

## ELECTRONIC SUPPLEMENTARY INFORMATION

### DFT Investigation of NH<sub>3</sub> Physisorption on CuSO<sub>4</sub> Impregnated SiO<sub>2</sub>

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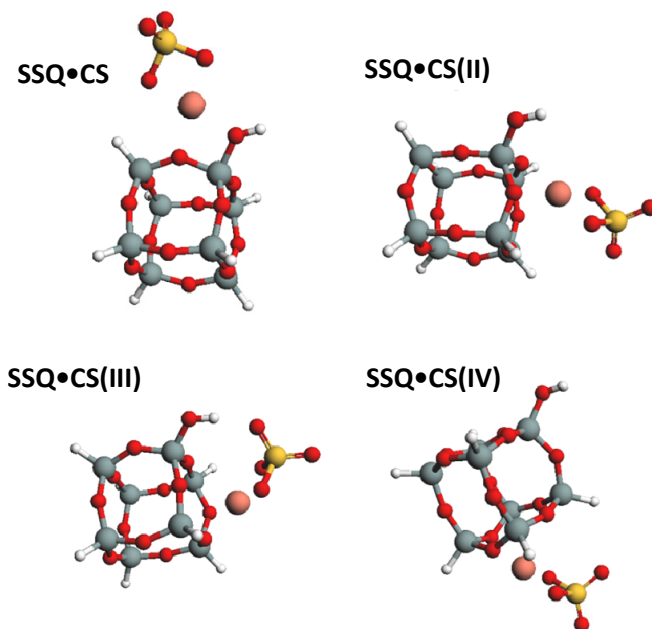
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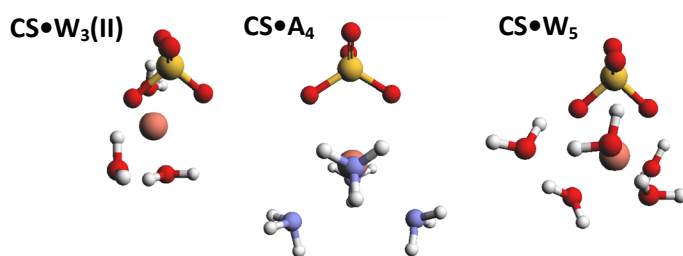
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**Table S-1** Energy of complexation (for reaction  $\text{SSQ} + \text{CS} \rightarrow \text{SSQ}\cdot\text{CS}$ ) of optimized SiO<sub>2</sub>/CuSO<sub>4</sub> complexes with variable final position of the Cu<sup>2+</sup> and SO<sub>4</sub><sup>2-</sup> ions. See Fig. S-1 for the corresponding geometries.

SiO <sub>2</sub> /CuSO <sub>4</sub> Structure	Total Energy (kJ/mol)
SSQ•CS	-213.4
SSQ•CS(II)	-167.9
SSQ•CS(III)	-209.1
SSQ•CS(IV)	-165.7



**Fig. S-1** 3D Optimized geometries of various configurations of SSQ•CS.



**Fig. S-2** Structures demonstrating the influence of hydrogen bonding in CuSO<sub>4</sub> complexed with (from left to right) 3 H<sub>2</sub>O, 4 NH<sub>3</sub>, or 5 H<sub>2</sub>O.