

## **Supporting information**

# **Effect of air humidity on the removal of carbon tetrachloride from air using Cu-BTC Metal-Organic Framework**

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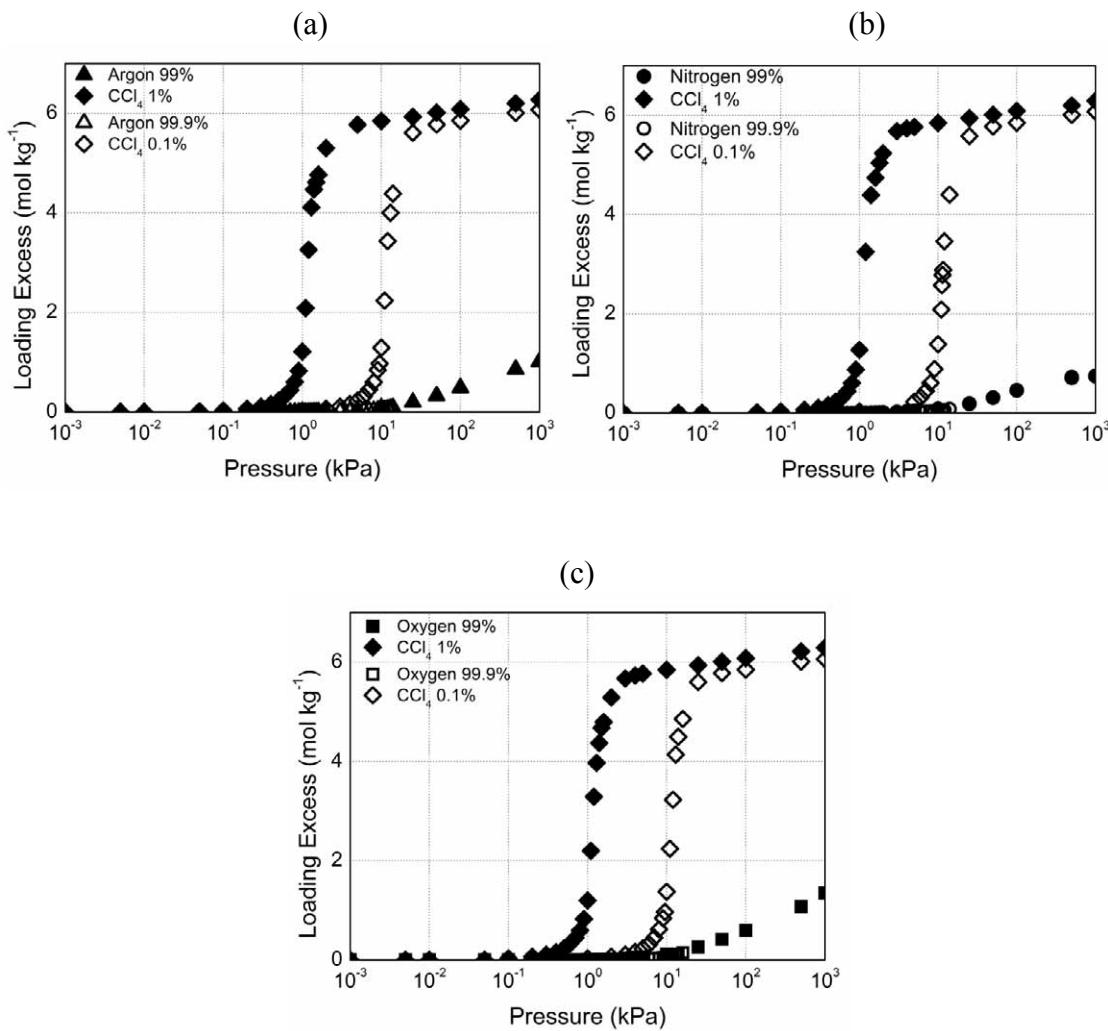
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**Table 1S:** Computed surface area of Cu-BTC using different types of molecules. The surface area is given in m<sup>2</sup>/g.

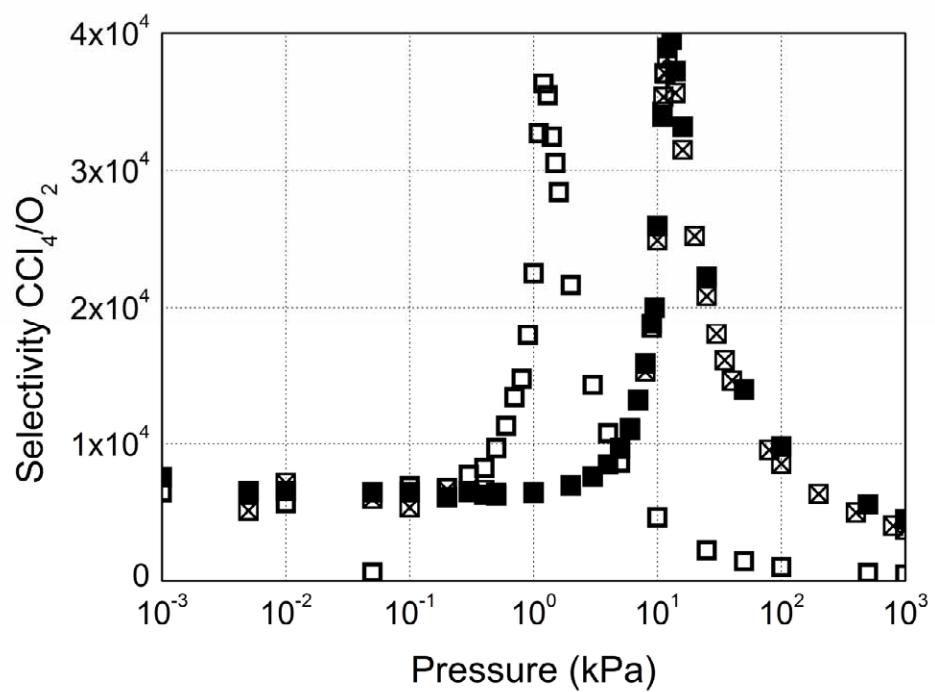
Hydrogen	Oxygen	Nitrogen	Argon	CCl <sub>4</sub>	Experimental <sup>1</sup>
2563.92	2513.30	2344.28	2296.84	1432.47	1958

**Table 2S:** Lennard-Jones parameters and partial charges used for Cu-BTC. The Lennard-Jones parameters were taken from the DREIDING generic force field <sup>2</sup>, except copper that was taken from the UFF force field <sup>3</sup>. The atomic charges for Cu-BTC were taken from reference <sup>4</sup>.

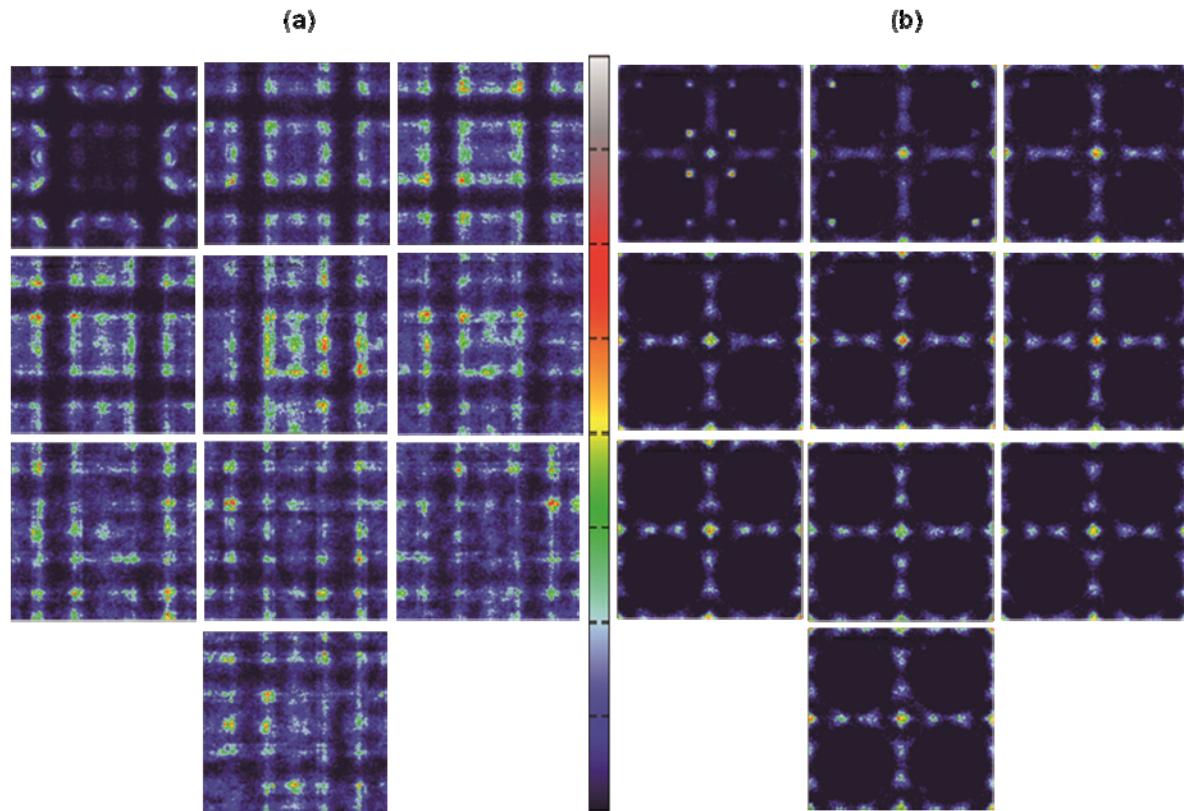
Cu-BTC			
Atoms	Charge [e <sup>-</sup> ]	ε/k <sub>B</sub> [K]	σ [Å]
Cu	1.0	2.518	3.114
O <sub>b</sub>	-0.6	48.19	3.03
C <sub>a</sub>	0.7	47.86	3.47
C <sub>b</sub>	0.0	47.86	3.47
C <sub>c</sub>	-0.15	47.86	3.47
H	0.15	7.65	2.85



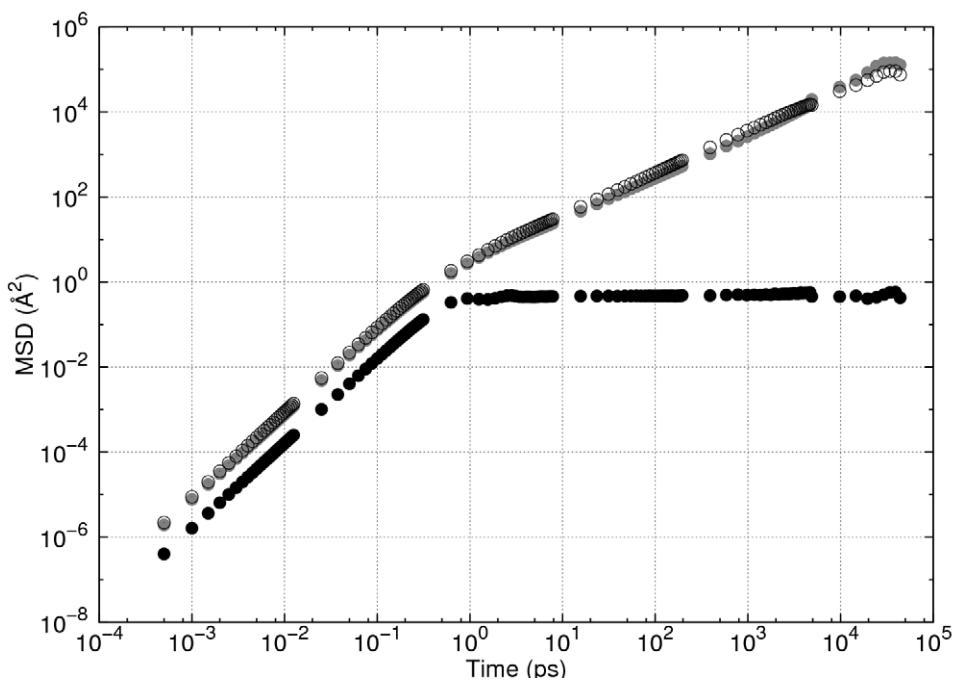
**Figure 1S:** Adsorption isotherms of (a) Ar/CCl<sub>4</sub>, (b) N<sub>2</sub>/CCl<sub>4</sub>, and (c) O<sub>2</sub>/CCl<sub>4</sub> binary mixtures in Cu-BTC. The adsorption isotherms of argon (triangles), nitrogen (circles), oxygen (squares), and carbon tetrachloride (rhombos) were computed at 298 K and at bulk partial fugacity ratio of 99:1 (full symbols) and 99.9:0.1 (empty symbols). Computed error bars are within the symbol size.



**Figure 2S:** Adsorption selectivity of carbon tetrachloride over oxygen obtained from the computed adsorption isotherms of  $\text{O}_2/\text{CCl}_4$  mixtures at a bulk partial fugacity ratio of 99:1 (empty symbols), 99.9:0.1 (full symbols), and 20.979:0.1 (crossed symbols)



**Figure 3S:** Average occupation profiles of (a) water and (b) carbon tetrachloride in Cu-BTC. The profiles were obtained from the molecular simulations of five-component mixtures of carbon tetrachloride in air with relative humidity from 10%-30% (top left-right) to 100% (bottom). The same color gradation (from dark blue to white) is employed in all figures, although the total number of molecules present in the unit cell is different for each calculation.



**Figure 4S:** Representation of the Mean Square Displacement (MSD) of oxygen (gray circles), nitrogen (empty circles), and carbon tetrachloride (black circles) as pure components.

- 1 S. Q. Ma and H. C. Zhou, *Chemical Communications*, 2010, **46**, 44-53.
- 2 S. L. Mayo, B. D. Olafson and W. A. Goddard, *Journal of Physical Chemistry*, 1990, **94**, 8897-8909.
- 3 A. K. Rappe, C. J. Casewit, K. S. Colwell, W. A. Goddard and W. M. Skiff, *Journal of the American Chemical Society*, 1992, **114**, 10024-10035.
- 4 D. Farrusseng, C. Daniel, C. Gaudillere, U. Ravon, Y. Schuurman, C. Mirodatos, D. Dubbeldam, H. Frost and R. Q. Snurr, *Langmuir*, 2009, **25**, 7383-7388.