

Revisiting isorecticular MOFs of alkaline earth metals: A comprehensive study on phase stability, electronic structure, chemical bonding, and optical properties of A-IRMOF-1 (A = Be, Mg, Ca, Sr, Ba)

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Page2, Table S1. Optimized bond length (Å) and bond angles (°) for A-IRMOF-1 series

Page3, Fig. S1, The calculated partial density of states (PDOS) for Mg-IRMOF-1.

Page4, Fig. S2, The calculated partial density of states (PDOS) for Ca-IRMOF-1.

Page5, Fig. S3, The calculated partial density of states (PDOS) for Sr-IRMOF-1.

Page6, Fig. S4, The calculated partial density of states (PDOS) for Ba-IRMOF-1.

Page7, Fig. S5, Calculated charge density (a), charge transfer (b), and electron localization function, ELF (c) plots for Mg-IRMOF-1 in the (110) plane.

Page8, Fig. S6, Calculated charge density (a), charge transfer (b), and electron localization function, ELF (c) plots for Ca-IRMOF-1 in the (110) plane.

Page9, Fig. S7, Calculated charge density (a), charge transfer (b), and electron localization function, ELF (c) plots for Sr-IRMOF-1 in the (110) plane.

Page10, Fig. S8, Calculated charge density (a), charge transfer (b), and electron localization function, ELF (c) plots for Ba-IRMOF-1 in the (110) plane.

Page11, Fig. S9, Calculated Bader charges (BC), bond overlap populations (BOP) and Mulliken effective charges (MEC) for the A-IRMOF-1 series (A = Be, Mg, Ca, Sr, Ba).

Page12, Fig. S10, The calculated optical properties for Mg-IRMOF-1.

Page13, Fig. S11, The calculated band structure for Mg-IRMOF-1.

Page14, Fig. S12, The calculated optical properties for Ca-IRMOF-1.

Page15, Fig. S13, The calculated band structure for Ca-IRMOF-1.

Page16, Fig. S14, The calculated optical properties for Sr-IRMOF-1.

Page17, Fig. S15, The calculated band structure for Sr-IRMOF-1.

Page18, Fig. S16, The calculated optical properties for Ba-IRMOF-1.

Page19, Fig. S17, The calculated band structure for Ba-IRMOF-1.

Table S1. Optimized bond length (Å) and bond angles (°) for A-IRMOF-1 (A = Be, Mg, Ca, Sr, or Ba) at their equilibrium volumes^a

A	C1-C2 (Å)	C2-C3 (Å)	C3-C3 (Å)	C1-O2 (Å)	A-O1 (Å)	A-O2 (Å)	O2-A-O2 (°)
Be	<i>1.490</i> , 1.491 <1.468>	<i>1.404</i> , 1.404 <1.388>	<i>1.391</i> , 1.391 <1.377>	<i>1.274</i> , 1.274 <1.262>	<i>1.717</i> , 1.717	<i>1.634</i> , 1.635 <1.602>	<i>102.905</i> , 102.921 <102.812>
Mg	<i>1.496</i> , 1.496 <1.472>	<i>1.404</i> , 1.404 <1.388>	<i>1.391</i> , 1.391 <1.377>	<i>1.277</i> , 1.276 <1.265>	<i>1.989</i> , 1.989	<i>1.959</i> , 1.960 <1.903>	<i>108.332</i> , 108.339 <108.210>
Ca	<i>1.499</i> , 1.500 <1.476>	<i>1.404</i> , 1.403 <1.389>	<i>1.391</i> , 1.391 <1.377>	<i>1.277</i> , 1.277 <1.266>	<i>2.270</i> , 2.270	<i>2.242</i> , 2.242 <2.125>	<i>112.455</i> , 112.459 <111.751>
Sr	<i>1.502</i> , 1.502	<i>1.403</i> , 1.403	<i>1.392</i> , 1.392	<i>1.276</i> , 1.276	<i>2.433</i> , 2.434	<i>2.406</i> , 2.406	<i>114.125</i> , 114.125
Ba	<i>1.504</i> , 1.504	<i>1.403</i> , 1.403	<i>1.392</i> , 1.392	<i>1.276</i> , 1.276	<i>2.607</i> , 2.607	<i>2.579</i> , 2.580	<i>115.537</i> , 115.529

^a The bond lengths and angles in italic and bold fonts are from Γ -point and $2\times 2\times 2$ **k**-point calculations, respectively. Data in <braces> are from ref 9.

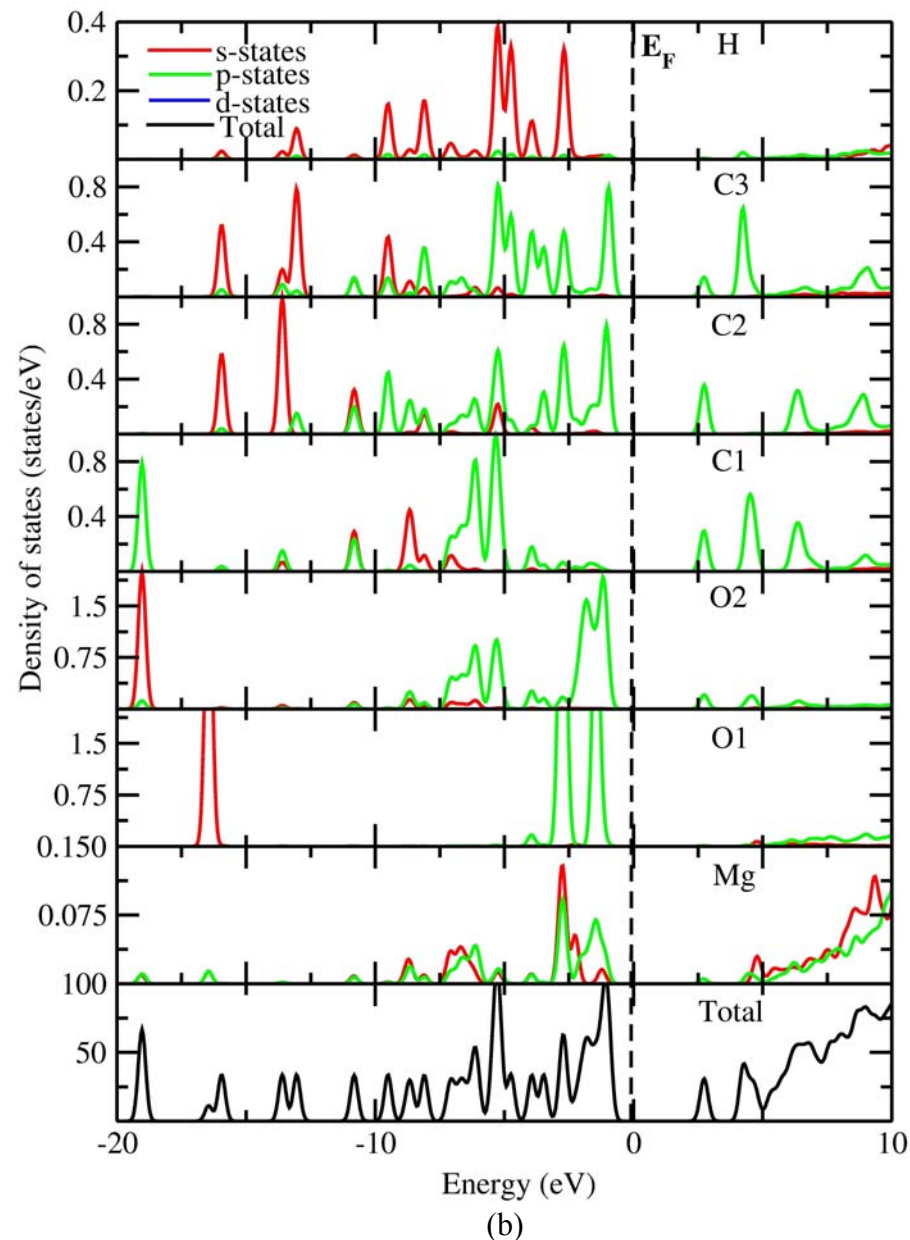
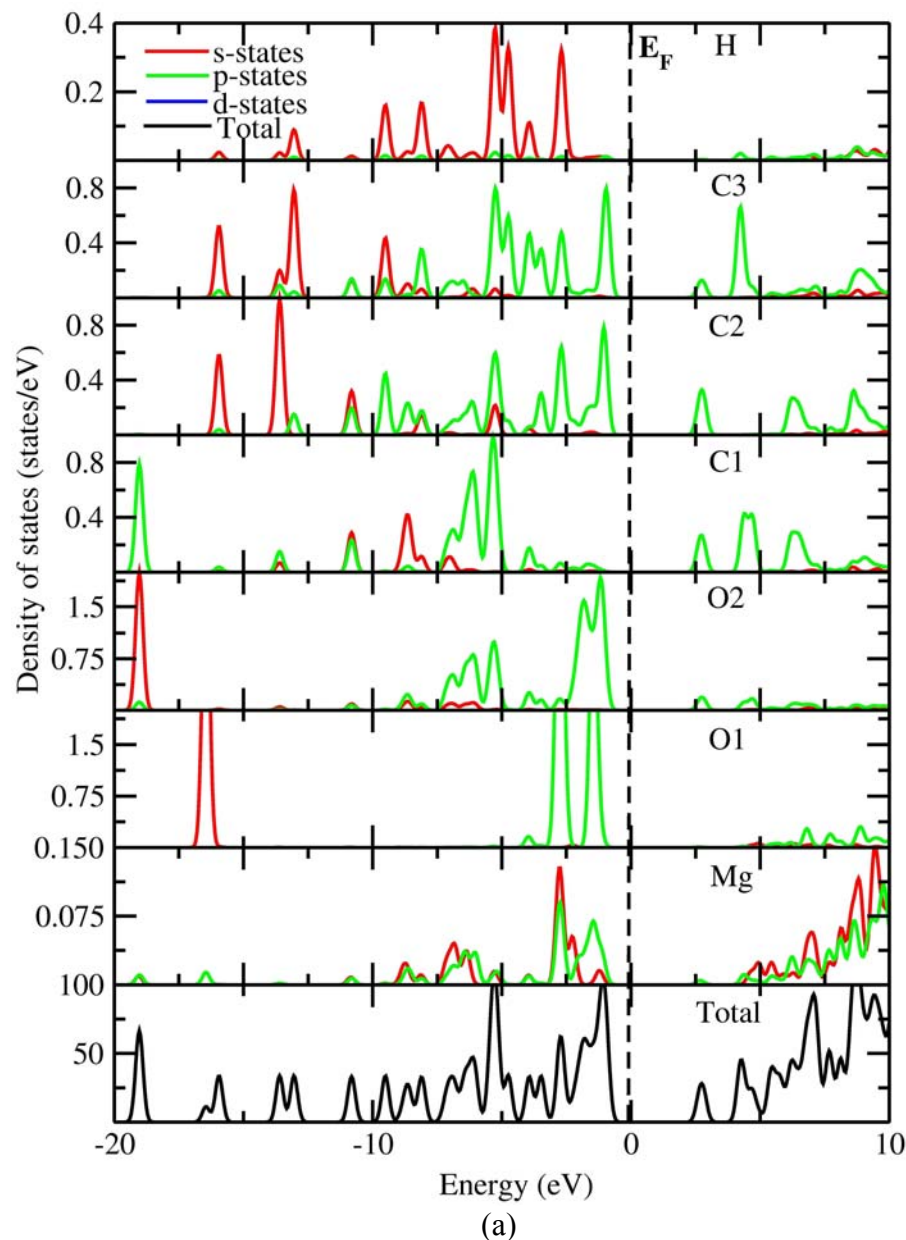


Figure S1. Calculated partial density of states (PDOS) for Mg-IRMOF-1 obtained from (a) Γ point ($1 \times 1 \times 1$) only $\mathbf{k}$ -mesh and (b) $3 \times 3 \times 3$ $\mathbf{k}$ -mesh using the Monkhorst–Pack scheme.

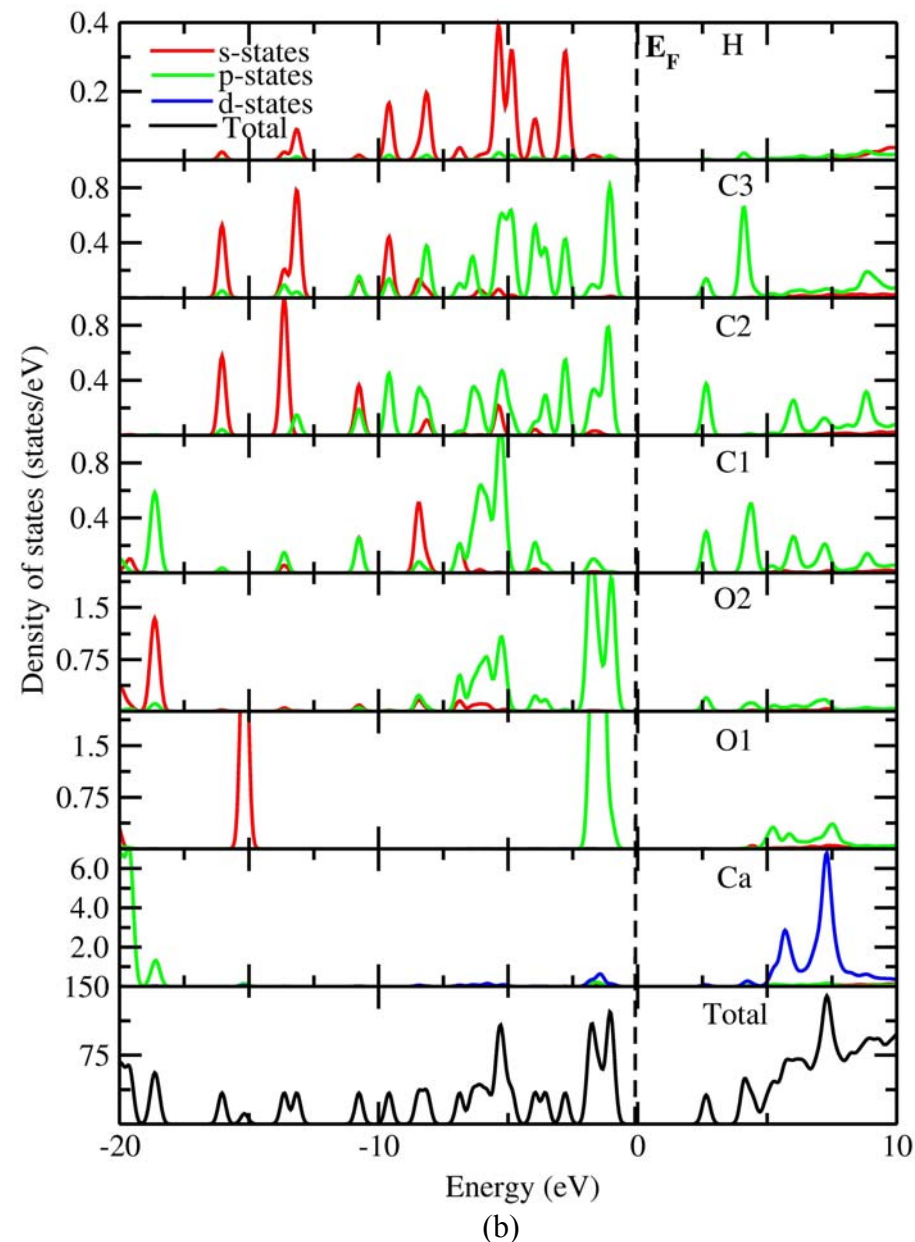
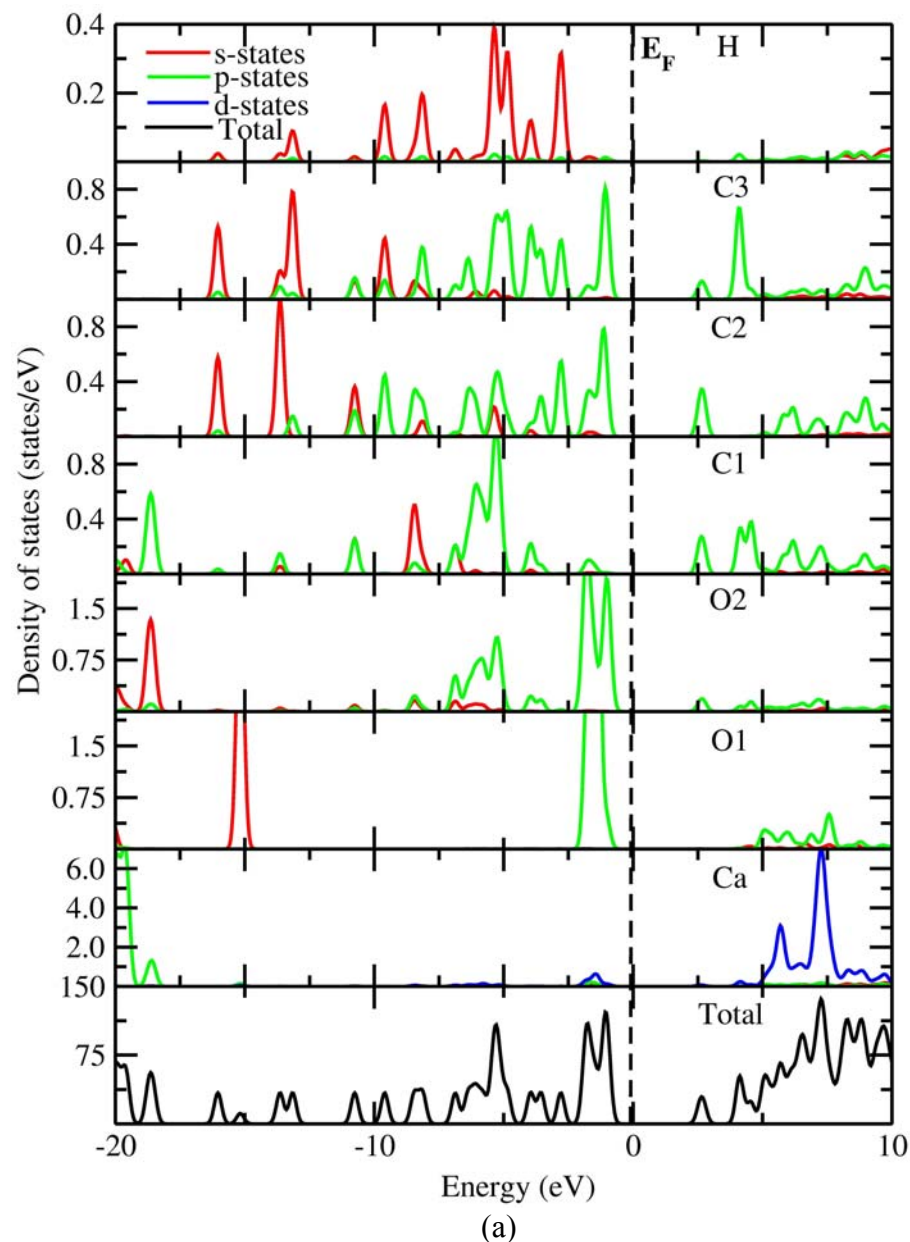


Figure S2. Calculated partial density of states (PDOS) for Ca-IRMOF-1 obtained from (a) Γ point ($1 \times 1 \times 1$) only $\mathbf{k}$ -mesh and (b) $3 \times 3 \times 3$ $\mathbf{k}$ -mesh using the Monkhorst–Pack scheme.

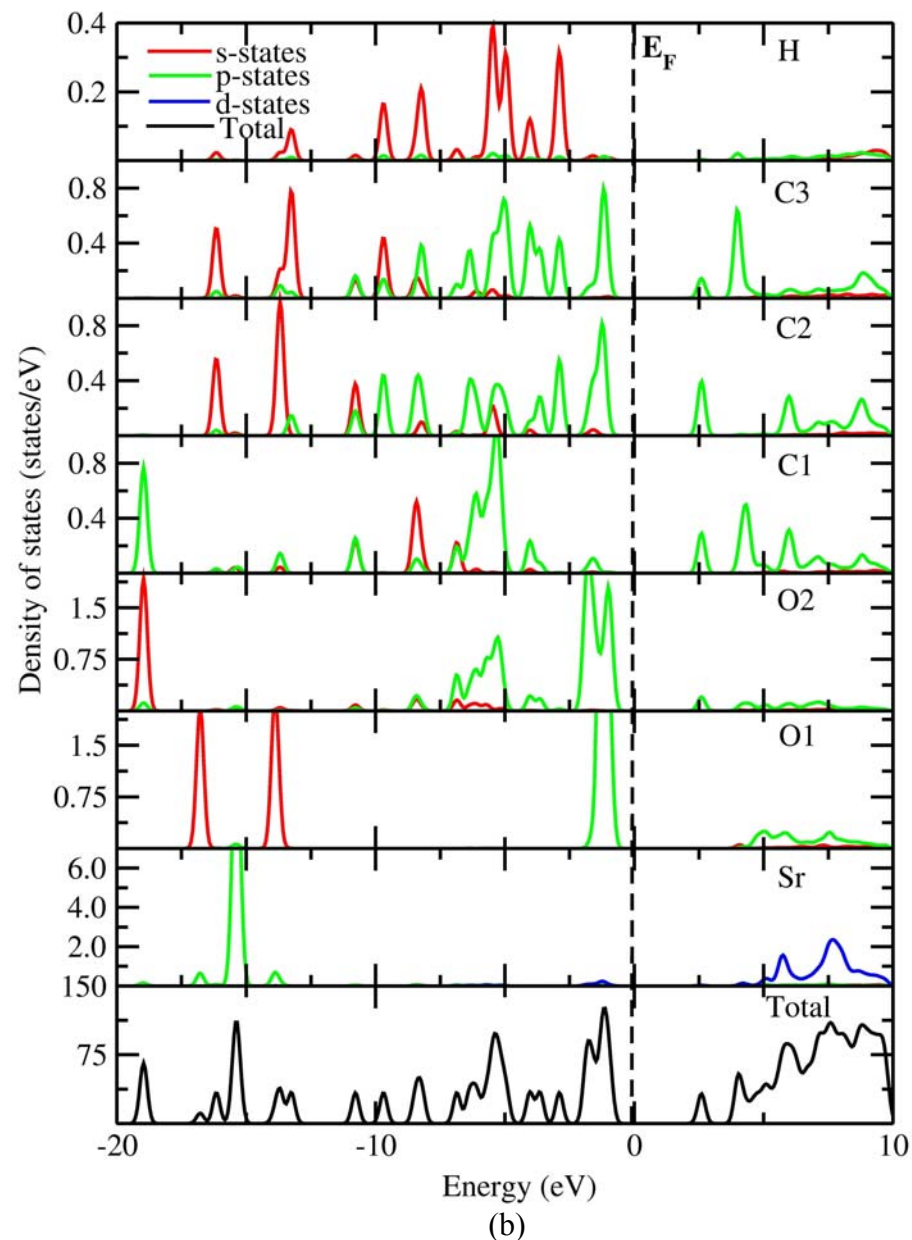
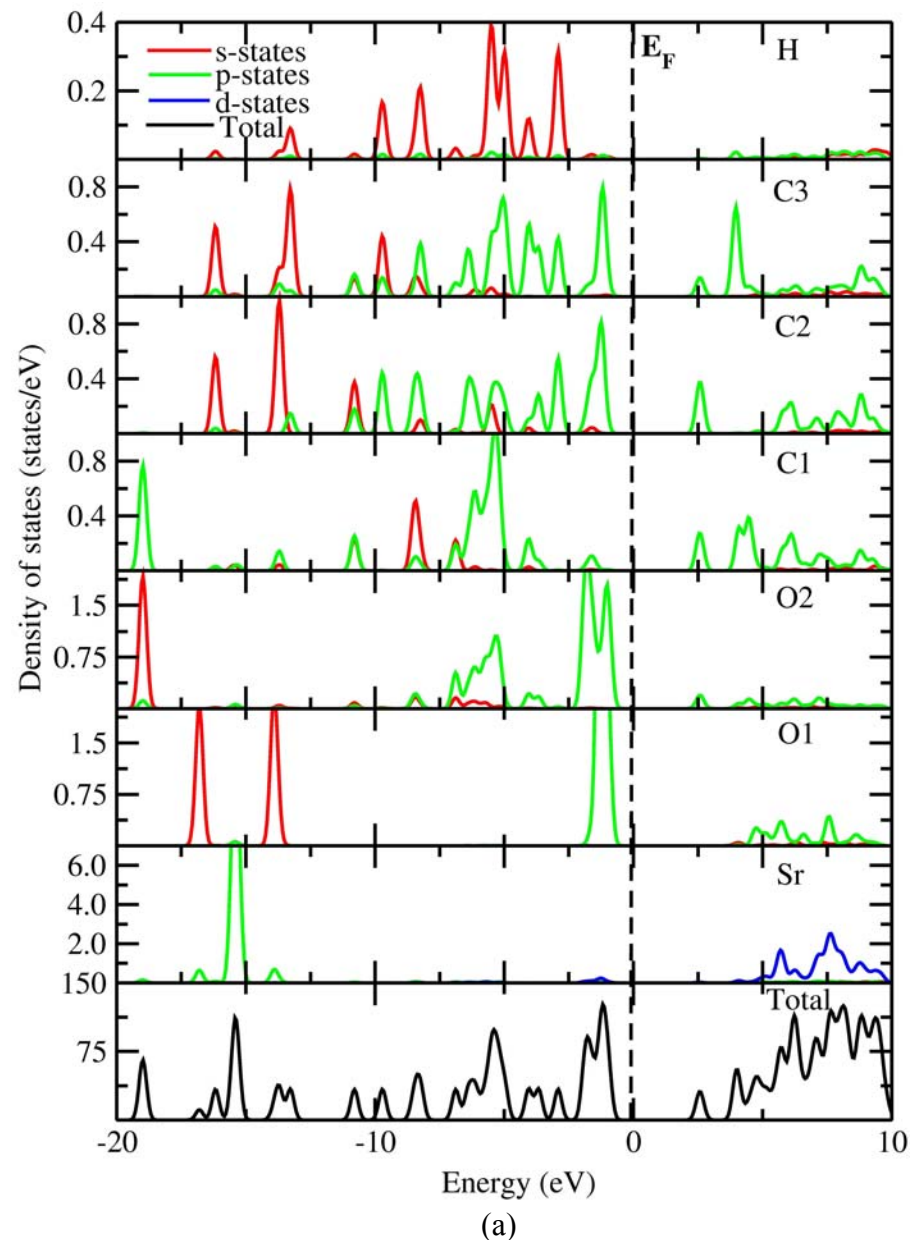


Figure S3. Calculated partial density of states (PDOS) for Sr-IRMOF-1 obtained from (a) Γ point ($1 \times 1 \times 1$) only $\mathbf{k}$ -mesh and (b) $3 \times 3 \times 3$ $\mathbf{k}$ -mesh using the Monkhorst–Pack scheme.

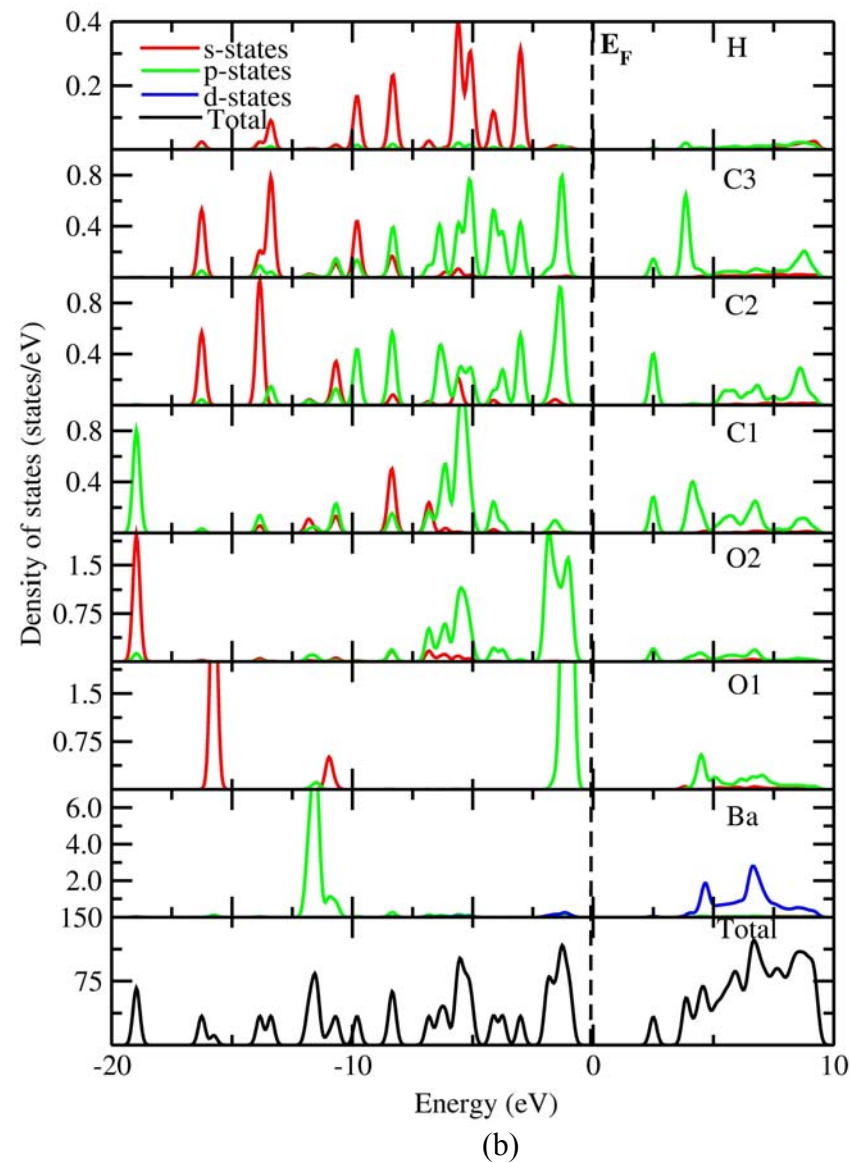
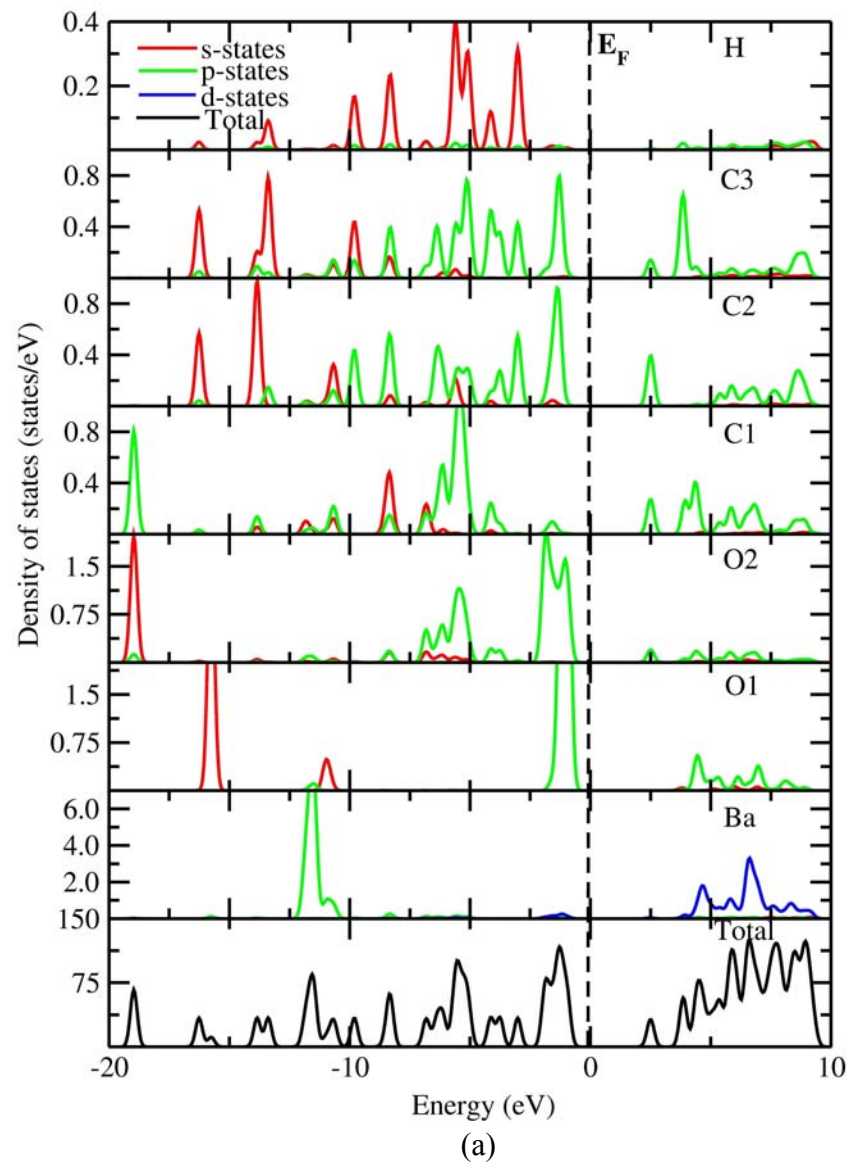
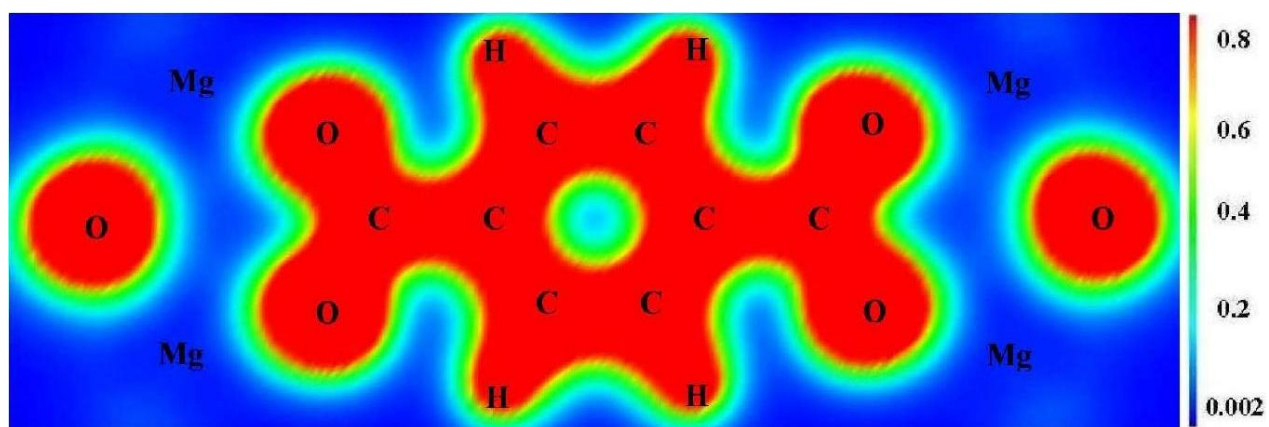
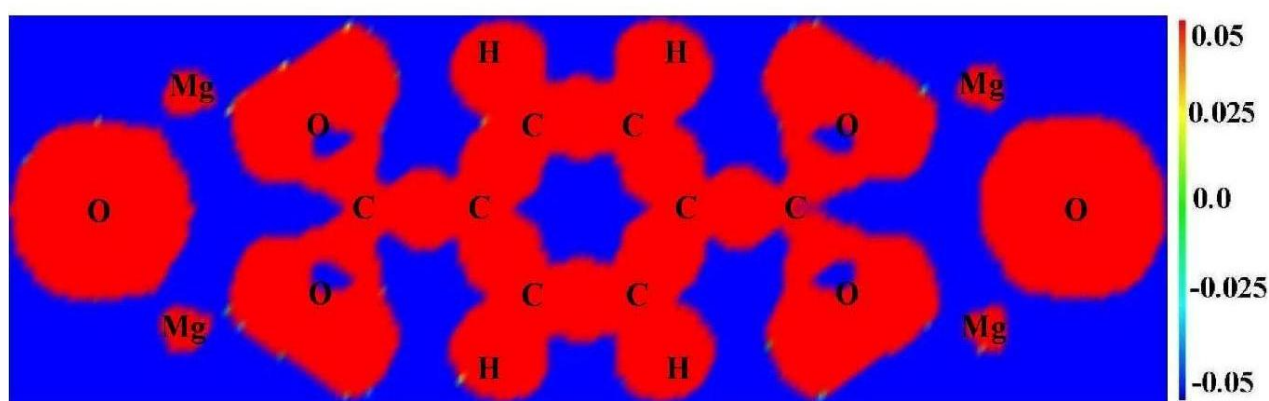


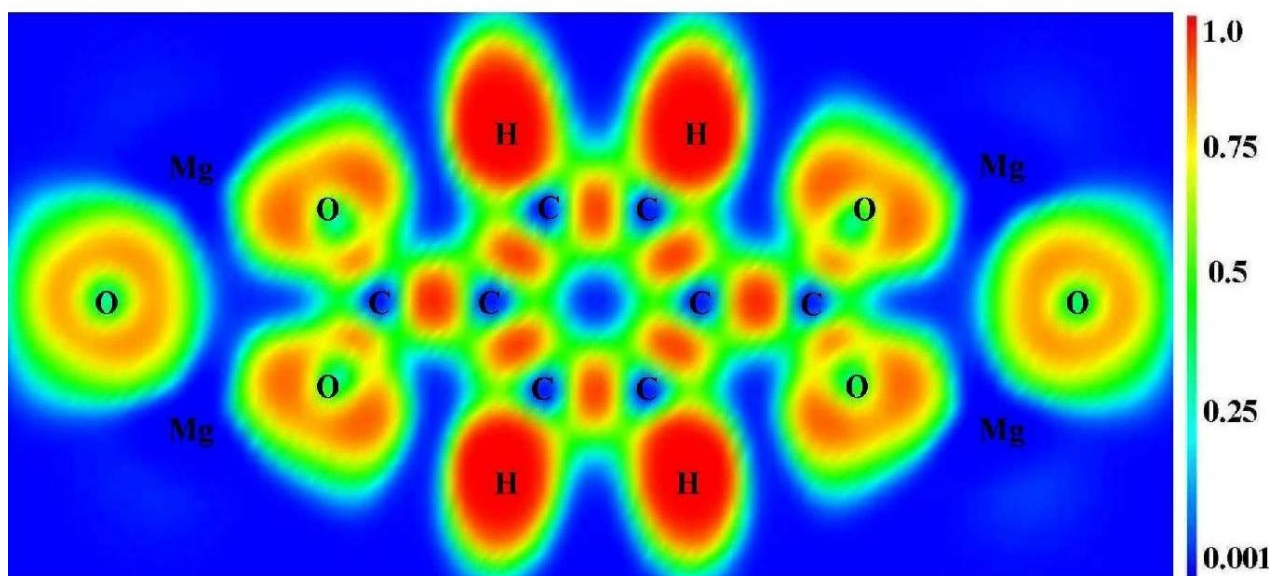
Figure S4. Calculated partial density of states (PDOS) for Ba-IRMOF-1 obtained from (a) Γ point ($1 \times 1 \times 1$) only $\mathbf{k}$ -mesh and (b) $3 \times 3 \times 3$ $\mathbf{k}$ -mesh using the Monkhorst–Pack scheme.



(a)



(b)



(c)

Figure S5. Calculated charge density (a), charge transfer (b), and electron localization function, ELF (c) plots for Mg-IRMOF-1 in the (110) plane.

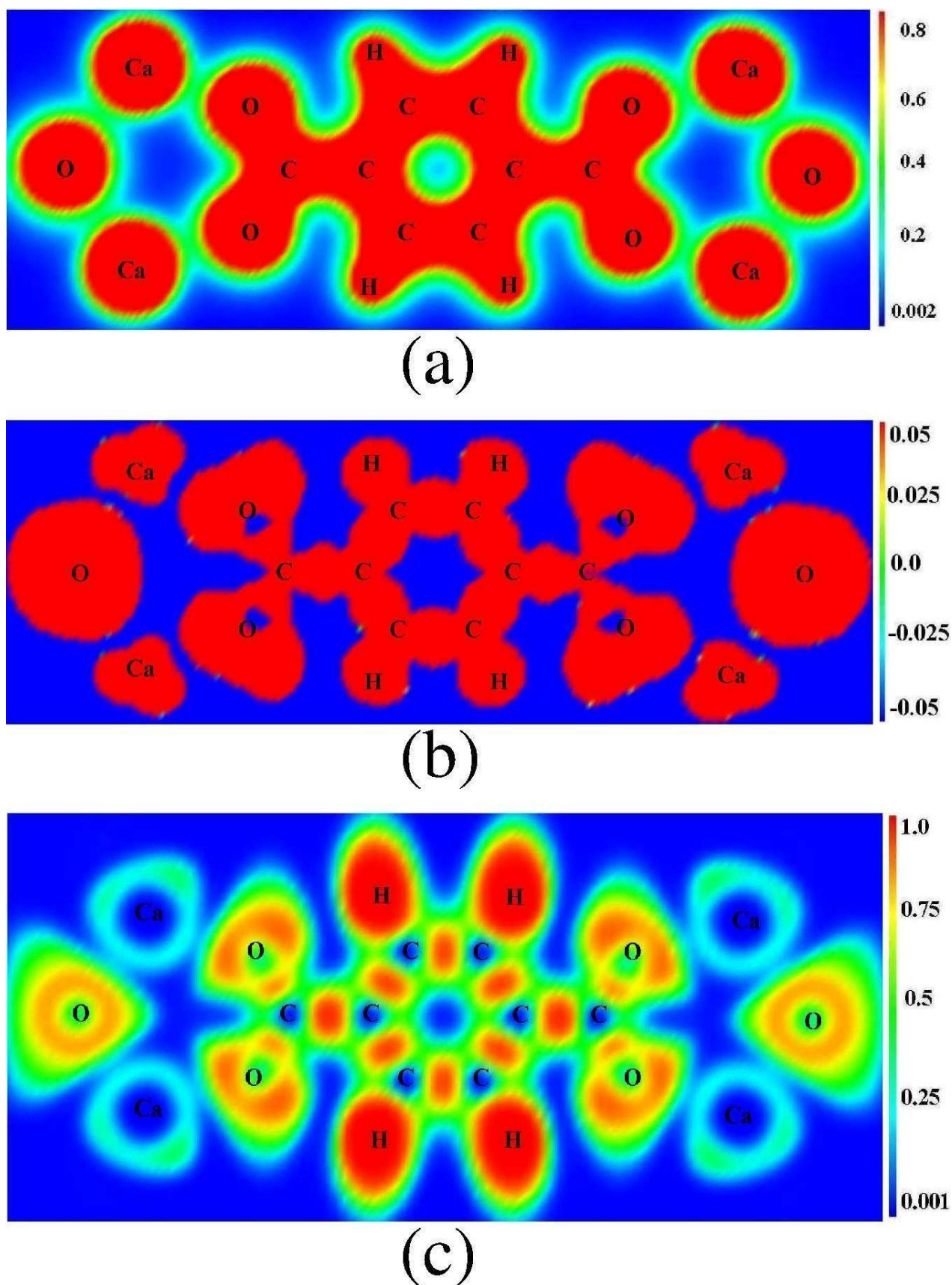
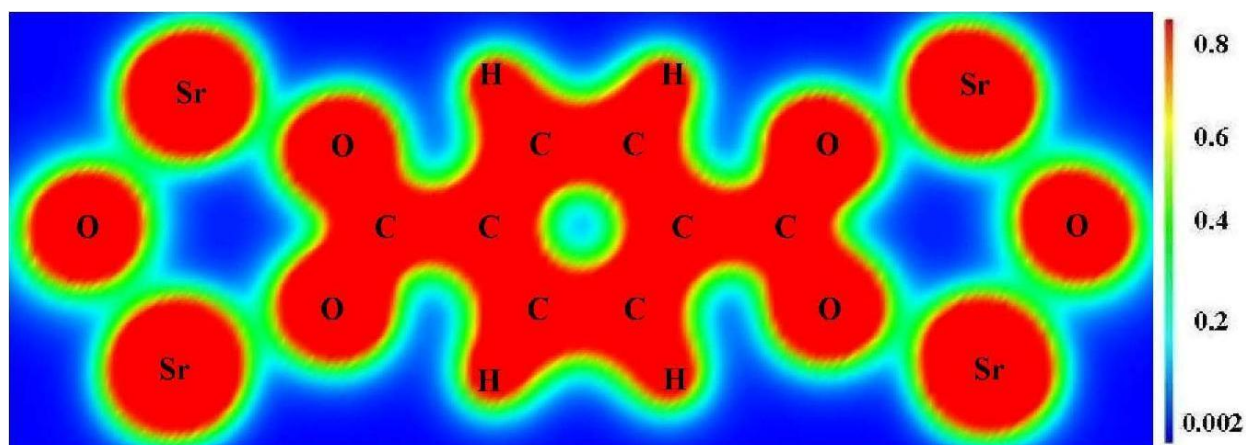
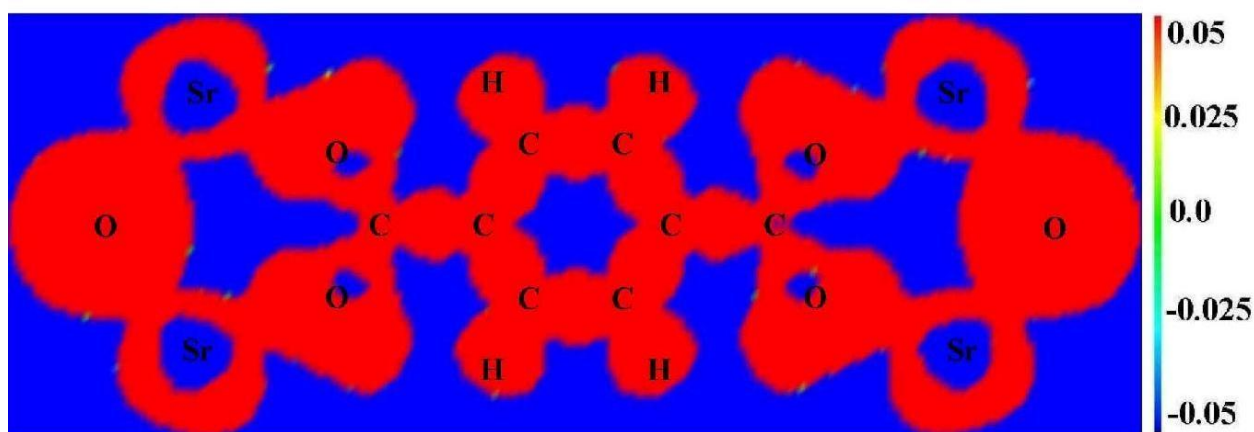


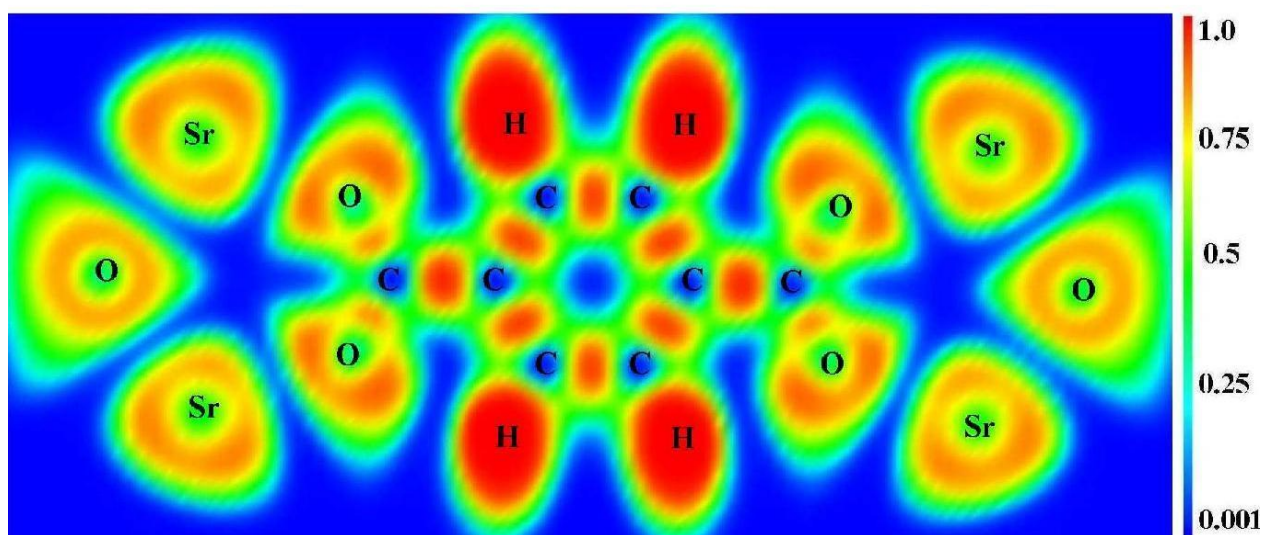
Figure S6. Calculated charge density (a), charge transfer (b), and electron localization function, ELF (c) plots for Ca-IRMOF-1 in the (110) plane.



(a)

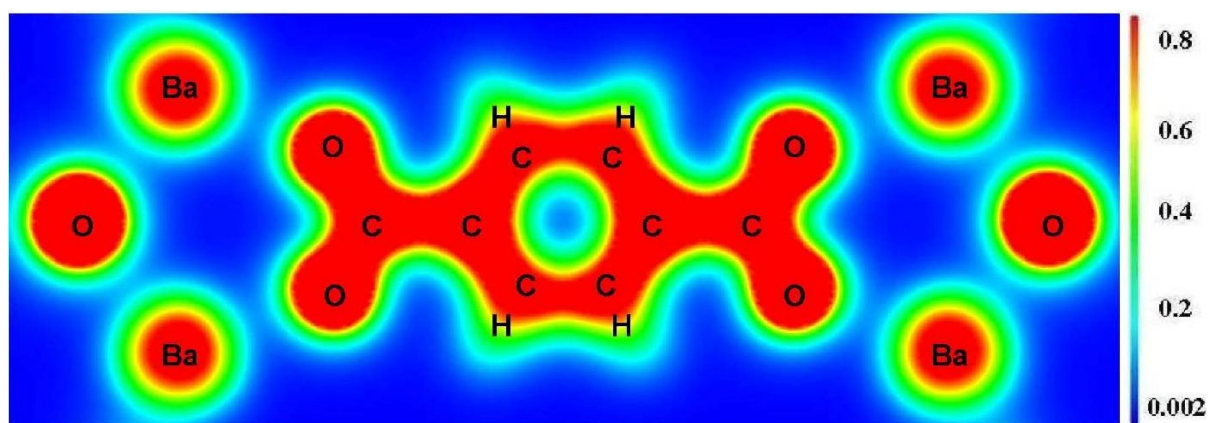


(b)

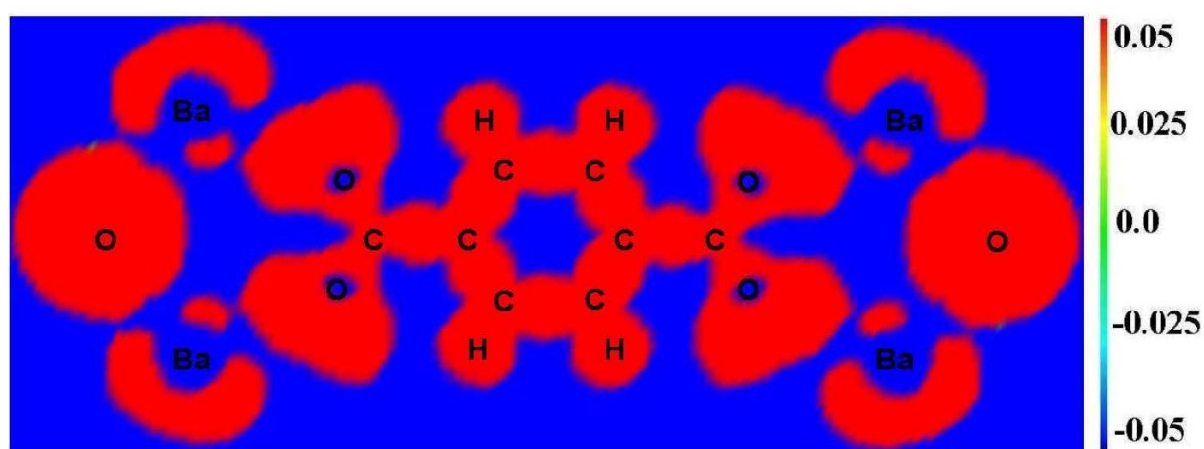


(c)

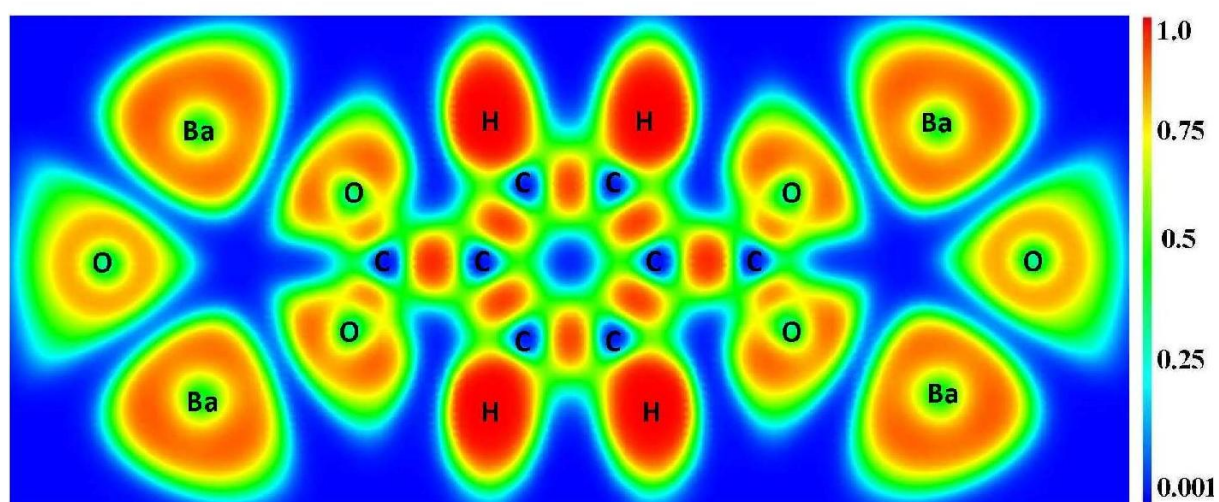
Figure S7. Calculated charge density (a), charge transfer (b), and electron localization function, ELF (c) plots for Sr-IRMOF-1 in the (110) plane.



(a)



(b)



(c)

Figure S8. Calculated charge density (a), charge transfer (b), and electron localization function, ELF (c) plots for Ba-IRMOF-1 in the (110) plane.

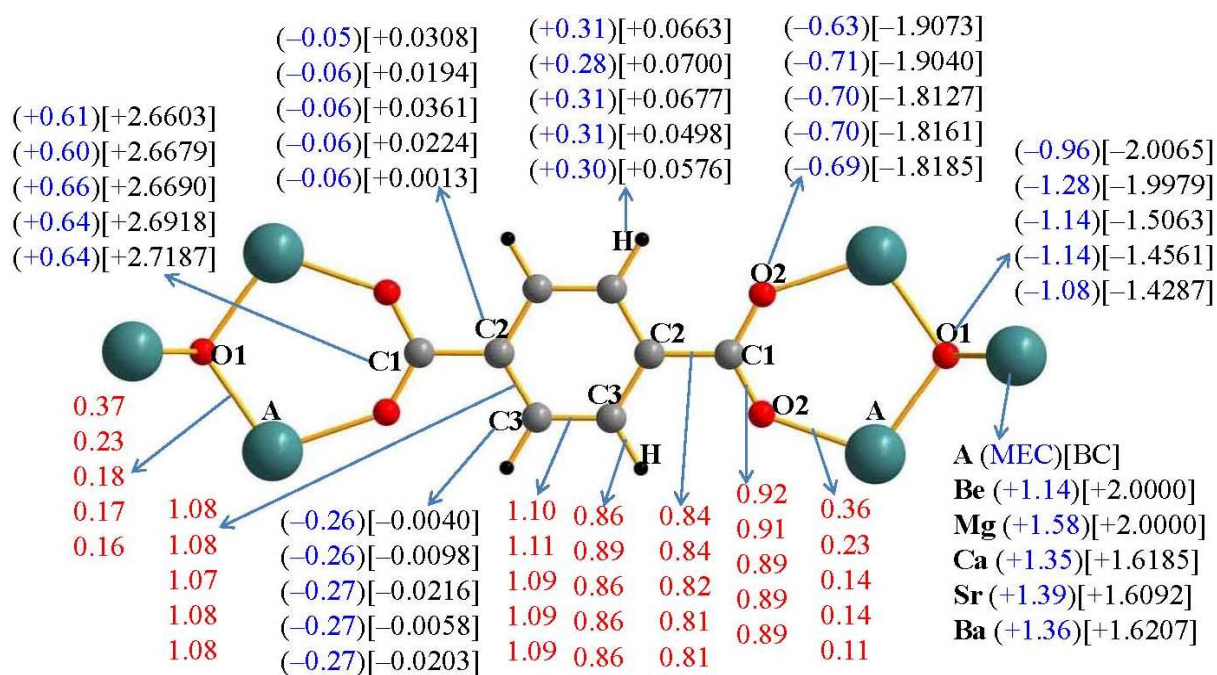


Figure S9. Calculated Bader charges (BC; black color in brackets), bond overlap populations (BOP; red color) and Mulliken effective charges (MEC; blue color in parentheses) for the A-IRMOF-1 series (A = Be, Mg, Ca, Sr, Ba).

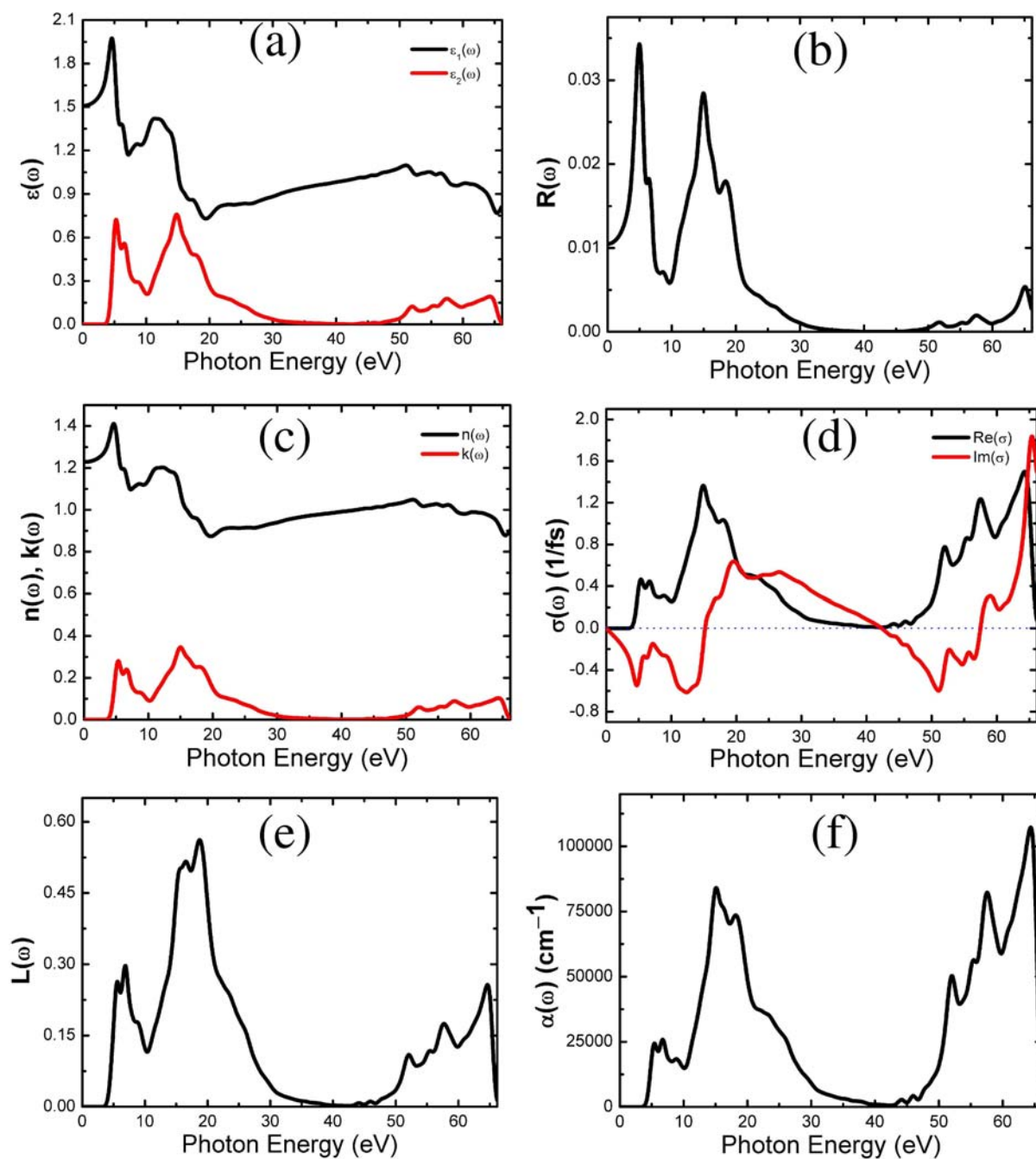


Figure S10. Calculated optical properties for Mg-IRMOF-1: (a) dielectric function $\epsilon(\omega)$, (b) reflectivity $R(\omega)$, (c) refractive index $n(\omega)$; extinction coefficient $k(\omega)$, (d) optical conductivity $\sigma(\omega)$, (e) energy loss function $L(\omega)$, and (f) absorption $\alpha(\omega)$.

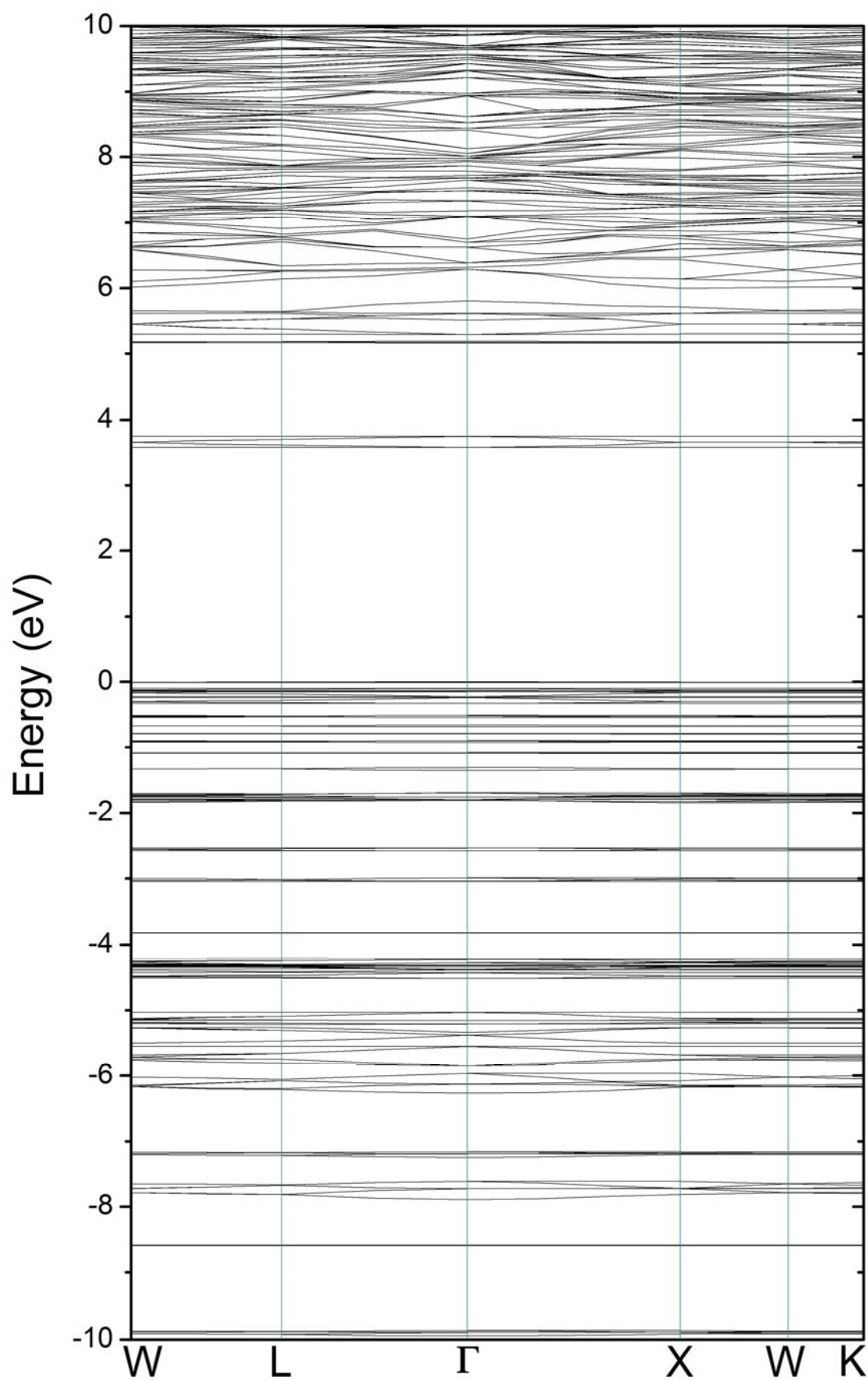


Figure S11. Band structure of Mg-IRMOF-1. The Fermi level is set to zero.

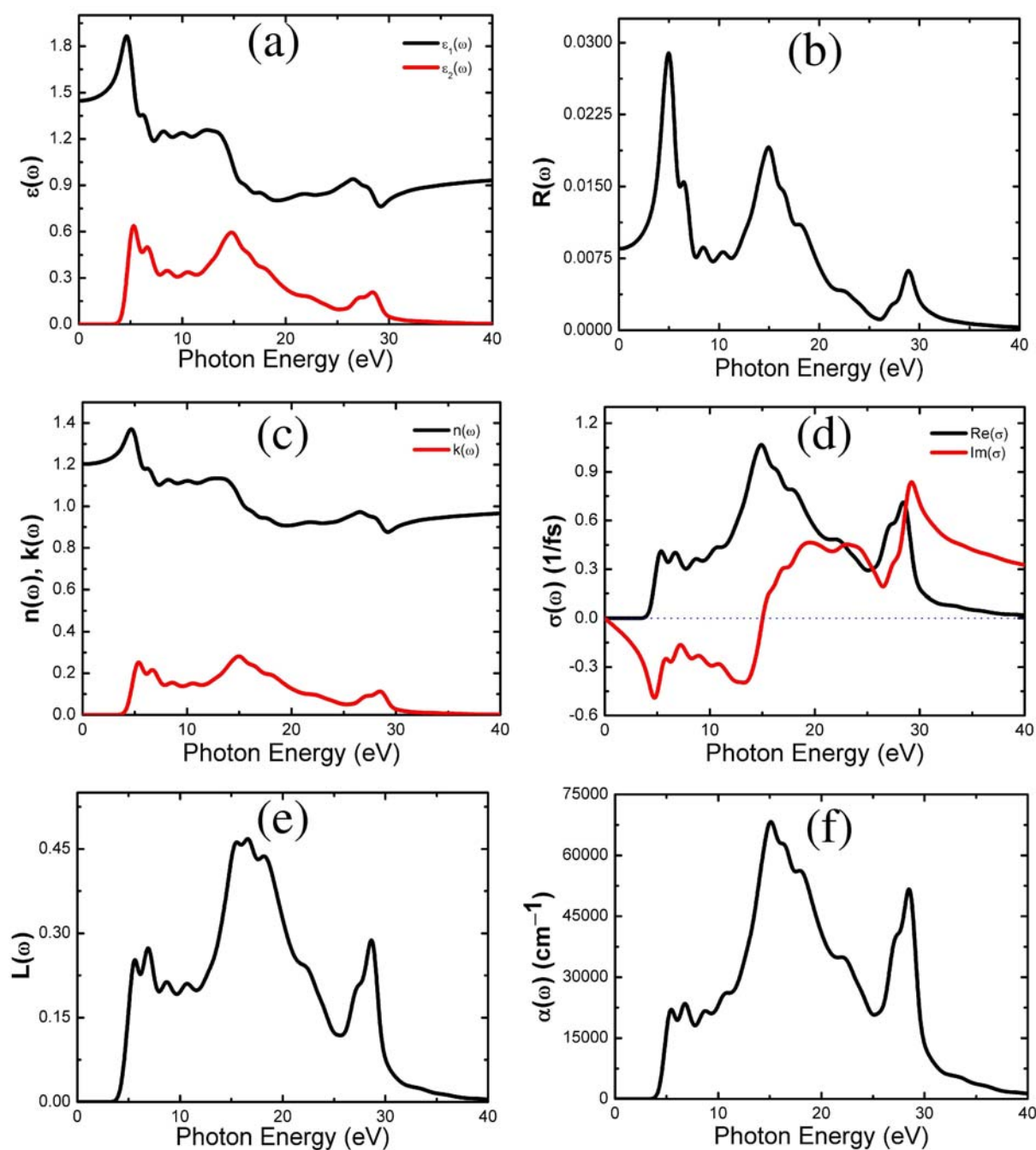


Figure S12. Calculated optical properties for Ca-IRMOF-1: (a) dielectric function $\epsilon(\omega)$, (b) reflectivity $R(\omega)$, (c) refractive index $n(\omega)$; extinction coefficient $k(\omega)$, (d) optical conductivity $\sigma(\omega)$, (e) energy loss function $L(\omega)$, and (f) absorption $\alpha(\omega)$.

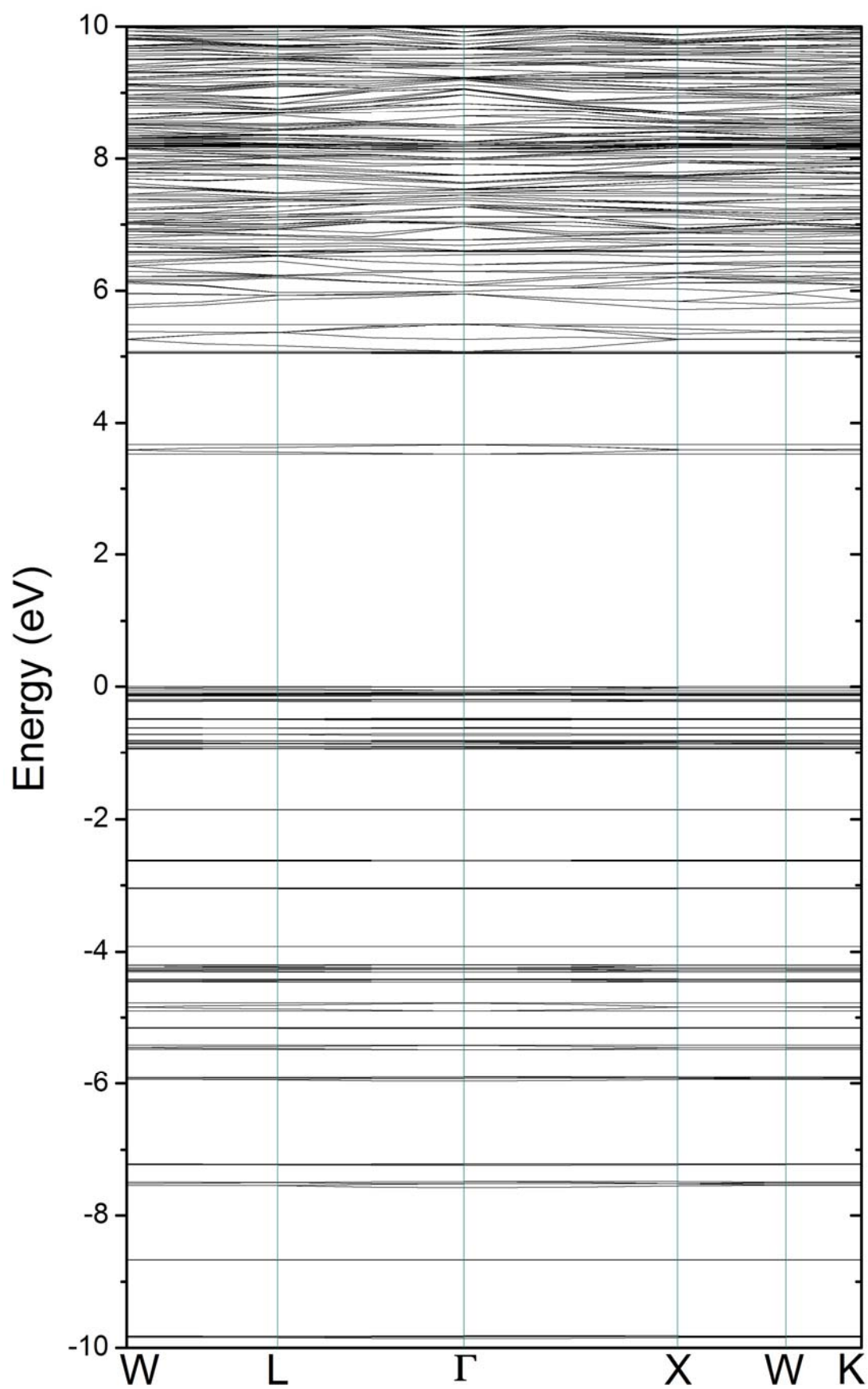


Figure S13. Band structure of Ca-IRMOF-1. The Fermi level is set to zero.

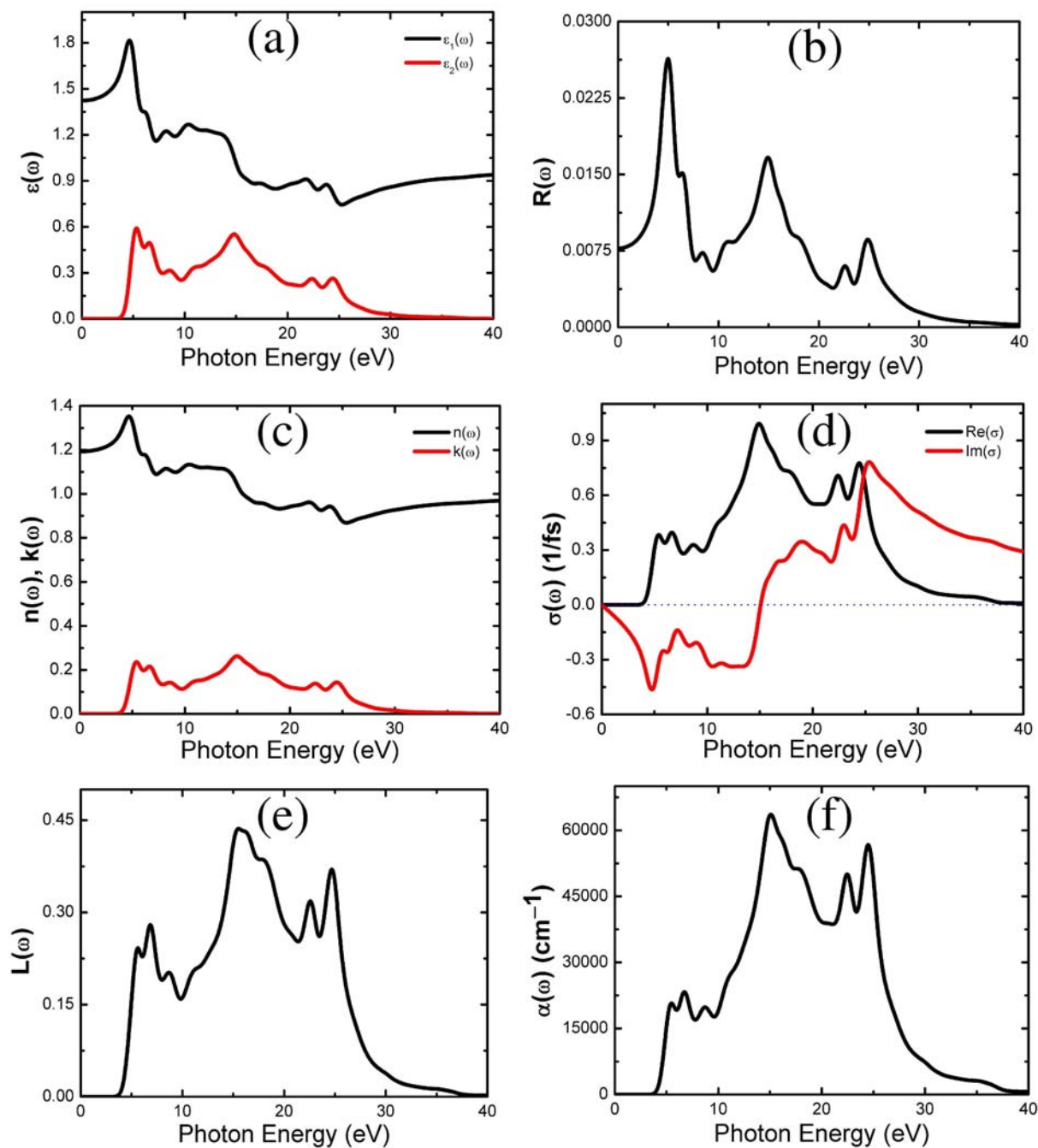


Figure S14. Calculated optical properties for Sr-IRMOF-1: (a) dielectric function $\epsilon(\omega)$, (b) reflectivity $R(\omega)$, (c) refractive index $n(\omega)$; extinction coefficient $k(\omega)$, (d) optical conductivity $\sigma(\omega)$, (e) energy loss function $L(\omega)$, and (f) absorption $\alpha(\omega)$.

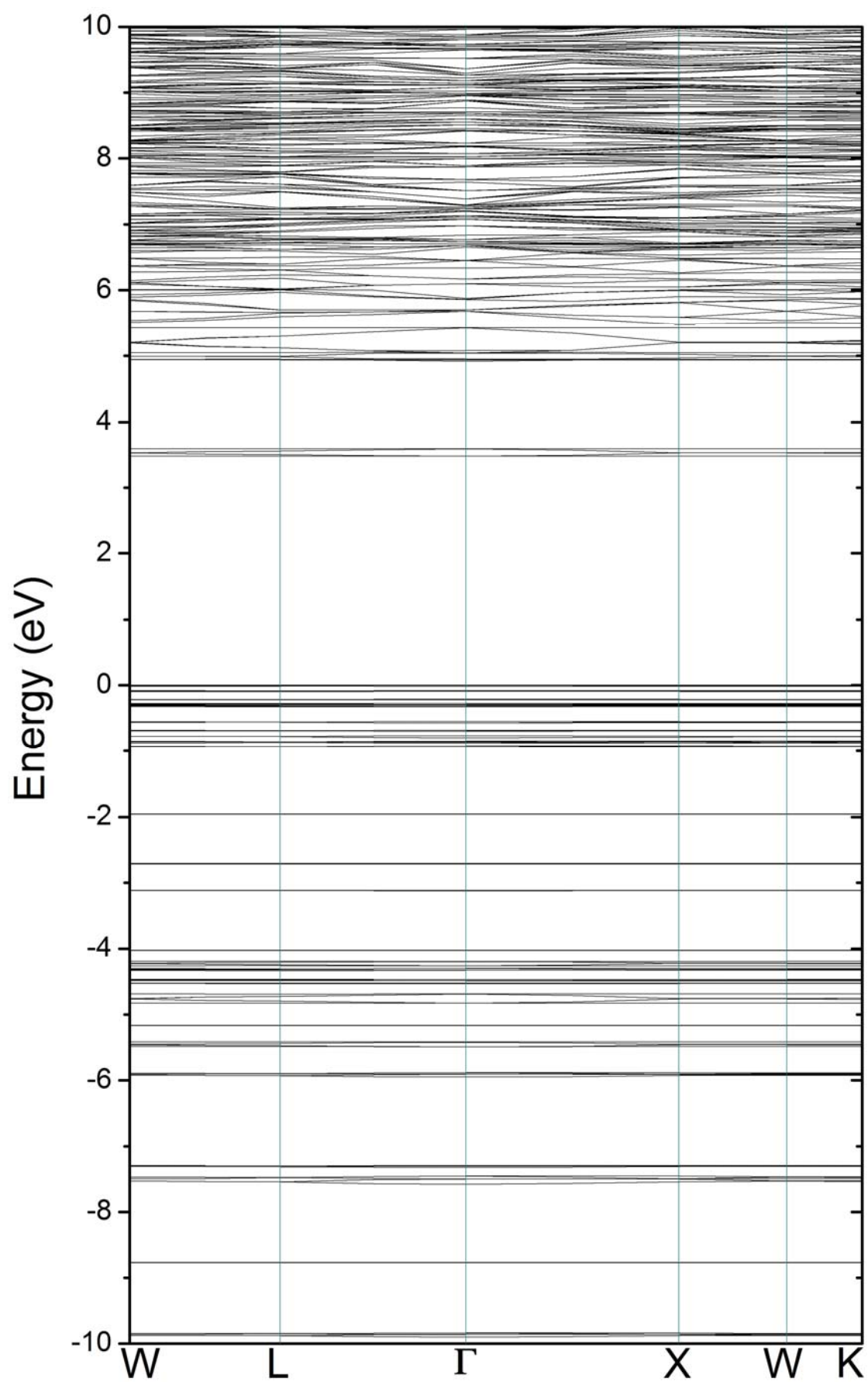


Figure S15. Band structure of Sr-IRMOF-1. The Fermi level is set to zero.

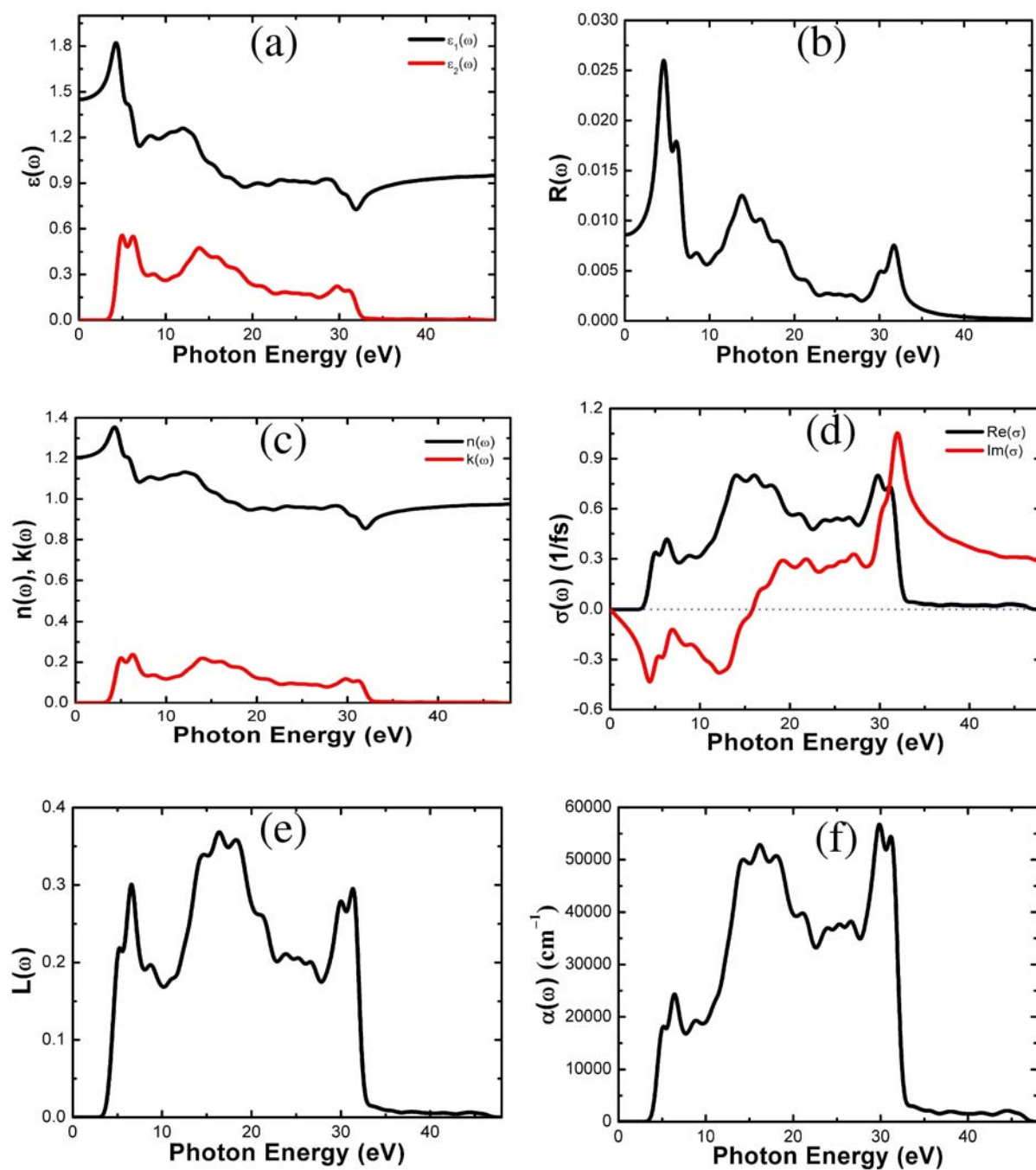


Figure S16. The calculated optical spectra for Ba-IRMOF-1: (a) dielectric function $\epsilon(\omega)$; (b) reflectivity $R(\omega)$; (c) refractive index $n(\omega)$ and extinction coefficient $k(\omega)$; (d) optical conductivity $\sigma(\omega)$; (e) energy loss function $L(\omega)$; (f) absorption coefficient $\alpha(\omega)$ (cm^{-1}).

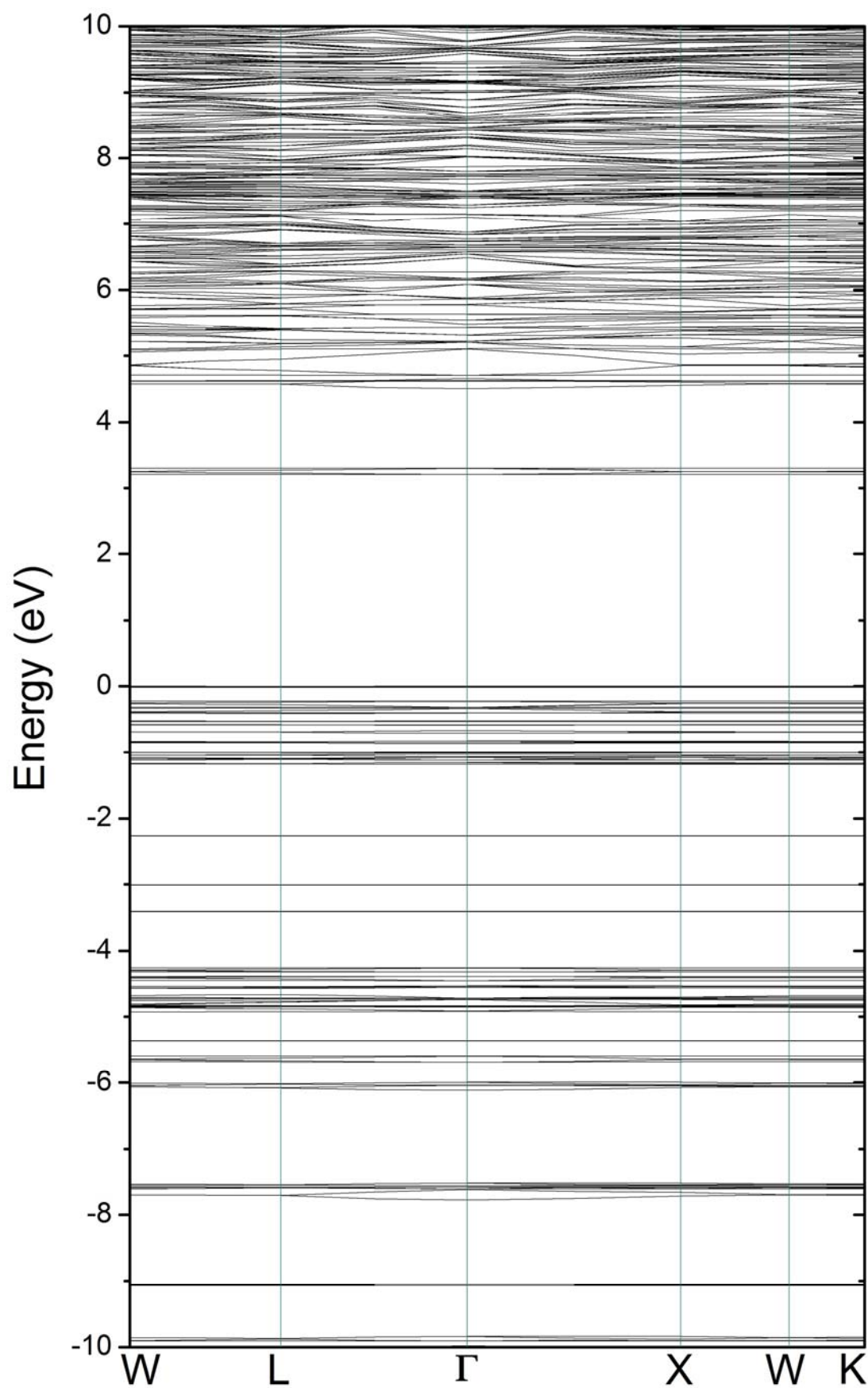


Figure S17. The calculated electronic band structure of Ba-IRMOF-1. The Fermi level is set to zero and placed in the valence band maximum.