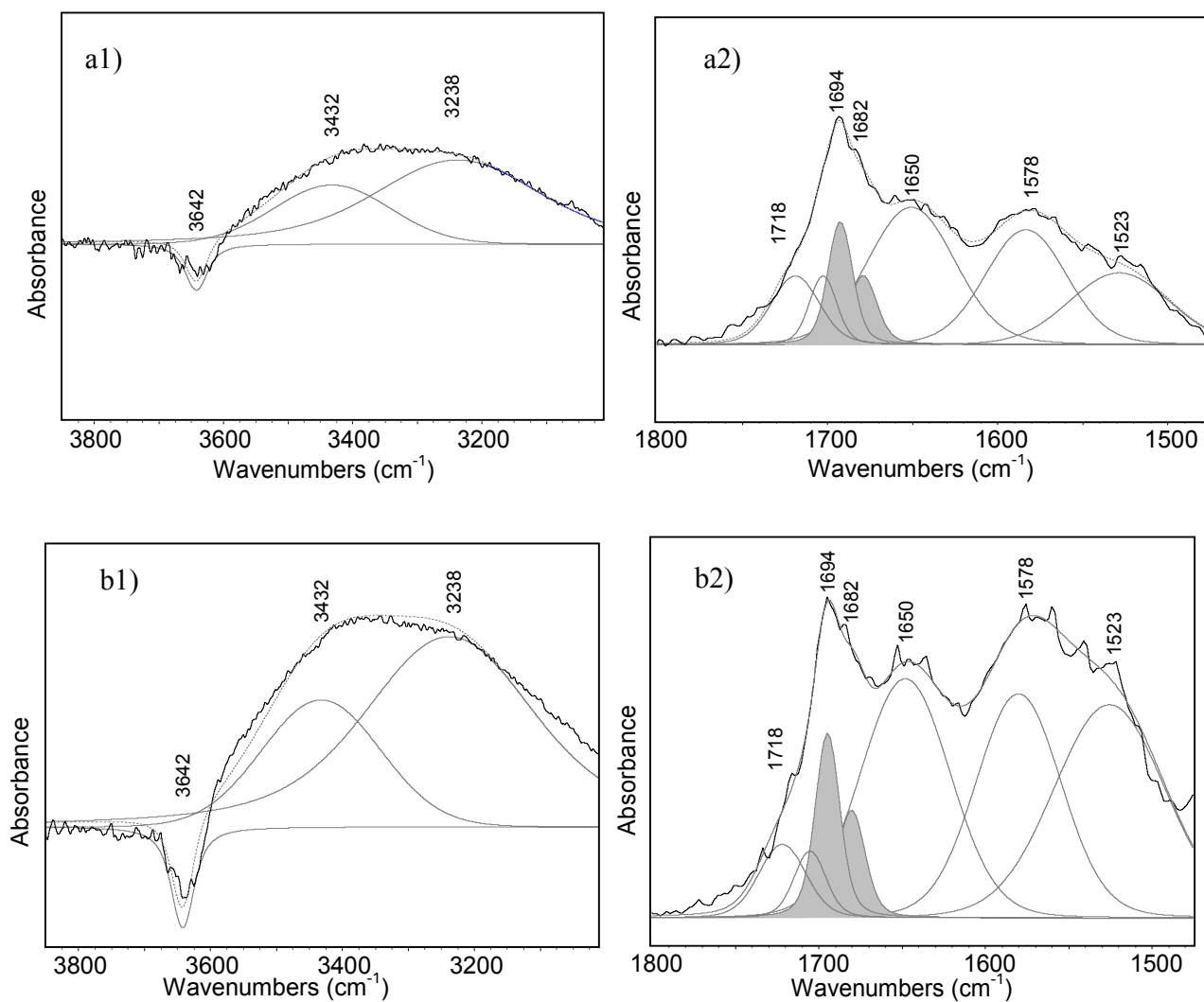
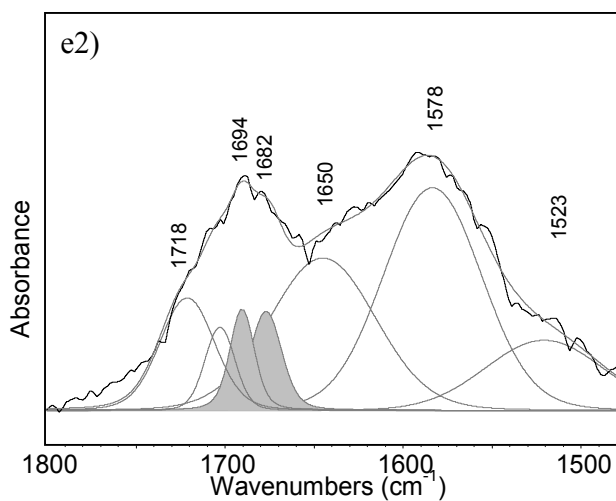
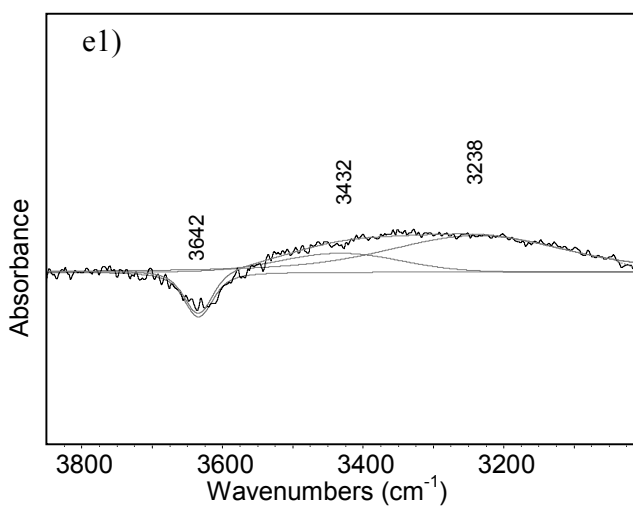
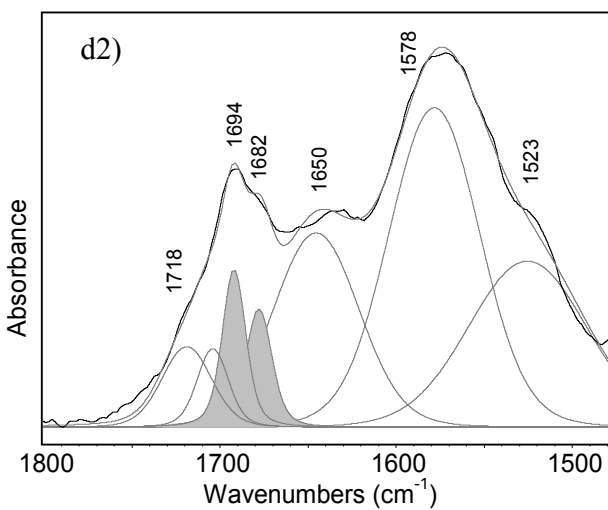
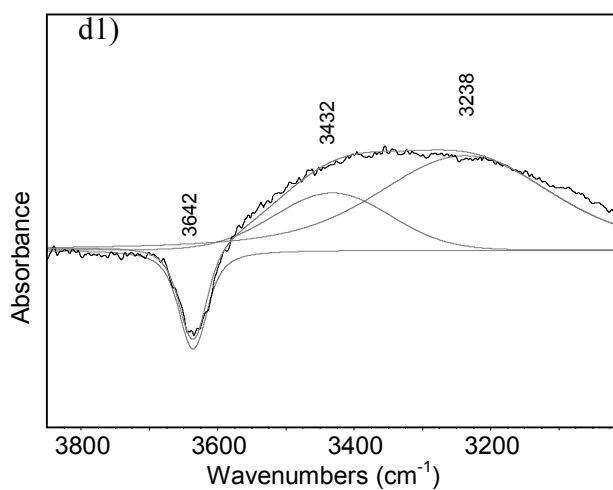
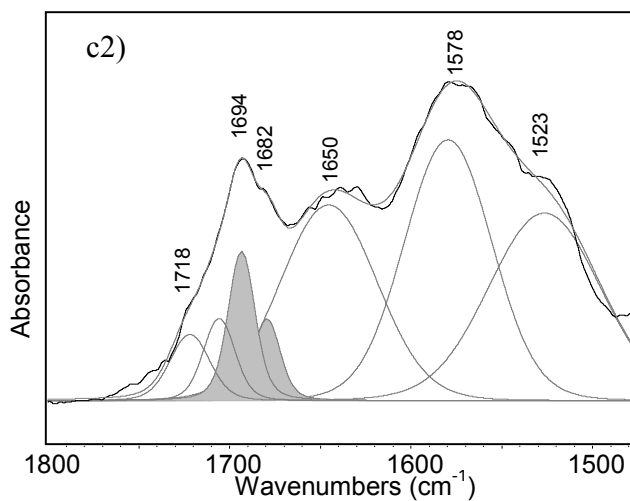
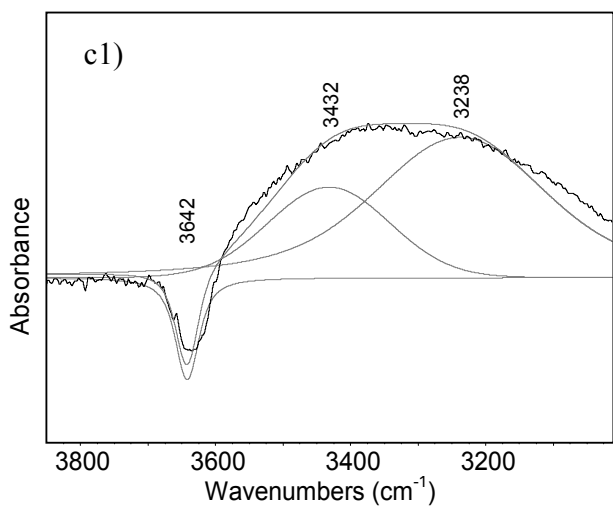


## Electronic Supplementary information

### UV illumination conditions

Figure 1 shows the spectrum and its deconvolution after two minutes of cyclohexane photo-oxidation at all temperatures, both in the high and low frequency range.





60C

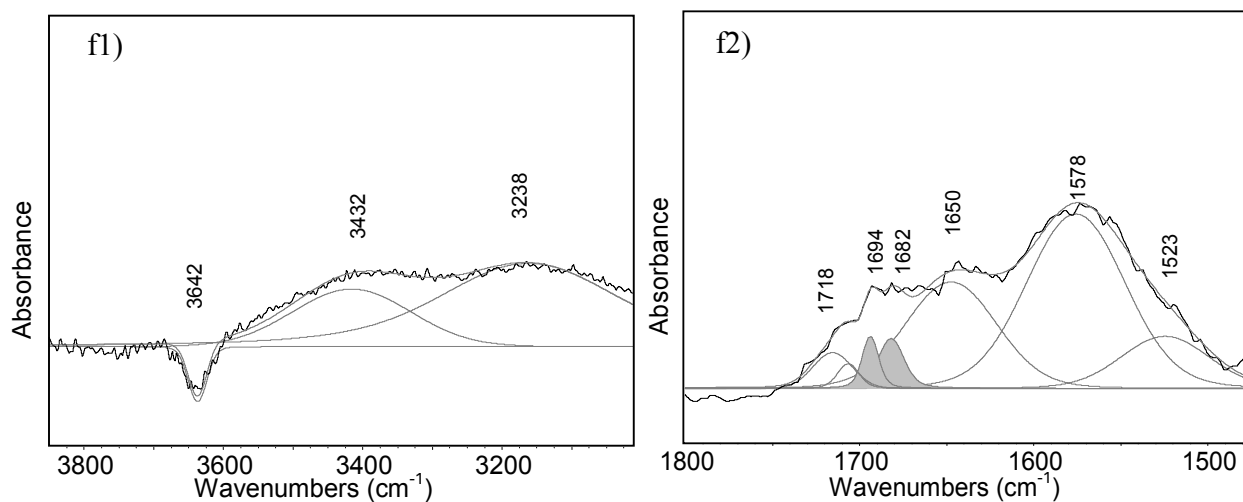


Figure 1. Spectral deconvolution after two minutes of cyclohexane photo-oxidation on TiO<sub>2</sub> at (a) 23°C, (b) 30°C, (c) 40°C, (d) 50°C, (e) 55°C and (f) 60°C for the 1) high wavenumbers region and 2) low wavenumbers region. The spectra are represented in common scale of absorbance among the same wavenumber region. The black line represents the original spectra, the grey bands correspond to the deconvoluted peaks and the dashed grey line represents the sum of the deconvoluted peaks. The two grey-shaded peaks correspond to adsorbed cyclohexanone.

Table 1 shows the correlation matrix of the 11 parameters that could be fitted, without becoming limited by the boundaries that were set to avoid the values moving into physically unacceptable ranges. None of the absolute values of the correlation coefficients between different parameters exceeds the value of 0.99 indicating the absence of a strong correlation between any of the estimated parameters<sup>1</sup>. The quite high correlation coefficient of 0.97 between  $k_{1\_296}$  and  $K_{H_2O\_296}$  may have contributed to the relatively large uncertainty in the calculated  $K_{H_2O\_296}$ . The parameter  $k_{4\_296}$  is associated with an even larger uncertainty, which results from the low sensitivity of the model to this parameter.

Table 1. Correlation matrix of the estimated microkinetic model parameters, calculated under UV illumination conditions.

Parameter	cf	cf	$\Delta H_{ads}$	K	$\Delta H_{ads}$	K	$\Delta H_{ads}$	$k_{1\_296}$	$k_{3\_296}$	$E_{a3}$	$k_{4\_296}$
	CyhOads	H2Oads	Cyh	CyhO_296	CyhO	H2O_296	H2O				
$cf_{CyhOads}$	1										
$cf_{H2Oads}$	-0.82	1									
$\Delta H_{ads_{Cyh}}$	-0.77	0.90	1								
$K_{CyhO\_296}$	-0.24	-0.20	-0.24	1							
$\Delta H_{ads_{CyhO}}$	-0.59	0.70	0.76	-0.46	1						
$K_{H2O\_296}$	0.64	-0.92	-0.90	0.49	-0.77	1					
$\Delta H_{ads_{H2O}}$	-0.66	0.80	0.88	-0.35	0.84	-0.89	1				
$k_{1\_296}$	0.66	-0.94	-0.93	0.43	-0.72	0.97	-0.83	1			
$k_{3\_296}$	-0.32	-0.11	-0.21	0.74	-0.22	0.40	-0.30	0.37	1		
$E_{a3}$	0.12	-0.17	-0.08	-0.07	0.09	0.05	0.12	0.12	-0.47	1	
$k_{4\_296}$	0.15	-0.27	-0.26	0.52	-0.24	0.36	-0.25	0.36	0.23	0.04	1

### Dark conditions

Table 2 shows the correlation matrix of the 4 parameters that were fitted to the ATR-FTIR results under dark conditions. Since the absolute values of all the correlation coefficients between different parameters are significantly lower than 0.99, it can be concluded that there is no strong dependency between any of the estimated parameters.

Table 2. Correlation matrix of the microkinetic model parameters, calculated under dark conditions.

Parameter	K	$\Delta H_{ads}$	$k_{4\_296}$	$E_{a4}$
	CyhO_296	CyhO		
$K_{CyhO\_296}$	1			
$\Delta H_{ads_{CyhO}}$	-0.89	1		
$k_{4\_296}$	0.60	-0.56	1	
$E_{a4}$	-0.42	0.54	-0.77	1

## References

- 1 Caracotsios, M., Personal Communication, 1997