# **Supporting Information for**

# The Hydrogen Evolution Reaction in a Room Temperature Ionic Liquid: Mechanism and Electrocatalyst trends

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# **1.** The details of voltammetry over a range of H[NTf<sub>2</sub>] concentration, metals and scan rate studies (Experiment data (-) and optimization of theoretical curve (o)) vs Ag/Ag<sup>+</sup> reference redox couple

1.1.1. H[NTf<sub>2</sub>] reduction only on Pt electrode in [C<sub>2</sub>mim][NTf<sub>2</sub>]



# 1.1.2. Hydrogen oxidation only on Pt electrode in [C<sub>2</sub>mim][NTf<sub>2</sub>]



# 1.2 H[NTf<sub>2</sub>] concentration study on gold in [C<sub>2</sub>mim][NTf<sub>2</sub>])

#### 1.2.1. 89mM H[NTf<sub>2</sub>] on gold in [C<sub>2</sub>mim][NTf<sub>2</sub>])



### 1.2.2. 33mM H[NTf<sub>2</sub>] on gold in [C<sub>2</sub>mim][NTf<sub>2</sub>])



Potential V vs  $H_2 / H^+$ 

## 1.2.3. 12mM H[NTf<sub>2</sub>] on gold in [C<sub>2</sub>mim][NTf<sub>2</sub>])



## 1.3 H[NTf<sub>2</sub>] concentration study on molybdenum in [C<sub>2</sub>mim][NTf<sub>2</sub>])

#### 1.3.1. 95mM H[NTf<sub>2</sub>] on molybdenum in [C<sub>2</sub>mim][NTf<sub>2</sub>])



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#### 1.3.2. 30mM H[NTf<sub>2</sub>] on molybdenum in [C<sub>2</sub>mim][NTf<sub>2</sub>])



#### 1.3.3. 19mM H[NTf<sub>2</sub>] on molybdenum in [C<sub>2</sub>mim][NTf<sub>2</sub>])



## 1.4. scan rate study of H[NTf<sub>2</sub>] reduction on Nickel in [C<sub>2</sub>mim][NTf<sub>2</sub>])





## 1.5. scan rate study of H[NTf<sub>2</sub>] reduction on Titanium in [C<sub>2</sub>mim][NTf<sub>2</sub>])



Ti Scan rate: 600mV/s Potential window (V): (-2.6, -0.6) D (HNTf2): 2.5e-7 cm<sup>2</sup>s<sup>-1</sup> D (H<sub>2</sub>): 5.5e-6 cm<sup>2</sup>s<sup>-1</sup> r T<sub>i</sub>: 2.5e-3 cm  $\alpha$ : 0.2  $k^{o}$  : 3.5e-8 cm s<sup>-1</sup> C: 88 mM E<sub>f</sub> : -0.3506 V



Ti Scan rate: 800mV/s Potential window (V): (-2.6, -0.6) D (HNTf2): 2.5e-7 cm<sup>2</sup>s<sup>-1</sup> D (H<sub>2</sub>): 5.5e-6 cm<sup>2</sup>s<sup>-1</sup> r T<sub>i</sub>: 2.5e-3 cm  $\alpha$ : 0.2  $k^{o}$  : 3.5e-8 cm s<sup>-1</sup> C: 88 mM E<sub>f</sub> : -0.3506 V



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