

Role of the local orderon the thermal *cis*-to-*trans* isomerisation kinetics of azo-dyes in nematic liquid crystals

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Supplementary Information

Table S1. First-order rate constants, k , and half-life times, $t_{1/2}$, for the thermal *cis*-to-*trans* isomerisation process of **AZO6** in different isotropic solvents and host nematogens at different temperatures. The experimental error associated to temperature and first-order rate constant was of ± 0.1 K and less than 10 %, respectively.

Solvent	Temperature / K	$10^5 \times k / \text{s}^{-1}$	$t_{1/2} / \text{h}$
acetonitrile	298	1.47	13
	308	4.83	4.0
	313	9.08	2.1
	318	15.9	1.2
	328	47.3	0.41
ethanol	298	2.30	8.4
	308	7.68	2.5
	313	13.3	1.4
	318	25.2	0.76
	328	71.0	0.27
toluene	298	1.95	9.9
	308	6.50	3.0
	313	11.6	1.7
	318	20.5	0.94
	328	60.2	0.32
cyclohexane	298	2.38	8.1
	308	7.45	2.6
	313	13.3	1.4

	318	23.8	0.81
	328	66.2	0.29
5CB nematic	294	2.93	6.6
	298	4.28	4.5
	301	5.67	3.4
	303	6.88	2.8
	305	8.23	2.3
5CB isotropic	313	17.8	1.1
	318	30.4	0.63
	323	51.1	0.38
	328	85.2	0.22
	333	143	0.13
ZLI-1695nematic	298	6.12	3.1
	303	10.4	1.8
	308	18.2	1.1
	313	30.3	0.63
	318	50.2	0.38
	323	80.1	0.24
MC4C6nematic	306	9.01	2.1
	309	12.0	1.6
	311	14.6	1.3
	313	17.5	1.1
MC4C6 isotropic	323	46.5	0.41
	328	76.4	0.25
	333	127	0.15
	338	210	0.09
MC5CHXnematic	308	16.3	1.2
	313	27.3	0.70
	318	44.1	0.44
	323	70.4	0.27