

Electrocatalytic activity of nanoporous Pd and Pt: effect of structural features

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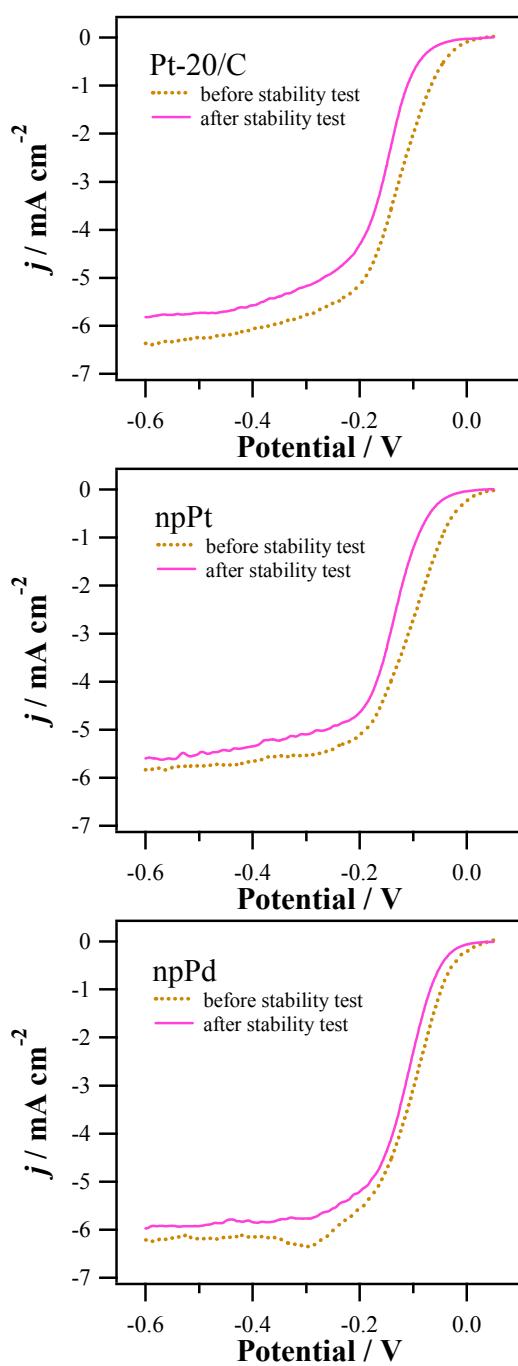


Fig. S1. RDE voltammetry curves obtained for the ORR before and after 1000 repetitive RDE runs in an O₂-saturated 0.1 M NaOH solution using Pt-20/C, npPt and npPd electrodes (scan rate of 10 mV s⁻¹, rotation speed of 2500 rpm).

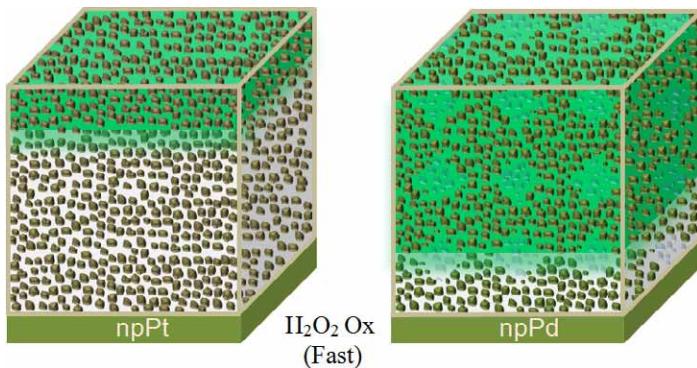


Fig. S2. Schematic view of npPt and npPd for less sluggish H₂O₂ oxidation. npPt uses rather shallow region of the deposited layer close to the solution due to the hindered mass transport through the nanopores. In contrast, npPd utilizes the metal layer up to greater depth due to more facilitated H₂O₂ transport through the macropores interconnecting the nanopores.