(dimethoxyethan)<sub>2</sub>}]<sub>2</sub>**  
 Note that atomic charges with hydrogens are non-vanishing and summed into the attached carbon atoms.

|     |          |      |          |     |          |      |          |
|-----|----------|------|----------|-----|----------|------|----------|
| 1Na | .156597  | 31H  | .000000  | 61C | -.031762 | 91C  | -.041327 |
| 2O  | -.600824 | 32H  | .000000  | 62C | .312250  | 92H  | .000000  |
| 3C  | -.024160 | 33H  | .000000  | 63C | .300060  | 93C  | -.066469 |
| 4C  | -.041288 | 34H  | .000000  | 64C | .283897  | 94C  | -.021034 |
| 5C  | -.027989 | 35H  | .000000  | 65C | .297786  | 95H  | .000000  |
| 6C  | -.066458 | 36H  | .000000  | 66C | .291617  | 96C  | -.061849 |
| 7C  | -.061845 | 37H  | .000000  | 67C | .273595  | 97C  | -.028005 |
| 8C  | -.021376 | 38H  | .000000  | 68C | .272017  | 98H  | .000000  |
| 9C  | -.021030 | 39H  | .000000  | 69C | .307736  | 99H  | .000000  |
| 10C | -.061624 | 40H  | .000000  | 70C | -.024047 | 100C | -.021379 |
| 11C | .319763  | 41H  | .000000  | 71C | .007416  | 101H | .000000  |
| 12C | -.042658 | 42H  | .000000  | 72C | -.061615 | 102H | .000000  |
| 13C | .007414  | 43H  | .000000  | 73C | .009115  | 103H | .000000  |
| 14C | .009118  | 44H  | .000000  | 74H | .000000  | 104H | .000000  |
| 15C | -.031774 | 45H  | .000000  | 75H | .000000  | 105H | .000000  |
| 16C | .312336  | 46H  | .000000  | 76H | .000000  | 106H | .000000  |
| 17O | -.457486 | 47H  | .000000  | 77H | .000000  | 107H | .000000  |
| 18C | .300058  | 48H  | .000000  | 78H | .000000  | 108H | .000000  |
| 19C | .283903  | 49H  | .000000  | 79H | .000000  | 109H | .000000  |
| 20O | -.458500 | 50H  | .000000  | 80H | .000000  | 110H | .000000  |
| 21C | .297792  | 51H  | .000000  | 81H | .000000  |      |          |
| 22C | .291609  | 52H  | .000000  | 82H | .000000  |      |          |
| 23O | -.467870 | 53O  | -.600851 | 83H | .000000  |      |          |
| 24C | .273573  | 54Na | .156432  | 84H | .000000  |      |          |
| 25C | .271950  | 55C  | .319771  | 85H | .000000  |      |          |
| 26O | -.446884 | 56O  | -.457479 | 86H | .000000  |      |          |
| 27C | .307745  | 57O  | -.458508 | 87H | .000000  |      |          |
| 28H | .000000  | 58O  | -.467853 | 88H | .000000  |      |          |
| 29H | .000000  | 59O  | -.446884 | 89H | .000000  |      |          |
| 30H | .000000  | 60C  | -.042725 | 90H | .000000  |      |          |

**Table S2 Mulliken atomic spin densities for [Fluorenone<sup>-•</sup>{Na<sup>+</sup>(dimethoxyethan)<sub>2</sub>}]<sub>2</sub>**

|     |                 |      |                 |     |                 |      |                 |
|-----|-----------------|------|-----------------|-----|-----------------|------|-----------------|
| 1Na | <b>-.002399</b> | 31H  | <b>.002086</b>  | 61C | <b>-.026853</b> | 91C  | <b>-.035931</b> |
| 2O  | <b>.239565</b>  | 32H  | <b>.002109</b>  | 62C | <b>.003019</b>  | 92H  | <b>-.004706</b> |
| 3C  | <b>.094146</b>  | 33H  | <b>-.007048</b> | 63C | <b>.000212</b>  | 93C  | <b>-.064346</b> |
| 4C  | <b>-.035943</b> | 34H  | <b>.001021</b>  | 64C | <b>.000105</b>  | 94C  | <b>-.040127</b> |
| 5C  | <b>.143915</b>  | 35H  | <b>-.005068</b> | 65C | <b>.000306</b>  | 95H  | <b>-.005068</b> |
| 6C  | <b>-.064352</b> | 36H  | <b>-.000151</b> | 66C | <b>.000232</b>  | 96C  | <b>-.064549</b> |
| 7C  | <b>-.064545</b> | 37H  | <b>.000169</b>  | 67C | <b>.000414</b>  | 97C  | <b>.143911</b>  |
| 8C  | <b>.142852</b>  | 38H  | <b>.001039</b>  | 68C | <b>-.000033</b> | 98H  | <b>.000857</b>  |
| 9C  | <b>-.040123</b> | 39H  | <b>.000033</b>  | 69C | <b>.001633</b>  | 99H  | <b>.002086</b>  |
| 10C | <b>.102752</b>  | 40H  | <b>-.000671</b> | 70C | <b>.094144</b>  | 100C | <b>.142861</b>  |
| 11C | <b>.307343</b>  | 41H  | <b>.000007</b>  | 71C | <b>.120613</b>  | 101H | <b>.001021</b>  |
| 12C | <b>-.024503</b> | 42H  | <b>-.000012</b> | 72C | <b>.102762</b>  | 102H | <b>.002109</b>  |
| 13C | <b>.120609</b>  | 43H  | <b>-.000078</b> | 73C | <b>.118429</b>  | 103H | <b>-.007134</b> |
| 14C | <b>.118419</b>  | 44H  | <b>-.000378</b> | 74H | <b>-.000150</b> | 104H | <b>-.007049</b> |
| 15C | <b>-.026842</b> | 45H  | <b>-.000027</b> | 75H | <b>.000168</b>  | 105H | <b>-.000173</b> |
| 16C | <b>.003025</b>  | 46H  | <b>.000253</b>  | 76H | <b>.001036</b>  | 106H | <b>.000820</b>  |
| 17O | <b>.000220</b>  | 47H  | <b>-.000019</b> | 77H | <b>.000033</b>  | 107H | <b>-.000204</b> |
| 18C | <b>.000213</b>  | 48H  | <b>-.000688</b> | 78H | <b>-.000672</b> | 108H | <b>-.000205</b> |
| 19C | <b>.000105</b>  | 49H  | <b>.000178</b>  | 79H | <b>.000007</b>  | 109H | <b>.000822</b>  |
| 20O | <b>.000304</b>  | 50H  | <b>.000006</b>  | 80H | <b>-.000012</b> | 110H | <b>-.000173</b> |
| 21C | <b>.000306</b>  | 51H  | <b>-.000017</b> | 81H | <b>-.000078</b> |      |                 |
| 22C | <b>.000231</b>  | 52H  | <b>-.000005</b> | 82H | <b>-.000378</b> |      |                 |
| 23O | <b>.000232</b>  | 53O  | <b>.239557</b>  | 83H | <b>-.000027</b> |      |                 |
| 24C | <b>.000414</b>  | 54Na | <b>-.002391</b> | 84H | <b>.000255</b>  |      |                 |
| 25C | <b>-.000032</b> | 55C  | <b>.307371</b>  | 85H | <b>-.000019</b> |      |                 |
| 26O | <b>.000247</b>  | 56O  | <b>.000221</b>  | 86H | <b>-.000688</b> |      |                 |
| 27C | <b>.001633</b>  | 57O  | <b>.000304</b>  | 87H | <b>.000177</b>  |      |                 |
| 28H | <b>-.004706</b> | 58O  | <b>.000230</b>  | 88H | <b>.000006</b>  |      |                 |
| 29H | <b>.000857</b>  | 59O  | <b>.000248</b>  | 89H | <b>-.000017</b> |      |                 |
| 30H | <b>-.007134</b> | 60C  | <b>-.024529</b> | 90H | <b>-.000005</b> |      |                 |

## 6. The Calculation of $D_i$ of the intramolecular interaction in terms of point-dipole approximation

Dipolar spin-spin interaction between electron spins,  $S_1$  and  $S_2$  is given by

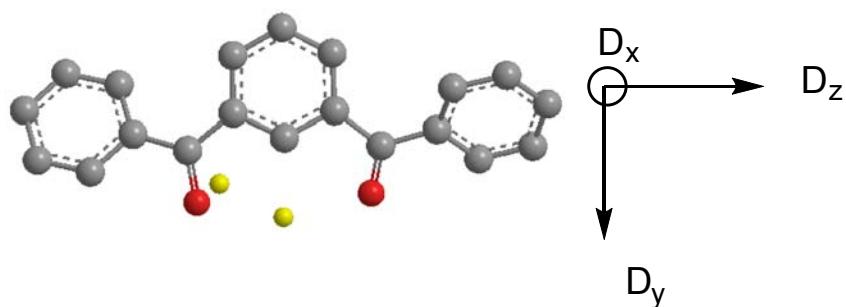
$$W_{ss} = \left( \frac{\mu_0}{4\pi} \right) (g\beta)^2 \left\{ \frac{S_1 \bullet S_2}{r_{12}^3} - \frac{3(S_1 \bullet r_{12})(S_2 \bullet r_{12})}{r_{12}^5} \right\}, \quad (\text{S1})$$

$\mathbf{r}_{12}$  is the distance vector from an electron i to j,  $\mu_0$  is the permeability of free space, and  $\beta$  is Bohr magneton. The  $D$  tensor corresponding to the dipolar interaction can be calculated in terms of a wave function  $\Psi_T$ ,

$$\mathbf{D} = \int \psi_T^* W_{ss} \psi_T d\nu_1 d\nu_2. \quad (\text{S2})$$

The calculation of the  $\mathbf{D}$  tensor for the intramolecular triplet state was made by the point-dipole approximation with the optimized molecular structure and the coefficients of  $2p_{x,y,z}$  of the LUMOs and next LUMOs with  $\alpha$  spins. We assume that each  $2p_{x,y,z}$  orbital is approximated in terms of a pair of points separated by 0.68 Å from the position of the carbon atom along the x, y, z directions. The MOs were calculated by the DFT method (USVWN/sto-3g). The  $D$  and  $E$  values are calculated to be  $-0.00427 \text{ cm}^{-1}$  and  $0.00108 \text{ cm}^{-1}$ , respectively. The sign of the  $D$  value is negative. The principal direction for  $D_z$  is close to along the long-axis of the molecule.

### 6.1 The Calculation of the $D_{12}$ tensors for the doublet, quartet, and quintet clusters in terms of point-dipole approximation



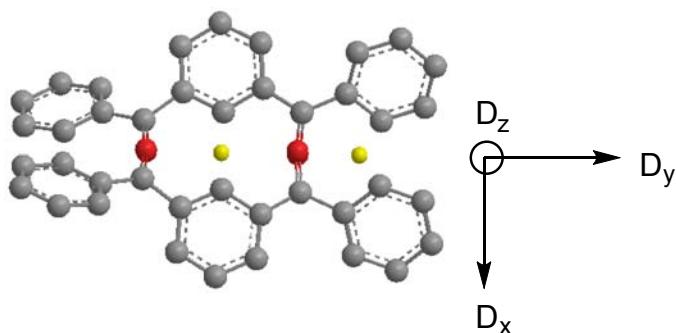
**Fig. S8.** The orientation of the principal axes of the  $\mathbf{D}_1$  tensor for the intramolecular triplet state of *m*-benzoylbenzene dianion.

The  $\mathbf{D}_{12}$  tensors for the quartet and quintet states of the molecular clusters were estimated by a semi-empirical method with both the observed  $D$  value for the triplet cluster and the theoretical  $D$  value obtained by the point-dipole approximation approach, in which the optimized molecular structure and the calculated  $\alpha$  spin density distribution are employed. The spin density distributions were calculated by the DFT method (UB3LYP/6-31G\*//UB3LYP/3-21G). The calculated  $\mathbf{D}_{12}^c$  was obtained by the following equation,

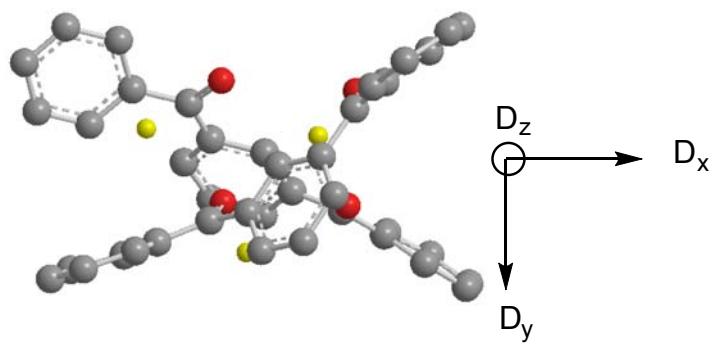
$$\mathbf{D}_{12}^c = \sum_{i,j} \rho_i W_{ss} \rho_j \quad (\text{S3})$$

where  $\rho_i$  denotes the spin density on the  $i_{\text{th}}$  atom in one dibenzoylbenzene moiety and  $\rho_j$  the spin density on the  $j_{\text{th}}$  atom in the other dibenzoylbenzene moiety. The calculated results for the high-spin clusters are in the following:

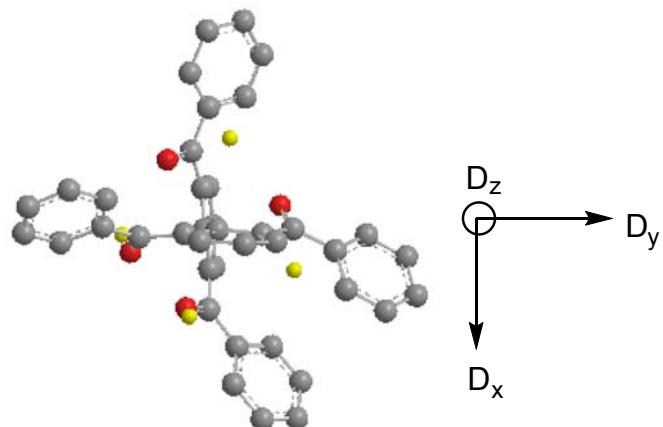
$$\begin{aligned} \mathbf{D}_{12}^c(\text{triplet cluster}) &= -0.0041 \text{ cm}^{-1}, \\ \mathbf{D}_{12}^c(\text{quartet cluster}) &= -0.0064 \text{ cm}^{-1}, \\ \mathbf{D}_{12}^c(\text{quintet cluster}) &= -0.0070 \text{ cm}^{-1}. \end{aligned}$$



**Fig. S9** The principal axes of the  $\mathbf{D}$  tensor for the intermolecular dianionic triplet state.



**Fig. S10** The principal axes of the  $\mathbf{D}_{12}$  tensor for the tetraanionic quartet state.  
The direction of  $D_z$  is along the direction connected between the dibenzoylbenzene  
molecules.



**Fig. S11** The principal axes of the  $\mathbf{D}_{12}$  tensor for the tetraanionic quintet state.  
The direction of  $D_z$  is along the direction connected between the dibenzoylbenzene  
molecules.

Table S3. The  $D_{12}$ -values for the triplet, quartet, and quintet clusters calculated by Equation S3 in terms of the optimized molecular structures and the calculated spin density distributions.

| $D_{12} / \text{cm}^{-1}$ | Calculated |           |         |
|---------------------------|------------|-----------|---------|
|                           | $S = 1$    | $S = 3/2$ | $S = 2$ |
|                           | -0.0041    | -0.0064   | -0.0070 |

The  $\mathbf{D}_{12}$  tensors for the quartet and quintet clusters were estimated by the following equations which include  $\mathbf{D}_{12}$  derived from the observed  $D$  value of the dianionic triplet cluster and the calculated  $D_{12}^c$ -values.

$$\begin{aligned} D_{12}(\text{quartet cluster}) &= D_{12}(\text{triplet cluster}) \times \frac{D_{12}^c(\text{quartet cluster})}{D_{12}^c(\text{triplet cluster})} \\ D_{12}(\text{quintet cluster}) &= D_{12}(\text{triplet cluster}) \times \frac{D_{12}^c(\text{quintet cluster})}{D_{12}^c(\text{triplet cluster})}. \end{aligned} \quad (\text{S4})$$

The results are as follows:

$$\begin{aligned} D_{12}(\text{quartet}) &= -0.0281 \text{ cm}^{-1}, \\ D_{12}(\text{quintet}) &= -0.0307 \text{ cm}^{-1}. \end{aligned}$$

Table S4. The  $D_{12}$ -values for the high-spin clusters estimated by Equations S4.

| $D_{12} / \text{cm}^{-1}$ | Calculated |           |         |
|---------------------------|------------|-----------|---------|
|                           | $S = 1$    | $S = 3/2$ | $S = 2$ |
|                           | -0.018     | -0.0281   | -0.0307 |

## 6.2 The **D** and **E** values for the quartet and quintet molecular clusters

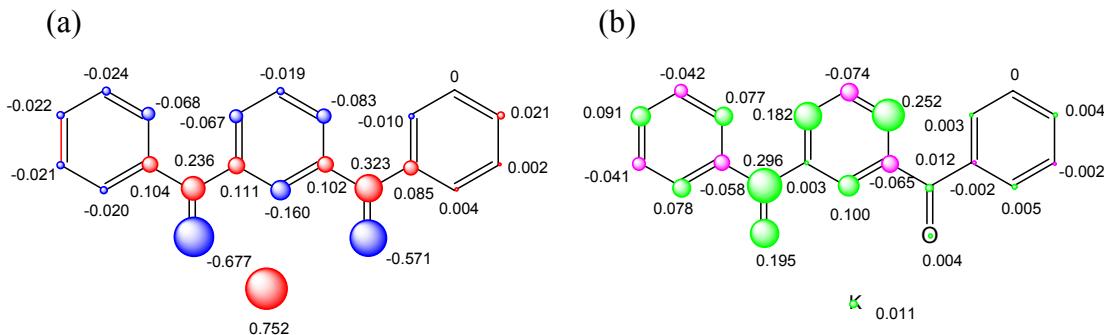
The **D** tensor for the quartet cluster is calculated by using **D<sub>1</sub>** and **D<sub>12</sub>** (quartet cluster). The *D* and *E* values are -0.0135 cm<sup>-1</sup> and 0.00097 cm<sup>-1</sup>, respectively.

The **D** tensor for the quintet cluster is calculated by Equation 12 with **D<sub>1</sub>** and **D<sub>12</sub>** (quintet cluster). The *D* and *E* values are -0.0096 cm<sup>-1</sup> and 0.0 cm<sup>-1</sup>, respectively.

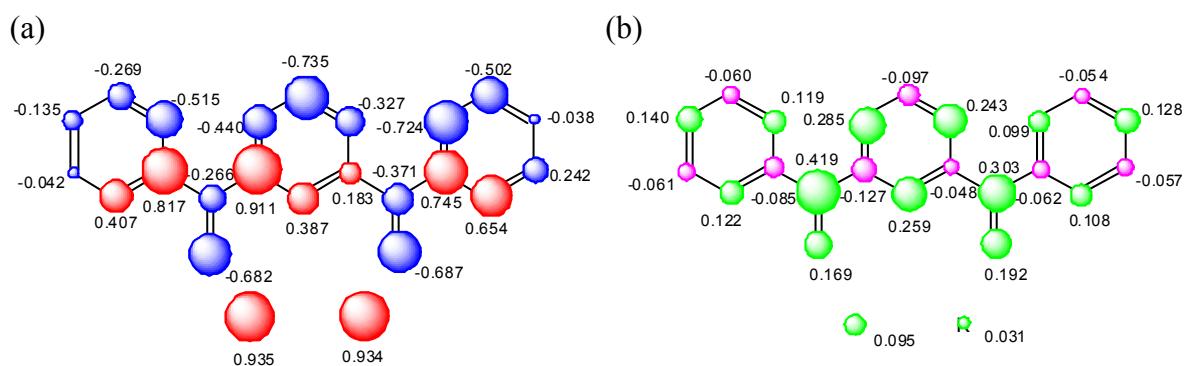
Table S5. The calculated *D* and *E* values for the quartet and quintet clusters compared with the observed values.

|                             | Observed | Calculated |
|-----------------------------|----------|------------|
| <i>S</i> = 3/2              |          |            |
| <i>D</i> / cm <sup>-1</sup> | 0.0066   | -0.0135    |
| <i>E</i> / cm <sup>-1</sup> | <0.0002  | 0.00097    |
| <i>S</i> = 2                |          |            |
| <i>D</i> / cm <sup>-1</sup> | 0.00585  | -0.0096    |
| <i>E</i> / cm <sup>-1</sup> | <0.0002  | 0.0        |

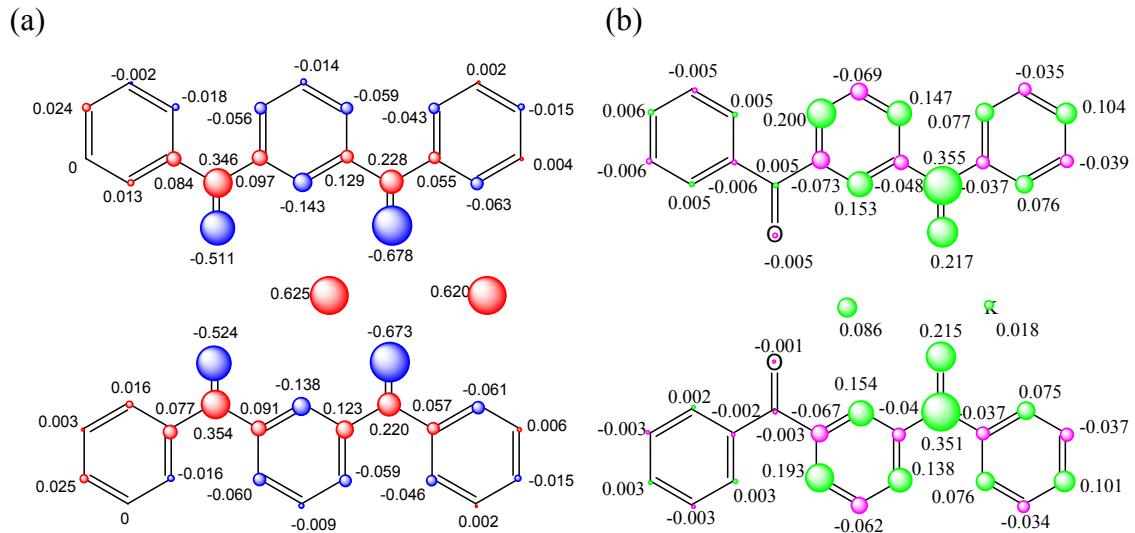
**7. The charge density and spin density for  $1^{\bullet}\text{K}^+$  in the doublet state,  $1^{2(-\bullet)}\text{K}_2$  in the triplet ground state,  $1^{\bullet}(\text{K}^+)_21^{\bullet}$  in the triplet state,  $1^{2(-\bullet)}(\text{K}^+)_31^{(-\bullet)}$  in the quartet state,  $1^{2(-\bullet)}(\text{K}^+)_41^{2(-\bullet)}$  in the quintet state, as calculated by the DFT quantum chemical calculations (UB3LYP/6-31G\*//UB3LYP/3-21G)**



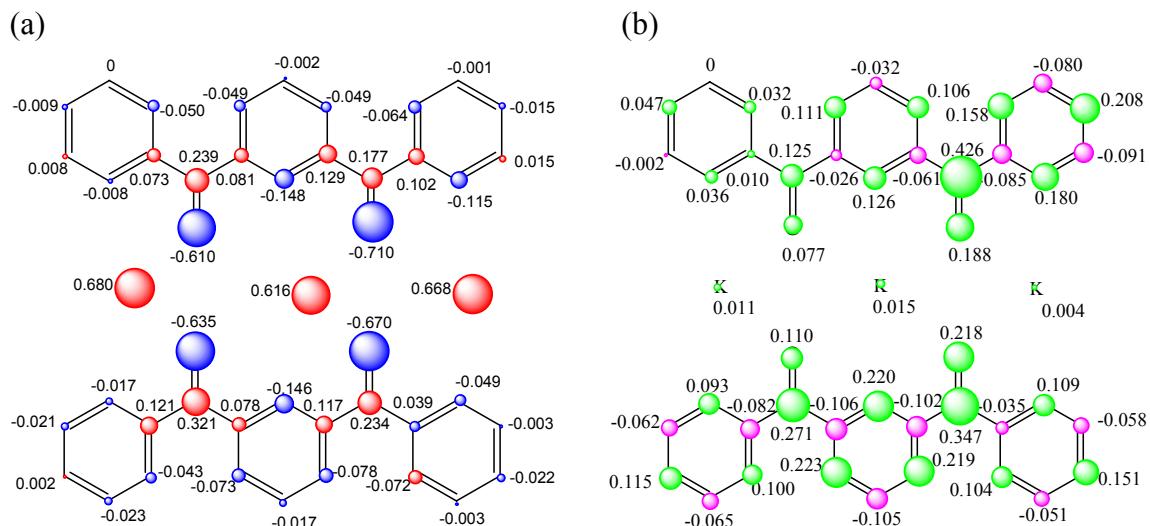
**Fig. S12** (a) The charge density and (b) the spin density for  $1^{\bullet}\text{K}^+$  in the doublet state, as calculated by the DFT quantum chemical calculations (UB3LYP/6-31G\*//UB3LYP/3-21G). The red and blue colors indicate positive and negative charge densities, respectively. The green and purple colors indicate  $\alpha$  and  $\beta$  spin densities, respectively.



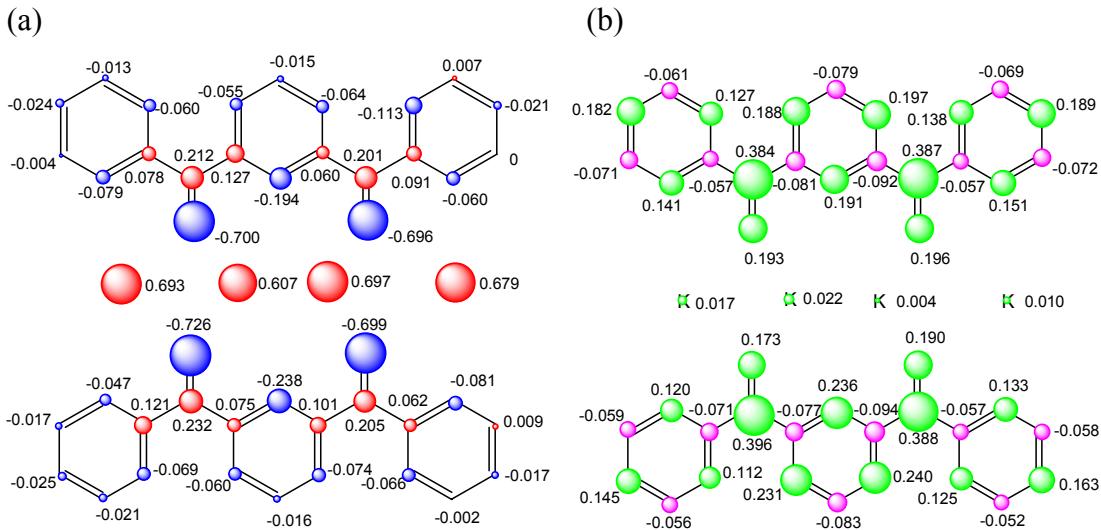
**Fig. S13** (a) The charge density and (b) the spin density for  $1^{2(-\bullet)}\text{K}_2$  in the triplet ground state, as calculated by the DFT quantum chemical calculations (UB3LYP/6-31G\*//UB3LYP/3-21G). The red and blue colors indicate positive and negative charge densities, respectively. The green and purple colors indicate  $\alpha$  and  $\beta$  spin densities, respectively.



**Fig. S14** (a) The charge density and (b) the spin density for  $\mathbf{1}^-(\mathbf{K}^+)_2 \mathbf{1}^-$  in the triplet state calculated by DFT method (UB3LYP/6-31G\*//UB3LYP/3-21G). The red and blue colors indicate positive and negative charge densities, respectively. The green and purple colors indicate  $\alpha$  and  $\beta$  spin densities, respectively.

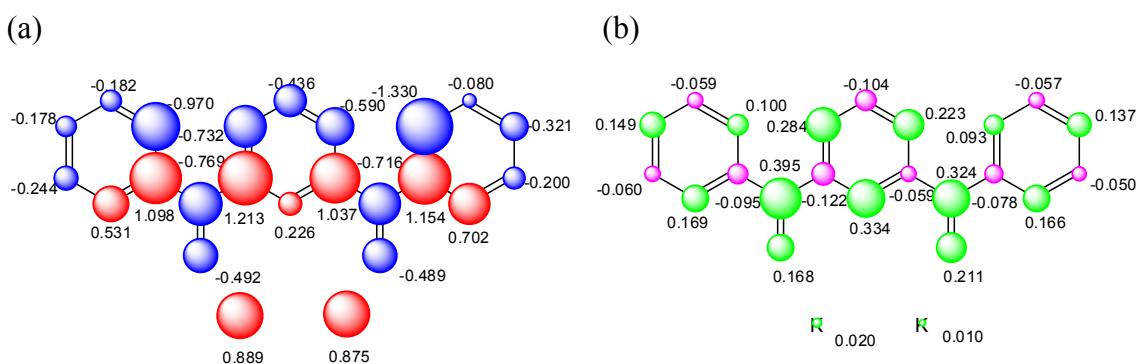


**Fig. S15** (a) The charge density and (b) the spin density for  $\mathbf{1}^2(-\bullet)(\mathbf{K}^+)_3 \mathbf{1}^{(-\bullet)}$  in the quartet state calculated by DFT method (UB3LYP/6-31G\*//UB3LYP/3-21G). The red and blue colors indicate positive and negative charge densities, respectively. The green and purple colors indicate  $\alpha$  and  $\beta$  spin densities, respectively.



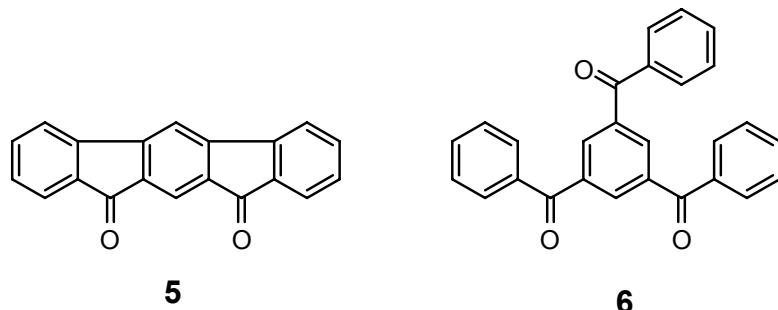
**Fig. S16** (a) The charge density and (b) the spin density for  $\mathbf{1}^{2(-\bullet)}(\mathbf{K}^+)_4\mathbf{1}^{2(-\bullet)}$  in the quintet state calculated by DFT method (UB3LYP/6-31G\*//UB3LYP/3-21G). The red and blue colors indicate positive and negative charge densities, respectively. The green and purple colors indicate  $\alpha$  and  $\beta$  spin densities, respectively.

### 8. The charge density distribution and the spin density distribution for $\mathbf{1}^{2(-\bullet)}(\mathbf{K}^+)_2$ in the triplet state calculated by DFT method (UB3LYP/6-311++G\*\* //UB3LYP/6-311G\*)



**Fig. S17** (a) The charge density distribution and (b) the spin density distribution for  $\mathbf{1}^{2(-\bullet)}(\mathbf{K}^+)_2$  in the triplet state calculated by DFT method (UB3LYP/6-311++G\*\* //UB3LYP/6-311G\*). The red and blue colors indicate positive and negative charge densities, respectively. The green and purple colors indicate  $\alpha$  and  $\beta$  spin densities, respectively.

**9. The molecular structures of fluorenone-based diketone **5** and 1,3,5- tribenzoyl benzene **6****



**Fig. S18** Fluorenone-based diketone **5** and 1,3,5- tribenzoyl benzene **6**.

**10. Atomic coordinates of the optimized structures for  $1^{\bullet}\text{K}^+$  and the molecular high $\theta$ spin clusters obtained by the DFT quantum chemical calculations**

**10.1 Atomic coordinates of the optimized structure for  $1^{\bullet}\text{K}^+$  in the spin-doublet state obtained by the DFT quantum chemical calculation (UB3LYP/6-311G\*)**

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -0.075918 | 0.114960  | 0.353161  |
| C | 1.186456  | -0.486313 | 0.087203  |
| C | 1.113522  | -1.820805 | -0.411820 |
| C | -0.093305 | -2.381678 | -0.796573 |
| C | -1.306499 | -1.675389 | -0.700716 |
| C | -1.286301 | -0.409487 | -0.095001 |
| C | 2.349591  | 0.351865  | 0.295442  |
| C | 3.751554  | -0.120821 | 0.208791  |
| C | 4.699506  | 0.491208  | 1.054107  |
| C | 6.039203  | 0.130323  | 1.018765  |
| C | 6.488289  | -0.837424 | 0.117389  |
| C | 5.574419  | -1.425389 | -0.754527 |
| C | 4.227522  | -1.073490 | -0.712728 |
| C | -2.456416 | 0.510081  | -0.061593 |
| C | -3.844345 | -0.006149 | 0.099460  |
| C | -4.116809 | -1.221494 | 0.743878  |
| C | -5.432590 | -1.641830 | 0.921737  |
| C | -6.485981 | -0.861119 | 0.451641  |
| C | -6.224002 | 0.353532  | -0.186113 |
| C | -4.914267 | 0.782437  | -0.350285 |
| O | -2.268145 | 1.727548  | -0.184042 |
| O | 2.156578  | 1.587694  | 0.617307  |
| H | -0.090072 | 1.001626  | 0.970704  |
| H | 2.011125  | -2.418881 | -0.501766 |
| H | -0.097934 | -3.387290 | -1.206973 |
| H | -2.218534 | -2.087836 | -1.114847 |
| H | 4.350853  | 1.256275  | 1.737602  |

|   |           |           |           |
|---|-----------|-----------|-----------|
| H | 6.741791  | 0.607641  | 1.696162  |
| H | 7.536704  | -1.117256 | 0.086234  |
| H | 5.913004  | -2.154460 | -1.485302 |
| H | 3.556399  | -1.506393 | -1.444090 |
| H | -3.300260 | -1.823181 | 1.124130  |
| H | -5.634647 | -2.577840 | 1.432111  |
| H | -7.510422 | -1.193425 | 0.586547  |
| H | -7.044239 | 0.964509  | -0.548986 |
| H | -4.694743 | 1.730503  | -0.827241 |
| K | 0.326934  | 2.818209  | -0.613880 |

**10.2 Atomic coordinates of the optimized structure for  $\text{1}^{2-\bullet}(\text{K}^+)_2$  in the triplet state obtained by the DFT quantum chemical calculation (UB3LYP/6-311G\*)**

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.181073  | 0.008300  | 0.286863  |
| C | -1.073942 | 0.633613  | 0.091712  |
| C | -1.024704 | 2.035695  | -0.161037 |
| C | 0.217519  | 2.664904  | -0.318426 |
| C | 1.428177  | 1.977133  | -0.287594 |
| C | 1.438942  | 0.580131  | -0.012371 |
| C | -2.244382 | -0.246808 | 0.161349  |
| C | -3.615844 | 0.131103  | -0.198000 |
| C | -4.676280 | -0.676599 | 0.281822  |
| C | -6.000461 | -0.396208 | -0.024324 |
| C | -6.326526 | 0.688191  | -0.841874 |
| C | -5.295539 | 1.471968  | -1.362683 |
| C | -3.967754 | 1.201714  | -1.053667 |
| C | 2.523081  | -0.378279 | -0.191957 |
| C | 3.943530  | -0.012649 | -0.147400 |
| C | 4.442708  | 1.152803  | 0.475569  |
| C | 5.805225  | 1.425304  | 0.508418  |
| C | 6.721766  | 0.544485  | -0.069336 |
| C | 6.249269  | -0.627093 | -0.666555 |
| C | 4.890372  | -0.905342 | -0.698365 |
| O | 2.202861  | -1.616333 | -0.409655 |
| O | -2.045930 | -1.474144 | 0.603003  |
| H | 0.201418  | -1.023568 | 0.597774  |
| H | -1.926665 | 2.628875  | -0.219314 |
| H | 0.231265  | 3.732053  | -0.526302 |
| H | 2.340899  | 2.499333  | -0.546051 |
| H | -4.422587 | -1.538987 | 0.886695  |
| H | -6.787870 | -1.032070 | 0.371376  |
| H | -7.361735 | 0.906225  | -1.084235 |
| H | -5.527411 | 2.296008  | -2.031557 |
| H | -3.194274 | 1.795633  | -1.521642 |
| H | 3.758774  | 1.832225  | 0.970824  |
| H | 6.157898  | 2.327447  | 1.001144  |
| H | 7.785297  | 0.761013  | -0.043666 |
| H | 6.951217  | -1.329201 | -1.108862 |
| H | 4.520042  | -1.819084 | -1.147443 |
| K | -0.089524 | -2.339640 | -1.324743 |
| K | -1.262008 | -0.552271 | 2.810091  |

**10.3 Atomic coordinates of the optimized structure for  $1^{\bullet}(\text{K}^+)_21^{\bullet}$  in the triplet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G)**

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 0.303191  | -2.943909 | -0.422439 |
| C | 1.561481  | -3.381919 | -0.018953 |
| C | 1.670639  | -4.299217 | 1.045478  |
| C | 0.489926  | -4.789634 | 1.627849  |
| C | -0.762463 | -4.331993 | 1.236713  |
| C | -0.891582 | -3.324042 | 0.235976  |
| C | 2.722719  | -2.697624 | -0.655360 |
| C | 4.007281  | -3.420727 | -0.858905 |
| C | 5.154909  | -2.651442 | -1.109455 |
| C | 6.379941  | -3.267963 | -1.339524 |
| C | 6.467513  | -4.664794 | -1.343437 |
| C | 5.325238  | -5.438040 | -1.122917 |
| C | 4.097342  | -4.821034 | -0.880642 |
| C | -2.096115 | -2.636425 | -0.178032 |
| C | -3.435289 | -3.002372 | 0.326015  |
| C | -3.738676 | -3.212952 | 1.692696  |
| C | -5.056545 | -3.389340 | 2.116939  |
| C | -6.114639 | -3.359153 | 1.201024  |
| C | -5.835062 | -3.148611 | -0.157599 |
| C | -4.522764 | -2.964411 | -0.586448 |
| 8 | -2.014485 | -1.654941 | -1.065145 |
| 8 | 2.609040  | -1.499693 | -1.014652 |
| H | 0.210066  | -2.331169 | -1.308761 |
| H | 2.639900  | -4.604607 | 1.417459  |
| H | 0.558268  | -5.534218 | 2.414018  |
| H | -1.648388 | -4.752036 | 1.694903  |
| H | 5.055445  | -1.572980 | -1.118928 |
| H | 7.264294  | -2.668523 | -1.521379 |
| H | 7.420725  | -5.147537 | -1.526201 |
| H | 5.389463  | -6.519536 | -1.146512 |
| H | 3.205653  | -5.415688 | -0.729570 |
| H | -2.941005 | -3.190357 | 2.425335  |
| H | -5.261485 | -3.535518 | 3.172322  |
| H | -7.135296 | -3.501123 | 1.536052  |
| H | -6.643885 | -3.144475 | -0.881578 |
| H | -4.300724 | -2.809753 | -1.637577 |
| C | 0.211830  | 2.924677  | 0.404048  |
| C | 1.449938  | 3.432589  | 0.020656  |
| C | 1.517007  | 4.419680  | -0.983118 |
| C | 0.314987  | 4.903475  | -1.525379 |
| C | -0.916898 | 4.377412  | -1.155586 |
| C | -1.001628 | 3.304876  | -0.219718 |
| C | 2.641535  | 2.752919  | 0.602813  |
| C | 3.897600  | 3.511784  | 0.850129  |
| C | 5.076671  | 2.774400  | 1.044884  |
| C | 6.277100  | 3.423210  | 1.312673  |
| C | 6.308471  | 4.819011  | 1.411055  |
| C | 5.134881  | 5.558475  | 1.246488  |
| C | 3.931581  | 4.909913  | 0.966150  |
| C | -2.178857 | 2.552810  | 0.161016  |

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | -3.534771 | 2.902438  | -0.307054 |
| C | -3.861039 | 3.198842  | -1.652611 |
| C | -5.188767 | 3.359768  | -2.051037 |
| C | -6.234851 | 3.228810  | -1.130237 |
| C | -5.932799 | 2.932307  | 0.207368  |
| C | -4.610337 | 2.763255  | 0.610084  |
| O | -2.054195 | 1.524333  | 0.988038  |
| O | 2.578042  | 1.529879  | 0.879690  |
| H | 0.147023  | 2.252548  | 1.249368  |
| H | 2.471699  | 4.783293  | -1.340214 |
| H | 0.349928  | 5.698742  | -2.262568 |
| H | -1.821092 | 4.794400  | -1.579502 |
| H | 5.020762  | 1.694594  | 0.982121  |
| H | 7.185827  | 2.849161  | 1.451154  |
| H | 7.242523  | 5.326405  | 1.623514  |
| H | 5.155532  | 6.637616  | 1.343074  |
| H | 3.015995  | 5.476994  | 0.857831  |
| H | -3.071766 | 3.257289  | -2.392334 |
| H | -5.410510 | 3.573967  | -3.091327 |
| H | -7.263391 | 3.359650  | -1.445056 |
| H | -6.732516 | 2.849503  | 0.936745  |
| H | -4.370803 | 2.544200  | 1.645871  |
| K | -3.918762 | -0.072051 | -0.059947 |
| K | 0.294991  | -0.001516 | -0.087493 |

**10.4 Atomic coordinates of the optimized structure for  $1^{2-\bullet}(K^+)_31^{-\bullet}$  in the quartet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G)**

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 2.340607  | -1.797773 | -0.218769 |
| C | 3.697013  | -1.533276 | -0.381569 |
| C | 4.304851  | -1.839485 | -1.590545 |
| C | 3.535105  | -2.380455 | -2.616393 |
| C | 2.166682  | -2.538297 | -2.477208 |
| C | 1.527930  | -2.215405 | -1.271537 |
| C | 4.359190  | -0.726936 | 0.730278  |
| C | 5.758275  | -1.057966 | 1.212597  |
| C | 6.466998  | -0.091429 | 1.923702  |
| C | 7.739460  | -0.364105 | 2.396271  |
| C | 8.309023  | -1.609580 | 2.168849  |
| C | 7.604470  | -2.580404 | 1.471102  |
| C | 6.333422  | -2.306578 | 0.992887  |
| C | 0.044271  | -2.103874 | -1.090149 |
| C | -0.877845 | -3.207908 | -1.477941 |
| C | -0.425843 | -4.522045 | -1.686120 |
| C | -1.304150 | -5.540642 | -1.993899 |
| C | -2.670288 | -5.295934 | -2.095536 |
| C | -3.140892 | -4.011058 | -1.898810 |
| C | -2.258322 | -2.979296 | -1.601283 |
| O | -0.408052 | -1.051926 | -0.373043 |
| O | 3.729324  | 0.218321  | 1.213893  |
| H | 1.895442  | -1.651802 | 0.755648  |
| H | 5.355255  | -1.633526 | -1.749024 |
| H | 4.006509  | -2.622470 | -3.561311 |

|   |           |           |           |
|---|-----------|-----------|-----------|
| H | 1.580993  | -2.883453 | -3.319732 |
| H | 6.009311  | 0.874841  | 2.099607  |
| H | 8.287621  | 0.392221  | 2.943420  |
| H | 9.303971  | -1.825256 | 2.539248  |
| H | 8.046508  | -3.553677 | 1.300666  |
| H | 5.783498  | -3.068728 | 0.455955  |
| H | 0.629108  | -4.739197 | -1.593339 |
| H | -0.926166 | -6.544419 | -2.146810 |
| H | -3.353823 | -6.100564 | -2.330237 |
| H | -4.199547 | -3.798610 | -1.989018 |
| H | -2.646304 | -1.973144 | -1.512215 |
| C | -1.968420 | 2.155875  | 0.689176  |
| C | -1.708754 | 3.280174  | -0.103912 |
| C | -2.804241 | 4.074270  | -0.463123 |
| C | -4.077319 | 3.726038  | -0.036024 |
| C | -4.316756 | 2.575558  | 0.695117  |
| C | -3.256568 | 1.731863  | 1.068697  |
| C | -0.313604 | 3.445293  | -0.658897 |
| C | 0.346528  | 4.766443  | -0.745080 |
| C | 1.495616  | 4.909630  | -1.538666 |
| C | 2.167715  | 6.112936  | -1.620987 |
| C | 1.719213  | 7.218338  | -0.912174 |
| C | 0.590604  | 7.095430  | -0.115953 |
| C | -0.084345 | 5.892044  | -0.028054 |
| C | -3.352166 | 0.460147  | 1.817328  |
| C | -4.482474 | -0.462579 | 1.738549  |
| C | -5.488717 | -0.435062 | 0.736699  |
| C | -6.467790 | -1.396747 | 0.668150  |
| C | -6.509000 | -2.459667 | 1.569291  |
| C | -5.543318 | -2.522027 | 2.555035  |
| C | -4.554627 | -1.552593 | 2.647776  |
| O | -2.257693 | 0.061575  | 2.474352  |
| O | 0.246158  | 2.377175  | -1.252812 |
| H | -1.140798 | 1.564495  | 1.052583  |
| H | -2.661922 | 4.947789  | -1.082485 |
| H | -4.914668 | 4.358968  | -0.306578 |
| H | -5.326090 | 2.342727  | 0.998547  |
| H | 1.844638  | 4.049233  | -2.092939 |
| H | 3.049479  | 6.195065  | -2.244614 |
| H | 2.243990  | 8.161361  | -0.977154 |
| H | 0.235058  | 7.949019  | 0.448029  |
| H | -0.953609 | 5.816634  | 0.609578  |
| H | -5.479273 | 0.352752  | 0.000287  |
| H | -7.214831 | -1.335125 | -0.114838 |
| H | -7.281639 | -3.210720 | 1.500422  |
| H | -5.563895 | -3.326878 | 3.280993  |
| H | -3.862759 | -1.577033 | 3.477654  |
| K | 1.317645  | 0.799045  | 0.262141  |
| K | -2.037751 | -2.041542 | 1.246099  |
| K | -1.523744 | 0.743594  | -1.702700 |

**10.5 Atomic coordinates of the optimized structure for  $1^{2-\bullet}(K^+)_4 1^{2-\bullet}$  in the quintet state obtained by the DFT quantum chemical calculation (UB3LYP/3-21G)**

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 1.188174  | 1.451536  | -1.556073 |
| C | 1.131157  | 2.847668  | -1.380442 |
| C | 2.152904  | 3.616579  | -1.992917 |
| C | 3.135694  | 2.982704  | -2.758568 |
| C | 3.151648  | 1.593886  | -2.948056 |
| C | 2.161531  | 0.791455  | -2.326458 |
| C | -0.062107 | 3.362366  | -0.700022 |
| C | -0.040587 | 4.536115  | 0.154855  |
| C | -1.287342 | 5.063111  | 0.605261  |
| C | -1.340748 | 6.113816  | 1.515480  |
| C | -0.162692 | 6.680875  | 2.021652  |
| C | 1.073734  | 6.174408  | 1.592366  |
| C | 1.141016  | 5.125245  | 0.682956  |
| C | 2.009312  | -0.673719 | -2.440475 |
| C | 3.155864  | -1.565544 | -2.432157 |
| C | 4.499309  | -1.163287 | -2.152432 |
| C | 5.517315  | -2.101759 | -2.002558 |
| C | 5.261566  | -3.473649 | -2.132846 |
| C | 3.954048  | -3.890740 | -2.424200 |
| C | 2.927712  | -2.967331 | -2.561873 |
| O | 0.783166  | -1.174970 | -2.326429 |
| O | -1.178043 | 2.640139  | -0.780436 |
| H | 0.481042  | 0.851628  | -0.999069 |
| H | 2.135770  | 4.697148  | -1.917708 |
| H | 3.884210  | 3.588793  | -3.259758 |
| H | 3.879019  | 1.145705  | -3.614169 |
| H | -2.193130 | 4.648389  | 0.176414  |
| H | -2.304466 | 6.510280  | 1.821357  |
| H | -0.205254 | 7.502556  | 2.726856  |
| H | 1.992626  | 6.600949  | 1.981994  |
| H | 2.105468  | 4.726654  | 0.391693  |
| H | 4.734648  | -0.108405 | -2.061984 |
| H | 6.525826  | -1.760293 | -1.789545 |
| H | 6.061312  | -4.196193 | -2.023776 |
| H | 3.744745  | -4.948577 | -2.551408 |
| H | 1.928304  | -3.292022 | -2.824826 |
| C | -1.558814 | -1.030763 | 1.137538  |
| C | -2.850116 | -1.572377 | 0.893129  |
| C | -3.343632 | -2.517009 | 1.841070  |
| C | -2.566126 | -2.884788 | 2.938984  |
| C | -1.284234 | -2.362406 | 3.146621  |
| C | -0.748068 | -1.420169 | 2.227631  |
| C | -3.478241 | -1.294884 | -0.396391 |
| C | -4.920787 | -1.388341 | -0.606403 |
| C | -5.394550 | -1.482994 | -1.938550 |
| C | -6.753964 | -1.544925 | -2.219292 |
| C | -7.695687 | -1.498351 | -1.183277 |
| C | -7.248524 | -1.376552 | 0.137952  |
| C | -5.887823 | -1.317762 | 0.426565  |
| C | 0.559915  | -0.776408 | 2.384965  |
| C | 1.686412  | -1.439801 | 3.028806  |
| C | 1.835614  | -2.850281 | 3.148302  |
| C | 3.010994  | -3.410635 | 3.636673  |
| C | 4.089203  | -2.602797 | 4.029089  |

|   |           |           |           |
|---|-----------|-----------|-----------|
| C | 3.963435  | -1.209599 | 3.927983  |
| C | 2.794801  | -0.636847 | 3.436185  |
| O | 0.756546  | 0.427602  | 1.849860  |
| O | -2.689827 | -1.001550 | -1.426433 |
| H | -1.158180 | -0.329769 | 0.412095  |
| H | -4.306646 | -2.984270 | 1.679130  |
| H | -2.963841 | -3.604255 | 3.648872  |
| H | -0.715567 | -2.642002 | 4.024546  |
| H | -4.657797 | -1.515425 | -2.731519 |
| H | -7.088366 | -1.631071 | -3.248370 |
| H | -8.756272 | -1.545136 | -1.401288 |
| H | -7.968868 | -1.312747 | 0.946849  |
| H | -5.563232 | -1.181487 | 1.451013  |
| H | 1.023687  | -3.495201 | 2.832873  |
| H | 3.096282  | -4.490374 | 3.707391  |
| H | 4.999354  | -3.048068 | 4.412880  |
| H | 4.778191  | -0.569758 | 4.254137  |
| H | 2.681587  | 0.441185  | 3.387832  |
| K | -1.221729 | 2.103172  | 1.791241  |
| K | -1.419632 | 0.785515  | -2.688930 |
| K | 2.848747  | -0.355251 | 0.406470  |
| K | -0.744705 | -2.770503 | -1.010437 |

**10.6 Atomic coordinates of the optimized structure of the dimeric cluster for [Fluorenone<sup>-•</sup> {Na<sup>+</sup>(dimethoxyethan)<sub>2</sub>}]<sub>2</sub> calculated by UB3LYP/3-21G**

|    |             |             |             |
|----|-------------|-------------|-------------|
| Na | -0.14575806 | -1.60830379 | 0.25451523  |
| O  | 1.50755692  | -0.25549775 | -0.66209173 |
| C  | 2.58084798  | 0.19007227  | 2.25483227  |
| C  | 3.30852485  | 0.42336926  | 3.42311835  |
| C  | 4.71727705  | 0.48276424  | 3.40034628  |
| C  | 5.42019701  | 0.28522524  | 2.20591521  |
| C  | 6.39320993  | -0.29949975 | -0.93472177 |
| C  | 6.49284601  | -0.56664574 | -2.30434871 |
| C  | 5.33477306  | -0.74555469 | -3.08511066 |
| C  | 4.06613874  | -0.65472472 | -2.51423979 |
| C  | 2.76778293  | -0.22459073 | -0.29347277 |
| C  | 3.27517390  | 0.01146727  | 1.05115533  |
| C  | 4.70913315  | 0.04397427  | 1.03070033  |
| C  | 5.13294172  | -0.20650972 | -0.34760576 |
| C  | 3.94658089  | -0.37843275 | -1.14317274 |
| C  | 2.45317483  | -3.61276460 | -0.34873176 |
| O  | 1.40076196  | -3.43992662 | 0.66549522  |
| C  | 1.94378388  | -3.28966069 | 2.02083135  |
| C  | 0.74415594  | -3.14999056 | 2.94011140  |
| O  | 0.04547095  | -1.91978574 | 2.56139827  |
| C  | -1.28002608 | -1.82363582 | 3.18956637  |
| C  | 0.24178395  | -2.09390163 | -3.05667162 |
| O  | -0.25523403 | -2.64237070 | -1.79561675 |
| C  | -1.23825812 | -3.71682072 | -1.93128169 |
| C  | -1.40945113 | -4.28699160 | -0.52384174 |
| O  | -1.92836308 | -3.27922559 | 0.41242924  |
| C  | -3.38674212 | -3.10373569 | 0.30193123  |

|    |             |             |             |
|----|-------------|-------------|-------------|
| H  | 1.50397193  | 0.07714027  | 2.27747035  |
| H  | 2.78756595  | 0.54672527  | 4.36782312  |
| H  | 5.26058197  | 0.66716129  | 4.32042694  |
| H  | 6.50502682  | 0.31535625  | 2.20117640  |
| H  | 7.29087782  | -0.16980273 | -0.33878377 |
| H  | 7.46921778  | -0.64227372 | -2.76932168 |
| H  | 5.43544912  | -0.96058470 | -4.14356089 |
| H  | 3.18005395  | -0.80514973 | -3.12163568 |
| H  | 1.91733396  | -3.58233166 | -1.29488468 |
| H  | 2.96610403  | -4.57383871 | -0.21493076 |
| H  | 3.17209387  | -2.78987861 | -0.31163475 |
| H  | 2.53055000  | -4.17437649 | 2.30518532  |
| H  | 2.56789684  | -2.39280176 | 2.07424331  |
| H  | 1.07079995  | -3.10559273 | 3.98772526  |
| H  | 0.07439994  | -4.00830889 | 2.79841137  |
| H  | -1.19218612 | -1.66960180 | 4.27254725  |
| H  | -1.74745011 | -0.96897274 | 2.70693231  |
| H  | -1.86194110 | -2.72636962 | 2.97477126  |
| H  | -0.57296908 | -1.69857478 | -3.67293477 |
| H  | 0.79528695  | -2.85591769 | -3.62312269 |
| H  | 0.90496993  | -1.29281473 | -2.73158169 |
| H  | -2.17580914 | -3.32019377 | -2.33798671 |
| H  | -0.85807508 | -4.50428343 | -2.59841466 |
| H  | -0.42416504 | -4.53158855 | -0.12505077 |
| H  | -2.06170797 | -5.17084169 | -0.53997278 |
| O  | -1.50751603 | 0.25599027  | 0.66224325  |
| Na | 0.14560895  | 1.60876024  | -0.25440177 |
| C  | -2.76771998 | 0.22489028  | 0.29354224  |
| O  | -1.40088904 | 3.44060445  | -0.66553676 |
| O  | -0.04618206 | 1.91979420  | -2.56120181 |
| O  | 0.25456497  | 2.64353442  | 1.79566228  |
| O  | 1.92856395  | 3.27920723  | -0.41255075 |
| C  | -3.27501512 | -0.01182173 | -1.05101371 |
| C  | -3.94657016 | 0.37906227  | 1.14309835  |
| C  | -2.45278502 | 3.61374545  | 0.34920424  |
| C  | -1.94477904 | 3.28916931  | -2.02039671 |
| C  | -0.74574608 | 3.14933825  | -2.94041777 |
| C  | 1.27919090  | 1.82406425  | -3.18965864 |
| C  | -0.24114905 | 2.09406543  | 3.05681920  |
| C  | 1.23888397  | 3.71683335  | 1.93121934  |
| C  | 1.41010094  | 4.28710127  | 0.52383322  |
| C  | 3.38695884  | 3.10357428  | -0.30258477 |
| C  | -2.58059216 | -0.19111173 | -2.25453472 |
| C  | -4.70897388 | -0.04432773 | -1.03063571 |
| C  | -4.06621504 | 0.65589827  | 2.51405025  |
| C  | -5.13287592 | 0.20677628  | 0.34752825  |
| H  | -1.91640413 | 3.58379340  | 1.29506135  |
| H  | -2.96594501 | 4.57467127  | 0.21522924  |
| H  | -3.17160296 | 2.79073238  | 0.31282625  |
| H  | -2.53210402 | 4.17344046  | -2.30498171 |
| H  | -2.56856704 | 2.39201927  | -2.07273173 |
| H  | -1.07312310 | 3.10390925  | -3.98775673 |
| H  | -0.07633106 | 4.00812626  | -2.79992771 |
| H  | 1.19117498  | 1.66956127  | -4.27256107 |

|   |             |             |             |
|---|-------------|-------------|-------------|
| H | 1.74713898  | 0.96980029  | -2.70682478 |
| H | 1.86073196  | 2.72715831  | -2.97535181 |
| H | 0.57423896  | 1.69813025  | 3.67183423  |
| H | -0.79392910 | 2.85568333  | 3.62450027  |
| H | -0.90473706 | 1.29328823  | 2.73178840  |
| H | 2.17617583  | 3.31899428  | 2.33734131  |
| H | 0.85991794  | 4.50452328  | 2.59877038  |
| H | 0.42484796  | 4.53207731  | 0.12518224  |
| H | 2.06265092  | 5.17072916  | 0.53993726  |
| C | -3.30818415 | -0.42506373 | -3.42273664 |
| H | -1.50371408 | -0.07828073 | -2.27711082 |
| C | -5.41995430 | -0.28623775 | -2.20576859 |
| C | -5.33488512 | 0.74691427  | 3.08480740  |
| H | -3.18016696 | 0.80660528  | 3.12143326  |
| C | -6.39318228 | 0.29996327  | 0.93453324  |
| C | -4.71694231 | -0.48444375 | -3.40003276 |
| H | -2.78716302 | -0.54895669 | -4.36733675 |
| H | -6.50478411 | -0.31638774 | -2.20109081 |
| C | -6.49290609 | 0.56765831  | 2.30404639  |
| H | -5.43563223 | 0.96236432  | 4.14316511  |
| H | -7.29081106 | 0.16998927  | 0.33859724  |
| H | -5.26017904 | -0.66935474 | -4.32005215 |
| H | -7.46930790 | 0.64344031  | 2.76893139  |
| H | -3.68065596 | -2.73889375 | -0.68554276 |
| H | -3.89638400 | -4.05132866 | 0.52178723  |
| H | -3.65797901 | -2.35013175 | 1.03940833  |
| H | 3.65786791  | 2.34996033  | -1.04016471 |
| H | 3.89661694  | 4.05112219  | -0.52259678 |
| H | 3.68119287  | 2.73868632  | 0.68478322  |