## Molecular Simulation and Macroscopic Modeling of the Diffusion of Hydrogen, Carbon Monoxide and Water in Heavy n-Alkane Mixtures

Zoi A. Makrodimitri,<sup>1</sup> Dominik J.M. Unruh<sup>2</sup> and Ioannis G. Economou<sup>1,3,\*</sup>

 <sup>1</sup>National Center for Scientific Research "Demokritos", Institute of Physical Chemistry,
Molecular Thermodynamics and Modelling of Materials Laboratory, GR – 153 10 Aghia Paraskevi Attikis, Greece

<sup>2</sup>Shell Global Solutions International BV, PO Box 38000, 1030 BN Amsterdam, The Netherlands

<sup>3</sup>The Petroleum Institute, Department of Chemical Engineering, PO Box 2533, Abu Dhabi, UAE

\*corresponding author: economou@chem.demokritos.gr

Supplementary Material

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Table	<b>S1.</b> Exper	iment	al values and M	1D predi	ctions at 3.4	MPa for	the 1	melt densi	ty of	the <i>n</i> -
alkane	mixtures.	The	"experimental"	density	of <i>n</i> -alkane	mixtures	was	obtained	from	linear
correla	ation of the	expe	rimental density	values o	f the pure <i>n</i> -a	lkanes.				

T (K)	Density (g/cm <sup>3</sup> )					
	"Experimental"	% Abs Dev				
		Mixture A				
473	0.733	$0.744 \pm 0.002$	1.5			
493	0.719	$0.732\pm0.001$	1.8			
513	0.709	$0.718\pm0.002$	1.3			
		Mixture B				
473	0.722	$0.733 \pm 0.001$	1.5			
493	0.707	$0.721 \pm 0.001$	2.0			
513	0.696	$0.708\pm0.001$	1.7			
		Mixture C				
473	0.692	$0.704\pm0.002$	1.7			
493	0.677	$0.690\pm0.001$	1.9			
513	0.663	$0.679\pm0.001$	2.4			
		Mixture D				
473	0.668	$0.680\pm0.001$	1.8			
493	0.652	$0.665\pm0.001$	2.0			
513	0.637	$0.651\pm0.001$	2.2			
		<b>Mixture E</b>				
473	0.719	$0.729\pm0.001$	1.4			
493	0.704	$0.718\pm0.001$	2.0			
513	0.693	$0.705\pm0.001$	1.7			
		<b>Mixture</b> F				
473	0.688	$0.702 \pm 0.001$	2.0			
493	0.672	$0.688\pm0.001$	2.4			
513	0.659	$0.674\pm0.001$	2.3			
		Mixture G				
473	0.692	$0.697\pm0.001$	0.7			
493	0.675	$0.689\pm0.001$	2.1			
513	0.661	$0.676\pm0.002$	2.3			
		Mixture H				
473	0.692	$0.703 \pm 0.002$	1.6			
493	0.675	$0.692\pm0.001$	2.5			
513	0.663	$0.675 \pm 0.006$	1.8			

	$H_2$	СО	H <sub>2</sub> O	
Solvent		β		VD
n-C <sub>8</sub>	0.0278	0.0187	0.0213	136.8
<i>n</i> -C <sub>12</sub>	0.0268	0.0141	0.0172	203.4
<i>n</i> -C <sub>16</sub>	0.0213	0.0104	0.0133	265.2
<i>n</i> -C <sub>20</sub>	0.0194	0.0102	0.0131	344.4
<i>n</i> -C <sub>28</sub>	0.0131	0.0060	0.0078	447.7
<i>n</i> -C <sub>64</sub>	0.0066	0.0025	0.0036	1034.3
<i>n</i> -C <sub>96</sub>	0.0051	0.0019	0.0027	1575.7

**Table S2.** Re-adjusted values of constants  $\beta$  and  $V_D$  in eq. (4) for H<sub>2</sub>, CO and H<sub>2</sub>O in various *n*-alkanes.

 $\beta$  in m<sup>2</sup>·(mol/s)/cm<sup>3</sup>·K<sup>1/2</sup> and  $V_D$  in cm<sup>3</sup>/mol.



**Fig. S1.** Simulated density of  $n-C_{12} - n-C_{96}$  mixture as a function of  $n-C_{12}$  mole fraction.



**Fig. S2.** Molecular simulation data and RHS theory correlations for the diffusivities of  $H_2$ , CO and  $H_2O$  in (a) mixture A (close symbols) and B (open symbols), (b) mixture C (close symbols) and D (open symbols), (c) mixture E (close symbols) and F (open symbols) and (d) mixture G (close symbols) and H (open symbols).