

Electronic Supplementary Information (ESI)

Molecular Wheel to Monocyclic Ring Transition in Boron-Carbon Mixed Clusters $C_2B_6^-$ and $C_3B_5^-$

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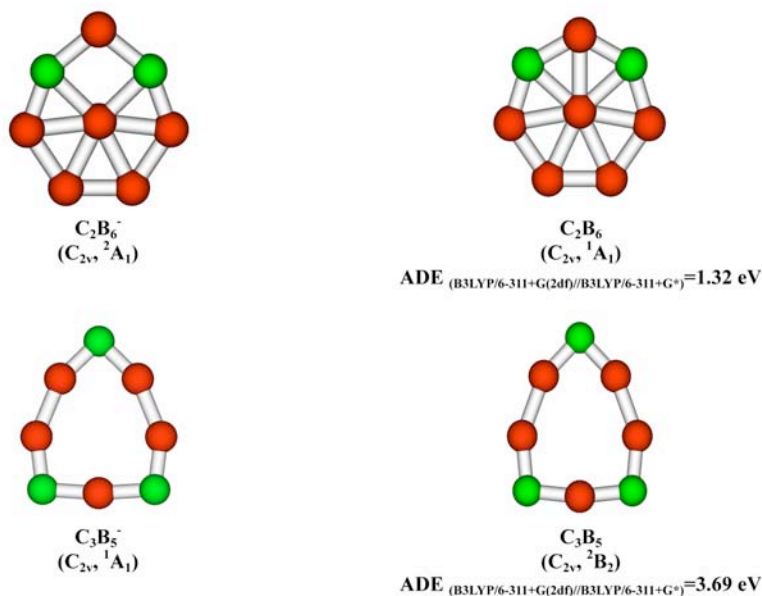


Fig. 1S Adiabatic Detachment Energies (ADE) of the $C_2B_6^-$ and $C_3B_5^-$ clusters

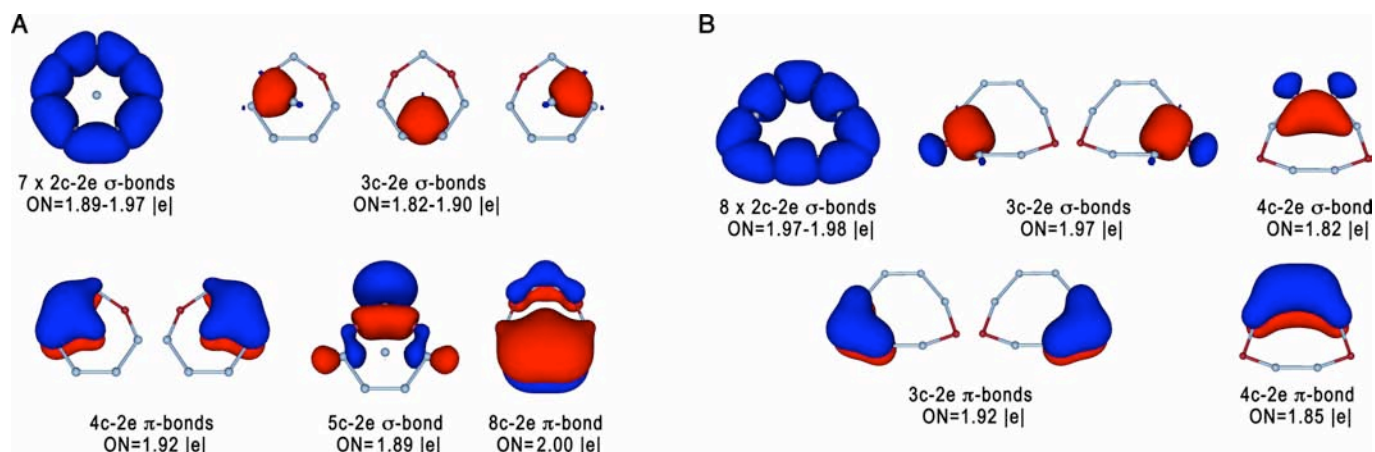


Fig. 2S The chemical bonding patterns revealed by the AdNDP analysis for the $C_2B_6^{2-}$ cluster at the optimized geometries of the global minimum wheel-type structure I.1 (**A**) and lowest ring isomer I.6 (**B**) of the $C_2B_6^{2-}$ cluster. The extra electron was added in order to avoid complications of the chemical bonding picture caused by spin polarization in the open-shell $C_2B_6^{2-}$ cluster. The 2c-2e C-B and B-B bonds of both isomers are superimposed on a single molecular framework.

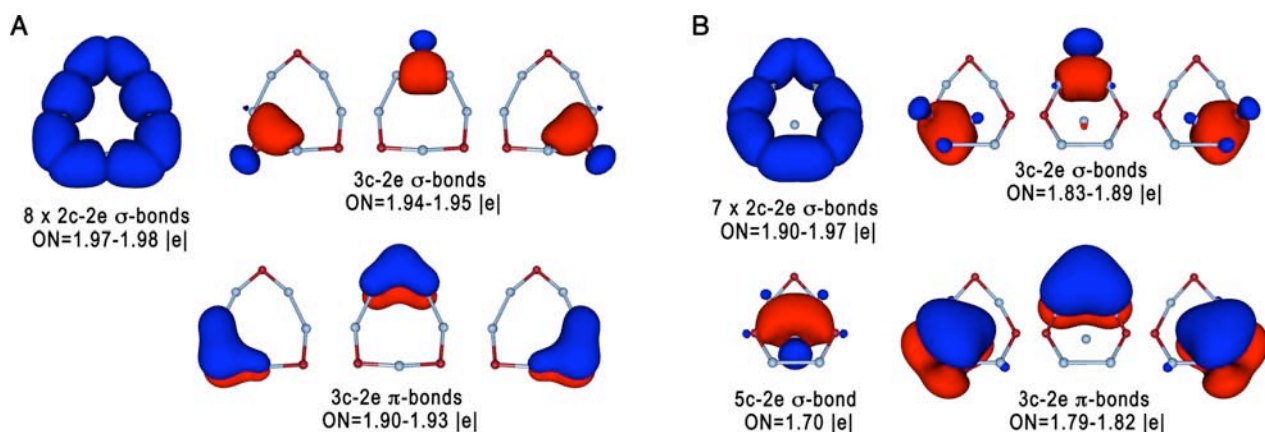


Fig. 3S The chemical bonding patterns revealed by the AdNDP analysis for the global minimum ring isomer II.1 (**A**) and the lowest wheel-type isomer II.5 (**B**) of the $C_3B_5^{2-}$ cluster. The 2c-2e C-B and B-B bonds of both isomers are superimposed on a single molecular framework.

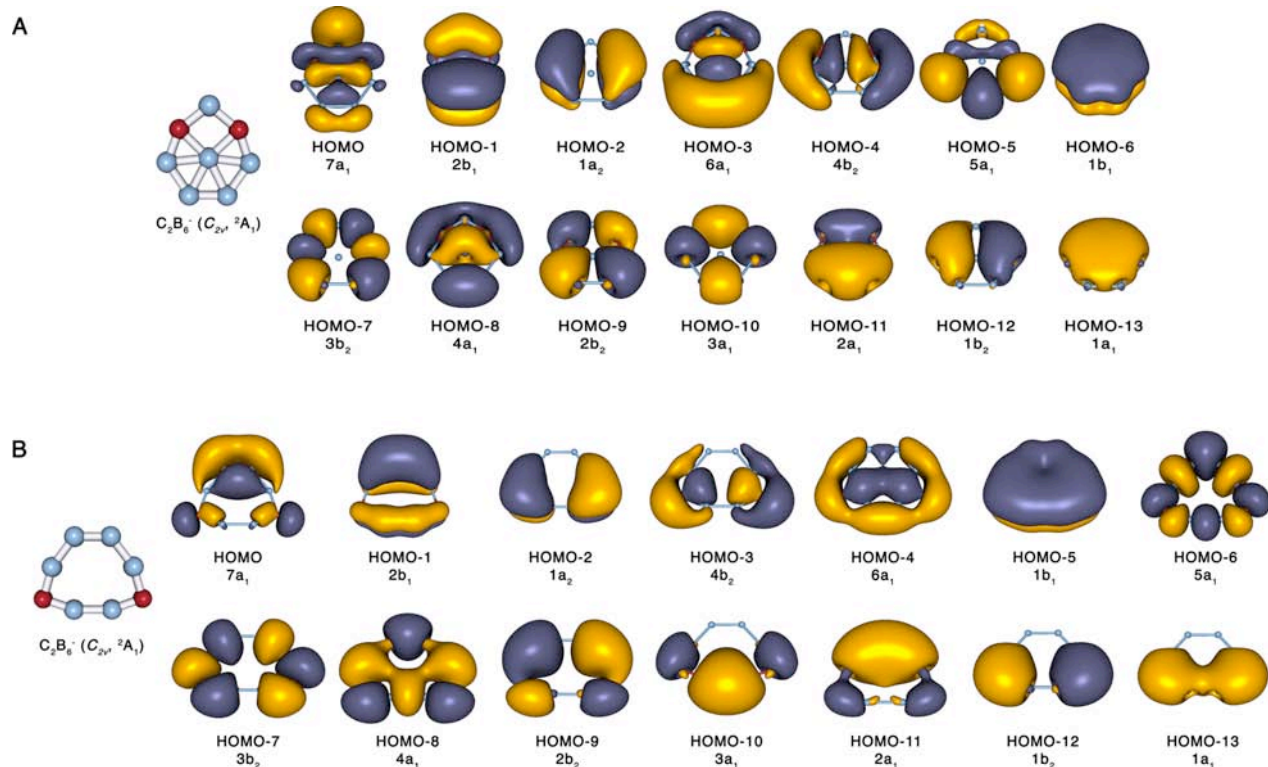


Fig. 4S Valence canonical molecular orbitals (B3LYP/6-311+G*) of the global minimum wheel-type structure I.1 (**A**) and the lowest-lying ring isomer I.6 (**B**) of the C_2B_6 cluster.

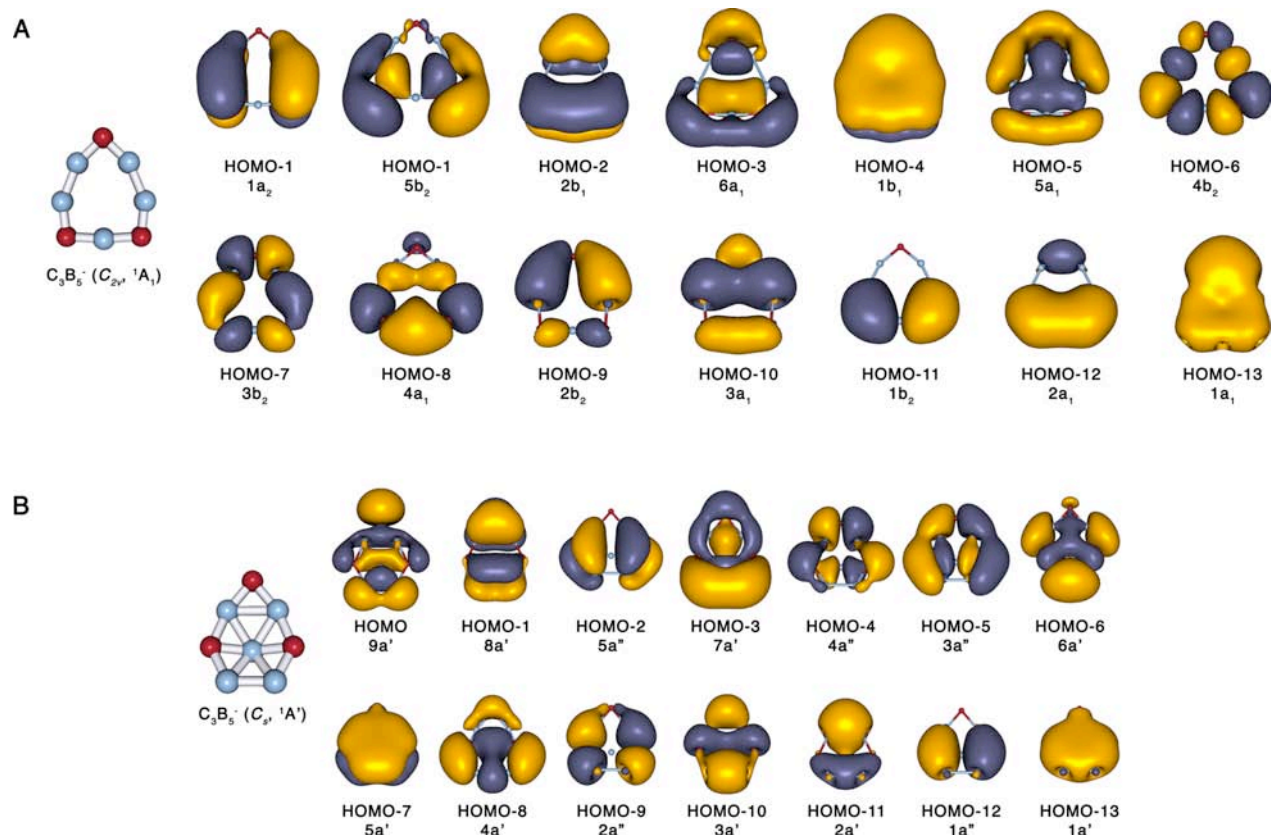


Fig. 5S Valence canonical molecular orbitals (B3LYP/6-311+G*) of the global minimum ring structure II.1 (**A**) and the lowest-lying wheel isomer II.5 (**B**) of the C_3B_5 cluster. The order of canonical molecular orbitals for the isomer II.1 (**A**) at B3LYP/6-311+G* is different from the order of the spectroscopy states calculated by TD-B3LYP, ROVGF and RCCSD(T) presented in Table 2.