

# Supportin Information on A Theoretical Study on Excited State Double Proton Transfer Reaction of 7-Azaindole Dimer: *Ab Initio* Potential Energy Surface and its Empirical Valence Bond Model

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## Atom indices of a 7-AI monomer

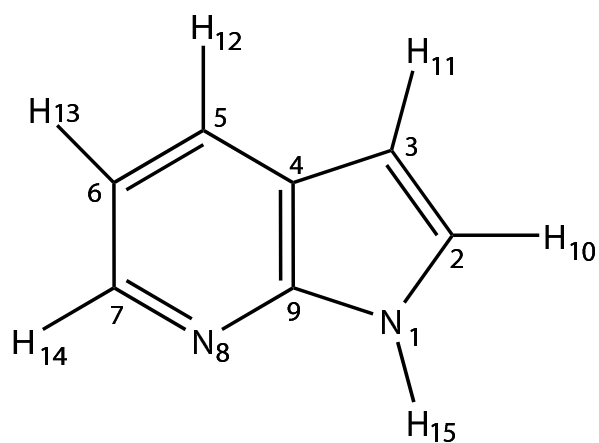


Figure S1: Atom indices of a 7-AI monomer. For a tautomer, the transferring H is labeled as 15.

## Definition of internal coordinates

Table S1: Definition of internal coordinates [ () for monomers of a tautomer. ]

coordinate	type	induces of atoms				
$r_1$	bond	1	2			
$r_2$	bond	2	3			
$r_3$	bond	3	4			
$r_4$	bond	4	5			
$r_5$	bond	5	6			
$r_6$	bond	6	7			
$r_7$	bond	7	8			
$r_8$	bond	8	9			
$r_9$	bond	9	4			
$r_{10}$	bond	9	1			
$r_{11}$	bond	2	10			
$r_{12}$	bond	3	11			
$r_{13}$	bond	5	12			
$r_{14}$	bond	6	13			
$r_{15}$	bond	7	14			
$\theta_{16}$	angle	1	2	10		
$\theta_{17}$	angle	2	3	11		
$\theta_{18}$	angle	4	5	12		
$\theta_{19}$	angle	5	6	13		
$\theta_{20}$	angle	6	7	14		
$\phi_{21}$	dihedral	9	1	2	10	
$\phi_{22}$	dihedral	1	2	3	11	
$\phi_{23}$	dihedral	9	4	5	12	
$\phi_{24}$	dihedral	4	5	6	13	
$\phi_{25}$	dihedral	5	6	7	14	
$\theta_{26}$	angle	9	1	2		
$\theta_{27}$	angle	1	2	3		
$\theta_{28}$	angle	2	3	4		
$\theta_{29}$	angle	3	4	9		
$\theta_{30}$	angle	4	9	1		
$\theta_{31}$	angle	7	8	9		
$\theta_{32}$	angle	8	9	4		
$\theta_{33}$	angle	9	4	5		
$\theta_{34}$	angle	4	5	6		
$\theta_{35}$	angle	5	6	7		
$\theta_{36}$	angle	6	7	8		
$\phi_{37}$	dihedral	9	1	2	3	
$\phi_{38}$	dihedral	1	2	3	4	
$\phi_{39}$	dihedral	2	3	4	9	
$\phi_{40}$	dihedral	3	4	9	1	
$\phi_{41}$	dihedral	4	9	1	2	
$\phi_{42}$	dihedral	7	8	9	4	
$\phi_{43}$	dihedral	8	9	4	5	
$\phi_{44}$	dihedral	9	4	5	6	
$\phi_{45}$	dihedral	4	5	6	7	
$\phi_{46}$	dihedral	5	6	7	8	
$\phi_{47}$	dihedral	6	7	8	9	
$\phi_{48}$	dihedral	1	9	4	5	
$\phi_{49}$	dihedral	8	9	4	3	
$r_{50}$	bond	1 ( 8)	15 (15)			
$\theta_{51}$	angle	9 ( 9)	1 ( 8)	15 (15)		
$\phi_{52}$	dihedral	4 ( 4)	9 ( 9)	1 ( 8)	15 (15)	

## The internal coordinates using in Eq. (5)

The internal coordinates using in Eq. (5) (natural internal coordinates) is given as follows:

$$q_i = r_i \quad (\text{for } i = 1, \dots, 15), \quad (\text{S1})$$

$$q_i = \theta_i \quad (\text{for } i = 16, \dots, 20), \quad (\text{S2})$$

$$q_i = \phi_i \quad (\text{for } i = 21, \dots, 25), \quad (\text{S3})$$

$$q_{26} = 0.6324\theta_{26} - 0.5117\theta_{27} + 0.1954\theta_{28} + 0.1954\theta_{29} - 0.5116\theta_{30}, \quad (\text{S4})$$

$$q_{27} = 0.3717\theta_{27} - 0.6015\theta_{28} + 0.6015\theta_{29} - 0.3717\theta_{30}, \quad (\text{S5})$$

$$q_{28} = 0.5774\theta_{31} - 0.2887\theta_{32} - 0.2887\theta_{33} + 0.5774\theta_{34} - 0.2887\theta_{35} - 0.2887\theta_{36}, \quad (\text{S6})$$

$$q_{29} = 0.5000\theta_{32} - 0.5000\theta_{33} + 0.5000\theta_{35} - 0.5000\theta_{36}, \quad (\text{S7})$$

$$q_{30} = 0.4082\theta_{31} - 0.4082\theta_{32} + 0.4082\theta_{33} - 0.4082\theta_{34} + 0.4082\theta_{35} - 0.4082\theta_{36}, \quad (\text{S8})$$

$$q_{31} = 0.6324\phi_{37} - 0.5117\phi_{38} + 0.1954\phi_{39} + 0.1954\phi_{40} - 0.5116\phi_{41}, \quad (\text{S9})$$

$$q_{32} = 0.3717\phi_{38} - 0.6015\phi_{39} + 0.6015\phi_{40} - 0.3717\phi_{41}, \quad (\text{S10})$$

$$q_{33} = 0.5774\phi_{42} - 0.2887\phi_{43} - 0.2887\phi_{44} + 0.5774\phi_{45} - 0.2887\phi_{46} - 0.2887\phi_{47}, \quad (\text{S11})$$

$$q_{34} = 0.5000\phi_{43} - 0.5000\phi_{44} + 0.5000\phi_{46} - 0.5000\phi_{47}, \quad (\text{S12})$$

$$q_{35} = 0.4082\phi_{42} - 0.4082\phi_{43} + 0.4082\phi_{44} - 0.4082\phi_{45} + 0.4082\phi_{46} - 0.4082\phi_{47}, \quad (\text{S13})$$

$$q_{36} = 0.7071\phi_{48} - 0.7071\phi_{49}, \quad (\text{S14})$$

where  $\{r_i\}$ ,  $\{\theta_i\}$  and  $\{\phi_i\}$  are given in Table S1.

## The definition of $\vartheta$ in Eq. (10)

The definition of  $\vartheta$  in Eq. (10) is an angle between two planes each of which includes two protons and the averaged position of C and N atoms of each monomer.

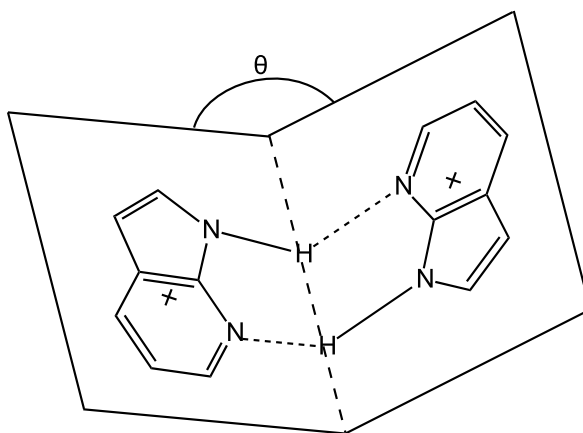


Figure S2: Graphical definition of  $\vartheta$  ( $\theta$  in the figure). “H” are the transferring H atoms ( $H_{29}$  and  $H_{30}$ ), and “ $\times$ ” are the averaged positions of C and N atoms of each monomer, respectively.

## The internal coordinates at the equilibrium geometries in Eq. (5)

Table S2: The internal coordinates at the equilibrium geometries. [unit: a.u. for length, radian for angle]

coordinates	A	A*	T	T*
$q_1^X$	2.5925372	2.4518200	2.6275132	2.4308217
$q_2^X$	2.5667306	2.7180162	2.5654832	2.7399253
$q_3^X$	2.7218150	2.7696040	2.7353528	2.8063599
$q_4^X$	2.6708075	2.7061183	2.5859045	2.7161306
$q_5^X$	2.6135025	2.7064303	2.7208457	2.7266622
$q_6^X$	2.6841343	2.6052897	2.5779752	2.5511454
$q_7^X$	2.4917979	2.5687652	2.6002479	2.6132843
$q_8^X$	2.5383235	2.5426987	2.5467361	2.6155990
$q_9^X$	2.6263435	2.5862249	2.7650053	2.5638161
$q_{10}^X$	2.5811553	2.7053270	2.4459721	2.6498177
$q_{11}^X$	2.0231191	2.0231007	2.0289360	2.0289667
$q_{12}^X$	2.0249011	2.0243302	2.0238963	2.0252201
$q_{13}^X$	2.0321786	2.0257161	2.0312925	2.0238338
$q_{14}^X$	2.0299443	2.0325190	2.0277068	2.0300379
$q_{15}^X$	2.0336426	2.0341529	2.0252684	2.0284492
$q_{16}^X$	2.1059131	2.1557256	2.0682843	2.1283642
$q_{17}^X$	2.1972201	2.1816701	2.2284846	2.2324353
$q_{18}^X$	2.1227640	2.1615093	2.1240036	2.1524406
$q_{19}^X$	2.1169384	2.0951695	2.1058832	2.0970655
$q_{20}^X$	2.0861695	2.0634408	2.1483318	2.1217328
$q_{21}^X$	3.1415927	3.1415927	3.1415927	3.1415927
$q_{22}^X$	3.1415927	3.1415927	3.1415927	3.1415927
$q_{23}^X$	3.1415927	3.1415927	3.1415927	3.1415927
$q_{24}^X$	3.1415927	3.1415927	3.1415927	3.1415927
$q_{25}^X$	3.1415927	3.1415927	3.1415927	3.1415927
$q_{26}^X$	-0.0227403	0.0454976	-0.1901739	-0.1421547
$q_{27}^X$	0.0119608	0.0027608	-0.0128891	-0.0153030
$q_{28}^X$	-0.1225686	-0.2819491	0.0009346	-0.1315385
$q_{29}^X$	0.0461145	0.0560220	-0.0058846	-0.0022476
$q_{30}^X$	-0.1192271	-0.1390426	0.0154060	0.0098503
$q_{31}^X$	0.0000000	0.0000000	0.0000000	0.0000000
$q_{32}^X$	0.0000000	0.0000000	0.0000000	0.0000000
$q_{33}^X$	0.0000000	0.0000000	0.0000000	0.0000000
$q_{34}^X$	0.0000000	0.0000000	0.0000000	0.0000000
$q_{35}^X$	0.0000000	0.0000000	0.0000000	0.0000000
$q_{36}^X$	0.0000000	0.0000000	0.0000000	0.0000000

## Partial charges in Eq. (8)

Table S3: Partial charges [unit:a.u.]

charge	A	A*	T	T*
$Q_1^{0,X}$	-0.169834	-0.082821	-0.581416	-0.490232
$Q_2^{0,X}$	-0.263313	-0.066338	0.074665	0.074726
$Q_3^{0,X}$	-0.327666	-0.020124	-0.407244	-0.066384
$Q_4^{0,X}$	0.095316	0.036474	0.162125	0.081614
$Q_5^{0,X}$	-0.151021	-0.339202	-0.112857	-0.235512
$Q_6^{0,X}$	-0.204233	-0.271592	-0.202760	-0.241682
$Q_7^{0,X}$	0.170251	0.069170	-0.187674	-0.243371
$Q_8^{0,X}$	-0.478876	-0.637976	-0.068190	-0.137908
$Q_9^{0,X}$	0.264236	0.263393	0.275304	0.237133
$Q_{10}^{0,X}$	0.223053	0.180178	0.156244	0.123750
$Q_{11}^{0,X}$	0.193080	0.170111	0.157577	0.123871
$Q_{12}^{0,X}$	0.155419	0.145060	0.148979	0.145596
$Q_{13}^{0,X}$	0.118982	0.134074	0.152760	0.171421
$Q_{14}^{0,X}$	0.086008	0.117481	0.175848	0.193623
$Q_{15}^{0,X}$	0.288597	0.302112	0.256638	0.263355

## The dumping function in Eq. (7)

The dumping function used in Eq. (7) is as follows:

$$f(r_{\alpha\beta}) = \begin{cases} \left(\frac{r_{\alpha\beta}}{s_{\alpha\beta}}\right)^4 - 2\left(\frac{r_{\alpha\beta}}{s_{\alpha\beta}}\right)^2 + 2\left(\frac{r_{\alpha\beta}}{s_{\alpha\beta}}\right) & (r_{\alpha\beta} < s_{\alpha\beta}), \\ 1 & (r_{\alpha\beta} \geq s_{\alpha\beta}), \end{cases} \quad (\text{S15})$$

where

$$s_{\alpha\beta} = A(a_{\alpha}a_{\beta})^{1/6}. \quad (\text{S16})$$

Here  $\{a_{\alpha}\}$  denote the “volume” of each atom derived from the atomic polarizability and are given for the atomic species as follows;  $a_C = 1.405 \text{ \AA}^3$ ,  $a_H = 0.514 \text{ \AA}^3$ , and  $a_N = 1.105 \text{ \AA}^3$ .  $A$  is a scaling factor and given as  $A = 2.6$ .



## Parameters used in the PEF

Table S4: Parameters used in the PEF (unit:a.u.).

term	AA*	TT*
$V_0^X$	-378.693877071	-378.666779221
$D_e^X$	$2.556996307 \times 10^{-1}$	$2.408089647 \times 10^{-1}$
$\alpha^X$	1.002760889	1.019491804
$r^X$	1.927998115	1.934760239
$\alpha_\phi^X$	$2.414311060 \times 10^{-1}$	$2.777469250 \times 10^{-1}$
$\theta^X$	2.176783012	2.057857084
$\alpha_\phi^X$	$2.113007700 \times 10^{-2}$	$4.205127250 \times 10^{-2}$
$\phi^X$	3.141592654	3.141592654
$V_0^{X*}$	-378.525263596	-378.568184655
$D_e^{X*}$	$2.391871087 \times 10^{-1}$	$2.562908779 \times 10^{-1}$
$\alpha^{X*}$	1.013914033	1.003143599
$r^{X*}$	1.940549747	1.926430678
$\alpha_\theta^{X*}$	$2.386010320 \times 10^{-1}$	$2.676720880 \times 10^{-1}$
$\theta^{X*}$	2.130919718	2.108576850
$\alpha_\phi^{X*}$	$5.529704930 \times 10^{-2}$	$3.437585380 \times 10^{-2}$
$\phi^{X*}$	3.141592654	3.141592654
$V_2^{A^*A} (V_1^{T^*T})$	2.262166652	2.731388624
$\zeta_2^{A^*A} (\zeta_1^{T^*T})$	1.461380588	1.491409338
$\tilde{V}_2^{A^*A} (\tilde{V}_1^{T^*T})$	$1.240963675 \times 10^{-1}$	$1.258089837 \times 10^{-1}$
$\tilde{\zeta}_2^{A^*A} (\tilde{\zeta}_1^{T^*T})$	$6.322459320 \times 10^{-1}$	$6.869198610 \times 10^{-1}$
$V_4^{A^*A} (V_3^{T^*T})$	2.282898550	2.734431873
$\zeta_4^{A^*A} (\zeta_3^{T^*T})$	1.461331602	1.491443718
$\tilde{V}_4^{A^*A} (\tilde{V}_3^{T^*T})$	$1.248733091 \times 10^{-1}$	$1.258842582 \times 10^{-1}$
$\tilde{\zeta}_4^{A^*A} (\tilde{\zeta}_3^{T^*T})$	$6.375272414 \times 10^{-1}$	$6.651187377 \times 10^{-1}$
$a_{\phi_2}^{A^*A} (a_{\phi_1}^{T^*T})$	$2.480882688 \times 10^{-1}$	$2.313455213 \times 10^{-1}$
$a_{\phi_4}^{A^*A} (a_{\phi_3}^{T^*T})$	$2.642175920 \times 10^{-1}$	$2.207843240 \times 10^{-1}$
$V_{\text{corr}0}^{X^*X}$	$-1.667780276 \times 10^{-2}$	$-1.621894161 \times 10^{-2}$
$(a_{\text{corr}}^{X^*X})^{1/6}$	4.377336791	4.618272738
$a_\emptyset$	$1.570261956 \times 10^{-1}$	$1.518469875 \times 10^{-1}$
$V^{\text{NN}}$	$7.500000000 \times 10^{-5}$	$7.500000000 \times 10^{-5}$
$\zeta^{\text{NN}}$	$1.000000000 \times 10^{+1}$	$1.000000000 \times 10^{+1}$
$R^{\text{NN}}$	5.000000000	5.000000000
$a_{\text{mod}}$	$1.000000000 \times 10^{-1}$	$1.000000000 \times 10^{-1}$

## The terms in the off-diagonal element in Eq. (14)

The terms in the off-diagonal element, Eq. (14), is as follows.

$$a'_1 = 1.362672513 \times 10^{-1}, \quad (\text{S17})$$

$$a'_2 = 1.047629760 \times 10^{-3} \xi_1^2 - 2.848680071 \times 10^{-2} \xi_1 + 4.827950385 \times 10^{-1}, \quad (\text{S18})$$

$$a'_3 = 1.853914153 \times 10^{-2}, \quad (\text{S19})$$

$$a''_1 = -4.453415570 \times 10^{-2} \xi_1^2 + 3.721369096 \times 10^{-3} \xi_1 - 1.335184827 \times 10^{-1}, \quad (\text{S20})$$

$$a''_2 = -1.194755309 \times 10^{-1} \xi_1^2 - 1.355779708 \times 10^{-2} \xi_1 + 5.613355470 \times 10^{-1}, \quad (\text{S21})$$

$$a''_3 = -1.523253963 \times 10^{-2}. \quad (\text{S22})$$

If  $a''_2 \leq 0$ , the second term of Eq. (14) is neglected, *i.e.*,  $a''_1 = 0$ .

## The force constant matrices in Eq. (5)

The force constant matrices in Eq. (5) are in another file. In the file, the parameters are tabulated as follows.

```
#
# ----- force constant matrix for monomer A, a^{A}_{i,j} -----
#
#j\i      1      2      3      4
  1  0.4768785930D+00  0.5564599340D-01 -0.1796669960D-01  0.7318808400D-02
  2  .....
  .      .      .      .
 36  .....

#j\i      5      6      7      8
  1  .....
  .      .      .      .
  .      .      .      .

#j\i      33     34     35     36
  1  .....
  .      .      .      .
  .      .      .      .
 36  .....

# -----
#
# ----- force constant matrix for monomer A*, a^{A*}_{i,j} -----
#
#j\i      1      2      3      4
  1  0.6822566280D+00  0.5550027470D-01  0.2299298890D-01 -0.3148648080D-01
  .      .      .      .
  .      .      .      .

# -----
#
# ----- force constant matrix for monomer T, a^{T}_{i,j} -----
#
#j\i      1      2      3      4
  1  .....
  .      .      .      .
  .      .      .      .

# -----
#
# ----- force constant matrix for monomer T*, a^{T*}_{i,j} -----
#
#j\i      1      2      3      4
  1  .....
  .      .      .      .
  .      .      .      .
```

## The CRK matrices in Eqs. (7)-(8)

The CRK matrices in Eqs. (7)-(8) are in another file.  
In the file, the parameters are tabulated as follows.

```

#
# ----- charge response kernel for monomer A, K^{A}_{i,j} -----
#
#j\i      1      2      3      4      5
1 -0.4149651000D+01  0.9245640000D+00  0.1956897000D+01 -0.1654392000D+01  0.3792570000D+00
2 .....
. . . . .
15 .....

#j\i      6      7      8      9     10
1 .....
. . . . .
. . . . .

#j\i     11     12     13     14     15
1 .....
. . . . .
. . . . .
15 .....

#
# -----
#
# ----- charge response kernel for monomer A*, K^{A*}_{i,j} -----
#
#j\i      1      2      3      4      5
1 .....
. . . . .
. . . . .

#
# -----
#
# ----- charge response kernel for monomer T, K^{T}_{i,j} -----
#
#j\i      1      2      3      4      5
1 .....
. . . . .
. . . . .

#
# -----
#
# ----- charge response kernel for monomer T*, K^{T*}_{i,j} -----
#
#j\i      1      2      3      4      5
1 .....
. . . . .
. . . . .

```