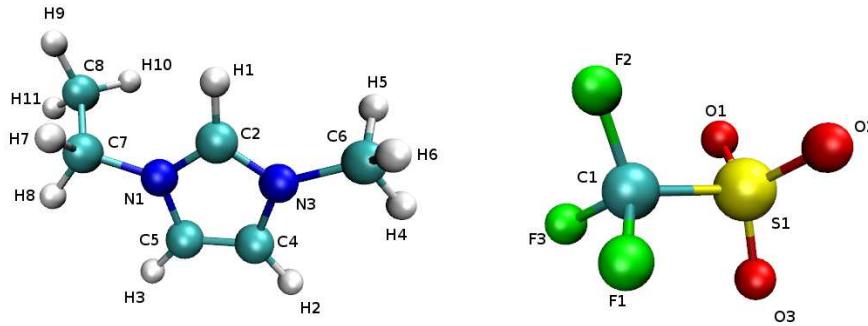


SUPPLEMENTARY MATERIAL



atom	partial charges $q_{i\beta}$ of the cation / e		polarizability $\alpha_{i\beta} / \text{\AA}^3$
	charge-scaled system ($S^{\text{eff}}=0.74$)	polarizable system ($\alpha = 100\%$)	
C6	0.09176	0.12400	1.28860
H4	0.04736	0.06400	(0.41342)
H5	0.04736	0.06400	(0.41342)
H6	0.04736	0.06400	(0.41342)
N1	-0.19758	-0.26700	0.97157
C2	0.30118	0.40700	1.28860
N3	-0.19758	-0.26700	0.97157
C4	0.07770	0.10500	1.28860
C5	0.07770	0.10500	1.28860
H1	0.07178	0.09700	(0.41342)
H2	0.06956	0.09400	(0.41342)
H3	0.06956	0.09400	(0.41342)
C7	0.09620	0.13000	1.28860
H7	0.04070	0.05500	(0.41342)
H8	0.04070	0.05500	(0.41342)
C8	-0.04366	-0.05900	1.28860
H9	0.03330	0.04500	(0.41342)
H10	0.03330	0.04500	(0.41342)
H11	0.03330	0.04500	(0.41342)
sum	0.74000	1.0000	

atom	partial charges $q_{i\beta}$ of the anion / e		polarizability $\alpha_{i\beta} / \text{\AA}^3$
	charge-scaled system ($S^{\text{eff}}=0.74$)	polarizable system ($\alpha = 100\%$)	
C1	0.25900	0.35000	1.28860
F1	-0.11840	-0.16000	0.44475
F2	-0.11840	-0.16000	0.44475
F3	-0.11840	-0.16000	0.44475
S1	0.75480	1.0200	2.47445
O1	-0.46620	-0.63000	0.85197
O2	-0.46620	-0.63000	0.85197
O3	-0.46620	-0.63000	0.85197
sum	-0.74000	-1.0000	

FIG. 13.

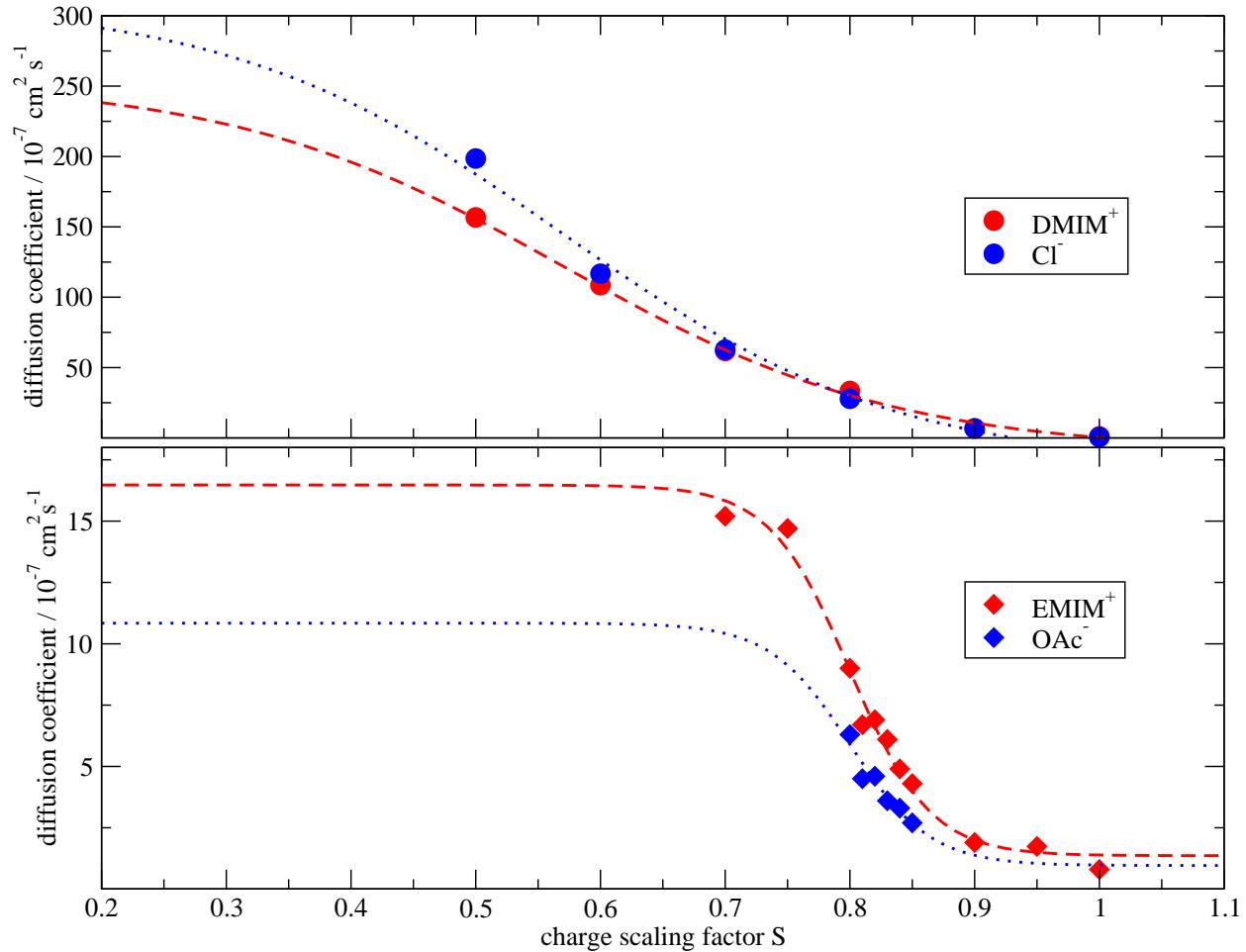


Fig. 13: Diffusion coefficients of molecular ionic liquids from Ref. 75 and 56.

	A_0	A	k	α_0
DMIM^+	254	-264	7.46	0.57
Cl^-	311	-332	7.46	0.57
EMIM^+	16.47	-15.11	31.25	0.80
OAc^-	10.84	-9.88	31.25	0.80

TABLE II. Fit parameters of the logistic function for the above displayed diffusion coefficients. Please note, that the prediction of diffusion coefficients for DMIM chloride above a scaling factor $\gg 1.0$ leads to wrong results due to $A_0 < |A|$.