

The Leapfrog Principle for Boron Fullerenes: A Theoretical Study of Structure and Stability of B₁₁₂

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SUPPLEMENTARY INFORMATIONS

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Table . Symmetries and energies of valences molecular orbitals for B₁₁₂ isomers computed at B3LYP/SVP level.

Valence orbitals of <i>D</i> ₂ -B ₁₁₂		Valence orbitals of <i>T</i> _d - B ₁₁₂	
Mos	Energies(eV)	Mos	Energies(eV)
70 <i>b</i> ₁	-3.3470 eV	19 <i>a</i> ₁	-3.238 eV
70 <i>b</i> ₃	-3.529 eV	42 <i>t</i> ₂	-3.325 eV
70 <i>b</i> ₂	-3.719 eV	41 <i>t</i> ₂	-3.529 eV
69 <i>b</i> ₁	-5.462 eV	24 <i>e</i>	-5.266 eV
73 <i>a</i>	-5.486 eV	7 <i>a</i> ₂	-5.450 eV
69 <i>b</i> ₂	-5.502 eV	29 <i>t</i> ₁	-5.682 eV
72 <i>a</i>	-5.584 eV	40 <i>t</i> ₂	-6.111 eV
69 <i>b</i> ₃	-5.624 eV	23 <i>e</i>	-6.264 eV
71 <i>a</i>	-5.725 eV	28 <i>t</i> ₁	-6.431 eV
68 <i>b</i> ₂	-5.836 eV	39 <i>t</i> ₂	-6.921 eV
70 <i>a</i>	-5.954 eV	18 <i>a</i> ₁	-6.968 eV
68 <i>b</i> ₁	-6.054 eV	38 <i>t</i> ₂	-7.262 eV
68 <i>b</i> ₃	-6.097 eV	37 <i>t</i> ₂	-7.475 eV
67 <i>b</i> ₂	-6.326 eV	22 <i>e</i>	-7.489 eV
67 <i>b</i> ₁	-6.391 eV	17 <i>a</i> ₁	-7.537 eV
67 <i>b</i> ₃	-6.443 eV	27 <i>t</i> ₁	-7.540 eV
69 <i>a</i>	-6.513 eV	21 <i>e</i>	-7.807 eV
66 <i>b</i> ₃	-7.046 eV	36 <i>t</i> ₂	-7.837 eV
66 <i>b</i> ₂	-7.163 eV	26 <i>t</i> ₁	-7.929 eV
68 <i>a</i>	-7.165 eV	25 <i>t</i> ₁	-8.290 eV
66 <i>b</i> ₁	-7.257 eV	20 <i>e</i>	-8.563 eV
67 <i>a</i>	-7.334 eV	24 <i>t</i> ₁	-8.773 eV
65 <i>b</i> ₁	-7.335 eV	19 <i>e</i>	-8.783 eV
65 <i>b</i> ₂	-7.354 eV	6 <i>a</i> ₂	-8.974 eV
66 <i>a</i>	-7.498 eV	35 <i>t</i> ₂	-9.023 eV
65 <i>b</i> ₃	-7.525 eV	16 <i>a</i> ₁	-9.504 eV
64 <i>b</i> ₂	-7.527 eV	34 <i>t</i> ₂	-9.592 eV
63 <i>b</i> ₂	-7.549 eV	23 <i>t</i> ₁	-9.728 eV
64 <i>b</i> ₃	-7.559 eV	33 <i>t</i> ₂	-10.484 eV
63 <i>b</i> ₃	-7.586 eV	18 <i>e</i>	-10.511 eV
65 <i>a</i>	-7.595 eV	32 <i>t</i> ₂	-10.896 eV
64 <i>b</i> ₁	-7.663 eV	17 <i>e</i>	-10.943 eV
64 <i>a</i>	-7.705 eV	31 <i>t</i> ₂	-11.220 eV

The first three orbitals are virtuals orbitals and the forth orbital is HOMO.