

Electronic Supplementary Information (ESI) for
Pathway Analysis of Super-exchange Electronic Couplings in
Electron Transfer Reactions by Using a Multiconfiguration
Self-consistent Field Method

Hirotaka Nishioka* and Koji Ando

Department of Chemistry, Graduate School of Science,
Kyoto University, Sakyo-ku, Kyoto 606-8502, Japan

* All the correspondence should be addressed to H. Nishioka

e-mail : nishioka@kuchem.kyoto-u.ac.jp

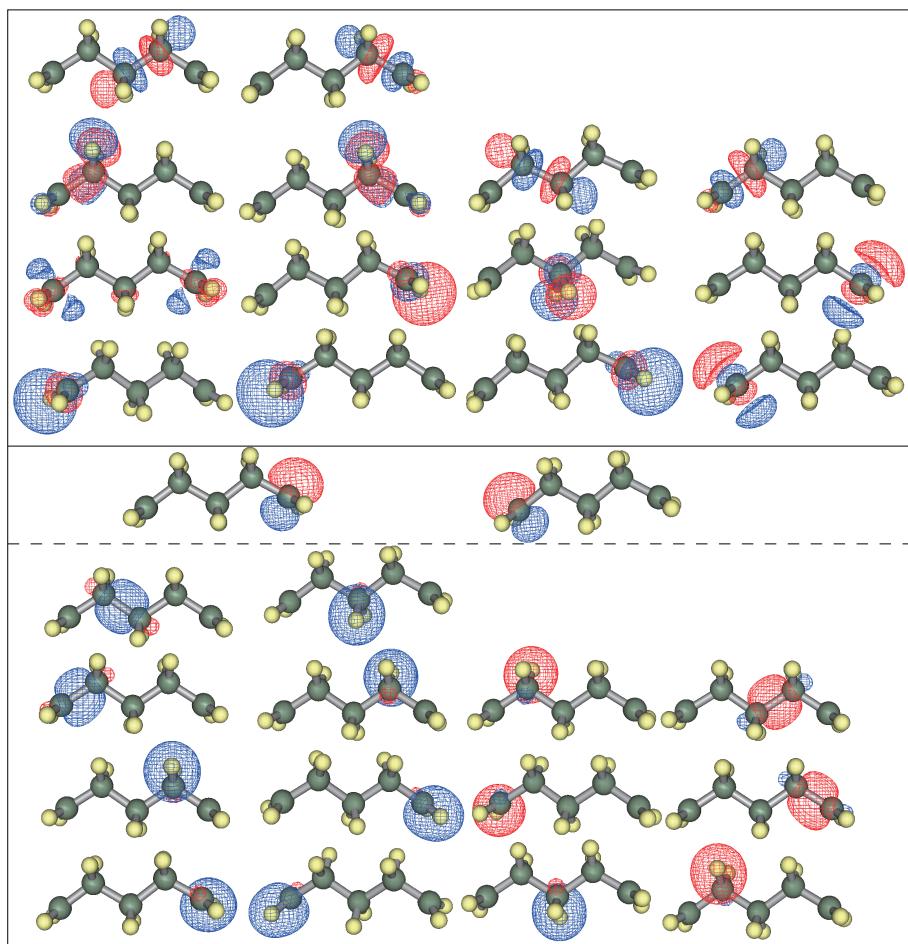
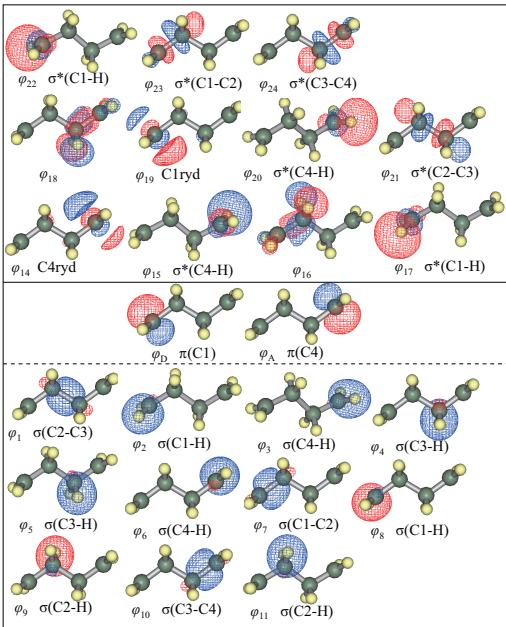


Fig. S1 Same as Fig. 3 but for pentane-1,5-diyil anion.

(a) Foster-Boys localization scheme



(b) Pipek-Mezey localization scheme

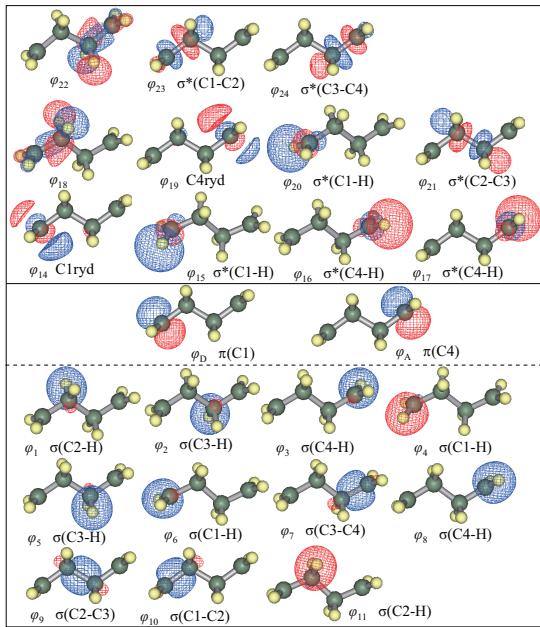


Fig. S2 Same as Fig. 3 but with (a) the Foster-Boys localization scheme and (b) the Pipek-Mezey localization scheme.

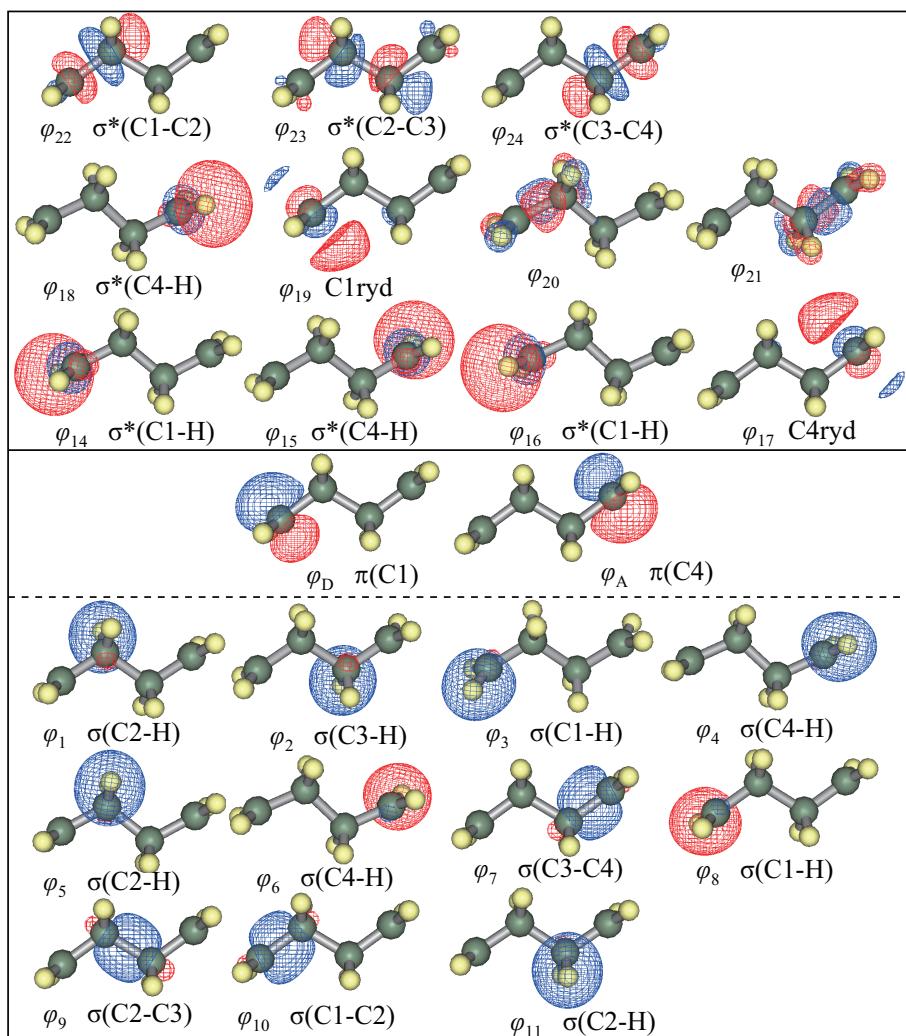


Fig. S3 Same as Fig. 3 but with the 6-31G(d) basis set.

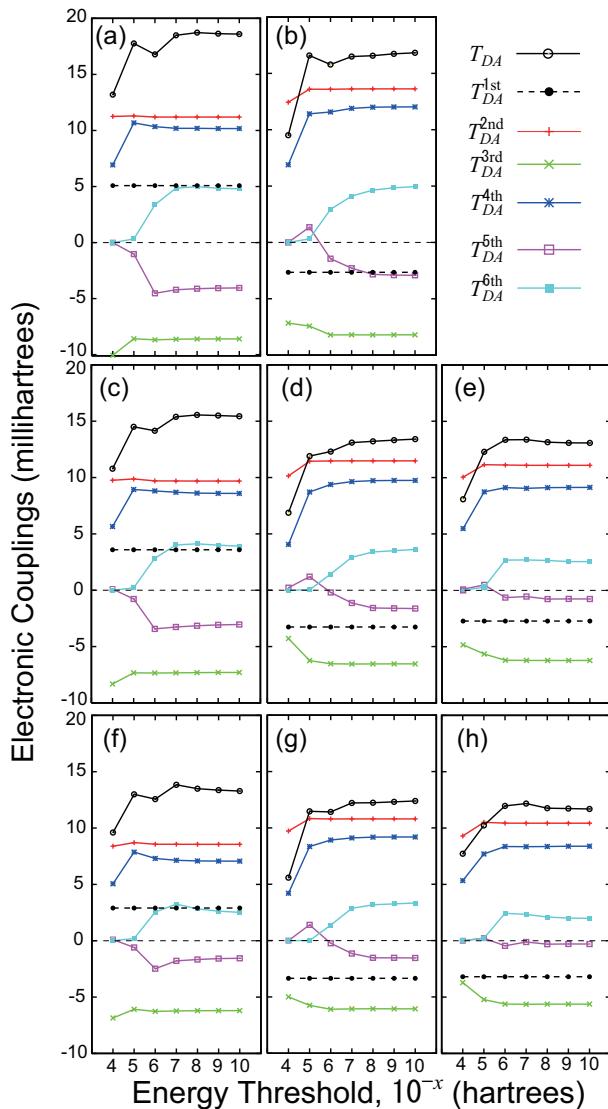


Fig. S4 Same as Fig. 4 but with the basis set / localization scheme of (a) 3-21G / Pipek-Mezey, (b) 3-21G / Foster-Boys, (c) 6-31G(d) / Pipek-Mezey, (d) 6-31G(d) / Foster-Boys, (e) 6-31G(d) / Edmiston-Ruedenberg, (f) cc-pVDZ / Pipek-Mezey, (g) cc-pVDZ / Foster-Boys, and (h) cc-pVDZ / Edmiston-Ruedenberg.

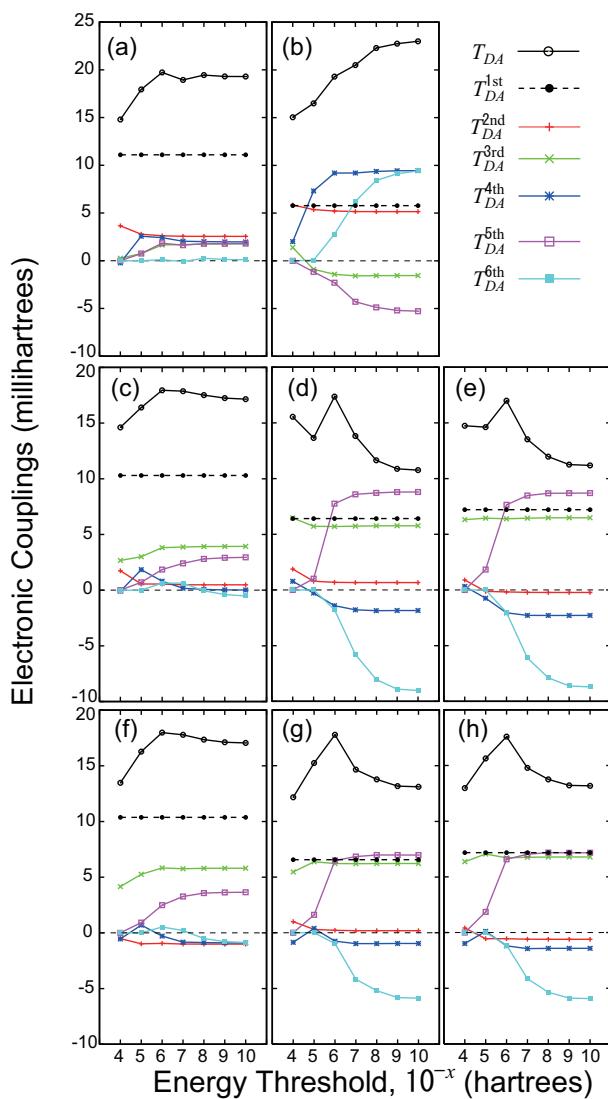


Fig. S5 Same as Fig. S4 but for anion coupling of pentane-1,5-diyl.

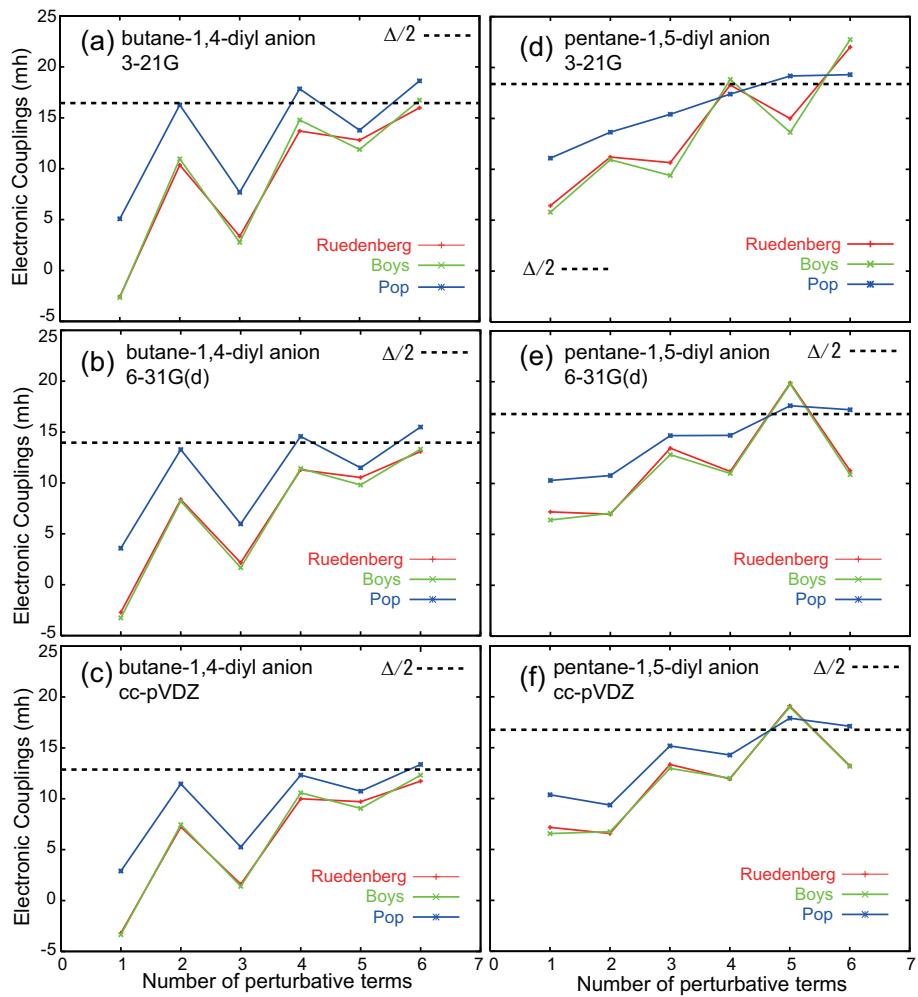


Fig. S6 Dependence of T_{DA} from eqn 15 on the number of perturbative terms with the energy threshold of 10^{-9} hartree.

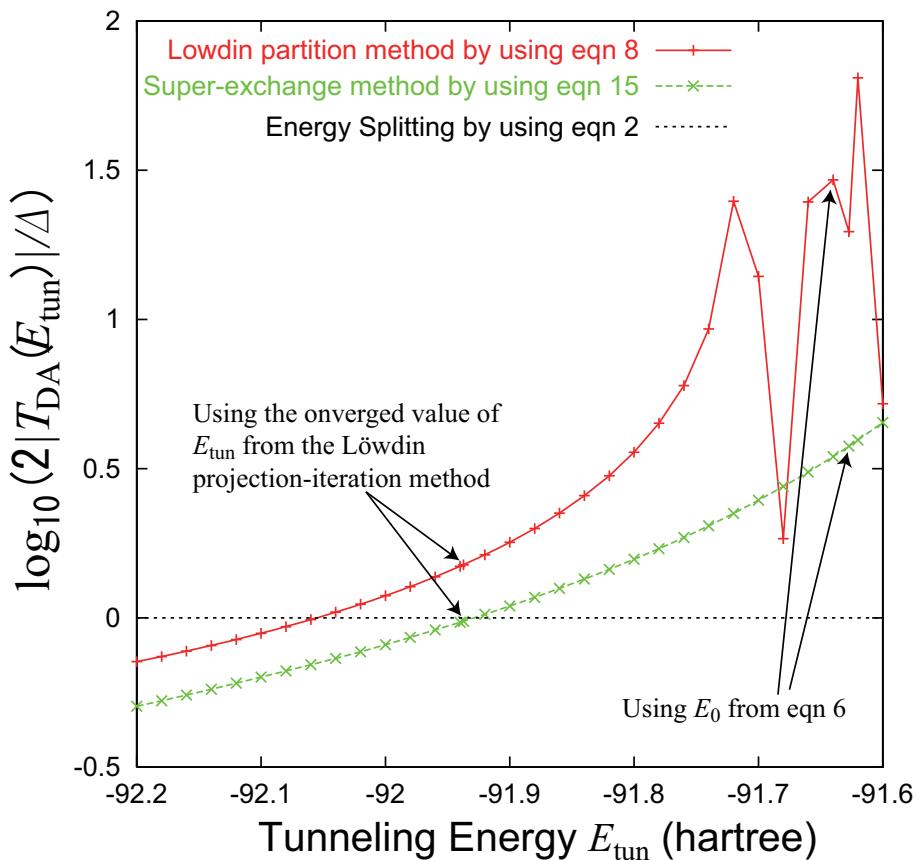


Fig. S7 Dependence of H_{DA}^{eff} from eqn 8 and T_{DA} from eqn 15 on the tunneling energy E_{tun} for butane-1,4-diyl anion. The Ruedenberg localization scheme with the 3-21G basis set was used. The energy threshold of 10^{-8} hartree was employed for the higher-order super-exchange calculation (eqn 15).

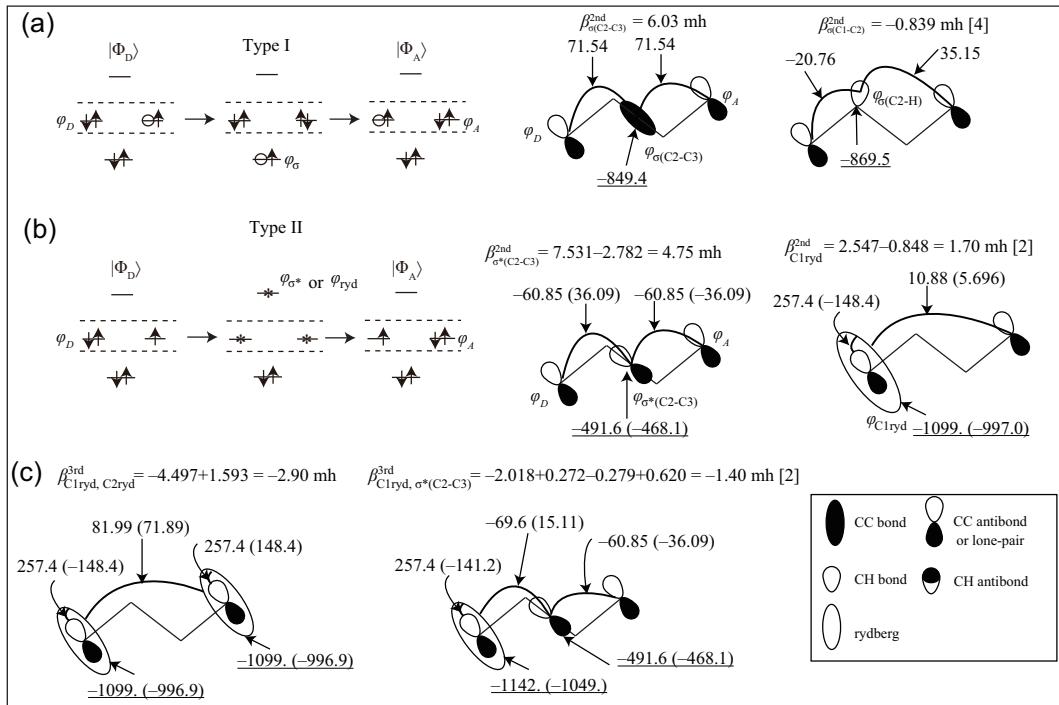


Fig. S8 Same as Fig. 6 but with the Pipek-Mezey localization scheme.

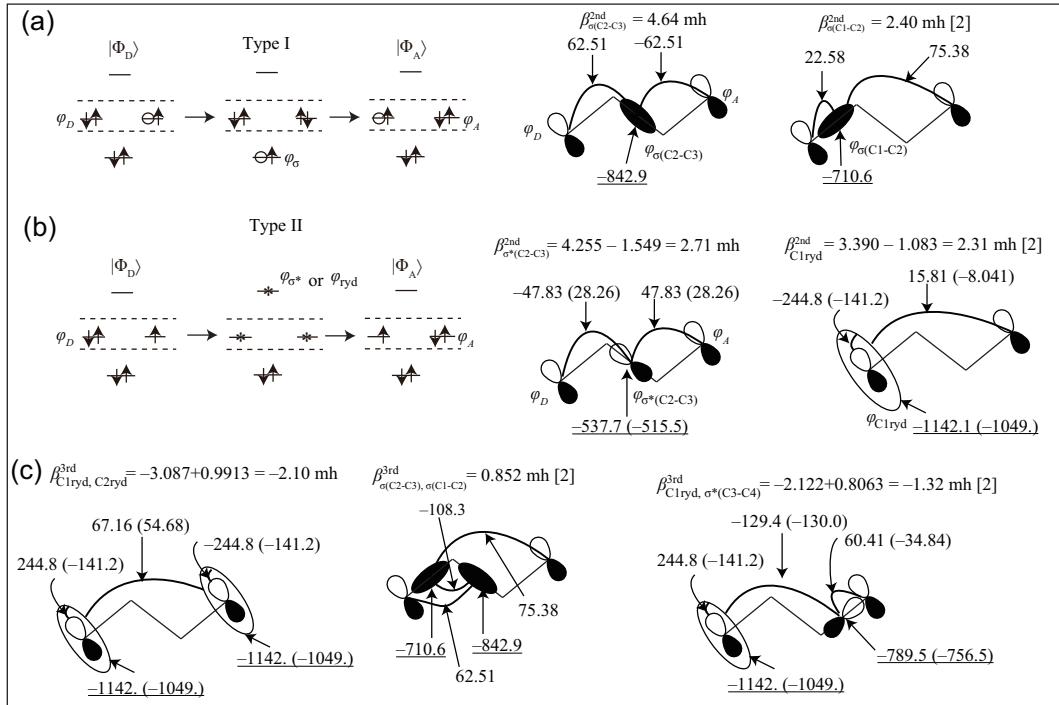


Fig. S9 Same as Fig. 6 but with the Foster-Boys localization scheme.

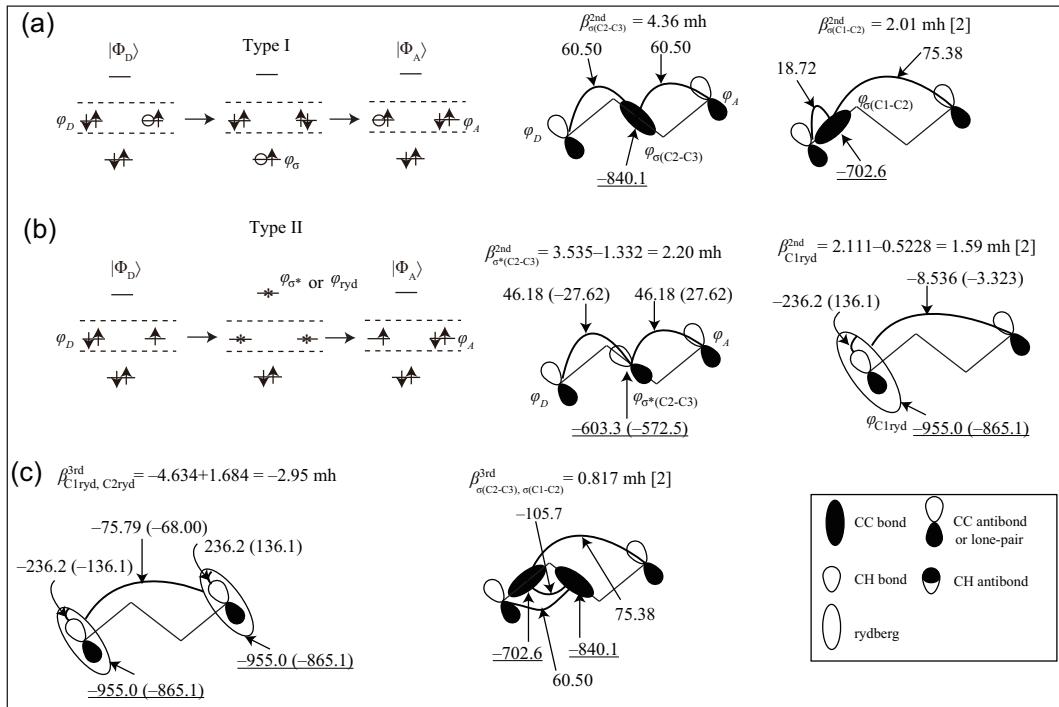


Fig. S10 Same as Fig. 6 but with the 6-31G(d) basis set.

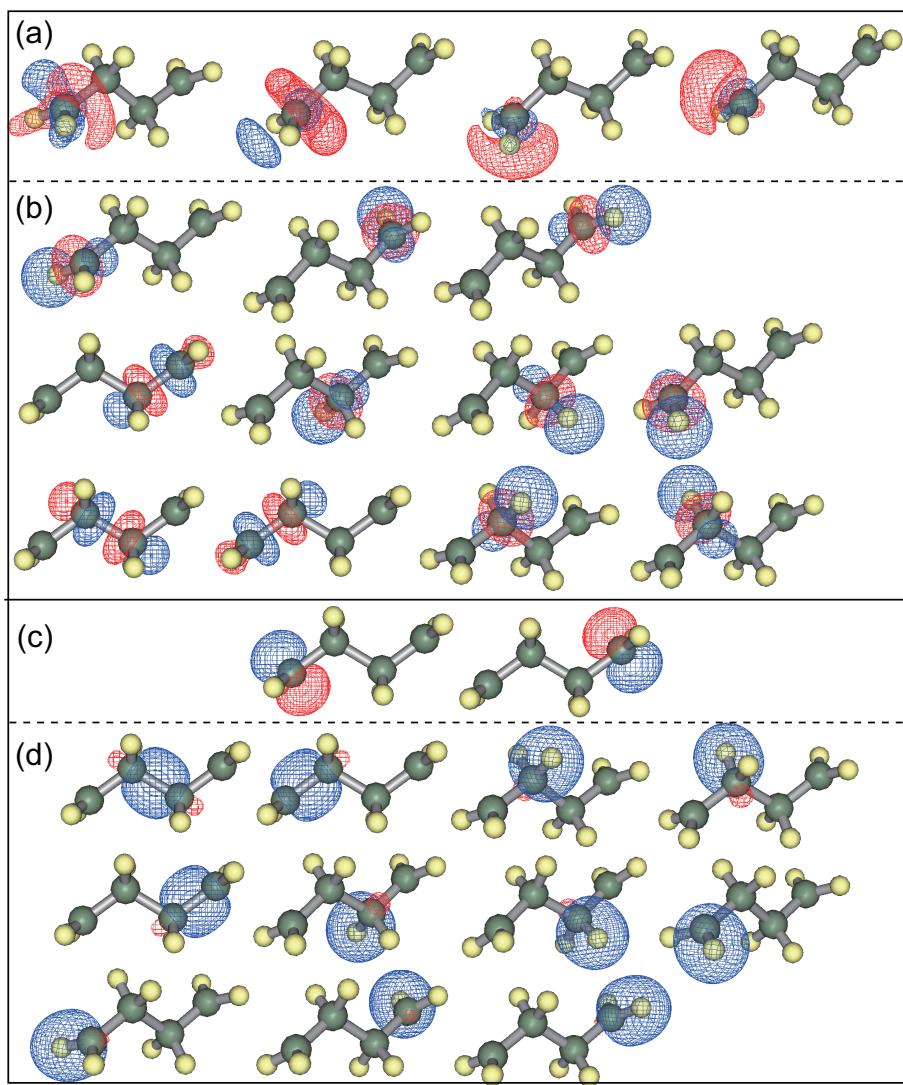


Fig. S11 The NBOs with the 3-21G basis set for neutral triplet state of butane-1,4-diyl. (a) NBOs correspond to the extra-valence-shell orbitals at C1. The orbital groups (b) (c), and (d) correspond to (a), (b), and (c) in Figure 3.